



## Isopropanol production from carbon dioxide in *Cupriavidus necator* in a pressurized bioreactor

Lucile Garrigues, Louna Maignien, Eric Lombard, Jasbir Singh, Stéphane  
Guillouet

### ► To cite this version:

Lucile Garrigues, Louna Maignien, Eric Lombard, Jasbir Singh, Stéphane Guillouet. Isopropanol production from carbon dioxide in *Cupriavidus necator* in a pressurized bioreactor. *New Biotechnology*, 2020, 56, pp.16-20. 10.1016/j.nbt.2019.11.005 . hal-02518899

**HAL Id: hal-02518899**

**<https://hal.inrae.fr/hal-02518899>**

Submitted on 21 Jul 2022

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial 4.0 International License

1    **Isopropanol production from carbon dioxide in *Cupriavidus***  
2    ***necator* in a pressurized bioreactor.**

3    Lucile Garrigues<sup>a</sup>, Louna Maignien<sup>a</sup>, Eric Lombard<sup>a</sup>, Jasbir Singh<sup>b</sup>, Stéphane E. Guillouet<sup>a</sup>

4    <sup>a</sup>Laboratoire d'Ingénierie des Systèmes Biologiques et des Procédés (LISBP), Université de  
5    Toulouse, CNRS, INRA, INSA, 135 avenue de Rangueil, 31077 Toulouse CEDEX 04, France

6    <sup>b</sup>H.E.L. limited, 9-10 Capital Business Park, Manor Way, Borehamwood, Hertfordshire  
7    WD6 1GW, UK

8

9    **CORRESPONDING AUTHOR**

10    Stéphane E. GUILLOUET

11    stephane.guillouet@insa-toulouse.fr

## ABSTRACT

A bioreactor was designed to provide high gas mass transfer to reach cell and product titres in the  $\text{g L}^{-1}$  level from  $\text{CO}_2$  for realistic, laboratory scale, engineered autotrophic strain evaluation. The design was based on independent  $\text{CO}_2$ ,  $\text{H}_2$  and air inputs and the ability to operate at high pressures. The bioreactor configuration and cultivation strategy enabled growth of *Cupriavidus necator* strains for long periods, to reach over  $3 \text{ g L}^{-1}$  dry cell weight. No negative impact of the high pressure was observed on viability of the strains up to more than 4 bar overpressure. The cultivation was then carried out using an engineered isopropanol producing strain; in this case,  $3.5 \text{ g L}^{-1}$  isopropanol was obtained from  $\text{CO}_2$  as the sole carbon source. This is the first reported demonstration of a successful production from engineered bacteria of product in the  $\text{g L}^{-1}$  range on  $\text{CO}_2$ , raising the prospect of future development of  $\text{CO}_2$ -based bioprocesses.

## KEYWORDS

*Ralstonia eutropha*; *Cupriavidus necator*; isopropanol; carbon dioxide; hydrogen; bioreactor design; gas fermentation; autotrophy; pressure

## ABBREVIATIONS

P(3HB): poly(3-hydroxybutyrate); TSB: tryptic soy broth; DCW: dry cell weight.

## INTRODUCTION

Developing bioconversion processes based on CO<sub>2</sub> has gained increasing interest in the microbial production of biofuel and chemical synthons. To develop future bioprocesses for bio-based chemical synthesis using CO<sub>2</sub>, the bacterium *Cupriavidus necator* is particularly attractive due to its efficient utilization of this sustainable carbon source. This facultative chemolithoautotroph, formerly also classified as *Ralstonia eutropha*, is the best-reported microorganism based on CO<sub>2</sub> as the sole carbon source. It has the ability to grow at high rates (in the range of 0.3-0.4 h<sup>-1</sup>) to high cell densities using gas mixtures of CO<sub>2</sub>/H<sub>2</sub>/O<sub>2</sub> [1]. It has also demonstrated higher energy efficiencies in producing biomass and bioproduct, poly(3-hydroxybutyrate) (P(3HB)) compared to plants or microalgae [2]. *C. necator* has also been extensively studied for P(3HB) production in heterotrophy [3-6] and autotrophy [1]. This natural ability to store excess carbon in the form of P(3HB) is of great interest as numerous interesting chemicals share the same production pathway precursor acetyl-coA, which implies that fewer genetic modifications would be required to divert P(3HB) precursors into the final product [7, 8].

Recently, *C. necator* was successfully engineered for the production of various compounds under autotrophic conditions from a CO<sub>2</sub>/H<sub>2</sub>/O<sub>2</sub> mixture. Up to 250 mg L<sup>-1</sup> isopropanol [8], 180 mg L<sup>-1</sup> methyl ketones [9], 4.4 mg L<sup>-1</sup> alka(e)nes [7] and 6 mg L<sup>-1</sup> α-humulene [10] were obtained under autotrophic conditions. The limited amount of product reported here is due to a combination of strain efficiency and unsuitability of the laboratory-scale bioreactor for gas fermentation. The previous cultivation systems using classical bioreactor flushed with commercial gas mixtures have demonstrated production levels suitable only for proof of concept studies. In order to further develop the engineered strains for future CO<sub>2</sub>-based

bioprocesses, laboratory-scale bioreactor systems have to be designed specifically for gas fermentation, allowing higher titres of biomass to be achieved and to characterize the strains under more realistic conditions in terms of product titres, productivity and yields.

