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Florian Monlau, Eric Trably, Abdellatif Barakat, Jérôme Hamelin, Jean-Philippe Steyer, et al.. Two-Stage Alkaline-Enzymatic Pretreatments To Enhance Biohydrogen Production from Sunflower Stalks. Environmental Science and Technology, 2013, 47 (21), pp.12591 - 12599. 10.1021/es402863v . hal-02647473

HAL Id: hal-02647473 https://hal.inrae.fr/hal-02647473v1

Submitted on 8 Aug 2023

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Two-stage alkaline-enzymatic pretreatments to enhance biohydrogen production from sunflower stalks

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¹⁰ Due to their rich composition in carbohydrates, lignocellulosic residues represent an interesting source of ¹¹ biomass to produce biohydrogen by dark fermentation. Nevertheless, pretreatments should be applied to ¹² enhance the solubilization of holocelluloses and increase their further conversion into biohydrogen. The ¹³ aim of this study was to investigate the effect of thermo-alkaline pretreatment alone and combined with ¹⁴ enzymatic hydrolysis to enhance biohydrogen production from sunflower stalks. Low increase of ¹⁵ hydrogen potentials from 2.3 ± 0.9 mL $H_2.g^{-1}$ VS to 4.4 ± 2.6 mL $H_2.g^{-1}$ and 20.6 ± 5.6 mL $H_2.g^{-1}$ VS ¹⁶ were observed with raw sunflower stalks and after thermo-alkaline pretreatment at 55°C, 24h, 4% NaOH ¹⁷ and 170°C, 1 h, 4% NaOH, respectively. Enzymatic pretreatment alone showed an enhancement of the ¹⁸ biohydrogen yields to 30.4 ml H_2 g⁻¹ initial VS whereas it led to 49 mL H_2 g⁻¹ initial VS and 59.5 mL H_2 ¹⁹ g⁻¹ initial VS when combined with alkaline pretreatment at 55°C and 170°C, respectively. Interestingly, a ²⁰ diauxic effect was observed with sequential consumption of sugars by the mixed cultures during dark ²¹ fermentation. Glucose was first consumed and, once glucose was completely exhausted, xylose was ²² utilized by the microorganisms, mainly related to *Clostridium* species.

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5

24 Graphical abstract

37 Introduction:

³⁸ Development of renewable energy such as hydrogen has recently taken a considerable place in the scientific community due to oil crisis ³⁹ and the wish of mitigating global warming by reducing greenhouse gas emission. Biohydrogen production by dark fermentation of ⁴⁰ lignocellulosic substrates appears to be one of the most promising ways, especially when using agricultural residues since they are cheap, ⁴¹ abundant and provide new incomes to farmers through a profitable reuse of the residues.^{1,2} Among them, sunflower stalks are present in ⁴² large quantities and represent renewable and low-cost raw materials for the production of biofuels such as biohydrogen.³ In particular, ⁴³ holocelluloses (hemicelluloses and cellulose) which are polymers of pentose and hexose simple sugars can be efficiently converted into ⁴⁴ biohydrogen by dark fermentation.⁴ Fermentative biohydrogen can be produced by either pure or mixed bacterial cultures. Even though ⁴⁵ pure cultures exhibit higher hydrogen yields, the high costs for maintaining these pure cultures restrict the development of industrial ⁴⁶ applications. Mixed cultures such as anaerobic sludge are easier to be used since they do not require sterile conditions, present a high ⁴⁷ metabolic flexibility and can therefore transform a wide range of feedstock.⁵ However, direct conversion of lignocellulosic substrates into ⁴⁸ hydrogen is recalcitrant due to the intrinsic structure of the organic materials, and so far, only very low hydrogen yields of 1, 3 and 9 ml ⁴⁹ H₂ g⁻¹VS have been reported for wheat straws, cornstalks and corn straws, respectively.¹ The main bottleneck of using lignocellulosic ⁵⁰ biomass by dark fermentation is to convert holocelluloses into readily fermentable products as previously reported.^{2.6} Recent studies^{1.6} ⁵¹ suggested that hydrogen yields correlate strongly with the content in easily accessible sugars of the organic substrates. Hydrolysis of ⁵² ceclulosic biomass is thus required since most of the

