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ACCEPTED MANUSCRIPT

Biohydrogen production at pH below 3.0: Is it possible?

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Abstract

Biological hydrogen production was investigated in continuous acidogenic reactors fed with sucrose at 30°C without pH control. In the first experimental phase, three reactors were compared: a structured fixed-bed (FB), a granular UASB (UG) and a flocculent UASB (UF-1). They were run at 3.3 h HRT and 33 gCOD L⁻¹d⁻¹ OLR. Hydrogen production occurred throughout the experimental period with an average effluent pH of only 2.8. The FB, UG and UF-1 reactors presented volumetric hydrogen production rates (VHPR) of 95 ± 69, 45 ± 37 and 54 ± 32 mLH₂ L⁻¹h⁻¹, respectively; and H₂ yields (HY) of 1.5 ± 0.8, 0.8 ± 0.6 and 1.2 ± 0.7 molH₂ mol⁻¹ sucrose_{consumed}, respectively. The UF-1 reactor showed intermediate VHPR and HY, but no declining trend, contrary to what was observed in the FB reactor. Thus, aiming at continuous and long-term H₂ production, a flocculent UASB was applied in the second experimental phase. In this phase, the HRT of the acidogenic reactor, which was named UF-2, was raised to 4.6 h, resulting in an OLR of 25 gCOD L⁻¹d⁻¹. The VHPR and the HY increased considerably to 175 ± 44 mLH₂ L⁻¹h⁻¹ and 3.4 ± 0.7 molH₂ mol⁻¹ sucrose_{consumed}, respectively. These improvements were accompanied by greater sucrose removal, higher suspended biomass concentration, less production of lactate and more of acetate, and high ethanol concentration.

Contradicting the current published literature data that reports strong inhibition of H₂ production by dark fermentation at pH less than 4.0, the UF-2 reactor presented stable, long-term H₂ production with satisfactory yields at pH 2.7 on average. 16S rDNA sequencing revealed that two sequences assigned as *Ethanoligenens* and *Clostridium* accounted for over 70% of the microbiota in all the reactors. The non-necessity of adding alkalizing agents and the successful H₂ production under very acid conditions, demonstrated in this study, open a new field of investigation in biological hydrogen production by dark fermentation towards a more sustainable and feasible technology.

Keywords: acidogenic reactor; acid-tolerant bacteria; biohydrogen; dark fermentation; hydrogen; pH

1. Introduction

In recent years, more attention has been given to the potential for hydrogen production by dark fermentation (DF). Hydrogen is produced concomitantly with volatile fatty acids (VFA) through acidogenesis during anaerobic treatment, and its recovery is a way of extracting additional energy in wastewater treatment plants. The technology is still evolving and stable, long-term H₂ production is challenging due to changes in bacterial metabolic pathways and the concomitant existence of H₂-producing and H₂-consuming microorganisms inside the acidogenic reactors. Current efforts are towards optimization of the operating parameters (e.g. reactor designs, environmental conditions, bacterial consortia, substrates) in order to achieve a sustainable H₂ net production.

In the DF processes, no more than 4 mol of H₂ per mol of hexose is attainable due to the production of products other than gas. The foregoing notwithstanding, usual H₂ yields are lower, due to the utilization of the substrate in a variety of pathways that produce less or no H₂ and for biomass growth, also due to microbial H₂ consumption.

Environmental pH plays a crucial role in hydrogen yields. A neutral pH, besides being onerous to maintain, can favour methanogen growth and be detrimental to the achievement of phase separation.

On the other hand, pH values less than 4.5 lead to changes in the metabolic pathways, towards the production of compounds more reduced than the VFA (solvents such as acetone and alcohols, and lactic acid) (Bahl *et al.*, 1982; Lay, 2000; Mizuno *et al.*, 2000; Kim *et al.*, 2004); increased concentrations of undissociated forms of organic acids, which affect microbial growth (Dabrock *et al.*, 1992; Yokoi *et al.*, 1995; Chen *et al.*, 2005; Ruggeri *et al.*, 2015); possible inhibition of hydrogenase activity (Micolucci *et al.*, 2014; Ghimire *et al.*, 2015; Ruggeri *et al.*, 2015; Roy and Das, 2016) as well as ferredoxin's capacity to donate electrons for the protons (Ruggeri *et al.*, 2015). In general, the desirable pH for hydrogen-producing reactors ranges from 4.5 to 6.5. However, even in this pH range, H₂-consuming microorganisms such as homoacetogenic and H₂-oxidizing methanogens can be found (Lee *et al.*, 2010).

The main drawback to controlling the pH in acidogenic reactors lies in the increased costs. Due to the constant CO₂ and acid production, the addition of alkalis to the reactors is usually needed. Ghimire *et al.* (2015) state that the use of an excessive amount of pH regulators can decrease the economics and sustainability of the process, as well as increase the salt concentration of the DF effluents.

The capacity of acid-tolerant facultative or anaerobic bacteria to produce H₂ under extremely acid conditions (pH<3.5) has not yet been investigated in acidogenic reactors, but only in other environments. In the study by Noguchi *et al.* (2010), it was found that live cultures of *Escherichia coli* survived at external pH values of 2.5 and 2.0 due to the activity of the [NiFe]-hydrogenase Hyd-3.

The reduction of H⁺ into H₂ to control the internal pH in extremely acidic environments such as the stomach is a strategy also reported for *Helicobacter pylori* (Bhattacharyya *et al.*, 2000). The capacity to grow in very acid environments has been demonstrated for other H₂-producing bacteria, such as *Sarcina ventriculi* and *Clostridium acidisoli*. *S. ventriculi* is a bacterium found in various environments (soil, mud, rabbit and guinea pig stomach contents, elephant dung, human feces and the surface of cereal seeds) and can grow at pH of 2.0-2.5 (Canale-Parola, 1986). However, Goodwin and Zeikus (1987) found that its metabolism shifted from H₂-acetate to ethanol production when the pH decreased from 7.0 to 3.0. Kuhner *et al.* (2000) first isolated *Clostridium akagii* and *Clostridium*

acidisoli from acid soils (pH ~3.0) and cultured them at pH 3.7-7.1 and 3.6-6.9, respectively. Their capacity to produce H₂ from carbohydrates was demonstrated at pH 5.5 and 6.8, but it was not assayed for other pH values.

Bearing in mind that the application of DF for H₂ recovery is only feasible if the environmental balance is beneficial and the economic costs are kept to a minimum, and, that there is a potential for H₂ production by acid-tolerant bacteria, the acidogenic reactors were run without addition of pH regulators in the present study. As reactor design and the biomass retention mechanism (biofilm, flocs or granules) affect the biological dynamics, and thus net hydrogen production, different configurations of reactors were evaluated.

2. Material and methods

2.1. Reactor configurations and inoculum

An up-flow structured fixed-bed reactor, a granular UASB reactor and a flocculent UASB reactor were used. The reactors were made of acrylic, having internal diameters of 6.3 cm, and with total and working volumes of approx. 2.5 and 2.2 L, respectively (Figure 1). The source of inoculum was granular sludge from a single stage UASB reactor treating poultry slaughterhouse wastewater (Pereiras, São Paulo, Brazil). The granules were completely disrupted with a blender prior to inoculating the structured fixed-bed and flocculent UASB reactors. The structured fixed-bed reactor design (Picanço *et al.*, 2001), as an alternative to the packed-bed reactor, prevents channelling and clogging. Polyethylene cylinders were chosen as the support material in the structured fixed-bed reactor (porosity = 82%), as Ferraz Júnior *et al.* (2015) found that the reactor filled with polyethylene obtained higher H₂ production and yield, also greater abundance of H₂-producing bacteria, as compared to the reactors filled with expanded clay, coal and porous ceramics.

Figure 1 – Schematic diagram of the acidogenic reactors. 1: distribution chamber, 2: reactional zone, 3: headspace, 4: biogas sampling, 5: biogas outlet

The initial concentration of total volatile solids (TVS) was 15 g/l. No sludge pretreatment was used. This allows the survival of non-spore forming H_2 -producers and makes the inoculation more practical and viable.

2.2. Substrate

The reactors were fed with sucrose-based wastewater composed of demerara sugar (Native®) and a nutrient's solution in the following concentrations ($mg\ L^{-1}$): demerara sugar (4450), NH_4Cl (170), $CaCl_2 \cdot 2H_2O$ (8), KH_2PO_4 (37), $MgSO_4 \cdot 4H_2O$ (9), $FeCl_3 \cdot 4H_2O$ (2), $CoCl_2 \cdot 6H_2O$ (2), $MnCl_2 \cdot 4H_2O$ (0.5), $CuCl_2 \cdot 2H_2O$ (0.03), $ZnCl_2$ (0.05), H_3BO_3 (0.05), $(NH_4)_6Mo_7O_{24} \cdot 4H_2O$ (0.09), $Na_2SeO_3 \cdot 5H_2O$ (0.1), $NiCl_2 \cdot 6H_2O$ (0.05), EDTA (1), HCl 36% ($1\ \mu L\ L^{-1}$).

