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1 **Gastro-intestinal *in vitro* digestions of protein emulsions monitored by pH-stat: influence**  
2 **of structural properties and interplay between proteolysis and lipolysis**

3

4 D. J. L. Mat<sup>a,b</sup>, I. Souchon<sup>a</sup>, C. Michon<sup>b</sup>, S. Le Feunteun<sup>a,\*</sup>

5

6 <sup>a</sup> UMR GMPA, AgroParisTech, INRA, Université Paris-Saclay, 78850, Thiverval-Grignon, France

7 <sup>b</sup> UMR Ingénierie Procédés Aliments, AgroParisTech, INRA, Université Paris-Saclay, 91300,  
8 Massy, France

9

10 \* Corresponding author: [steven.le-feunteun@inra.fr](mailto:steven.le-feunteun@inra.fr)

11

12

13

14 **Abstract**

15 This study describes an experimental design, based on pH-stat, to rapidly screen and assess  
16 food formulation effects on the degrees of hydrolysis (DH) of both proteins and lipids  
17 throughout *in vitro* gastro-intestinal digestions. This approach was used to quantitatively  
18 compare and hierarchize key structure parameters of protein emulsions. Six matrices (15  
19 wt% whey proteins, 0 or 10 wt% oil), each differing by at least one structure characteristic,  
20 were studied. The physical state of the bulk and the oil droplet size were the major  
21 structural levers to modulate the hydrolysis of proteins (final DH between 51.7 and 58.3%)  
22 and lipids (final DH between 46.9 and 72.7%), with non-trivial interplays between proteolysis  
23 and lipolysis. Additionally, pH-stat measurements in presence of a pancreatic lipase inhibitor  
24 proved to be an efficient way to widen the scope of the proposed experimental approach to  
25 foods that are intrinsically made of both proteins and lipids.

26

27 **Keywords**

28 Food; Structure; Emulsion gel; Chemical-physics; Hydrolysis; Orlistat

29

30 **Abbreviations**

31 LCP: Liquid Continuous Phase; GCP: Gelled Continuous Phase; LFE: Liquid Fine Emulsion;

32 GCE: Gelled Coarse Emulsion; GCEi: Gelled Coarse Emulsion with a modified o/w interface;

33 GFE: Gelled Fine Emulsion

34

35        **1. Introduction**

36 Food structure is characterized by different spatial scales, from nano to macroscopic levels.  
37 Soluble molecules are typically in the nanometer scale and are generally embedded in  
38 structures organized at larger scales: biopolymer networks (with subunit-structure of about  
39 10-100 nm), droplets (0.1-5  $\mu\text{m}$ , typically), etc. To better understand how these structures  
40 break down during digestion and assess their impact on the release of nutrients, both *in vivo*  
41 and *in vitro* experiments can be undertaken. While *in vivo* studies are mandatory to  
42 unambiguously demonstrate that food structure can modulate the fate of nutrients in the  
43 host, *in vitro* experiments can be used to predict outcomes of *in vivo* digestions (Bohn et al.,  
44 2018) and more thoroughly investigate the underlying mechanisms.

45 When considering lipids, which are dispersed as oil droplets in most foods, both *in vivo* and  
46 *in vitro* studies have shown that smaller droplet size leads to faster lipolysis kinetics (Armand  
47 et al., 1999; McClements & Li, 2010), since they develop a larger interfacial area for lipase  
48 adsorption. The type of emulsifiers at the interface is another key factor to consider as it  
49 may control not only the ability for lipases to access their substrate (Mun, Decker, &  
50 McClements, 2007), but also the emulsion stability. In the changing conditions of the  
51 digestive tract, the composition of the interface may change and droplets interact with each  
52 other. This can lead to droplet flocculation and/or coalescence during both the gastric  
53 (Golding et al., 2011; Sarkar, Goh, Singh, & Singh, 2009) and intestinal (Giang et al., 2015,  
54 2016; Li, Ye, Lee, & Singh, 2013; Sarkar, Horne, & Singh, 2010) phases of digestion, and lower  
55 the rate of lipid absorption as estimated from the postprandial blood triglyceride  
56 concentrations (Golding et al., 2011; Keogh et al., 2011).

57 The kinetics of protein digestion also depends on food structure. The macrostructure is of  
58 course a key parameter, with a slower rate of proteolysis generally observed for solid foods.

59 For instance, dairy proteins have been reported to be much more rapidly metabolized in  
60 mini-pigs when eaten in the liquid state compared to rennet gels of identical composition  
61 (Barbé et al., 2013). Nanostructure and microstructure effects are also important to  
62 consider. For example, when used as emulsifiers, dairy whey proteins show an enhanced  
63 susceptibility to hydrolysis by pepsin because of their modified conformation upon  
64 adsorption at the oil interface (Macierzanka, Sancho, Mills, Rigby, & Mackie, 2009). Similarly,  
65 thermal denaturation and/or aggregation of  $\beta$ -lactoglobulin, the main protein of dairy whey,  
66 can lead to an improved hydrolysis by both pepsin (Guo, Fox, Flynn, & Kindstedt, 1995;  
67 Singh, Øiseth, Lundin, & Day, 2014) and intestinal proteases (Stănciuc, van der Plancken,  
68 Rotaru, & Hendrickx, 2008). Differently structured whey protein gels can therefore lead to  
69 highly contrasting digestion profiles (Macierzanka et al., 2012).

70 Few studies have also focused on food emulsions rich in proteins and on the interplay  
71 between lipolysis and proteolysis. Gelatin gels have been shown to slow down the rate of  
72 pancreatic lipolysis of embedded oil droplets because of a reduced lipase diffusion into the  
73 protein network (Sarkar et al., 2015). Others have investigated the influence of the structure  
74 of whey protein emulsion gels on lipid digestion, with interesting findings on the relations  
75 between their mechanical properties and their rate of disintegration upon simulated gastric  
76 contractions (Guo, Ye, Bellissimo, Singh, & Rousseau, 2017). Beyond confirming a slower  
77 lipolysis in presence of a surrounding protein network, it has been shown that the release of  
78 oil droplets from whey protein emulsion gels is delayed for hard gels compared to soft ones  
79 (Guo, Ye, Lad, Dalgleish, & Singh, 2014b), leading to a reduced rate of lipid hydrolysis (Guo,  
80 Bellissimo, & Rousseau, 2017; Guo, Ye, Lad, Dalgleish, & Singh, 2016). For emulsion gels of  
81 identical composition (10% whey proteins, 20% oil), the same team also observed that the  
82 size of the dispersed oil droplets can modify the structural and mechanical properties of the

83 gels, and hence their digestion kinetics, with less coalescence and phase separation when  
84 embedded droplets were small (1  $\mu\text{m}$  vs 6 and 12  $\mu\text{m}$ ) (Guo, Ye, Lad, Dalglish, & Singh,  
85 2014a).