Bioreactor designs for gas fermentation have previously been reported for cultivation of  $H_2$  oxidising bacteria such as *C. necator*. They generally consist of gas mixtures containing  $CO_2$ ,  $H_2$  and  $O_2$  added through a single sparger. The optimal gas composition employed in the literature for autotrophic growth of *C. necator* is typically  $H_2/O_2/CO_2=7:1:1$  [11-13],  $H_2/O_2/CO_2=7:2:1$  [12], or  $H_2/O_2/CO_2=6:2:1$  [14, 15]. However, these gas mixtures lie within the explosive composition range (which is normally regarded as being above 6%  $O_2$  in the presence of  $H_2$ ) and limits the cell densities reached. In order to avoid this  $O_2$  limitation during autotrophic growth of *C. necator*, pure oxygen has been fed to the bioreactor [1]. This was combined with a basket-type agitator which generated huge increases in the  $k_{La}$  (up to  $2970\ h^{-1}$ ), allowing a high cell density of  $91\ g\ L^{-1}$  and a residual biomass containing 68% of PHB ( $29\ g\ L^{-1}$  of catalytic biomass concentration) to be reached.

The design of a small laboratory-scale bioreactor dedicated to  $CO_2$  fermentation of  $H_2$ -oxidizing bacteria is described here. The challenges tackled were (i) the danger of gas explosion above the limit of 6%  $O_2$  in the presence of excess  $H_2$  and (ii) the gas transfer efficiency at small scale. Reducing the size of a bioreactor is always more challenging for parameters such as gas transfer and necessitates the application of process engineering and biochemical engineering principles. The engineering solution to poor gas transfer is usually limited to the volumetric gas transfer coefficient,  $k_{La}$ , increase through changes in sparger and stirring arrangement as mentioned above. However, this offers relatively limited scope for improvement. An effective alternative is to operate the bioreactor at elevated pressure as this

can, in principle, increase the gas transfer rate several fold without changes to sparging or agitation. This concept has already been successfully applied in bioreactors with various microorganisms, such as *Arxula adeninivorans*, *Corynebacterium glutamicum*, *C. necator*, *Escherichia coli*, and *Saccharomyces cerevisiae* [16-20]. There is, however, a lack of information on its impact on cell viability and productivity with engineered microorganisms during long fermentations runs.

Moreover, in order to have more flexibility in terms of gas composition and therefore higher gas solubility, the bioreactor gas-supply system was designed with three independent gas lines supplying three spargers (CO<sub>2</sub>, H<sub>2</sub> and air). A safety system was designed to control O<sub>2</sub> level in the headspace in order to avoid an explosive mixture within the bioreactor. Experimental validation was performed with autotrophic growth and production of isopropanol with an engineered *C. necator* strain.

## **MATERIALS AND METHODS**

### **Strains**

In this study, the reference strain used was *C. necator* Re2133, deficient for P(3-HB) biosynthesis (H16 ΔphaCAB) obtained from [21]. The isopropanol strain tested in this study was *C. necator* Re2133 bearing the plasmid pEG7b carrying a synthetic isopropanol operon under the constitutive pTac promoter [22]. The strains were stored at -80°C in liquid TBS medium (dextrose-free tryptic soy broth, Becton Dickinson, France) with 20 % glycerol (v/v).

### **Media**

The rich medium consisted of 2.75% (w/v) dextrose-free tryptic soy broth (TSB, Becton Dickinson, France). The minimal medium A used for preculture in flask was previously

described by [21] and [7]. Fructose ( $20 \text{ g L}^{-1}$ ) was used as the only carbon source.  $\text{NH}_4\text{Cl}$  ( $0.5 \text{ g L}^{-1}$ ) was used as the nitrogen source to reach a biomass concentration of about  $1 \text{ g L}^{-1}$ . The minimal medium B used for the cultures in the bioreactor consisted of  $0.19 \text{ g L}^{-1}$  nitrilotriacetic acid,  $0.06 \text{ g L}^{-1}$  ferrous ammonium citrate,  $0.5 \text{ g L}^{-1}$  of  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ ,  $0.01 \text{ g L}^{-1}$  of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ , and  $1 \text{ ml}$  of trace element solution. The composition of the latter was  $0.3 \text{ g L}^{-1}$   $\text{H}_3\text{BO}_3$ ,  $0.2 \text{ g L}^{-1}$   $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ,  $0.1 \text{ g L}^{-1}$   $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ ,  $0.03 \text{ g L}^{-1}$   $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ ,  $0.03 \text{ g L}^{-1}$   $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$ ,  $0.02 \text{ g L}^{-1}$   $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ , and  $0.01 \text{ g L}^{-1}$   $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ .  $(\text{NH}_4)_2\text{SO}_4$  ( $2 \text{ g L}^{-1}$ ) was used as nitrogen source to reach a biomass concentration of about  $4 \text{ g L}^{-1}$ . After autoclaving the above medium,  $40 \text{ ml}$  of a sterile phosphate solution containing  $224 \text{ g L}^{-1}$   $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$  and  $37.5 \text{ g L}^{-1}$   $\text{KH}_2\text{PO}_4$  was aseptically added to the fermenter.