2.3. Operating conditions

In the first experimental phase, in which different reactors were evaluated (Table 1), the mean hydraulic retention time (HRT) was 3.3 h. This corresponded to an organic loading rate (OLR) of $33.1\ gCOD\ L^{-1}d^{-1}$.

One configuration was chosen to be applied in the next experimental phase, in order to keep the investigation on continuous hydrogen production. In this phase, a different start-up was applied: after inoculation, the reactor was operated at HRT in the 2.8-6.1 h range for 80 days. It was verified that higher hydrogen production was obtained at HRT between 4 and 5 h (data not shown). Thereafter, the HRT was adjusted to 4.6 h in Phase 2. This corresponded to an OLR of $25.0\ gCOD\ L^{-1}d^{-1}$.

According to the design and/or inoculum structure and to the experimental phase, the reactors were named as follows: (i) structured fixed-bed reactor: FB; (ii) granular UASB reactor: UG; (iii) flocculent UASB reactors applied in experimental phases 1 and 2: UF-1 and UF-2, respectively (Table 1).

Table 1 – Reactor configurations and operating conditions

Reactor	Reactor design	Inoculum structure	Biomass retention	Experimental phase	HRT - h	OLR - gCOD L ⁻¹ d ⁻¹
FB	Structured fixed-bed	Disaggregated granules	Biofilm and flocs	1	3.3	33.1
UG	UASB	Intact granules	Granules	1	3.3	33.1
UF-1	UASB	Disaggregated granules	Flocs	1	3.3	33.1
UF-2	UASB	Disaggregated granules	Flocs	2	4.6	25.0

The reactors were fed continuously and the temperature was maintained at 30±2 °C. The influent pH was naturally neutral, 6.5 on average, and the pH in the reactors was not controlled.

2.4. Analyses

The biogas flow rate was measured using Milligas counter gas meters (Ritter®). The composition, in terms of H₂, CH₄ and CO₂, was analysed using Shimadzu GC-2010 gas chromatograph with the following specifications: thermal conductivity detector; argon as carrier gas; Carboxen 1010 capillary column; initial detector and injector temperatures of 200 and 230 °C, respectively; oven temperature of 130-135 °C; flow rate of 12 mL min⁻¹; and, sample volume of 300 µL.

Sucrose (glucose and fructose) and organic acids (lactic, formic, acetic, propionic, isobutyric, butyric, isovaleric, valeric) were determined using Shimadzu System UV/DAD (210 nm) high performance liquid chromatography (HPLC) with Refractive Index (in series) detectors, Aminex HPX-87H column, 0.005M H₂SO₄ solution as eluent, flow of 0.5 mL min⁻¹, oven temperature of 43 °C, and 100

μL of sample injection. Ethanol was determined using Shimadzu GC-2010 gas chromatograph with a flame ionization detector (FID), flow of 1.5 mL min⁻¹ with ultra-pure hydrogen as the carrier gas, injector and detector temperature of 250 °C and 280 °C, respectively.

Total COD of the affluent, soluble COD of the effluent (filtered in 1.2 μm membrane) and volatile suspended solids (VSS) concentration in the effluent were analysed according to APHA *et al.* (2005). The pH was measured using a pHmeter (Hach equipment).

Statistical analyses were done using Statistica 13 software. Normal distribution of the results was checked using the Shapiro-Wilk test before applying the other tests. The 95% confidence level was adopted for all tests.

2.5. Theoretical calculations of the percentage of acidified sucrose and of hydrogen yield, by the different metabolic routes

The simplified stoichiometric equations (Equations 1, 2, 3, 4, 5) were used to calculate the molar ratio between sucrose consumed and acids produced ([sucrose]/[acid]) and between hydrogen gas and acids produced ([H₂]/[acid]). These equations show calculated acidified sucrose (in mmol L⁻¹) = S (in mmol sucrose mmol⁻¹ acid) x acid concentration (in mmol acid). It is shown that S = 0.25 via lactate, 0.25 via acetate and/or formate, 0.25 via propionate, 0.50 via butyrate, and 0.50 via valerate. The percentage of acidified sucrose for each acid is its respective calculated acidified sucrose divided by the total calculated acidified sucrose.

To determine the HY percentage, the maximum yield or consumption by each route was calculated, as follows: theoretical HY (in mmol H₂ mmol⁻¹ sucrose_{consumed}) = H (in mmol H₂ mmol⁻¹ acid) x acid yield (in mmol acid mmol⁻¹ sucrose_{consumed}). It is shown that H = 0 via lactate, 2 via acetate and/or formate, -1 via propionate, 2 via butyrate, and -1 via valerate. The HY percentage is the theoretical

HY from each acid divided by the sum of the theoretical HY from acetate and/or formate and butyrate.

Via lactate: $1 \text{ C}_{12}\text{H}_{22}\text{O}_{11} + 1 \text{ H}_2\text{O} = 4 \text{ CH}_3\text{CH}(\text{OH})\text{COO}^- + 4 \text{ H}^+$ (Eq. 1)

Via acetate and/or formate*: $1 \text{ C}_{12}\text{H}_{22}\text{O}_{11} + 5 \text{ H}_2\text{O} = 4 \text{ CH}_3\text{COO}^- + 4 \text{ H}^+ + 4 \text{ CO}_2 + 8 \text{ H}_2$ (Eq. 2)

*In the mixed-acid fermentation, Equation 2 derives from the sum of the reaction of acetate and formate formation ($1 \text{ C}_{12}\text{H}_{22}\text{O}_{11} + 5 \text{ H}_2\text{O} = 4 \text{ CHOO}^- + 4 \text{ CH}_3\text{COO}^- + 8 \text{ H}^+ + 4 \text{ H}_2$) followed by the reaction of formate cleavage ($4 \text{ CHOO}^- + 4 \text{ H}^+ = 4 \text{ CO}_2 + 4 \text{ H}_2$). Since [formate] ~ 0 , only [acetate] was included in the calculations.

Via propionate: $1 \text{ C}_{12}\text{H}_{22}\text{O}_{11} + 4 \text{ H}_2 = 4 \text{ CH}_3\text{CH}_2\text{COO}^- + 4 \text{ H}^+ + 3 \text{ H}_2\text{O}$ (Eq. 3)

Via butyrate: $1 \text{ C}_{12}\text{H}_{22}\text{O}_{11} + \text{H}_2\text{O} = 2 \text{ CH}_3\text{CH}_2\text{CH}_2\text{COO}^- + 2 \text{ H}^+ + 4 \text{ CO}_2 + 4 \text{ H}_2$ (Eq. 4)

Via valerate: $1 \text{ C}_{12}\text{H}_{22}\text{O}_{11} + 2 \text{ H}_2 = 2 \text{ CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{COO}^- + 2 \text{ H}^+ + 2 \text{ CO}_2 + 3 \text{ H}_2\text{O}$ (Eq. 5)

2.6 Molecular analysis

Biomass were collected from different heights from the FB, UG, UF-1 and UF-2 reactors by the end of operation. Cells were separated by centrifugation (6000 g, 10 min, 4 °C). Genomic DNA was extracted and purified using the protocol of Griffiths et al. (2000). The amount and purity of DNA in the extracts were measured by spectrophotometry (Infinite NanoQuant M200, Tecan). The extracted DNA was stored at -20 °C until further use. The 16S rDNA gene V4-5 region was amplified with the forward primer CTTTCCCTACACGACGCTCTTCCGATCTGTGYCAGCMGCCGCGGTAA and the reverse primer GGAGTTCAGACGTGTGCTCTTCCGATCTCCCGYCAATTCMTTTRAGT plus the respective linkers over 30 amplification cycles at an annealing temperature of 65 °C. In a second PCR reactor of 12 cycles, an index sequence was added. The resulting PCR products were purified and loaded onto the Illumina MiSeq cartridge for sequencing of paired 375-380 bp reads. Sequencing-

related work was done at the GeT PlaGe sequencing center of the genotoul life science network in Toulouse, France (get.genotoul.fr). Forward and reverse sequences were retained after assembly and quality checking using a slightly modified version of the Standard Operation Procedure for MiSeq data by Kozich et al. (2013) in Mothur version 1.35.0 (Schloss et al., 2009). SILVA SSU Ref NR 99, release 128, was used for alignment and as taxonomic outline (Pruesse et al., 2012). The sequences found in this study were submitted to the GenBank (accession numbers MF612196-MF613645). For the construction of a phylogenetic tree, the most abundant sequences found in the reactors were then compared with the available sequences in the GenBank database using the BLAST program (Altschul et al., 1990). Phylogenetic analyses of the sequences were performed using the Molecular Evolutionary Genetic Analysis (MEGA7) software (Kumar et al., 2016). Evolutionary distances were based on the Kimura model (Kimura, 1980) and tree reconstruction on the Neighbor-Joining method with bootstrap values calculated from 500 replicate runs.