86 This literature on the digestion of emulsion gels remain quite recent and is still scarce, but  
87 nicely illustrates how protein digestion can govern the way lipids entrapped in a protein  
88 network are released. Such findings are interesting to gain a better view of the fate of  
89 complex foods in the gastro-intestinal tract. Nonetheless, most of these studies have been  
90 conduct from a lipid digestion perspective, with no concomitant measurements of the  
91 protein digestion extent to more thoroughly quantify the interplays between protein and  
92 lipid hydrolysis. Previous studies of our group have shown that pH-stat can be used to  
93 monitor the degrees of hydrolysis of proteins during the gastric phase of *in vitro* digestions  
94 (Mat, Cattenoz, Souchon, Michon, & Le Feunteun, 2018), and of both proteins and lipids  
95 during the intestinal phase (Mat, Le Feunteun, Michon, & Souchon, 2016). Building upon  
96 these previous developments, our present work intends to show how pH-stat can be used,  
97 as a rather simple and high throughput approach, to quantitatively assess food formulation  
98 effects on the gastro-intestinal *in vitro* digestion of both their protein and lipid contents.  
99 More specifically, this study aimed at applying such approach to quantitatively compare the  
100 effects of three key structure parameters of protein emulsions: the state of proteins in the  
101 continuous phase (liquid and gelled state), the thickness and degree of denaturation of the  
102 protein layer at the oil/water interface, and the lipid droplet sizes (1 vs 20  $\mu\text{m}$ ). Additionally,  
103 because pH-stat is sensitive to both proteolysis and lipolysis in intestinal conditions, this  
104 method is generally considered inappropriate to the study of complex foods that are  
105 intrinsically made of both proteins and lipids. To overcome this issue, we also present a

106 means, based on the use of a lipase inhibitor during pH-stat intestinal monitoring, that  
107 appears suitable to evaluate the contribution of each reaction with such complex foods.

108

## 109 **2. Material and methods**

### 110 2.1. Material

111 Whey protein isolate powder (Prolacta 95, 95 wt% of proteins on dry powder) was obtained  
112 from Lactalis, France. Rapeseed oil (Fleur de colza, Lesieur, France) was purchased at a local  
113 supermarket. Pepsin (P6887), pancreatin (P7545), pancreatic lipase (L3126) and bile extract  
114 (B8631), all of porcine origin, were obtained from Sigma-Aldrich, France, as well as the  
115 Orlistat lipase inhibitor (O4139). Enzyme activities and bile salts concentrations were  
116 determined according to the protocols described in (Brodkorb et al., 2019; Minekus et al.,  
117 2014). Water was Milli-Q water and all other materials were of standard analytical grade.

118

### 119 2.2. Designed matrices

120 In total, 6 different matrices were designed for the purpose of this study: 4 protein  
121 emulsions and 2 protein-only matrices, the latter corresponding to the continuous phases of  
122 the emulsions. Their compositions are presented in Table 1, as well as their schematic  
123 structure and size characteristics.

124 A gelled coarse emulsion (GCE) was produced based on to the protocol previously described  
125 in (Mat et al., 2016). In short, an emulsion was prepared with 0.3 wt% of whey proteins and  
126 30 wt% of oil using a rotor-stator homogenizer (Polytron PT3100D, Kinematica AG,  
127 Switzerland) fitted with a PTDA32/2-B250 for 5 min at 10,000 rpm. It was then heated for 5  
128 min at 70 °C. This preliminary step was performed to ensure a certain amount of  
129 denaturation of the adsorbed proteins, and to test if their partial cross-linking at the

130 interface could influence lipolysis. The warm emulsion was mixed with a solution of 22.3  
131 wt% of proteins prepared beforehand to achieve the final composition of 15 wt% of proteins  
132 and 10 wt% of oil. The preparation was then heated for 30 min at 80 °C in a water bath to  
133 perform gelation.

134 An equally-composed liquid fine emulsion (LFE) was prepared as described in (Mat et al.,  
135 2016). In short, it consisted in emulsification by the same rotor-stator treatment, followed  
136 by sonication at 20 kHz and 130 W using a 13 mm probe (VCX 130, Sonic & Materials, UK) for  
137 10 min (effective time, with on/off cycles of 10 s) in order to further reduce the sizes of the  
138 oil droplets. The mean temperature in the emulsion was maintained below 30 °C with an ice  
139 bath.

140 The corresponding lipid-free matrices consisted in a 15 wt% of whey protein solution for the  
141 liquid one (liquid continuous phase, LCP), while its gelled counterpart (gelled continuous  
142 phase, GCP) was obtained after a heat treatment (80 °C, 10 min) of the same solution.

143 The other two emulsions were prepared in similar ways as GCE with slight variations. On the  
144 one hand, a gelled coarse emulsion with a modified oil/water interface (GCEi) was made  
145 with omission of the first heat treatment (at 70 °C), meaning that whey proteins were not  
146 heat-denatured before the dispersion of the droplets in the protein solution. On the other  
147 hand, a gelled fine emulsion (GFE), in which the pre-emulsion was produced with a higher  
148 quantity of whey proteins (6.0 wt% instead of 0.3 wt%) and a prolonged sonication step of  
149 15 min to produce oil droplets similar in size to those in LFE. These two additional emulsions  
150 were designed to change only one parameter at a time (interface structure and droplet size,  
151 respectively) when compared to GCE, and more properly assess the effect of the physical  
152 state of the continuous phase by comparing LFE with GFE.

153

154 2.3. Matrix characterization

155 The oil droplet size distributions in the emulsions were controlled with a laser light  
156 scattering particle size analyzer (Mastersizer 2000, Malvern, France), using refractive index  
157 of 1.47 and 1.33 for oil and water (dispersant), respectively. A value of 0.001 was set as the  
158 absorption of the emulsion. Droplet sizes (Table 1) are given as volume-weighted mean  
159 diameters, calculated using  $d_{4,3} = \sum n_i d_i^4 / \sum n_i d_i^3$ . Similar  $d_{4,3}$  values were obtained for LFE  
160 and GFE:  $1.22 \pm 0.06 \mu\text{m}$  and  $1.37 \pm 0.01 \mu\text{m}$ , respectively; and for GCEi and GCE:  $19.13 \pm$   
161  $0.03$  and  $18.83 \pm 0.64 \mu\text{m}$ , respectively.

162 The evolution of the rheological properties of the protein emulsions upon heating up to 80  
163 °C was also measured in triplicate according to the protocol used in (Mat et al., 2016). Final  
164 storage modulus values ( $G'$ ) are reported in Table 1.  $G'$  was similar for protein emulsions  
165 with large oil droplets (GCE and GCEi) and for their lipid-free counterpart (GCP), whereas it  
166 tended to be slightly higher for the protein emulsion containing small oil droplets (GFE)  
167 because of the increased interactions between dispersed and continuous phases.