## **Seed cultivations**

After thawing at room temperature, one glycerol stock was streaked on a TSB agar Petri dish. and incubated for  $36 \text{ h}$  at  $30^\circ\text{C}$ . One isolated colony was used to inoculate the first seed culture grown for  $24 \text{ h}$  with  $5 \text{ mL}$  of TSB supplemented with  $10 \text{ mg L}^{-1}$  gentamycin and  $200 \text{ mg L}^{-1}$  kanamycin, in a  $50 \text{ mL}$  baffled Erlenmeyer flask. This culture was used to inoculate a  $250 \text{ mL}$  Erlenmeyer baffled flask containing  $50 \text{ mL}$  of mineral medium A. The seed was cultivated at  $30^\circ\text{C}$ ,  $100 \text{ rpm}$  for  $18 \text{ h}$ . The culture was then centrifuged ( $5 \text{ min}$ ,  $10000g$ ), the supernatant was discarded and the cell pellet was resuspended in  $30 \text{ mL}$  sterile fresh medium B to eliminate the residual fructose. The suspension was then used to inoculate the bioreactor containing  $300 \text{ mL}$  of sterile mineral medium B.

## **Culture conditions and bioreactor system**

The cultures were performed in a  $500 \text{ ml}$  total volume Xplorer® (HEL Ltd, UK) bioreactor that was modified specifically for this study. The working liquid volume was  $330 \text{ mL}$ . The

bioreactor was first used with a commercial gas mixture containing H<sub>2</sub>/O<sub>2</sub>/CO<sub>2</sub>/N<sub>2</sub> (molar %: 60:2:10:28) (Air Liquide, France) provided into the medium through a sparger. For the experiments carried out with individual gases, the bioreactor was then set up with three spargers in order to deliver air, pure CO<sub>2</sub> and pure H<sub>2</sub> independently. A fourth inlet gas was added in the headspace for pure N<sub>2</sub> supply as a safety system. Each gas inlet was controlled by a gas mass flow meter. The bioreactor was equipped with pH, dissolved oxygen (DO), temperature and pressure controllers. The Xplorer® software handled the on-line monitoring and control systems of the reactor. Both CO<sub>2</sub> and O<sub>2</sub> in the outlet gases were analysed using a Tandem Pro gas analyzer (Magellan BioTech HEL Ltd , UK). The bioreactor was safely controlled by the O<sub>2</sub> concentration in the headspace in order to avoid any explosive risk. The headspace was flushed with pure N<sub>2</sub> if the O<sub>2</sub> surpassed 5% in the headspace. The DO level in the reactor was controlled below 5% of air saturation at atmospheric pressure by varying stirring speed and/or inlet air flow rate in order to restrict the known inhibition of *C. necator* hydrogenases by the dissolved O<sub>2</sub> concentration [23]. Temperature was maintained at 30°C. The pH was maintained at 7.0 by addition of a 14% (v/v) NH<sub>4</sub>OH solution.

### **Characterization of the biomass**

Cell growth was followed spectrophotometrically at 600 nm with a DR3900 spectrophotometer (Hach, Loveland, Colorado, USA) after calibration against cell dry weight measurements. For cell dry weight determination, culture medium was harvested and filtered through 0.2 µm pore-size polyamide membranes (Sartorius AG, Göttingen, Germany), which were then dried to a constant weight at 60°C under partial vacuum (200 mmHg, i.e. approximately 26.7 kPa).



Cell viability was measured with a BD Accuri C6® flow cytometer (BD Biosciences, US) after cell staining with propidium iodide (PI) dye (Molecular Probes, Invitrogen, USA) commonly used to monitor membrane integrity as indicator of viability [8]

## **Metabolites and statistical analyses**

Metabolites (isopropanol, acetone, pyruvic and acetic acids) were quantified by chromatography as described in [22].

## **Statistical analyses**

Concentrations of metabolites and biomass are given as the mean value of 2 to 3 independent analyses. The error on the specific growth rate (determined as being the slope of  $\ln[\text{Optical Density}] = f(t)$ ) was calculated as the standard deviation (SD) of the slope. The instantaneous specific growth rates were calculated upon fitting process of the experimental data and the derivative calculation. For the yield determination, the state variables were plotted pairwise in a scatter plot within the considered period of the culture. A linear regression was applied to determine the yield (as the slope) and the error (as the SD of the slope). For the calculation of the CO<sub>2</sub> to isopropanol conversion yield, the isopropanol amount was plotted versus the amount of consumed CO<sub>2</sub>, calculated from a classical mass balance from the analyses of inlet and outlet gas composition.

## **RESULTS & DISCUSSION**

### **Benefits of elevated pressure on autotrophic growth of *C. necator* Re2133 on a commercial gas mixture**

Autotrophic production of chemicals by *C. necator* was previously reported in a classical CSTR bioreactor where gases were supplied by flushing the reactor with a commercial gas

mixture containing H<sub>2</sub>/O<sub>2</sub>/CO<sub>2</sub>/N<sub>2</sub> mix (molar %: 60:2:10:28) [7, 8]. Here, this mixture was used first to characterize the growth of the reference strain *C. necator* Re2133 in the newly designed bioreactor (**Figure 1**). After 6 h cultivation, the maximum values of agitation and ventilation rates were reached. The biomass concentration was low at 0.2 g L<sup>-1</sup>. An alternative way to provide higher gas transfer is to operate the bioreactor at elevated pressure. Then the dissolved O<sub>2</sub> concentration was kept at 2% saturation, at various pressures in the reactor up to 3 bar overpressure. Growth was then extended to 15 h leading to a final biomass concentration of 0.5 g L<sup>-1</sup>. Growth followed an exponential profile throughout the cultivation with a maximum specific rate of 0.13 ± 0.01 h<sup>-1</sup>. Cell viability was analysed by flow cytometry after staining with propidium iodide. The non-PI stained cells were considered viable, and the percentage was >95% over the duration of the culture, even when overpressure reached 3 bar, indicating their robustness towards high pressure.