3. Results and Discussion

3.1. Volumetric hydrogen production rate and biogas composition

Figure 2 shows the volumetric hydrogen production rate (VHPR) and effluent pH of the FB, UG and UF-1 reactors. The mean effluent pH values were 2.8, 2.8 and 2.9 in the FB, UG and UF-1 reactors, respectively. As no buffers, acids or bases were added, pH reduction resulted from the production of organic acids and carbon dioxide. Despite the low pH, hydrogen production occurred throughout the experimental period.

Figure 2 – Volumetric hydrogen production rate and pH in the first experimental phase: (a) FB reactor, (b) UG reactor, (c) UF-1 reactor.

Punctual increases in pH, accompanied by reduction of sucrose removal, organic acid and H₂ production, were observed. This was more noticeable in the UF-1 reactor, when pH values were

above 3.4 on days 30, 80 and 133 (Figure 2). The lowest organic acid concentrations in effluent were also reported on these days and sucrose removal efficiency was null on days 80 and 133. On days 23, 29, 59, 60, 79 and 130, there were feeding problems in the UF-1 reactor due to clogging of tubes. It was likely that feeding reduction or interruption led to biomass decay, as could be inferred by the lower visible turbidity of the medium and reduction in effluent VSS concentrations after these events. Nevertheless, pH above 3.0 accompanied by a drastically reduced sucrose conversion, was also observed in the UG reactor notably on days 32, 85 and 136, and in the FB reactor on day 85. Consequently, effluent pH was increased due to dilution of medium with non-consumed affluent.

The VHPR were equivalent to: $95 \pm 69 \text{ mLH}_2 \text{ L}^{-1}\text{h}^{-1}$ in the FB reactor, $45 \pm 37 \text{ mLH}_2 \text{ L}^{-1}\text{h}^{-1}$ in the UG reactor, and $54 \pm 32 \text{ mLH}_2 \text{ L}^{-1}\text{h}^{-1}$ in the UF-1 reactor. The non-parametric Kruskal-Wallis ANOVA by Ranks test showed statistically significant differences regarding H_2 production ($p\text{-value} = 0.006$). Further analysis of multiple comparisons of mean ranks for all groups showed that H_2 production in reactors UG and UF-1 was not significantly different ($p\text{-value} = 0.575$). As shown in Figure 2, although the FB reactor achieved the highest VHPR at the beginning of the operation, it tended to decrease during the experimental period. This did not occur in the UG and UF-1 reactors. The UF-1 reactor showed superior stability during the entire period of operation, as indicated by its VHPR data. These were the only data that presented normal distribution.

A possible explanation for the higher initial VHPR in the FB reactor could be the lower biomass wash-out, owing to the presence of the support material. Biomass was observed to be "trapped" in the polyethylene cylinders, although it did not form a thick biofilm. The flocs formed in the sludge bed at the bottom of the FB reactor were visually larger than those from the UF-1 reactor. This was probably due to the shear stress and physical selection caused by the support material, which retained larger particles, while the smaller ones passed easily through the pores. Low interspecies distances are a key point of efficient interspecies hydrogen transfer between acetogenic bacteria and hydrogenotrophic methanogenic archaea in anaerobic aggregates, biofilms and granules (MacLeod et al., 1990; Davey and O'toole, 2000; Hulshoff Pol et al., 2004; Felchner-Zwirello et al. 2013). As stated by Dinamarca

et al. (2011), this mechanism also can play a relevant role in non-methanogenic mixed cultures, through hydrogen transfer between hydrogen producers and consumers, limiting sustainable hydrogen production due to homoacetogenesis. Therefore, in the present study, the biomass agglutination in larger flocs, granules and biofilm could have had an adverse effect on long-term H_2 . Acetate formation was observed to increase from day 77 in the FB reactor. However, this was not followed by an increase in hydrogen production, which can be an indicator of homoacetogenic activity. Penteadó et al. (2013) studied seven structured fixed-bed reactors having different sources of inoculum, fed with sucrose. VHPR decreased over time in all reactors, and it was observed that HY decreased as the percentage of acetic acid produced by homoacetogenesis increased. In addition, the filling in the FB and UG reactors with support material and granules, respectively, may have hindered the escape of the produced biogas, increasing the H_2 partial pressure in the medium, which inhibits its own production (Sikora *et al.*, 2013).

The granules inside the UG reactor were originally dark colored with an average diameter of 2.1 mm. They became whitened and smaller, with an average diameter of 1.5 mm by the end of operation. Floc formation and suspended biomass growth were also observed. On the 135th day of operation, the UG reactor lost most of its biomass due to a remarkable wash-out of the granules. The granule flotation likely occurred due to the adherence of gas bubbles to their surfaces, and reduction of their inner densities. This assumption is based on the fact that the environmental conditions were not favourable for the maintenance of the methanogenic microorganisms, leading to biomass decay in the inner layers of the granules. Then, in the outer layers, the granules became most colonized by acidogenic bacteria that survived in the acid environment. The increasing substitution of mixed-consortia granules by specific acidogenic bacteria inside the UG reactor probably had a positive effect on H_2 production, as indicated by the increased VHPR at the end of operation (Figure 2).

In the UF-1 reactor, it is probable that faster biomass decay and washing-out occurred at the beginning of the operation due to the larger contact surface of the biomass with the medium and the initial absence of a biomass retention mechanism. This may be the reason that H_2 production started

later in this reactor and with less intensity. Reyes *et al.* (2012) also observed a delay in H₂ production in the reactors inoculated with disintegrated granules. This production started after about 40-70 hours of continuous operation, compared to the reactors inoculated with intact granules, in which H₂ production started within the first 12 hours. On the other hand, the selection of bacteria resistant to the adverse conditions (low pH and high organic acids concentration) as well as the increasing biomass concentration due to the self-flocculation phenomenon provided a superior stability to the UF-1 reactor. The higher selectivity of the desired bacteria from the disaggregated granules was verified by Reyes *et al.* (2012), who found that this form of inoculation resulted in greater specific hydrogenogenic activity compared to that from intact granules.

The effluent VSS concentrations were (in mg L⁻¹): 89.9 ± 68.4 , 101.7 ± 95.9 and 90.9 ± 63.4 in the FB, UG and UF-1 reactors, respectively. In each reactor, the effluent VSS concentration correlated positively with the VHPR according to the Spearman non-parametric test ($\alpha = 5\%$). The values of the R correlation coefficients were 0.45 ($p = 0.0063$), 0.56 ($p = 0.0014$) and 0.40 ($p = 0.0198$) for the FB, UG and UF-1 reactors, respectively.

For the application in the second experimental phase, the UG reactor was considered less advantageous as it had the lowest VHPR and HY. The FB reactor, however, showed the highest VHPR and HY mean values, although with a tendency toward performance decrease over time. The UF-1 reactor showed intermediate VHPR and HY values, but no declining trend was observed. From the 80th day of operation, the H₂ yield in the UF-1 reactor also showed progressive improvement, contrasted to the FB reactor. Thus, aiming at continuous and long-term H₂ production, this configuration seemed to be the most adequate among those studied. It is also pertinent that the flocculated UASB reactor design has the greatest potential to use the entire reactor volume to be filled with active biomass, thus maximizing the reactor space utility and virtually increasing cell density in the reactor, without formation of close microbial associations such as biofilms and granules. For these reasons, this configuration was chosen for the second experimental phase.

The flocculated UASB applied in Phase 2 was identified as the UF-2 reactor and was operated at a higher HRT (4.6 h) and lower OLR (25.0 gCOD L⁻¹d⁻¹). It obtained constant and stable H₂ production, and achieved significant improvement over the previous experimental phase (Figure 3). The VHPR was very satisfactory, corresponding to 175 ± 44 mLH₂ L⁻¹h⁻¹. Continuous acid and CO₂ production in the reactor led to strong acidification of the effluent, and pH was self-adjusted to values consistently less than 3.0, with an average value of 2.7 (Figure 3). The H₂ production in the UF-2 reactor presented normal distribution and was statistically higher, according to Kolmogorov-Smirnov test (p-value < 0.001), compared to the other reactors.

Figure 3 – Volumetric hydrogen production rate and pH in the second experimental phase: UF-2 reactor

The growth of suspended biomass was much more noticeable than in the reactors during Phase 1, achieving an effluent concentration of 295 ± 275 mgVSS L⁻¹. There was also a significant correlation of VSS with VHPR at the 5% significance level (Spearman R = 0.33).

The biogas of the FB, UG, UF-1 and UF-2 reactors presented a H₂ content equal to (%): 59.6 ± 11.0, 62.1 ± 10.8, 62.2 ± 7.1 and 59.8 ± 5.9, respectively. The percentage of hydrogen in the biogas was not significantly different, according to the Kruskal-Wallis ANOVA (FB, UG, UF-1 reactors) and the Kolmogorov-Smirnov tests (UF-2 reactor vs. FB, UG, UF-1 reactors). Methane was not detected in the biogas in any of the reactors. This leads to the assumption that the environmental conditions established by the pH self-adjustment and low HRT were sufficient to completely inhibit methanogenesis. Operating with extreme pH values seems to be an efficient strategy for avoiding methanogenic activity, as verified by Wang et al. (2015), studying hydrogen production in waste activated sludge at pH 10.