⁵³ Among hydrolysis technologies that convert holocelluloses into soluble sugars, two conventional methods using mineral acids or ⁵⁴ enzymes are commonly used. Enzymatic hydrolysis is considered as an efficient, energy saving and environment-friendly process. ⁵⁵ Recently, hydrogen yields from poplar leaves were increased from 15.04 to 44.92 mL H₂ g⁻¹ dry materials after pretreatment with 2% ⁵⁶ Vicozyme L (mixture of arabinase, cellulase, β-glucanase, hemicellulase and xylanase).⁸ However, the accessibility of holocelluloses to ⁵⁷ the enzymes for further conversion into fermentable sugars is limited by the lignin content and the cellulose crystallinity.^{8,9,10} ⁵⁸ Consequently, additional pretreatments need to be applied prior to the enzymatic attack to reduce this strong physical barrier and enhance ⁵⁹ holocellulose hydrolysis.¹¹ Alkaline pretreatments are known to remove efficiently lignin and lignin-hemicellulose complex, and thus can ⁶⁰ increase enzymatic hydrolysis of holocelluloses.^{9,10} Combining alkaline pretreatment with enzymatic hydrolysis appears to be a ⁶¹ promising approach to enhance hydrogen production from lignocellulosic residues. So far, alkaline pretreatment followed by enzymatic ⁶² hydrolysis has been widely investigated for bioethanol production ^{3,12} and, to our knowledge, only one study investigated alkaline ⁶³ pretreatment combined with enzymatic hydrolysis for biohydrogen production of lignocellulosic residues and using anaerobic mixed ⁶⁴ cultures.¹³ In this study, alkaline pretreatment (4 % NaOH (w/v) at 100°C for 2h) was combined with enzymatic hydrolysis (cellulases ⁶⁵ from *Trichoderma Reesei* at 20 U g⁻¹) to enhance biohydrogen production from bagasse.¹³ A hydrogen yield of 300 mL H₂ g⁻¹ VS ⁶⁶ (Volatile Solids) was observed compared to 31 mL H₂ g⁻¹ VS with only enzymatic hydrolysis. Nevertheless, the use of hemicellulases ⁶⁷ was not considered in their study although fermentative biohydrogen can be produced from both hemicellul

68 The objectives of our study were: (1) to investigate the effect of thermo-alkaline pretreatment (55°C, 24h, 4% NaOH or 170°C, 1h, 4%

⁶⁹ NaOH) or two-stage thermo-alkaline and enzymatic pretreatments on biohydrogen production, (2) to monitor the dynamics of the ⁷⁰ bacterial communities supporting the fermentation process of the sugars released from enzymatic hydrolysis.

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72 Experimental

73 Feedstocks:

⁷⁴ "Serin" sunflower stalks were dried at 37°C for 48 h and milled to a particle size of 2-3 mm using a SM-100 cutting mill. Total Solids ⁷⁵ (TS) and Volatile Solids (VS) were analysed according to the APHA standard methods.¹⁴ All measurements were carried out in ⁷⁶ duplicates.