#### **Benefits of independent gas sparging on autotrophic growth of *C. necator* Re2133**

The relatively low final biomass concentration obtained even under high pressure was likely due to the low gas transfer capacity when using the mixture containing H<sub>2</sub>/O<sub>2</sub>/CO<sub>2</sub>/N<sub>2</sub> (molar %: 60:2:10:28). This composition was far from optimal for autotrophic growth of *C. necator*. However, they are all within the explosive range for H<sub>2</sub> and O<sub>2</sub>. The cultivation of the Re2133 strain was initiated by sparging the 3 independent gases in order to reach 88% H<sub>2</sub>, 2% O<sub>2</sub> and 2.5% CO<sub>2</sub> gas mixture in the reactor (62, 6.7 and 1.75 mL/min, for H<sub>2</sub>, air and CO<sub>2</sub>). It allowed ensuring an H<sub>2</sub>-rich (H<sub>2</sub> being the less soluble gas) and safe environment controlling the O<sub>2</sub> % in the headspace below the explosive range (6% of air saturation at atmospheric pressure). The level of dissolved O<sub>2</sub> was controlled to meet the increasing biological demand while maintaining the level at 2% saturation in order to limit the reported inhibition of the

hydrogenases [23, 24] by increasing the ratio air/H<sub>2</sub> flow rate within the range of 6.7-25 mL/min for air and 62-86 mL/min for H<sub>2</sub> (**Figure 2**). The duration of growth at atmospheric pressure was 15 h compared to 6 h with the reactor operating with the gas mix, illustrating the benefit of three independent spargers. Growth followed an exponential profile throughout the cultivation at atmospheric pressure with a maximum specific rate of  $0.11 \pm 0.01 \text{ h}^{-1}$ . At 15 h, inlet gas composition was 70% H<sub>2</sub>, 5% O<sub>2</sub> and 2.8% CO<sub>2</sub> (71, 26 and 2.8 mL/min for H<sub>2</sub>, air and CO<sub>2</sub>). When the inlet gas composition for O<sub>2</sub> reached the safety level (5%) defined to ensure a safe environment, the dissolved O<sub>2</sub> level was maintained by switching the strategy to a step-wise increase in total reactor pressure up to >3 bars. The growth profile was still exponential without any observed drop in specific rate ( $0.11 \pm 0.01 \text{ h}^{-1}$ ). A final pressure up to 4.2 bars was applied with a concomitant increase in H<sub>2</sub> (117 mL/min) and a decrease in O<sub>2</sub> level in the inlet gas (3.7%) in order to maintain the dissolved oxygen at a low level. The amount of CO<sub>2</sub> in the outlet gas was always found to be lower than in the inlet and never exceeded 1.6%. During this period, the specific growth was lower but constant at  $0.07 \pm 0.005 \text{ h}^{-1}$ . Cell viability was assessed by flow cytometry and did not show any fall in the percentage of PI unstained cells.

### **Isopropanol production under autotrophic condition**

Isopropanol autotrophic production was tested in the designed bioreactor using an engineered strain Re2133/pEG7b [22]. Cultivation was initiated by sparging the three independent gases in order to reach 87% H<sub>2</sub>, 2% O<sub>2</sub> and 3%CO<sub>2</sub> mixture in the reactor (59, 6.7 and 2 mL/min respectively for H<sub>2</sub>, air and CO<sub>2</sub>) allowing an H<sub>2</sub>-rich, O<sub>2</sub>-low environment to be reached. The level of dissolved O<sub>2</sub> was monitored throughout by increasing the ratio air/H<sub>2</sub> flow rate and the overpressure in the bioreactor in order to fulfill the increasing microbial O<sub>2</sub> demand