3.2. Hydrogen yield and sucrose removal

The results of sucrose removal and HY are plotted in box and whisker graphics (Figures 4 and 5), that show the distribution of data into quartiles, highlighting the mean (X). The lines extending vertically indicate variability outside the upper and lower quartiles.

Figure 4 – Sucrose removal

Figure 5 – H₂ yield

The mean sucrose removal in the UF-2 reactor was 81% while, in the FB, UG and UF-1 reactors, it was 64, 67 and 56%, respectively (Figure 4). However, as the OLR applied in the UF-2 reactor was less than in the other reactors, the mean volumetric sucrose removal rate was in the same range as the other reactors: 2.22, 2.38, 1.91 and 2.16 mmol sucrose_{consumed} L⁻¹h⁻¹ in the FB, UG, UF-1 and UF-2 reactors, respectively. Thus, the substantial increase in VHPR obtained in the UF-2 reactor was mainly due to the improvement in the H₂ yield. The mean HY of 1.50, 0.76 and 1.19 mol H₂ mol⁻¹ sucrose_{consumed} obtained in the FB, UG and UF-1 reactors, respectively, was surpassed by a level of 3.35 mol H₂ mol⁻¹ sucrose_{consumed} obtained in the UF-2 reactor (Figure 5). Statistical analyses revealed that H₂ yield in Phase 1 differed significantly among reactors (Kruskal-Wallis ANOVA, p-value = 0.002); however, the difference between the FB and UF-1 reactors was not significant (multiple comparisons of mean ranks, p-value = 0.413). Nevertheless, H₂ yield obtained in the UF-2 reactor was statistically higher than that obtained in the FB, UG and UF-1 reactors (Kolmogorov-Smirnov test, p-value < 0.001).

Table 2 shows the organic acid concentrations, and the percentages of respective acidified sucrose and H₂ yield. Comparing the effluent organic acid composition of the UF-2 reactor to the other reactors, it was concluded that there was a shift from less lactate to more acetate production, thus accounting for UF-2 reactor superior performance. Lactate production involves the consumption of NADH and pyruvate, reducing the potential production of H₂ by both the NADH-pathway and, mainly, substrate competition due to pyruvate consumption. On the other hand, the acetate production represented by

the reactions in Equation 2 is desired for both Clostridial- and Enterobacterial-type fermentation (mixed acid fermentation), as the acetate route provides the highest H₂ yield in Clostridial-type fermentation and acetate is produced along with formate in Enterobacterial-type fermentation. Since H₂ can be produced from formate cleavage, acetate indicates that the formate route took place in the mixed acid fermentation. Also, the concentrations of propionate and valerate, which are produced at the expense of H₂ consumption, were lower in the UF-2 reactor.

Table 2 – Organic acid concentrations and the respective percentages of acidified sucrose and H₂ yield

Parameter	Reactor	lactate	formate	acetate	propionate	butyrate	Valerate
mean (sd) - mmol L⁻¹	FB	5.8 (±5.5)	0.2 (±0.2)	3.9 (±3.9)	1.8 (±3.0)	1.0 (±1.6)	0.6 (±0.5)
	UG	8.0 (±7.5)	0.1 (±0.1)	3.7 (±2.0)	4.3 (±5.7)	1.7 (±2.0)	0.5 (±0.4)
	UF-1	7.0 (±6.8)	0.1 (±0.1)	3.2 (±1.9)	2.2 (±3.2)	0.7 (±1.0)	0.6 (±0.4)
	UF-2	3.6 (±1.3)	0.2 (±0.1)	7.8 (±2.7)	1.1 (±0.6)	1.1 (±0.6)	0.4 (±0.2)
calculated acidified sucrose (sd) - %	FB	42 (±33)	26 (±20)	11 (±17)	14 (±20)	7 (±5)	
	UG	44 (±33)	20 (±10)	18 (±21)	14 (±12)	5 (±3)	
	UF-1	44 (±32)	24 (±14)	16 (±22)	9 (±13)	7 (±4)	
	UF-2	25 (±8)	50 (±7)	7 (±3)	14 (±4)	5 (±2)	
calculated HY (sd) - %	FB	0 (±0)	74 (±26)	-13 (±28)	26 (±26)	-6 (±4)	
	UG	0 (±0)	75 (±19)	-27 (±33)	25 (±19)	-5 (±3)	
	UF-1	0 (±0)	81 (±19)	-25 (±38)	19 (±19)	-8 (±5)	
	UF-2	0 (±0)	88 (±3)	-6 (±3)	12 (±3)	-2 (±1)	

The reduced OLR and possible higher biomass concentration (indicated by higher VSS concentrations) in the UF-2 reactor resulted in lower specific organic loading (food/ microorganism ratio). Thus, the efficiency of the substrate conversion was increased, which was verified by the greater sucrose removal.

The overloading in Phase 1 seemed to be the main factor accounting for reduced hydrogen yields. According to Cohen et al. (1984), lactate pathway is energetically less favourable and its formation in acid digestion could be associated with an imbalance between electron donating and electron

accepting reactions, in conditions of high accessibility of the substrate, such as low HRT and shock loading. Apart from the influence of organic loading on metabolic routes, the *Lactobacillus* genus was found in greater relative abundance in the reactors of Phase 1 (section 3.3).

Propionate concentration was higher in the UG reactor (Table 2). Butyrate production was similar among the reactors, suggesting that activity of the butyrate-producers, was not severely affected by the different conditions. Since propionate production is not likely to occur under very acid conditions (Wang et al., 2006), it was likely that the bacteria arrangement in the granules kept the medium pH in microcolonies higher than in the external environment, allowing the activity of propionate-producing microorganisms.

The results presented in Table 2 are only for comparison, based on the equations shown in section 2.5. Many other pathways could have taken place in the reactors. The calculated acidified sucrose corresponded to 55%, 66%, 64% and 39% of the consumed sucrose in the FB, UG, UF-1 and UF-2 reactors, respectively. Naturally, part of the sucrose could have been used for cellular growth. Moreover, it is likely that other pathways leading to hydrogen formation were also present. The mean calculated HY in the FB, UG, UF-1 and UF-2 reactors was equivalent to 1.00, 0.78, 0.85 and 1.62 $\text{mmolH}_2 \text{ mmol}^{-1} \text{ sucrose}_{\text{consumed}}$, respectively, which corresponded to 67%, 103%, 71% and 48% of the measured HY, respectively.

Ethanol was measured in the effluent from the UF-2 reactor. Unfortunately, this measurement was not performed in the other reactors, due to technical problems. The average concentration was 11 mmol L^{-1} , which accounted for 27% of total soluble COD effluent, while COD from organic acids and sucrose were 33% and 24%, respectively. In the FB, UG and UF-1 reactors COD from organic acids and sucrose accounted for a greater proportion, being respectively: 33% and 41% (FB), 45% and 37% (UG), 32% and 53% (UF-1) of total soluble COD effluent. From COD balance analysis, it was inferred that ethanol concentrations in the reactors of Phase 1 did not reach such high levels as were reached in the UF-2 reactor.

The high concentrations of ethanol in the UF-2 reactor is contrary to what was first expected, because ethanol is a more reduced compound than organic acids and its formation is usually associated with HY reduction. Nevertheless, some pathways are proposed for ethanol formation along with hydrogen (Xu et al., 2008; Lee et al., 2009). Equation 6 shows the reaction proposed by Hwang et al. (2004) for bacterial conversion of glucose into ethanol, acetate and hydrogen. Although the ethanol-acetate pathway yields less hydrogen than the acetate-pathway (Equation 2), the hydrogen yield could be 4.0 mol of H₂ per mol of sucrose consumed (which is in the range achieved in the UF-2 reactor), considering sucrose as substrate. Since the hydrogen yield per mol of acetate produced is the same of as shown in Equation 2 (H = 2 mmol H₂ mmol⁻¹ acetate), the assumption of this reaction would not change the HY percentage values depicted in Table 2, whereas the acidified sucrose percentage would be higher from acetate (S = 0.50 mmol sucrose mmol⁻¹ acetate). However, as ethanol was not analysed in all effluents, however, it was not possible to account for it in the estimations presented in Table 2. Ethanol formation is in agreement with the findings of sequencing analyses (section 3.3), that revealed an abundance of microorganisms affiliated with *Ethanoligenens harbinense*.

3.3 Structure and composition of the microbial community in the FB, UG, UF-1 and UF-2 reactors

There were 123838 partial 16S ribosomal DNA gene sequences obtained from the microbial sequencing, of which 94-99% were assigned to the phylum Firmicutes in the reactors versus 17% in the inoculum. Sequences assigned to the domain Archaea were 9.6% of the inoculum and less than 0.1% of the reactors, indicating that the conditions applied in this study dispensed with an inoculum pretreatment. Based on the operational taxonomic units (OTU), the Shannon-diversity index was reduced from 4.0 in the inoculum to 1.2, 1.3, 1.4 and 0.7 in the FB, UG, UF-1 and UF-2 reactors, respectively, by the end of operation. The self-established harsh environment likely played a key role in the reduction of biomass diversity. An annotated abundance relative description is given in Table 3.