168

169 2.4. Gastro-intestinal *in vitro* digestions monitored by pH-stat

170 The digestion experiments were all performed in triplicate, and consisted in a 3-phase  
171 digestion based on the recommendations of the Infogest consortium (Brodkorb et al., 2019;  
172 Minekus et al., 2014), where details can be found on the composition of digestive fluids and  
173 enzyme activities. The only noticeable difference is that  $\text{NaHCO}_3$  salts were replaced by  $\text{NaCl}$   
174 at the same molar ratio in all electrolyte solutions to avoid unwanted pH drift (Mat et al.,  
175 2016), meanwhile maintaining the same ionic strength. Gastro-intestinal digestions were  
176 carried out in a 200 mL jacketed beaker maintained at 37 °C by water circulation using a

177 constant magnetic stirring (250 rpm). This set-up was mounted onto an automatic titration  
178 unit (Titroline7000, VWR, France).

179

180 *Oral phase:* Solid matrices were demolded and grinded with a domestic kitchen food  
181 chopper (Braun Turbo, 600W, type 4191, Spain) for 3 s at maximum power to produce  
182 submillimeter particles. 7.5 g of the grinded matrix were then mixed with 7.5 mL of  
183 simulated salivary fluid in the jacketed beaker and let to reach temperature equilibrium. This  
184 oral phase was only carried out for electrolyte concentration considerations, with no added  
185 enzymes at this stage.

186

187 *Gastric phase and pH-stat measurements:* The pH probe and the titration cone of the  
188 titration unit were put in place. 13.5 mL of gastric electrolytes were added and the pH was  
189 adjusted to 3.0 using HCl 1 N. Once pH and temperature equilibria were achieved, 1.5 mL of  
190 a pepsin solution (to reach 2,000 U/mL in the final mixture) were added and titration was  
191 immediately turned on in a pH-stat mode. It was programmed to maintain a constant pH  
192 value of 3.0 for 2 h using HCl 0.3 N as a titrant.

193

194 *Intestinal phase and pH-stat measurements:* At the end of the gastric phase, the beaker was  
195 reconnected to another titration unit (same reference). The volume of the chyme was  
196 completed with water up to 33 mL in order to always start the intestinal phase with the  
197 same volume, regardless of the volume of titrant added previously. 25.5 mL of intestinal  
198 electrolytes containing a pre-established amount of NaOH 1 N, to bring the pH to 7.0, were  
199 first added. 2.5 mL of bile solution (prepared beforehand by melting bile extract in a 55 °C  
200 water bath) were then added to the mix. Pre-established amounts of pancreatin and

201 pancreatic lipase powders (to achieve a trypsin activity of 100 U/mL and a lipase activity of  
202 2,000 U/mL in the final mixture), conserved at -20 °C in a tube, were rapidly rehydrated with  
203 an intestinal electrolyte solution containing a pre-established amount of NaOH 1 N to bring  
204 the pH of the solution to 7.0. The tube was left in the water bath to reach 37 °C, and after 5  
205 min, the pH was checked, and adjusted to 7.0 if needed. 5 mL of the enzymes solution were  
206 then added into the jacketed beaker to complete the intestinal fluid (33 mL). The titration  
207 program was immediately turned on for 2 h in a pH-stat mode to maintain a pH value of 7.0  
208 using NaOH 0.2 N as a titrant.

209 Three blank intestinal digestions, with no food, were also conducted as it appeared that the  
210 mixing of the intestinal solutions induced a small pH-stat signal, possibly induced by  
211 interactions between bile extract constituents and pancreatic enzymes. This contribution  
212 was subtracted from all the titration curves obtained during the intestinal digestion of the  
213 studied matrices.

214

215 *Intestinal phase in presence of Orlistat:* In order to test whether protein hydrolysis could be  
216 solely monitored during the intestinal digestion of GCE and LFE matrices (*i.e.* despite their  
217 high lipid content and the lipolytic activity of pancreatin), the action of a lipase inhibitor  
218 (tetrahydrolipstatin, branded as Orlistat) was tested during additional digestions  
219 experiments. For these experiments, the gastric phase was carried out as described above.  
220 However, 660 µL of the intestinal electrolyte solution were substituted by 660 µL of an  
221 Orlistat solution (4 mg/mL in DMSO) in order to achieve a final concentration of 40 µg/mL.  
222 Moreover, pancreatic lipase was not added in complement to pancreatin in order to limit  
223 the lipolytic activity of the intestinal secretions, while maintaining the same proteolytic  
224 activity.

225

## 226 2.5. Determination of the degree of hydrolysis

227 *Gastric proteolysis:* The degree of hydrolysis of proteins during the gastric phase ( $DH_{prot\_G}$ )

228 was estimated according to the relation previously described in (Mat et al., 2018):

$$DH_{prot\_G} = 100 \times \frac{V \times N}{m \times h_{tot}} \times \frac{1}{1 - \alpha_{COOH}} \quad (1)$$

229 where  $V$  is the volume of added HCl (mL),  $N$  is the normality of the acid,  $m_{prot}$  is the mass

230 of proteins (g),  $h_{tot} = 8.8$  meqv/g is the total number of peptide bonds in whey proteins

231 (Spellman, McEvoy, O’Cuinn, & FitzGerald, 2003), and  $\alpha_{COOH} = 0.080$  is the mean degree of

232 dissociation of the peptide carboxylic groups at pH = 3.0 and 37 °C (Mat et al., 2018).

233

234 *Intestinal proteolysis:* The degree of hydrolysis of proteins during the intestinal phase

235 ( $DH_{prot\_I}$ ) was estimated for the lipid-free matrices (LCP and GCP) and for LFE and GCE in

236 presence of Orlistat, according to the following relation:

$$DH_{prot\_I} = 100 \times \frac{V \times N}{m_{prot} \times h_{tot}} \times \frac{1}{\alpha_{NH_2}} + DH_{prot\_G}(end) \quad (2)$$

237 where  $V$  is the volume of added NaOH (mL),  $DH_{prot\_G}(end)$  is the degree of hydrolysis at

238 the end of the gastric phase, and  $\alpha_{NH_2}$  is the mean degree of dissociation of the  $\alpha$ -amino

239 groups. The value of  $\alpha_{NH_2}$  was estimated to be 0.1118 and 0.1111 from the results obtained

240 with LCP and GCP, respectively, using the same procedure as in (Mat et al., 2016) that relies

241 on independent determinations of the degree of hydrolysis of end samples, *i.e.* collected at

242 the end of the experiments, using the OPA (ortho-phthalaldehyde) method. A value of  $\alpha_{NH_2}$

243 = 0.1114 was therefore used in Eq. (2).