(**Figure 3**). The gas flow rates ranged between 59-200 mL/min for H<sub>2</sub>, 6.7 -75 mL/min for air and 2-6 mL/min for CO<sub>2</sub>. *C. necator* Re2133/pEG7b grew for 20 h at an average specific rate of  $0.051 \pm 0.006 \text{ h}^{-1}$ . Growth slowly decreased to  $0.023 \pm 0.001 \text{ h}^{-1}$  for the remaining period of the cultivation. The final biomass concentration reached at 85 h was  $2.9 \text{ g L}^{-1}$ . Compared to the reference strain Re2133, the lower growth rate was likely due to inhibition by isopropanol and acetone as previously observed with the same strain with fructose ( $0.05$  vs  $0.17 \text{ h}^{-1}$ , for Re2133 and Re2133/pEG7b respectively) [22]. Isopropanol and acetone were produced concomitantly to reach maximum concentrations in the liquid phase of 3 and  $0.65 \text{ g L}^{-1}$  respectively. Any isopropanol and acetone evaporated with the gas stream was trapped in water and quantified during the cultivation. Taking this evaporation into account, the maximal concentrations were equivalent to  $3.5$  and  $0.9 \text{ g L}^{-1}$  of isopropanol and acetone, respectively. Reproducibility was assessed with a second independent cultivation and led to the production of  $1.96 \text{ g L}^{-1}$  biomass,  $2.1 \text{ g L}^{-1}$  isopropanol and  $1 \text{ g L}^{-1}$  acetone. These performances were remarkable when compared to that obtained with fructose: Re2133/pEG7b was able to produce up to  $2.2 \text{ g L}^{-1}$  isopropanol and  $1 \text{ g L}^{-1}$  acetone in 88 h with fructose [22]. The overall specific production rate of isopropanol in *C. necator* Re2133/pEG7b from CO<sub>2</sub> was  $0.013 \text{ gg}_{\text{DCW}}^{-1} \text{ h}^{-1}$  ( $0.030 \text{ gg}_{\text{DCW}}^{-1} \text{ h}^{-1}$  with fructose) and peaked at  $0.053 \text{ gg}_{\text{DCW}}^{-1} \text{ h}^{-1}$  with CO<sub>2</sub> ( $0.085 \text{ gg}_{\text{DCW}}^{-1} \text{ h}^{-1}$  with fructose). Cell viability was assessed by flow cytometry after staining with PI during the cultivation. The percentage of PI unstained cells ranged from 82-98%, which was slightly lower cultivation of Re2133 in the pressurized bioreactor. Concomitantly with the reduction in the  $\mu_{\text{max}}$ , this could be attributed to an inhibitory effect of isopropanol on Re2133/pEG7b.

The proof of concept of producing isopropanol from CO<sub>2</sub> with an engineered *C. necator* strain was previously demonstrated: 250 mg L<sup>-1</sup> isopropanol was obtained from 1 g L<sup>-1</sup> biomass using a classical bioreactor and commercialized gas mix containing H<sub>2</sub>, O<sub>2</sub> and CO<sub>2</sub> (60:2:10) [8]. It is clearly demonstrated here that designing a pressurized bioreactor with independent gas feeds significantly improved gas transfer to reach higher biomass and consequently higher isopropanol concentrations of up to 3.5 g L<sup>-1</sup>. The calculation of the carbon balance in this fermentation showed that 70-80% of the CO<sub>2</sub> was recovered into biomass, isopropanol and acetone. The yield of CO<sub>2</sub> converted into isopropanol was 0.13 Cmol Cmol<sup>-1</sup> in this study, slightly lower than 0.2 Cmol Cmol<sup>-1</sup> reached previously. [8]. However, strain Re2133/pEG7b used here constitutively expressed the isopropanol synthetic operon whereas Re2133/pEG23 previously tested had an inducible isopropanol operon controlled by the arabinose promoter. Thus uncoupling of growth and production allowed the carbon flow to be better directed towards the isopropanol pathway, as reported with fructose [22].

## CONCLUSIONS

The design of a dedicated gas bioreactor incorporating the independent management of the three gases, CO<sub>2</sub>, H<sub>2</sub> and Air, coupled with the use of high pressure has enabled a dramatic improvement in gas transfer in a small scale bioreactor. *C. necator* showed an ability to sustain a pressure of over 4 bars, leading to a higher biomass concentration and an isopropanol production of up to 3.5 g L<sup>-1</sup>. This is the first reported demonstration successfully producing compounds at the g L<sup>-1</sup> scale from engineered autotrophic bacteria and raises the prospect of the near-future development of CO<sub>2</sub>-based commercial bioprocesses.

256    **ACKNOWLEDGMENT**

257    We thank M. Appleton (H.E.L., UK) for all the efforts and discussions provided on the design  
258    of the dedicated bioreactor.

259    This research did not receive any specific grant from funding agencies in the public,  
260    commercial, or not-for-profit sectors.