Representative sequences (abundance of more or equal to 1.0%) were selected from the acidogenic reactors to infer a phylogenetic tree (Figure 6).

According to the results of 16 rDNA sequencing, the main emerging classes were related to Bacilli and Clostridia, which represented approximately 23% and 74%, respectively, of the total sequences in the reactors of Phase 1, and 1% and 99% of the total sequences in the UF-2 reactor. Only two sequences, represented by OTU0002 and OTU0003, accounted for more than 70% of the total bacteria (Table 3). The alignment of the sequence of OTU0002 (*Ethanoligenes*) in BLAST revealed an identity of 99% to the *Ethanoligenens harbinense* strain YUAN-3. The same procedure applied to OTU0003 revealed it is 98% affiliated with *Clostridium acidisoli* (Figure 6). These results are in agreement with the literature that reports the ability of both *Ethanoligenens harbinense* and *Clostridium acidisoli* to grow and produce hydrogen under very acid conditions; specifically, pH below 4.0 (Kuhner et al. 2000; Xing et al., 2008; Carosia et al., 2017; Zhao et al., 2017). However, this has never been demonstrated for pH below 3.0.

Although the microbial structure is very similar among the reactors of Phase 1, it is not possible to conclude that biomass retention mechanism does not affect microbial composition, because the samples were only analyzed by the end of operation. As discussed in section 3.1, considerable suspended biomass grew in the FB and UG reactors, the latter as a consequence of granule wash-out and disruption. In the UF-2 reactor, the relative abundance of sequences affiliated with *E. harbinense* was the highest, corresponding to 81%. From these results, it is inferred that *E. harbinense* played the most relevant role in the reactor performance. However, the differences in terms of relative abundance should be interpreted with caution, considering that the 16S sequencing technique is subjected to errors in terms of quantification (Haas et al., 2011), the efficiency of DNA extraction can interfere with the results, and the microorganisms found were not necessarily active. While most of *Clostridium*, including *C. acidisoli* are able to sporulate (Kuhner et al., 2000), *Ethanoligenens* is not (Xing et al., 2006). Therefore, the high relative abundance of sequences related to *Clostridium* does

483 not mean that they were active in the same proportion. Also, the absolute abundance of each
484 microorganism is very relevant to the performance of the reactors, since the efficiency of sucrose
485 consumption was associated with increased production and yield of hydrogen (section 3.2).

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Table 3 – Comparative study of 16S rDNA sequencing (V4-5 region) using SINA (v1.2.11). Relative abundance > 1% is shown for the FB, UG, UF-1 and UF-2 reactors; and > 5% for the inoculum

Domain	OTU	Phylum	Class	Order	Family	Genus	Inoculum*	FB	UG	UF-1	UF-2
Bacteria	OTU0002	Firmicutes	Clostridia	Clostridiales	Ruminococcaceae	<i>Ethanoligenens</i>	0%	40%	43%	41%	81%
	OTU0003	Firmicutes	Clostridia	Clostridiales	Clostridiaceae_1	<i>unclassified</i>	0%	31%	31%	35%	15%
	OTU0009	Firmicutes	Bacilli	Lactobacillales	Lactobacillaceae	<i>Lactobacillus</i>	0%	27%	20%	13%	0%
	OTU0023	Firmicutes	Bacilli	Bacillales	Sporolactobacillaceae	<i>Sporolactobacillus</i>	0%	0%	3%	2%	1%
	OTU0001	Firmicutes	Bacilli	Lactobacillales	Lactobacillaceae	<i>unclassified</i>	0%	0%	0%	7%	0%
	OTU0171	Firmicutes	Negativicutes	Selenomonadales	Veillonellaceae	<i>Pectinatus</i>	0%	0%	0%	0%	1%
	OTU0008	Bacteroidetes	vadinHA17	unclassified	unclassified	<i>unclassified</i>	16%	0%	0%	0%	0%
	OTU0014	Bacteroidetes	vadinHA17	unclassified	unclassified	<i>unclassified</i>	6%	0%	0%	0%	0%
	OTU0146	Firmicutes	Clostridia	Clostridiales	Family XI	<i>Tissierella</i>	5%	0%	0%	0%	0%
Archaea	Otu002	Euryarchaeota	Methanomicrobia	Methanosarcinales	Methanosaetaceae	<i>Methanosaeta</i>	85%	0%	0%	0%	0%
	Otu004	Euryarchaeota	Methanobacteriales	Methanobacteriaceae	Methanobacterium	<i>Methanobacterium</i>	5%	0%	0%	0%	0%

* Domain *Bacteria* and *Archaea* represented 90.4% and 9.6% of total sequences, respectively, in the inoculum.

Figure 6 – Consensus phylogenetic tree based on 16S rDNA for bacteria domain obtained from the highly abundant OTUs found in the reactors. The tree is drawn to scale, with branch lengths in the same units as those of the evolutionary distances used to infer the phylogenetic tree. There was a total of 373 positions in the final dataset. Outgroup: *Methanosarcina acetivorans*.

C. acidisoli was isolated from acidic peat-bog soil and was grown at pH 3.6-6.9, with no distinct optimum between pH 3.6-6.6, in a temperature range of 5-37 °C, with an optimum of 25-30 °C (Kuhner et al., 2000). At pH 4.0, 5.5 and 6.5, glucose fermentation yielded lactate, acetate, butyrate, H₂ and CO₂ as end-products. At pH 5.5, the molar ratio of H₂ to lactate, acetate and butyrate produced was 6.4, 4.1 and 3.6, respectively, and the HY was 1.8 mmolH₂ mmol⁻¹ glucose_{consumed}. Lee et al. (2009) found great abundance of a species affiliated with *Clostridium* sp. HPB-16, which is phylogenetically close to *C. acidisoli*, during batch fermentation with hydrogen production at final pH of 3.5. Acetate and butyrate were the dominant organic products. These authors assumed that hydrogen was formed by the pyruvate decarboxylation-ferredoxin-hydrogenase pathway, which is the common mechanism for H₂ formation by the *Clostridium* and *Ethanoligenens* species.

Xing et al. (2006) isolated *E. harbinense* YUAN-3 from anaerobic activated sludge of molasses wastewater. They found that it grows in the pH range 3.5-9.0 at 20-44 °C, and the optima for growing were pH 4.5-5.0 and 35 °C. Acetate, ethanol, hydrogen and carbon dioxide were formed as end products of glucose fermentation. At 35 °C a hydrogen yield up to 2.8 molH₂ mol⁻¹ glucose was achieved, along with production of 1.1 mol ethanol and 0.7 mol acetate per mol of glucose. In the UF-2 reactor, the mean production was 1.1 mol ethanol per mol of sucrose (= 0.6 per mol of hexose) and 0.8 mol acetate per mol of sucrose (= 0.4 mol per mol of hexose). The differences in the yields of ethanol and acetate were expected because the fermentation in the UF-2 reactor was carried out by a microbial consortium, which means that many more pathways were possible, and relatively high amounts of lactate were formed (section 3.2). Nevertheless, the molar proportion of ethanol and acetate is similar between the UF-2 reactor (= 1.4 ethanol: acetate) and that reported by Xing et al.

(2006) (= 1.6 ethanol: acetate). Since the maximum achieved hydrogen yield with the pure culture of *E. harbinense* (Xing et al. 2006) was higher than the theoretical yield depicted in Equation 6 (section 3.2), it is probable that this bacterium is able to produce hydrogen and ethanol by pathways other than ethanol-acetate fermentation. Xu et al. (2008) also found an HY higher than 2 molH₂ mol⁻¹ glucose with the *Ethanoligenens harbinense* B49 strain. They suggested oxidative decarboxylation of pyruvate as the possible route for the hydrogen production observed, in accordance with Lee et al. (2009). However, the ethanol-type hydrogen production mechanism by *E. harbinense* is still unclear (Zhao et al., 2017).

Lactobacillus sp. ranged from 20% to 27% in the reactors of Phase 1 and was less than 1% in the UF-2 reactor. The most representative sequence of *Lactobacillus* was affiliated with *L. nagelii* (Figure 6), which is characterized as producing lactic acid from glucose without gas formation (Edwards *et al.*, 2000). Then, it is probable that the presence of *Lactobacillus* in the FB, UG and UF-1 reactors contributed to higher lactic acid formation and less hydrogen yield, due to the reduction of pyruvate availability for the H₂-producing pathways (section 3.2). The excretion of extracellular polymeric substances (EPS) by lactic acid bacteria protects them against hostile environments and favors the formation of flocs and biofilm (Rafrafi et al. 2013), which may have implied competitive advantages at higher OLR.

The presence of *Pectinatus* sp. (OTU0171) in the UF-2 reactor probably is associated with alcoholic fermentation, because this genus is usually found in beer spoilage (Chihib and Toloan, 1999).

3.4. Interaction among performance evaluation parameters

Table 4 shows the overall results obtained, indicating the minimums, maximums, means, standard deviations (SD) and coefficients of variation (CV).