244

245 *Intestinal lipolysis*: The degree of hydrolysis of lipids during the intestinal phase ( $DH_{lip}$ ) was  
246 estimated according to the following relation:

$$DH_{lip} = 100 \times \frac{\Delta V \times N \times M_{lip}}{m_{lip} \times 2} \times \frac{1}{\alpha_{COOH}} \quad (3)$$

247 where  $\Delta V$  (mL) is the difference between the volumes of NaOH added for the studied  
248 emulsion and its lipid-free counterpart (LCP or GCP),  $m_{lip}$  is the oil mass (g),  $M_{lip}$  is the  
249 molar weight of the triglycerides in the oil (calculated as 930 g/mol), and  $\alpha_{COOH}$  is the mean  
250 degree of dissociation of the free fatty acids' carboxylic group.  $\alpha_{COOH}$  was estimated to be  
251 0.77 according to complementary 12h long intestinal digestions, performed on reduced  
252 quantities (1.5 g) of a liquid emulsion made with 10 wt% of rapeseed oil and 0.1 wt% of  
253 whey proteins in order to reach 100% release of fatty acids (controlled by the appearance of  
254 a sustained plateau during pH-stat measurements).

255

## 256 2.6. Statistical analysis

257 One way ANOVA was used to compare the degree of hydrolysis between two matrices, or  
258 two categories of matrices (*e.g.* liquid vs gelled). The initial reaction rates and the final  
259 extent of hydrolysis were compared 3 min and 120 min after the start of the gastric and  
260 intestinal phases, respectively. Statistically significant effects were accepted at the 95%  
261 level. All statistical analyses were conducted using the statistics toolbox of Excel™.

262

### 263 3. Results & Discussion

#### 264 3.1. Effects of structural properties on gastric proteolysis

265 Fig. 1 shows the kinetics of protein hydrolysis by pepsin at pH = 3.0 for the 6 studied  
266 matrices measured by pH-stat during gastric digestion (Mat et al., 2018). Results are spitted  
267 into Fig. 1A and 1B for legibility, and those obtained for the gelled coarse emulsion (GCE) are  
268 duplicated in both subfigures for comparison purposes. The degree of hydrolysis (DH) of  
269 proteins measured after 3 and 120 min of reaction are also reported in Table 2.

270 A high initial reaction rate followed by a progressive slowdown was observed for all foods.  
271 Results also shows that the beginning of the reaction was slightly slower ( $P < 0.001$  at  $t = 3$   
272 min, Table 2) for 4 gelled matrices (GCP, GCE, GCEi, GFE) than for the 2 liquid ones (LCP and  
273 LFE). This can be directly related to the physical state of the protein phase since the  
274 substrate is readily accessible to pepsin in solutions, whereas its accessibility is initially  
275 limited to the external surface of gel fragments for solid matrices. It also appears that the  
276 slowdown was more pronounced for liquid matrices than for the solid ones, leading to final  
277 DH values (Table 2) significantly higher ( $P < 10^{-6}$ ) for solid matrices (5.5-6.0%) than for LCP  
278 and LFE (3.4-3.8%). This can be explained by an enhanced susceptibility of denatured  
279 proteins to peptic hydrolysis. Indeed,  $\beta$ -lactoglobulin, the major constituent of whey  
280 proteins, has been reported to be rather resistant to pepsin in its native form (Astwood,  
281 Leach, & Fuchs, 1996), and more sensitive to pepsinolysis after denaturation by heat  
282 treatments above 70 °C (Reddy, Kella, & Kinsella, 1988). Overall, the degrees of hydrolysis  
283 we measured are in line with the values found in the literature for pepsin digestion of whey  
284 proteins: 1.7% after 4 h with native proteins (Asselin, Hébert, & Amiot, 1989), between 3 to  
285 10% after 2 h with heat-treated proteins (Kim et al., 2007), or 7.9% after 3 h with whey  
286 protein gels (Luo, Boom, & Janssen, 2015).

287 Fig. 1A and 1B also show that the trends observed for protein emulsions closely followed  
288 those observed for their lipid-free counterparts (LCP and GCP), with undistinguishable  
289 kinetics and final degree of hydrolysis ( $P > 0.7$ ) for the 4 solid matrices (Table 2). This  
290 suggests that, whatever their size or interface (Table 1 and Fig. 1B), embedded oil droplets  
291 had a negligible influence on the peptic digestion of our emulsion gels. The final DH was  
292 slightly (by 0.4% on average) but statistically ( $P = 0.045$ ) higher for LCP than for LFE,  
293 however. This difference, which appeared in the early stages of the reaction (Fig. 1A), could  
294 reflect a decreased pepsin-substrate meeting probability in oil-droplet containing solutions,  
295 but additional data would be needed to confirm such an effect.

296 In summary, these results confirm the significant effect of whey protein denaturation on  
297 their hydrolysis by pepsin, and show a lesser influence of embedded oil droplets, whatever  
298 their size or interface, on our protein-rich emulsions (15 wt% proteins and 10 wt% oil).

299

300 3.2. Effects of structural properties on intestinal proteolysis as inferred from lipid-  
301 free matrices

302 The cumulative volume of NaOH recovered by pH-stat during the subsequent intestinal  
303 digestion are presented in Fig. 2A for all matrices. However, in the neutral conditions of  
304 intestinal digestion, both proteolysis and lipolysis reactions contribute to the pH-stat signal.  
305 This is the reason why the volume of added titrant was much larger for lipid-containing  
306 matrices.

307 To address the impact of the protein structure on the proteolysis kinetics, one should  
308 therefore focus on the comparison between LCP and GCP. In a similar but more substantial  
309 way than during the gastric phase (Fig. 1A), the reaction was initially faster ( $P = 10^{-5}$ ) for LCP  
310 than GCP (Fig. 2A), illustrating again the effect of substrate accessibility. By the end of the

311 experiments, a sustained plateau was clearly reached with both matrices, meaning that the  
312 reaction was complete in each case. The final volumes of added titrant were similar, though  
313 they tended ( $P = 0.08$ ) to be higher for GCP ( $2.79 \pm 0.05$  mL) than for LCP ( $2.66 \pm 0.03$  mL).  
314 The overall quantity of peptide bonds hydrolyzed during the intestinal phase were thus  
315 about the same for denatured and native whey proteins. This small difference, if any, is  
316 consistent with the previous report that heat denaturation of whey proteins has a much less  
317 pronounced effect on chymotrypsin action (major constituent of pancreatic proteases) than  
318 pepsin action (Reddy et al., 1988). It is noteworthy that we previously found a bigger  
319 difference in a previous work on comparable matrices (Mat et al., 2016). We since figured  
320 out that protons entrapped in too large gel fragments (e.g.  $> 1$  mm) during the gastric phase  
321 can slowly release during the intestinal phase, hence leading to an overtitration by pH-stat  
322 and an overestimation of the proteolysis extent. Special care was therefore taken in the  
323 present study to avoid such effects by grinding the solid matrices into submillimeter pieces  
324 (averaged minimal Feret diameters of 0.5 mm, data not shown) and wait enough time for pH  
325 adjustment at the transition between gastric and intestinal phases.