261

## 262 REFERENCES

- 263 1. Tanaka K, Ishizaki A, Kanamaru T, Kawano T. Production of poly(D-3-  
264 hydroxybutyrate) from CO<sub>2</sub>, H<sub>2</sub>, and O<sub>2</sub> by high cell density autotrophic cultivation  
265 of *Alcaligenes eutrophus*. *Biotechnol Bioeng* 1995;45:268-75.  
266 <https://doi.org/10.1002/bit.260450312>
- 267 2. Lu Y, Yu J. Comparison analysis on the energy efficiencies and biomass yields in  
268 microbial CO<sub>2</sub> fixation. *Process Biochem* 2017;62:151-160.  
269 <https://doi.org/10.1016/j.procbio.2017.07.007>
- 270 3. Reinecke F, Steinbuchel A. *Ralstonia eutropha* strain H16 as model organism for PHA  
271 metabolism and for biotechnological production of technically interesting  
272 biopolymers. *J Mol Microbiol Biotechnol* 2009;16:91-108.  
273 <https://doi.org/10.1159/000142897>
- 274 4. Budde CF, Riedel SL, Willis LB, Rha CK, Sinskey AJ. Production of poly(3-  
275 hydroxybutyrate-co-3-hydroxyhexanoate) from plant oil by engineered *Ralstonia*  
276 *eutropha* strains. *Appl Environ Microbiol* 2011;77:2847-54.  
277 <https://doi.org/10.1128/AEM.02429-10>.
- 278 5. Koller M, Bona R, Braunegg G, Hermann C, Horvat P, Kroutil M et al. Production of  
279 Polyhydroxyalkanoates from Agricultural Waste and Surplus Materials.  
280 *Biomacromolecules* 2005;6:561-565. <https://doi.org/10.1021/bm049478b>
- 281 6. Schlegel HG, *Alcaligenes Eutrophus* and Its Scientific and Industrial Career. In:  
282 Dawes E.A. (eds) *Novel Biodegradable Microbial Polymers*. NATO ASI Series  
283 (Series E: Applied Sciences), vol 186. Springer, Dordrecht 1990, p. 133-141.  
284 [https://doi.org/10.1007/978-94-009-2129-0\\_12](https://doi.org/10.1007/978-94-009-2129-0_12)
- 285 7. Crépin L, Lombard E, Guillouet SE. Metabolic engineering of *Cupriavidus necator* for  
286 heterotrophic and autotrophic alka(e)ne production. *Metab Eng* 2016;37:92-101.  
287 <https://doi.org/10.1016/j.ymben.2016.05.002>
- 288 8. Marc J, Grousseau E, Lombard E, Sinskey AJ, Gorret N, Guillouet SE. Over  
289 expression of GroESL in *Cupriavidus necator* for heterotrophic and autotrophic  
290 isopropanol production. *Metab Eng* 2017;42:74-84.  
291 <https://doi.org/10.1016/j.ymben.2017.05.007>
- 292 9. Muller J, MacEachran D, Burd H, Sathitsuksanoh N, C Bi, Yeh YC et al. Engineering  
293 of *Ralstonia eutropha* H16 for autotrophic and heterotrophic production of methyl  
294 ketones. *Appl Environ Microbiol* 2013;79:4433-9.  
295 <https://doi.org/10.1128/AEM.00973-13>
- 296 10. Krieg T, Sydow A, Faust S, Huth I, Holtmann D. CO<sub>2</sub> to Terpenes: Autotrophic and  
297 Electroautotrophic alpha-Humulene Production with *Cupriavidus necator*. *Angew*  
298 *Chem Int Ed Engl* 2018;57:1879-1882. <https://doi.org/10.1002/anie.201711302>.
- 299 11. Ishizaki A, Tanaka K. Batch culture of *Alcaligenes eutrophus* ATCC 17697 using  
300 recycled gas closed circuit culture system. *J Ferment Bioeng* 1990;69:170-174.  
301 <https://doi.org/10.1002/anie.201711302>
- 302 12. Takeshita T, Tanaka K. Influence of hydrogen limitation on gaseous substrate  
303 utilization in autotrophic culture of *Alcaligenes eutrophus* ATCC 17697. *J Ferment*  
304 *Bioeng* 1996;81:83-86. [https://doi.org/10.1016/0922-338X\(96\)83128-X](https://doi.org/10.1016/0922-338X(96)83128-X)
- 305 13. Volova TG, Kiselev EG, Shishatskaya, Zhila No, Boyandin ANR, Syrvacheva DA et  
306 al. Cell growth and P(3HB) accumulation from CO<sub>2</sub> of a carbon monoxide-tolerant

- hydrogen-oxidizing bacterium, *Ideonella* sp. O-1. *Appl Microbiol Biotechnol* 2011;92:1161-9. <https://doi.org/10.1016/j.biortech.2013.07.070>.
14. Ammann ECB, Reed LL, Durichek JJE.. Gas consumption and growth rate of *Hydrogenomonas eutropha* in continuous culture. *Appl Microbiol* 1968;16:822-826.
15. Volova TG, Voinov NA. Kinetic parameters of a culture of the hydrogen-oxidizing bacterium *Ralstonia eutropha* grown under conditions favoring polyhydroxybutyrate biosynthesis. *Appl Biochem Microbiol* 2003;39:166-170. <http://dx.doi.org/10.1016/j.resmic.2012.10.008>
16. Belo I, Pinheiro R, Mota M, Fed-batch cultivation of *Saccharomyces cerevisiae* in a hyperbaric bioreactor. *Biotechnol Progress* 2003;19:665-671. <https://doi.org/10.1021/bp0257067>
17. Knoll A, Bartsch S, Husemann B, Engel P, Schroer K, Ribeiro B et al. High cell density cultivation of recombinant yeasts and bacteria under non-pressurized and pressurized conditions in stirred tank bioreactors. *J Biotechnol* 2007;132:167-179. <https://doi.org/10.1016/j.jbiotec.2007.06.010>
18. Knabben I, Regestein L, Marquering F, Steinbusch S, Lara AR, Büchs J. High cell-density processes in batch mode of a genetically engineered *Escherichia coli* strain with minimized overflow metabolism using a pressurized bioreactor. *J Biotechnol* 2010;150:73-9. <https://doi.org/10.1016/j.jbiotec.2010.07.006>
19. Matsui T, Sato H, Yamamuro H, Shinzato N, Matsuda H, Misawa S et al. High cell density cultivation of recombinant *E. coli* for hirudin variant 1 production by temperature shift controlled by pUC18-based replicative origin. *Appl Microbiol Biotechnol* 2008; 80:779-783. <https://doi.org/10.1007/s00253-008-1611-2>
20. Yu J, Munasinghe P. Gas fermentation enhancement for chemolithotrophic growth of *Cupriavidus necator* on carbon dioxide. *Fermentation* 2018;4:63. <https://doi.org/10.3390/fermentation4030063>
21. Lu J, Brigham CJ, Gai CS, Sinskey AJ. Studies on the production of branched-chain alcohols in engineered *Ralstonia eutropha*. *Appl Microbiol Biotechnol* 2012;96:283-97. <https://doi.org/10.1007/s00253-012-4320-9>
22. Grousseau E, Lu JN, Gorret N, Guillouet SE, Sinskey AJ. Isopropanol production with engineered *Cupriavidus necator* as bioproduction platform. *Appl Microbiol Biotechnol* 2014;98:4277-4290. <https://doi.org/10.1007/s00253-014-5591-0>
23. Siegel RS, Ollis DF, Kinetics of Growth of the Hydrogen-Oxidizing Bacterium *Alcaligenes eutrophus* (ATCC 17707) in Chemostat Culture. *Biotechnol Bioeng* 1984; 26:764-770. <https://doi.org/10.1007/s00253-007-0983-z>
24. Cracknell JA, Wait AF, Lenz O, Friedrich B, Armstrong FA. A kinetic and thermodynamic understanding of O<sub>2</sub> tolerance in [NiFe]-hydrogenases. *Proc Nat Acad Sci USA* 2009;106:20681. <https://doi.org/10.1073/pnas.0905959106>