Table 4 – Performance evaluation parameters of all reactors

Parameter	Reactor	Minimum	Maximum	Mean	SD	CV
VHPR - mLH₂ L⁻¹h⁻¹	FB	4.9	259.3	94.9	68.6	72%
	UG	4.0	171.4	44.7	37.5	84%
	UF-1	0.4	114.0	53.7	32.2	60%
	UF-2	92.3	300.8	175.2	43.9	25%
H₂ in biogas - %	FB	40.5	85.4	59.6	11.0	18%
	UG	38.6	82.1	62.1	10.8	17%
	UF-1	47.5	77.5	62.2	7.1	11%
	UF-2	48.4	75.9	59.8	5.9	10%
HY - molH₂ mol⁻¹ sucrose_{consumed}	FB	0.10	3.16	1.50	0.83	55%
	UG	0.06	2.47	0.76	0.56	74%
	UF-1	0.11	3.05	1.19	0.71	60%
	UF-2	1.63	4.94	3.35	0.68	20%
Sucrose removal - %	FB	0.0	100.0	64.3	23.0	36%
	UG	18.7	95.7	66.8	21.4	32%
	UF-1	0.0	90.8	53.1	19.1	36%
	UF-2	56.1	99.7	80.3	9.9	12%

The VHPR and HY improvements in the UF-2 reactor are noteworthy, with respect to the others. These improvements were attributed mainly to the HRT increasing from 3.3 to 4.6 h and, therefore, the OLR decreasing from 33.1 to 25.0 gCOD L⁻¹d⁻¹, since these were the only operational parameters changed intentionally. Several operating indicators accompanied the UF-2 improvement in H₂ production. These indicators include: increased VSS concentration; higher sucrose removal; less production of lactate and more of acetate, and high production of ethanol; pH always below 3.0; and, longer chains of rods. Figure 7 presents a proposed model of the relationship among these parameters that led to higher H₂ production in the UF-2 reactor.

Figure 7 – Proposed model to explain changes in the UF-2 reactor that led to increased H₂ production

Based on the proposed model, we suggest that the increased HRT led to greater removal of sucrose, due to the longer contact time between the substrate and the biomass, and to a higher VSS

concentration (biomass) that resulted from the lower wash-out and longer time for bacterial growth. The higher HRT allowed the formation of long chains of bacteria, increasing their adaptability to the harsh environmental conditions (low pH) and contributing to the increased biomass in the reactor. The growth of acid tolerant bacteria such as *Ethanoligenens* was favoured and the competitive advantage of *Lactobacillus* was reduced. The higher VSS and HRT resulted in a lower specific organic loading rate, which enhanced the sucrose removal efficiency. The increased sucrose removal resulted in higher concentrations of fermentation products, such as acids, CO₂ and H₂. This latter directly reflected in higher VHPR. The high levels of acids and CO₂/ carbonic acid caused a reduction in the pH of the medium. The maintenance of a very acid environment and less relative abundance of *Lactobacillus* resulted in reduced lactate formation. The increased pyruvate availability to other H₂-producing pathways, such as acetate and ethanol formation, thus increased hydrogen yield and production.

On the other hand, increasing HRT over the suitable values is not recommended as it leads to OLR reduction. In addition to increasing reactor volume requirements, this can reduce volumetric substrate removal rates, reducing the attainable VHPR. Very low OLR can also lead to cellular decay, reducing the biomass concentration. In addition, reduced HRT values can increase the pH (through the consumption and release of CO₂, H⁺ and acids) and the H₂ in the medium, due to the mass transfer reduction caused the less turbulence. Increasing the pH and H₂ in the liquid medium then favours the growth of H₂-consuming bacteria; and, it can reduce the competitive advantage of the H₂-producing bacteria tolerant to very acid conditions.

3.5. Comparative studies

Hydrogen production in extremely acidic environments, average pH of 2.8 in the FB, UG, and UF-1 reactors, and of 2.7 in the UF-2 reactor, was unexpected. Extensive data in the literature indicate drastic reduction or cessation of hydrogen production by dark fermentation at pH values below 4.5-4.0 (Yokoi *et al.*, 1995; Lay, 2000; Mizuno *et al.*, 2000; Lee *et al.*, 2002; Kim *et al.*, 2004; Liu and

Shen, 2004; Hwang *et al.*, 2004; Chen *et al.*, 2005; Liu *et al.*, 2006; Chojnacka *et al.*, 2011; Ruggeri *et al.*, 2015). It is only in specific cases, in continuous acidogenic reactors, that H₂ production at pH values below 4.0 is reported (Xing *et al.*, 2008; Tähti *et al.*, 2013, Carosia *et al.*, 2017). The capacity of *Ethanoligenens harbinense* strain YUAN-3 to produce H₂ was evaluated by Xing *et al.* (2008) in a continuous stirred reactor at 35 °C for 21 days. The pH value was kept above 3.5 by a pH controller and they observed that H₂ production was not severely affected when the pH reached the minimum values (i.e., around 3.6), obtaining HY of approx. 1.5 molH₂ mol⁻¹glucose. Carosia *et al.* (2017) found bacteria similar to *Ethanoligenens harbinense* to be dominant bacteria in H₂-producing anaerobic fluidized bed reactors, inoculated with heat-treated sludge. Although buffers (hydrochloric acid and sodium bicarbonate) were added, effluent pH was approximately 3.7, and the optimum HY obtained was 0.76 molH₂ mol⁻¹glucose. Tähti *et al.* (2013) used an extreme thermophilic (70 °C) UASB reactor for H₂ production from glucose by mixed culture. However, a low HY was obtained, equivalent to 0.73 mol mol⁻¹glucose_{added}, which was accompanied by a decrease in pH to around 3.7. In the present study, despite the lowest pH values already being reported, the HY and VHPR obtained are in the highest-range. For comparison purposes, Table 5 shows the results obtained in the UF-2 reactor with results from other studies applying continuous hydrogen-producing reactors fed with sucrose-based wastewater, in the mesophilic range.

Table 5 – Comparison of hydrogen production in continuous acidogenic reactors using sucrose as substrate

Reactor type	OLR - gCOD L ⁻¹ d ⁻¹	Effluent pH	Temp - °C	H ₂ in biogas - %	VHPR* - mL H ₂ L ⁻¹ h ⁻¹	HY - mol H ₂ mol ⁻¹ sucrose	Ref.
stirred tank	48.6	5.5	26	63	542	3.9	Fang <i>et al.</i> (2002)
stirred tank	80	5.25	35	55	506	2.3	Kyazze <i>et al.</i> (2006)
granular UASB	7.1 - 37.4/ 8.5 - 128	4.4	38	57 - 37/ 44 - 42	50 - 190/ 33 - 202	2.9 - 2.0/ 1.6 - 1.0	Yu and Mu (2006)
granular UASB	4.4 - 30	4.0	30	26 - 50	4 - 122	0.5 - 3.3	Zhao <i>et al.</i> (2008)
UASB	12	4 - 4.5	35	45 (approx.)	12 (approx.)	0.3	Wang and Li (2010)
fixed-bed	24	4.4	25	46 - 56	73 - 125	0.9 - 1.4	Lima and Zaiat

							(2012)
fixed-bed	24	4.8	25	54 - 62	15.1 – 61.6	0.7 – 2.1	Penteado <i>et al.</i> (2013)
granular UASB	21.6	4.0	36	40 (approx.)	92 (approx.)	1.6 (approx.)	Ning <i>et al.</i> (2013)
structured fixed-bed	24.0	6.5	25	70	12 - 25	0.4 - 0.6	Anzola-Rojas and Zaiat, (2016)
flocculent UASB (UF-2)	25.0	2.7	30	60	175	3.4	This study

*The reference conditions adopted were 25 °C and 1 atm.

These results indicate that the formation of a very acidic environment allowed the growth of acid-tolerant bacteria that were able to produce H₂ under very acid conditions, especially *Clostridium* sp. and *Ethanoligenens* sp.

4. Conclusions

This study stands out as the first to demonstrate the real possibility for continuous, long-term, stable H₂ production at pH below 3.0, with a mean yield of 3.4 mols of H₂ per mol of sucrose consumed. Proper HRT and OLR were crucial for enhancing hydrogen production. This was associated with increased sucrose consumption, reduced lactate formation, high acetate and ethanol concentrations, reduction of relative abundance of *Lactobacillus* sp. and increase of *Ethanoligenens* sp.

The operating requirements were kept at minimum and the non-pH control, along with the production of H₂ in extremely acid environments, presents several operating and economic advantages, including: the non-addition of alkalizing agents, which contributes to reduction of the costs; elimination of the demand for sludge pretreatment, due to the naturally acid environment; and, the non-necessity of constant sludge removal, since higher biomass concentration leads to enhanced H₂ production. These results open a new field of investigation in biological hydrogen production by dark fermentation towards a more sustainable and feasible technology.