326 The evolution of the intestinal DH of proteins, which accounts for the hydrolysis achieved  
327 during the gastric phase (*i.e.* DH of 5.7% for GCP and of 3.8% for LCP at  $t = 0$  min of the  
328 intestinal phase), are presented in Fig. 2B. The final DH obtained for LCP ( $51.7 \pm 0.4\%$ ) was  
329 smaller ( $P = 0.024$ ) than the one measured for GCP ( $57.4 \pm 2.7\%$ ), hence confirming the  
330 higher overall susceptibility of gelled whey proteins (GCP) than native proteins (LCP) to  
331 digestive proteases. We may highlight in here that LCP and CGP matrices have been shown  
332 to induce contrasted impacts on both the gastrointestinal physiology and the intestinal  
333 microbiota in rats, associated to a higher protein content reaching the caecum for LCP

334 (Beaumont et al., 2017; Oberli et al., 2018). A higher resistance of this native protein  
335 solution to protease action therefore appears consistent with these *in vivo* findings.  
336 From Fig. 2A, we may finally note that the titration curves for GCE (pre-heated initial  
337 emulsion) and GCEi (unheated initial emulsion) were indistinguishable, hence suggesting  
338 that the higher quantity of proteins and cross-links at the oil droplet interface of GCE was of  
339 negligible influence on the overall digestion kinetics. We may indeed expect both liquid  
340 matrices, on the one hand, and the 4 solid matrices, on the other hand, to behave similarly  
341 from a pancreatic proteolysis point of view, as demonstrated during the gastric phase.  
342 However, it is not possible to ensure this statement from the analyses of the results of Fig.  
343 2A, nor to investigate the possible effects of oil droplets on the intestinal protein hydrolysis  
344 kinetics, because of the contribution of lipid hydrolysis.

345

### 346 3.3. pH-stat monitoring of intestinal digestion in the presence of a lipase inhibitor

347 Most foods are intrinsically made of both proteins and lipids, with no simple way of  
348 manufacturing a lipid-free equivalent, as performed in this study, to evaluate the  
349 contributions of proteolysis and lipolysis to the pH-stat signal in intestinal conditions. To  
350 overcome this limitation, we investigated another experimental strategy that relies on the  
351 use of a lipase inhibitor, tetrahydrolipstatin, branded as Orlistat. This molecule has been  
352 studied both *in vitro* and *in vivo* and proved very efficient in inhibiting the human pancreatic  
353 lipase (Carrière et al., 2001; Tiss, Lengsfeld, Carrière, & Verger, 2009; Wilcox, Brownlee,  
354 Richardson, Dettmar, & Pearson, 2014). It prevents the hydrolysis of triglycerides by  
355 covalently reacting with the active site of the lipase. According to the specificity of its  
356 mechanism of action, Orlistat is also assumed to solely inhibit lipases and leave protease  
357 action unhindered.

358 Fig. 3A and 3B present the titration kinetics obtained when Orlistat was added into the  
359 reaction mixture during intestinal digestions of LFE and GCE, respectively. Results were very  
360 close to the ones obtained with the lipid-free matrices, LCP and GCP, hence confirming that  
361 Orlistat is a very good inhibitor of intestinal lipolysis, with no, or very little, effects on  
362 proteases. This strategy therefore constitutes a very interesting alternative, which should be  
363 suitable to all types of food, to unveil the contributions of intestinal proteolysis and lipolysis  
364 when using pH-stat. However, the lipid hydrolysis reaction cannot be perfectly prevented, as  
365 evidenced by the slow but progressive increase towards the end of the experiments.  
366 Nonetheless, present results clearly show that this approach enables to reach the same  
367 general conclusions as those obtained from the lipid-free matrices strategy. The present  
368 approach can moreover provide a means to estimate the effects of lipids on the hydrolysis of  
369 proteins, as illustrated by the slower initial rate of proteolysis for GCE than for GCP ( $P <$   
370  $0.001$  at  $t = 3$  min) during the first hour (Fig. 3B).

371

### 372 3.3. Effects of structural properties on intestinal lipolysis

373 The lipolysis contribution to the intestinal titration was determined by subtracting the  
374 volumes of titrant added for the lipid-free matrices. The resulting curves were then  
375 converted into DH values, as presented in Fig. 4 and Table 2.

376 The intestinal titration curves were the same for GCE and GCEi (Fig. 2A), leading to similar  
377 final DH ( $46.9 \pm 8.8\%$  and  $48.1 \pm 2.2\%$ , respectively). Alongside to our conclusion on protein  
378 hydrolysis, the higher quantity proteins and of cross-links at the oil droplet interface of GCE  
379 also appeared to be of negligible influence on the lipolysis kinetics. This finding is also in  
380 excellent agreement with the previously reported limited effect of interfacial  $\beta$ -lactoglobulin

381 cross-linking on the *in vitro* intestinal digestion of liquid emulsions (Sandra, Decker, &  
382 McClements, 2008).

383 Compared to GCE and GCEi, both LFE and GFE showed marked enhanced lipolysis profiles  
384 (Fig. 4,  $P < 0.001$  at 120 min). The mean diameter of oil droplets in GFE and LFE was almost  
385 10 times smaller than in GCE and GCEi (Table 1), hence demonstrating the major influence of  
386 the interfacial surface on the lipid hydrolysis rate even when oil droplets are embedded  
387 within a protein gel. The most rapid initial reaction rate was obtained with LFE ( $P < 0.05$  from  
388 3 to 9 min) most certainly because of the higher accessibility of oil droplets in this liquid  
389 matrix when compared with GFE. However, the subsequent slowdown of lipid hydrolysis was  
390 less pronounced for GFE than for LFE, hence leading to statistically comparable final DH ( $P =$   
391 0.19). This suggests that the physical state of the continuous phase also has an influence on  
392 lipid hydrolysis. The entrapment of the oil droplets in the protein gel may help stabilizing  
393 them throughout the intestinal reactions, whereas droplets in LFE can more easily enter into  
394 contact, coalesce and cream, thereby almost disappearing from a reaction point of view  
395 (Giang et al., 2015). This was visually witnessed as no oil layer was formed at all with GFE,  
396 which contrast with what was observed with LFE, GCE, and SECc. The solid fine emulsion  
397 thus appeared to enable an initially more gradual, but overall comparable, release of  
398 lipolysis products than its liquid counterpart. Such finding might be of interest to better  
399 understand nutrient interactions within the gut, or in the formulation of lipid-based delivery  
400 systems.