## 346 **Figure legends**

347 **Figure 1:** (A) Evolution of the dissolved O<sub>2</sub> level (dashed line), the total gas flow rate (thin  
348 black line) and the overpressure (thick black line) in the bioreactor. (B) Evolution of the  
349 biomass concentration (black circles) and cell viability (open triangles)

350 **Fig.2:** Autotrophic growth of *C. necator* Re2133 in the gas bioreactor modified with 3  
351 independent gas spargers. (A) Evolution of relative pressure (thick line) and partial pressure  
352 of dissolved oxygen (dashed line). (B) Evolution of the inlet gas composition for H<sub>2</sub> (solid  
353 thick line), O<sub>2</sub> (dashed line) and CO<sub>2</sub> (solid thin line) .(C) Evolution of biomass  
354 concentrations (●) and % of PI unstained cells (Δ).

355 **Fig.3:** Autotrophic isopropanol production of *C. necator* Re2133/pEG7b in the gas bioreactor  
356 modified with 3 independent gas spargers. (A) Evolution of the relative pressure (thick line)  
357 and the partial pressure of dissolved O<sub>2</sub> (dashed line). (B) Evolution of the inlet gas  
358 composition in H<sub>2</sub> (solid thick line), O<sub>2</sub> (dashed line) and CO<sub>2</sub> (solid thin line). (C) Evolution  
359 of biomass concentrations (●), % of PI unstained cells (Δ) and isopropanol (□), acetone (▽)  
360 concentrations in the bioreactor liquid phase. Total isopropanol (■) and acetone (▼)  
361 production were calculated taking amounts evaporated into account.