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References

- Anzola-Rojas M.D.P., Zaiat M., 2016. A novel anaerobic down-flow structured-bed reactor for long-term stable H₂ energy production from wastewater. *Journal of Chemical Technology and Biotechnology*, 91, 1551-1561.
- Altschul S.F., Gish W., Miller W., Myers E.W., Lipman D.J., 1990. Basic local alignment search tool. *Journal of Molecular Biology*, 215, 403-410.
- APHA, AWWA, WEF, 2005. Standards methods for the examination of water and wastewater (21st ed.). American Public Health Association/ American Water Works Association/ Water Environmental Federation, Washington DC, USA.
- Bahl H., Andersch W., Gottschalk G., 1982. Continuous production of acetone and butanol by *Clostridium acetobutylicum* in a two-stage phosphate limited chemostat. *European Journal of Applied Microbiology and Biotechnology*, 15, 201-205.
- Bhattacharyya S., Krishnarnurthy P., Shaker R., Staff D., Vakil N., Phadnis S.H., Dunn B.E., 2000. *Helicobacter pylori* requires active hydrogenase enzyme to survive under acidic conditions. *Gastroenterology*, 118(4), A729.

- 657 Canale-Parola E., 1986. *Sarcina* In: *Bergey's Manual of Systematic Bacteriology*, Vol. 3, Second
 658 edition, 847-850. Edited by P. H. A. Sneath, N. S. Mair, M. E. Sharpe & J. G. Holt. Baltimore:
 659 Williams and Wilkins. 2009.
- 660
- 661 Carosia M.F., Reis C.M., Sakamoto I.K., Varesche M.B.A., Silva E.L., 2017. Influence of C/P and
 662 C/N ratios and microbial characterization in hydrogen and ethanol production in an anaerobic
 663 fluidized bed reactor. *International Journal of Hydrogen Energy*, 42(15), 9600-9610.
- 664
- 665 Chen W.M., Tseng Z.J., Lee K.S., Chang J.S., 2005. Fermentative hydrogen production with
 666 *Clostridium butyricum* CGS5 isolated from anaerobic sewage sludge. *International Journal of*
 667 *Hydrogen Energy*, 30, 1063–1070.
- 668
- 669 Chihib N.E., Tholozan J.L., 1999. Effect of rapid cooling and acidic pH on cellular homeostasis of
 670 *Pectinatus frisingensis*, a strictly anaerobic beer-spoilage bacterium. *International Journal of Food*
 671 *Microbiology*, 48, 191-202
- 672
- 673 Chojnacka A., Błaszczyk M.K., Szczęsny P., Latoszek K., Sumińska M., Tomczyk K., Zielenkiewicz
 674 U., Sikora A., 2011. Comparative analysis of hydrogen-producing bacterial biofilms and granular
 675 sludge formed in continuous cultures of fermentative bacteria. *Bioresource Technology*, 102 (21),
 676 10057-10064.
- 677
- 678 Cohen A., van Gemert A.M., Zoetemeyer R.J., Breure A.M., 1984. Main characteristics and
 679 stoichiometric aspects of acidogenesis of soluble carbohydrate containing wastewater. *Process*
 680 *Biochemistry*, 19, 228-232.
- 681
- 682 Dabrock B., Bahl H., Gottschalk G., 1992. Parameters affecting solvent production by *Clostridium*
 683 *pasteurianum*. *Applied and Environmental Microbiology*, 58(4),1233-1239.
- 684

- 685 Davey M. E., O'toole G.A., 2000. Microbial Biofilms: from Ecology to Molecular Genetics.
 686 *Microbiology and Molecular Biology Reviews*, 64(4), 847-867.
 687
- 688 Dinamarca C., Gañán M., Liu J., Bakke R., 2011. H₂ consumption by anaerobic non-methanogenic
 689 mixed cultures. *Water Science & Technology*, 63(8), 1582-1589.
 690
- 691 Edwards C.G., Collins M.D., Lawson P.A., Rodriguez A.V., 2000. *Lactobacillus nagelii* sp. nov., an
 692 organism isolated from a partially fermented wine. *International Journal of Systematic and*
 693 *Evolutionary Microbiology*, 50, 699-702
 694
- 695 Fang H.H., Liu H., Zhang T., 2002. Characterization of a hydrogen-producing granular sludge.
 696 *Biotechnology and Bioengineering*, 78(1), 44-52.
 697
- 698 Felchner-Zwirello M., Winter J., Gallert C., 2013. Interspecies distances between propionic acid
 699 degraders and methanogens in syntrophic consortia for optimal hydrogen transfer. *Applied*
 700 *Microbiology and Biotechnology*, 97(20), 9193-9205.
 701
- 702 Ferraz Júnior A.D.N., Etchebehere C., Zaiat M., 2015. Mesophilic hydrogen production in acidogenic
 703 packed-bed reactors (APBR) using raw sugarcane vinasse as substrate: Influence of support materials.
 704 *Anaerobe*, 34, 94-105.
 705
- 706 Ghimire A., Frunzo L., Pirozzi F., Trably E., Escudie R., Lens P.N.L., Esposito G., 2015. A review on
 707 dark fermentative biohydrogen production from organic biomass: Process parameters and use of by-
 708 products. *Applied Energy*, 144, 73-95.
 709
- 710 Goodwin S., Zeikus J.Z., 1987. Physiological adaptations of anaerobic bacteria to low pH: metabolic
 711 control of proton motive force in *Sarcina ventriculi*. *Journal of Bacteriology*, 169(5), 2150-2157.
 712

- 713 Griffiths R.I., Whiteley A.S., O'Donnell A.G., Bailey M.J., 2000. Rapid method for coextraction of
 714 DNA and RNA from natural environments for analysis of ribosomal DNA- and rRNA-based
 715 microbial community composition. *Applied and Environmental Microbiology*, 66(12), 5488-91.
 716
- 717 Haas B.J., Gevers D., Earl A.M., Feldgarden M., Ward D.V., Giannoukos G., Ciulla D., Tabbaa D.,
 718 Highlander S.K., Sodergren E., Methé B., DeSantis T.Z., Human Microbiome Consortium, Petrosino
 719 J.F., Knight R., Birren B.W., 2011. Chimeric 16S rRNA sequence formation and detection in Sanger
 720 and 454-pyrosequenced PCR amplicons. *Genome Research*, 21, 494–504.
 721
- 722 Hulshoff Pol L. W., de Castro Lopes S.I., Lettinga G., Lens P.N.L., 2004. Anaerobic sludge
 723 granulation. *Water Research*, 38(6), 1376-1389.
 724
- 725 Hwang M.H., Jang N.J., Hyun S.H., Kim I.S., 2004. Anaerobic bio-hydrogen production from ethanol
 726 fermentation: the role of pH. *Journal of Biotechnology*, 111, 297–309.
 727
- 728 Kim I.S., Hwang M.H., Jang N.J., Hyun S.H.; Lee S.T., 2004. Effect of low pH on the activity of
 729 hydrogen utilizing methanogen in bio-hydrogen process. *International Journal of Hydrogen Energy*,
 730 29(11), 1133–1140.
 731
- 732 Kimura M., 1980. A simple method for estimating evolutionary rate of base substitutions through
 733 comparative studies of nucleotide sequences. *Journal of Molecular Evolution*, 16, 111-120.
 734
- 735 Kozich, J.J., Westcott, S.L., Baxter, N.T., Highlander, S.K., Schloss, P.D., 2013. Development of a
 736 dual-index sequencing strategy and curation pipeline for analyzing amplicon sequence data on the
 737 MiSeq illumina sequencing platform. *Applied and Environmental Microbiology*, 79(17), 5112-5120.
 738
- 739 Kuhner C.H., Matthies C., Acker G., Schmittroth M., Goßner A.S., Drake H.L., 2000. *Clostridium*
 740 *akagii* sp. nov. and *Clostridium acidisoli* sp. nov.: acid-tolerant, N₂-fixing clostridia isolated from

- acidic forest soil and litter. *International Journal of Systematic and Evolutionary Microbiology*, 50, 873–881.
- Kumar S., Stecher G., Tamura K., 2016. MEGA7: Molecular Evolutionary Genetics Analysis Version 7.0 for Bigger Datasets. *Molecular Biology and Evolution*, 33(7), 1870-4.
- Kyazze G., Martinez-Perez N., Dinsdale R., Premier G.C., Hawkes F.R., Guwy A.J., Hawkes D.L., 2006. Influence of substrate concentration on the stability and yield of continuous biohydrogen production. *Biotechnology and Bioengineering*, 93(5), 971-9.
- Lay J.J., 2000. Modeling and optimization of anaerobic digested sludge converting starch to hydrogen. *Biotechnology and Bioengineering*, 68(3), 269-78.
- Lee H-S, Krajmalinik-Brown R., Zhang H., Rittmann B.E., 2009. An electron-flow model can predict complex redox reactions in mixed-culture fermentative bioH₂: microbial ecology evidence. *Biotechnology and Bioengineering*, 104, 687-697.
- Lee H-S., Vermaas W.F.J., Rittmann B.E., 2010. Biological hydrogen production: prospects and challenges. *Trends in Biotechnology*, 28(5), 262–271.
- Lee Y.H., Miyahara T., Noike T., 2002. Effect of pH on microbial hydrogen fermentation. *Journal of Chemical Technology and Biotechnology*, 77, 694-698.
- Lima D.M.F., Zaiat M., 2012. The Influence of the degree of back-mixing on hydrogen production in an anaerobic fixed-bed reactor. *International Journal of Hydrogen Energy*, 37, 9630-9635.
- Liu D., Liu D., Zeng R.J., Angelidaki I., 2006. Hydrogen and methane production from household solid waste in the two-stage fermentation process. *Water Research*, 40, 2230-2236.