401

## 402 **Conclusions**

403 In this study an integrated method of *in vitro* static digestion, respecting the Infogest  
404 recommendations, was presented. It relies on the use of the pH-stat titration during both

405 the gastric and intestinal phases of digestion. The method proved efficient at determining  
406 the kinetics of gastric proteolysis, intestinal lipolysis and intestinal proteolysis on both liquid  
407 and solid matrices, which may be considered as good models of complex foods rich in both  
408 proteins and lipids. The physical state of the proteins (particularly in the bulk) and the size of  
409 the oil droplet were identified as major structural parameters to modulate protein and lipid  
410 hydrolysis. Overall, pH-stat proved to be a simple, rapid, and very efficient method to  
411 quantitatively monitor gastro-intestinal proteolysis and lipolysis of food products. The use of  
412 Orlistat as an efficient inhibitor of lipolysis makes it possible to study the proteolysis of food  
413 matrices in which lipids are naturally present.

414

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418

419

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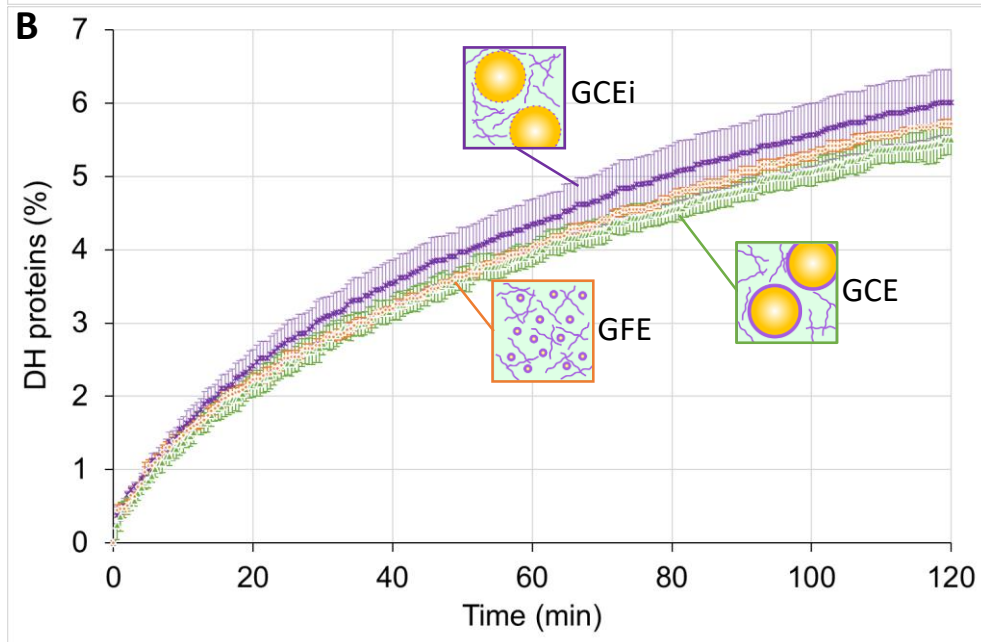
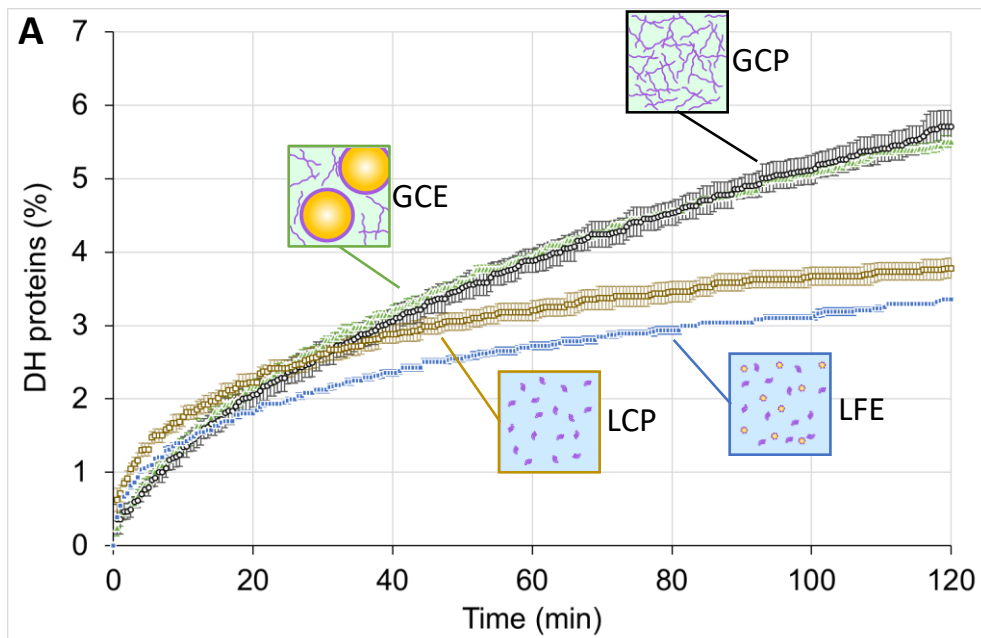
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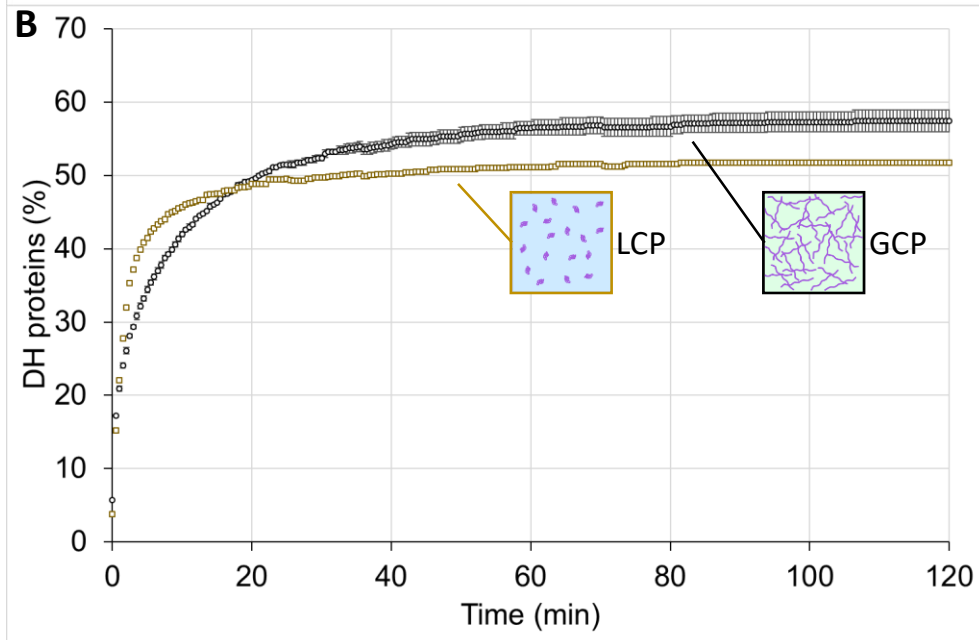
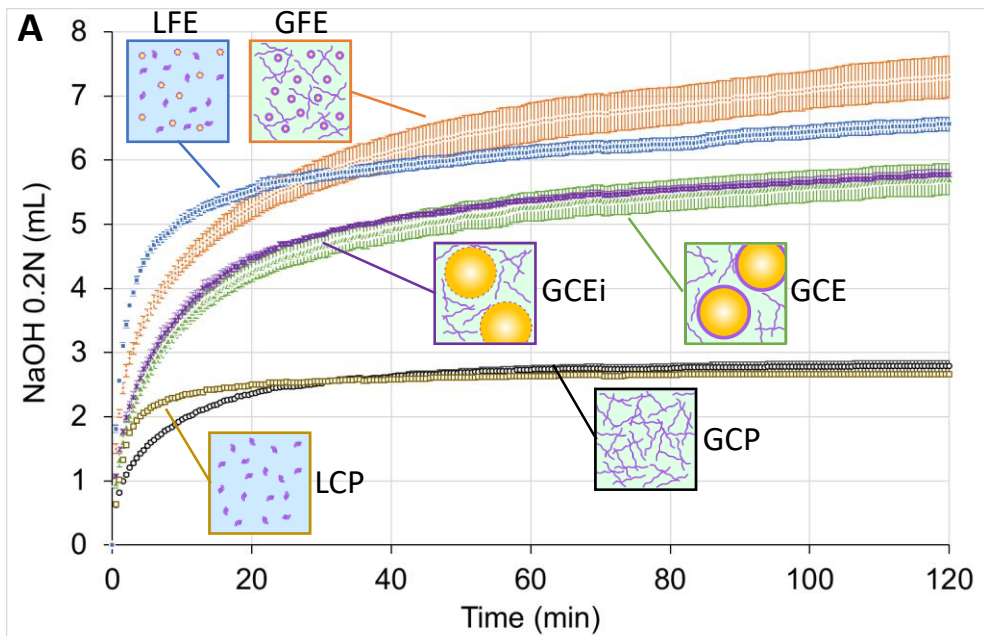
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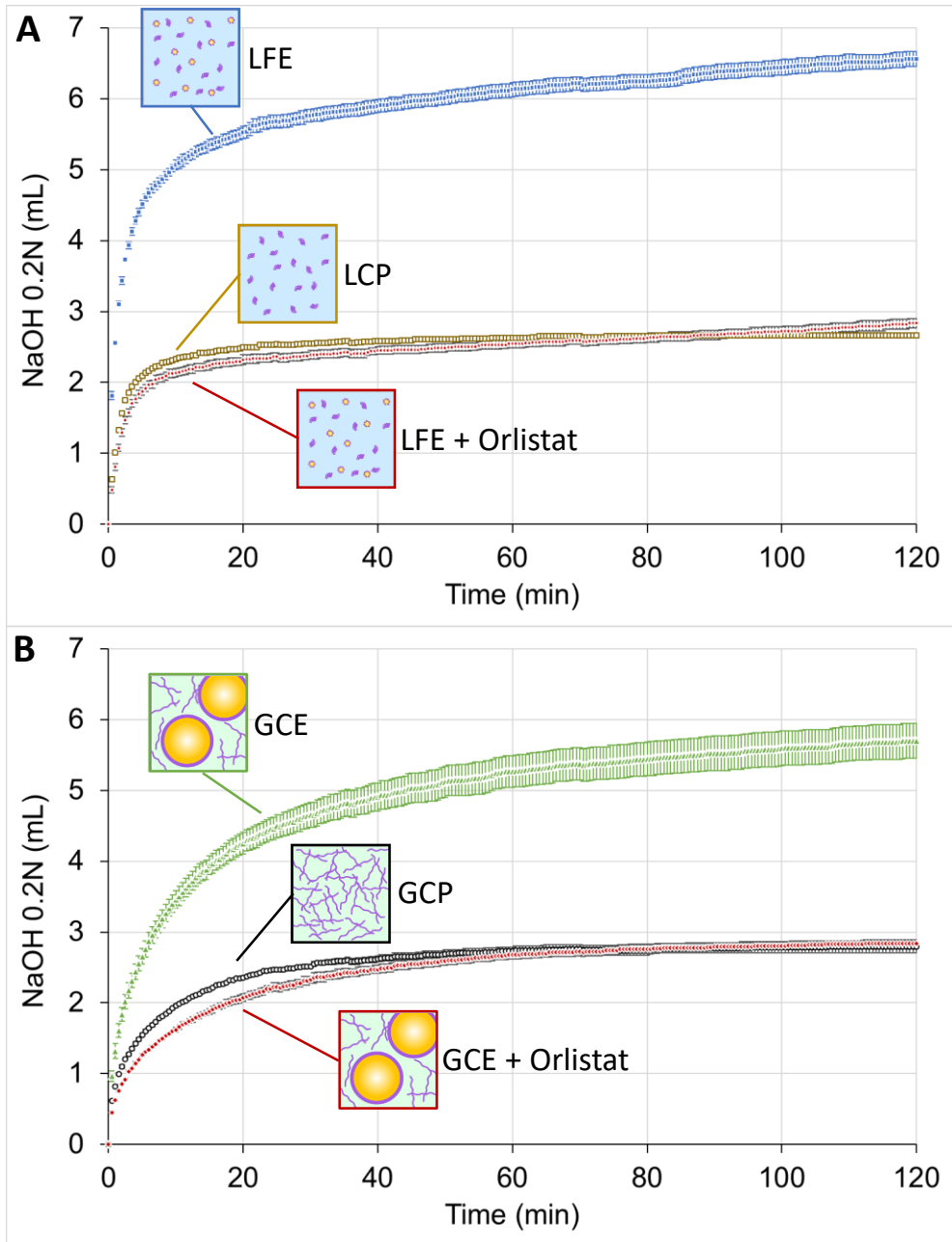
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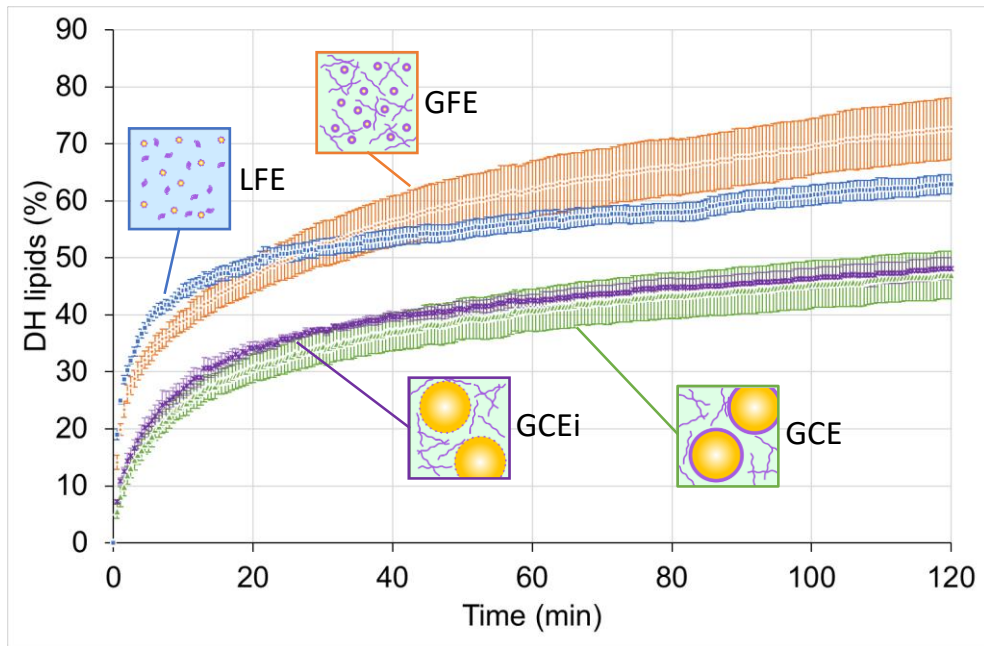
**Fig. 1** Evolution of the degree of hydrolysis (DH) of proteins measured by pH-stat during the course of *in vitro* gastric digestion (pH = 3.0). LCP stands for Liquid Continuous Phase (squares), GCP for Gelled Continuous Phase (circles), LFE for Liquid Fine Emulsion (squares), GCE for Gelled Coarse Emulsion (triangles, duplicated in A and B), GFE for Gelled Fine Emulsion (diamonds), and GCEi for Gelled Coarse Emulsion with a modified o/w interface (stars). Data represent means  $\pm$  SEM over at least 3 replicates.