Figure 1

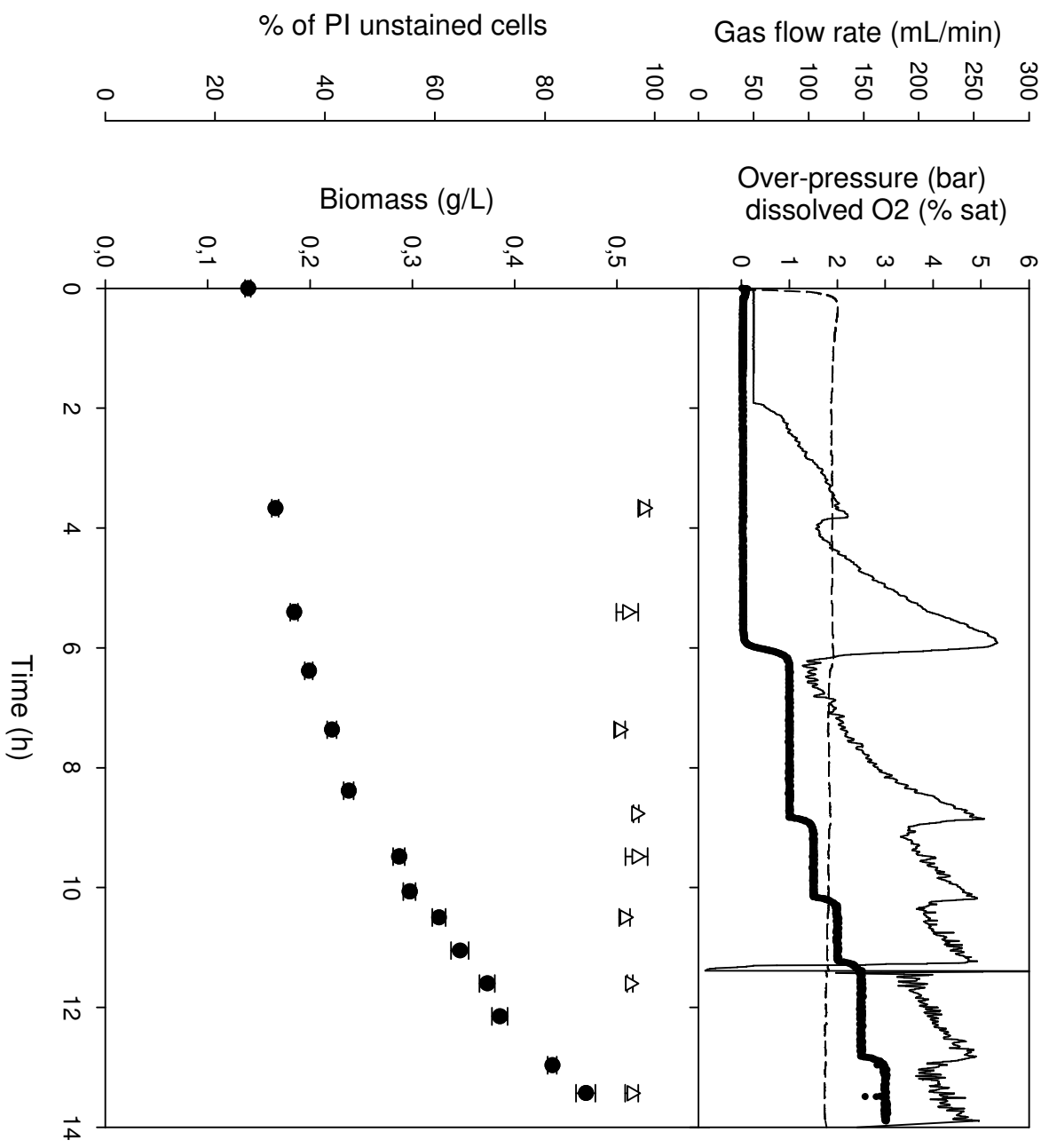


Figure 2

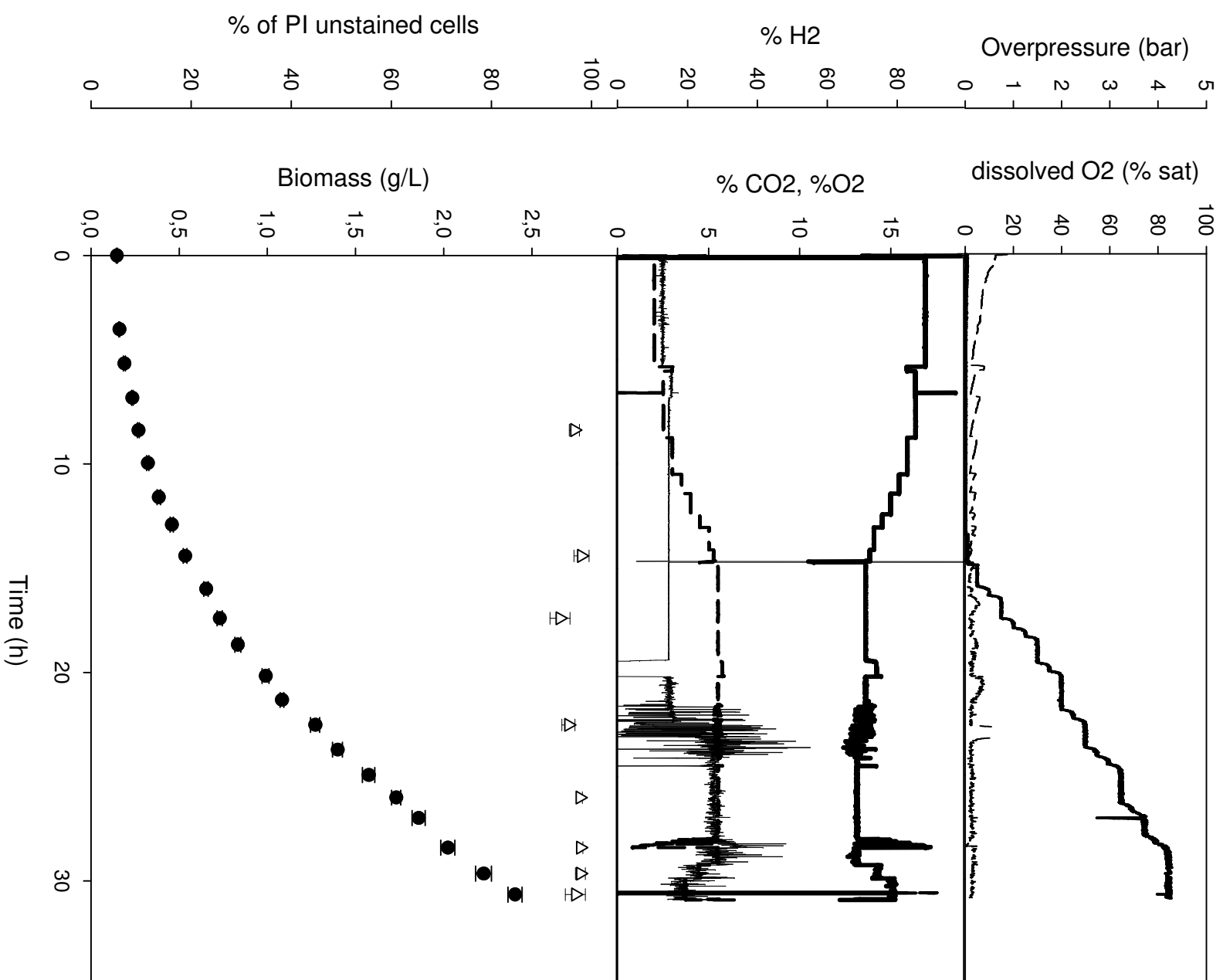


Figure 3