769

770 Liu G., Shen J., 2004. Effect of culture and medium conditions on hydrogen production from starch
771 using anaerobic bacteria. *Journal of Bioscience and Bioengineering*, 98, 251-256.

772

773 MacLeod F.A., Guiot S.R., Costerton J.W., 1990. Layered structure of bacterial aggregates produced
774 in an upflow anaerobic sludge bed and filter reactor. *Applied and Environmental Microbiology*, 56(6),
775 1598-607.

776

777 Micolucci F., Gottardo M., Bolzonella D., Pavan P., 2014. Automatic process control for stable bio-
778 hythane production in two-phase thermophilic anaerobic digestion of food waste. *International*
779 *Journal of Hydrogen Energy*, 39, 17563-17572.

780

781 Mizuno O., Shinia M., Suzuki K., Yaguguchi J., Noike T. 2000. Effect of pH on hydrogen production
782 from noodle manufacturing wastewater. *Environmental Engineering Research*, 37, 97-106.

783

784 Ning Y., Wang S., Jin D., Harada H., Shi X., 2013. Formation of hydrogen-producing granules and
785 microbial community analysis in a UASB Reactor. *Renewable Energy*, 53, 12-17.

786

787 Noguchi K., Riggins D.P., Eldahan K.C., Kitko R.D., Slonczewski J.L., 2010. Hydrogenase-3
788 contributes to anaerobic acid resistance of *Escherichia coli*. *Plos One*, 5(4), e10132 (1-7).

789

790 Penteado E.D., Lazaro C.Z., Sakamoto I.K., Zaiat M., 2013. Influence of seed sludge and
791 pretreatment method on hydrogen production in packed-bed anaerobic reactors. *International Journal*
792 *of Hydrogen Energy*, 38, 6137-6145.

793

794 Picanço A.P., Vallero M.V.G., Gianotti E.P., Zaiat M., Blundi C.E., 2001. Influence of porosity and
795 composition of supports on the methanogenic biofilm characteristics developed in a fixed bed
796 anaerobic reactor. *Water Science and Technology*, 44(4), 197-204.

- 797
- 798 Pruesse E., Peplies J., Glöckner, F.O., 2012. SINA: accurate high-throughput multiple sequence
799 alignment of ribosomal RNA genes. *Bioinformatics*, 28, 1823-1829
- 800
- 801 Rafrafi Y., Trably E., Hamelin J., Latrille E., Meynial-Salles I., Benomar S., Giudici-Orticoni M.,
802 Steyer J., 2013. Sub-dominant bacteria as keystone species in microbial communities producing bio-
803 hydrogen. *International Journal of Hydrogen Energy*, 38, 4975-4985.
- 804
- 805 Reyes J., Celis L.B., Alatríste-Mondragón F., Montoya L., Razo-Flores E., 2012. Different Start-Up
806 Strategies to Enhance Biohydrogen Production from Cheese Whey in UASB Reactors. *International*
807 *Journal of Hydrogen Energy* 37(7), 5591-5601.
- 808
- 809 Roy S., Das D., 2016. Biohythane production from organic wastes: Present state of art. *Environmental*
810 *Science and Pollution Research International*, 23(10), 9391-410.
- 811
- 812 Ruggeri B., Tommasi T., Sanfilippo S., 2015. BioH₂ & BioCH₄ through anaerobic digestion: From
813 research to full-scale applications. Springer-Verlag London, London, pp. 1-218.
- 814
- 815 Schloss P.D., Westcott S.L., Ryabin T., Hall J.R., Hartmann M., Hollister E.B., Lesniewski R.A.,
816 Oakley B.B., Parks D.H., Robinson C.J., Sahl J.W., Stres B., Thallinger G.G., Van Horn D.J., Weber
817 C.F., 2009. Introducing mothur: opensource, platform-independent, community-supported software
818 for describing and comparing microbial communities. *Applied and Environmental Microbiology*,
819 75(23), 7537-7541.
- 820
- 821 Sikora M.B., Jurkowski M., Zielenkiewicz U., 2013. Lactic acid bacteria in hydrogen-producing
822 consortia: On purpose or by coincidence? Chapter 21, In: Lactic Acid Bacteria - R & D for Food,
823 Health and Livestock Purposes, Dr. J. Marcelino Kongo (Ed.), InTech.
- 824

- 825 Tähti H., Kaparaju P., Rintala J., 2013. Hydrogen and methane production in extreme thermophilic
 826 conditions in two-stage (upflow anaerobic sludge bed) UASB reactor system. *International Journal of*
 827 *Hydrogen Energy*, 38, 4997–5002.
- 828
- 829 Wang D., Zeng G., Chen Y., Li X., 2015. Effect of polyhydroxyalkanoates on dark fermentative
 830 hydrogen production from waste activated sludge. *Water Research*, 73, 311-322.
- 831
- 832 Wang L., Li Y., 2010. Biohydrogen production and wastewater treatment by anaerobic fermentation
 833 with UASB. *International Conference on Mechanic Automation and Control Engineering* (Mace),
 834 Wuhan, China. P. 1683-1686.
- 835
- 836 Wang L., Zhou Q., Li F.T., 2006. Avoiding propionic acid accumulation in the anaerobic process for
 837 biohydrogen production. *Biomass and Bioenergy*, 30(2),177-182.
- 838
- 839 Xing D.F., Ren N.Q., Li Q.B., Lin M., Wang A., 2006. *Ethanoligenens harbinense* gen nov., sp. nov.,
 840 isolated from molasses wastewater. *International Journal of Systematic and Evolutionary*
 841 *Microbiology*, 56, 755-760
- 842
- 843 Xing D.F., Ren N.Q., Rittmann B.E., 2008. Genetic diversity of hydrogen-producing bacteria in an
 844 acidophilic ethanol-H₂-coproducing system, analyzed using the [Fe]-hydrogenase gene. *Applied and*
 845 *Environmental Microbiology*, 74, 1232-1239
- 846
- 847 Xu L., Ren N., Wang X., Jia Y.F., 2008. Biohydrogen production by *Ethanoligenens harbinense* B49:
 848 nutrient optimization. *International Journal of Hydrogen Energy*, 33, 6962-6967.
- 849
- 850 Yokoi H., Ohkawara T., Hirose J., Hayashi S., Takasaki Y., 1995. Characteristics of hydrogen
 851 production by aciduric *Enterobacter Aerogenes* Strain HO-39. *Journal of Fermentation and*
 852 *Bioengineering*, 80(6), 571–574.

853

854 Yu H.Q., Mu Y., 2006. Biological hydrogen production in a UASB reactor with granules. II: Reactor
855 performance in 3-year operation. *Biotechnology Bioengineering*, 94, 988–995.

856

857 Zhao B.H., Yue Z.B., Zhao Q.B., Mu Y., Yu H.Q., Harada H., Li Y.Y., 2008. Optimization of
858 hydrogen production in a granule-based UASB reactor. *International Journal of Hydrogen Energy*,
859 33(10), 2454-2461.

860

861 Zhao X., Xing D., Qi N., Zhao Y., Hu X., Ren N., 2017. Deeply mechanism analysis of hydrogen
862 production enhancement of *Ethanoligenens harbinense* by Fe^{2+} and Mg^{2+} : Monitoring at growth and
863 transcription levels. *International Journal of Hydrogen Energy*, 42(31), 19695-19700.

864

865

Figure 1 – Schematic diagram of the acidogenic reactors. 1: distribution chamber, 2: reactional zone, 3: headspace, 4: biogas sampling, 5: biogas outlet

Figure 2 – Volumetric hydrogen production rate and pH in the first experimental phase: (a) FB reactor, (b) UG reactor, (c) UF-1 reactor.

Figure 3 – Volumetric hydrogen production rate and pH in the second experimental phase: UF-2 reactor

Figure 4 – Sucrose removal

Figure 5 – H₂ yield

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0.020

Figure 7 – Proposed model to explain changes in the UF-2 reactor that led to increased H₂ production

Highlights

Acidogenic reactors were fed with sucrose (4.7 gCOD L^{-1}) without pH regulator.

Flocculent UASB showed stability superior to granular UASB and fixed bed reactors.

Acetate production replaced lactate when the HRT increased from 3.3 to 4.6 h.

Continuous long-term and stable H_2 production was achieved at pH always below 3.0.

H_2 production of $175 \text{ mLH}_2 \text{ L}^{-1} \text{ h}^{-1}$ and yield of $3.4 \text{ molH}_2 \text{ mol}^{-1} \text{ sucrose}$ were obtained.

Bacteria affiliated with *Ethanoligenens harbinense* were predominant.

ACCEPTED MANUSCRIPT