**Fig. 2** Results of the pH-stat titration during the course of *in vitro* intestinal digestion (pH = 7.0): (A) Titration curves; and (B) Evolution of the degree of hydrolysis (DH) of proteins for lipid-free matrices that accounts for the preceding gastric phase. LCP stands for Liquid Continuous Phase (squares), GCP for Gelled Continuous Phase (circles), LFE for Liquid Fine Emulsion (squares), GCE for Gelled Coarse Emulsion (triangles), GFE for Gelled Fine Emulsion (diamonds), and GCEi for Gelled Coarse Emulsion with a modified o/w interface (stars). Data represent means  $\pm$  SEM over at least 3 replicates.

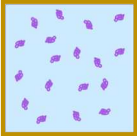
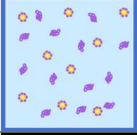
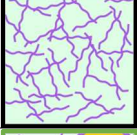
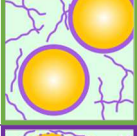
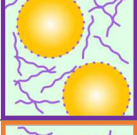
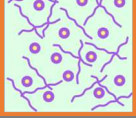


**Fig. 3** pH-stat titration curves during the course of *in vitro* intestinal digestion (pH = 7.0) for (A) liquid matrices: Liquid Fine Emulsion (LFE) in absence (squares) or presence (diamonds) of Orlistat to be compared with its lipid-free counterpart (LCP, squares); and (B) gelled matrices: Gelled Coarse Emulsion (SCE) in absence (triangles) or presence (diamonds) of Orlistat to be compared with its lipid-free counterpart (GCP, circles). Data represent means  $\pm$  SEM over at least 3 replicates.



**Fig. 4** Evolution of the degree of hydrolysis (DH) of lipids measured by pH-stat during the course of *in vitro* intestinal digestion (pH = 7.0). LFE stands for Liquid Fine Emulsion (squares), GCE for Gelled Coarse Emulsion (triangles), GFE for Gelled Fine Emulsion (diamonds), and GCEi for Gelled Coarse Emulsion with a modified o/w interface (stars). Data represent means  $\pm$  SEM over at least 3 replicates.

**Table 1:** Overview of the designed matrices. All matrices contain 15 wt% whey proteins. When lipids are present, they represent 10 wt% of the emulsion. Data represent Mean  $\pm$  SD over at least 3 replicates.

Matrix	Continuous phase	Protein state / heat treatment	G' for solid matrices (kPa)	d <sub>4,3</sub> (μm) of oil droplet	Schematic representation
LCP	Liquid	Native	–	–	
LFE	Liquid	Native	–	1.22 $\pm$ 0.06	
GCP	Gel	Gelation at 80 °C	39.0 $\pm$ 1.3	–	
GCE	Gel	Pre-emulsion at 70°C Gelation at 80 °C	46.9 $\pm$ 4.4	18.83 $\pm$ 0.64	
GCEi	Gel	Gelation at 80 °C	41.3 $\pm$ 0.5	19.13 $\pm$ 0.03	
GFE	Gel	Pre-emulsion at 70°C Gelation at 80 °C	68.8 $\pm$ 3.6	1.37 $\pm$ 0.01	

**Table 2:** Degrees of hydrolysis (DH) of proteins and lipids after 3 min and 120 min of gastric and intestinal digestion. Data represent Mean  $\pm$  SD over at least 3 replicates.

	DH proteins (%) Gastric phase		DH proteins (%) Intestinal phase		DH lipids (%) Intestinal phase	
	3 min	120 min	3 min	120 min	3 min	120 min
<b>LCP</b>	1.1 $\pm$ 0.2	3.8 $\pm$ 0.2	37.1 $\pm$ 0.3	51.7 $\pm$ 0.4 <sup>1</sup>	–	–
<b>LFE</b>	0.9 $\pm$ 0.1	3.4 $\pm$ 0.1	32.1 $\pm$ 1.7 <sup>2</sup>	54.8 $\pm$ 2.2 <sup>2</sup>	33.6 $\pm$ 1.3	62.9 $\pm$ 4.0
<b>GCP</b>	0.6 $\pm$ 0.1	5.7 $\pm$ 0.4	29.4 $\pm$ 0.6	57.4 $\pm$ 2.7 <sup>1</sup>	–	–
<b>GCE</b>	0.6 $\pm$ 0.2	5.5 $\pm$ 0.2	24.8 $\pm$ 0.4 <sup>2</sup>	58.3 $\pm$ 1.7 <sup>2</sup>	14.6 $\pm$ 3.9	46.9 $\pm$ 8.8
<b>GCEi</b>	0.8 $\pm$ 0.1	6.0 $\pm$ 1.1			16.6 $\pm$ 2.4	48.1 $\pm$ 2.2
<b>GFE</b>	0.7 $\pm$ 0.1	5.7 $\pm$ 0.1			28.8 $\pm$ 3.0	72.7 $\pm$ 10.0

<sup>1</sup> DH of proteins determined using the OPA method, and used to estimate the value of  $\alpha_{NH_2}$  in Eq. (2).

<sup>2</sup> DH of proteins determined by pH-stat in presence of Orlistat assuming a total lipase inhibition.