77 Thermo-alkaline and enzymatic pretreatments

78 Thermo-alkaline pretreatments were performed on sunflower stalks at an initial solid organic load of 35 g_{TS} L⁻¹. At 55°C, pretreatment 79 was performed in 500 mL flasks with 4% NaOH (g/100g_{TS}) for 24 hours, in a "Edmund Butler" heating shaker series SM-30-control with ⁸⁰ an agitation speed of 150 rpm. Pretreatment was performed at 170°C in a 1 L Zipperclave autoclave series 02-0378-1 (Autoclave France) si with 4% NaOH for 1 hour under an agitation speed of 150 rpm. After pretreatment, the liquors were separated from the solid fraction by se filtration through a 0.25 mm sieve. Biological Hydrogen Potentials (BHP) batch tests were carried out on total, liquid and solid fractions 83 of the residual solids after alkaline pretreatment (Fig. 1). The remaining liquors were kept at 4°C for further chemical analysis and the s4 solid fractions were washed several times with distilled water to remove water soluble compounds, and then dried at 40°C for 24 h. This 85 washing step allowed removing phenolic compounds which are known to inhibit or deactivate enzymes.¹⁵ Solid fractions were kept at ⁸⁶ 4°C for further chemical analysis and enzymatic hydrolysis. Concerning enzymatic hydrolysis, untreated and dried alkaline-pretreated 87 solid fractions of sunflower stalks were autoclaved during 20 min at 121°C. This sterilization step was required to avoid consumption of ^{ss} the sugars released during enzymatic hydrolysis by indigenous microorganisms present naturally in lignocellulosic substrates.¹⁶ 89 Enzymatic hydrolysis of the untreated and alkaline-pretreated solid fractions was performed at a solid organic load of 50 g remaining TS 90 L⁻¹ in flasks incubated at 35°C in an "Edmund Butler" heating shaker series SM-30-control under 150 rpm agitation. The pH was 91 buffered at 5 using 50 mM of 2-(N-morpholino) ethanesulfonic acid (MES) buffer. Cellulase from Trichoderma reesei (Sigma-Aldrich), 92 β-glucosidase from Aspergillus niger (Sigma-Aldrich) and xylanase from Thermomyces lanuginosus (Sigma-Aldrich) were added at a 93 concentration of 50 U g⁻¹ TS, 25 U g⁻¹ TS and 50 U g⁻¹ TS, respectively. After enzymatic hydrolysis, BHP tests were performed on the 94 total fraction (residual solids + liquor rich is xylose and glucose).

95 Biochemical analysis:

⁹⁶ All samples were milled using a cutting miller Ika Werke MF 10 basic, and a 1 mm mesh. Total Kjeldahl Nitrogen (TKN) was titrated ⁹⁷ using Buchi 370-K after mineralization of the samples. Proteins were determined by multiplying TKN by 6.25. The carbohydrate ⁹⁸ composition of the untreated and solid fractions of pretreated sunflower stalks were measured using strong acid hydrolysis adapted from ⁹⁹ Effland (1977).¹⁷ Samples (200 mg) were hydrolyzed with 12 M H₂SO₄ for 2h at room temperature. They were then diluted to reach an ¹⁰⁰ acid concentration of 1.5 M and kept at 100°C for 3 h. The insoluble residues were separated from the supernatant by filtration with a ¹⁰¹ GFA-WHATMAN glass fiber filter. The insoluble residues were washed with 100 mL of deionised water and then placed in a crucible. ¹⁰² The crucible and the glass fiber filters were dried at 100°C during 24 h and then placed at 550°C during 2 h to determine the Klason ¹⁰³ lignin content. The supernatant was filtrated with nylon filters (0.2 µm) and analyzed for quantification of sugars by High-Pressure ¹⁰⁴ Liquid Chromatography (HPLC).

¹⁰⁵ HPLC analysis was used to quantify monosaccharides (glucose, xylose, arabinose, glucuronic and galacturonic acids). The analysis was ¹⁰⁶ performed with a combined Water/Dionex system, using a Biorad HPX-87H column at 50°C and a refractive index detector. The eluent ¹⁰⁷ corresponded to a 0.005 M H₂SO₄ solution continuously pumped under a flow-rate of 0.3mL.min⁻¹. The system was calibrated with ¹⁰⁸ glucose, xylose, arabinose, glucuronic and galacturonic acids standards (Sigma–Aldrich).

¹⁰⁹ Cellulose and hemicellulose contents were estimated on the basis of monomeric sugar contents. Cellulose is a polymer of glucose while ¹¹⁰ hemicelluloses consist of branched chains of monomeric sugars composed mainly of pentoses such as xylose, arabinose. Consequently, ¹¹¹ the cellulose and hemicellulose contents were estimated as follows:

¹¹² Cellulose (% TS) = Glucose (% TS) / 1.11 (1)

Hemicelluloses (% TS) = [Xylose (% TS) + Arabinose (% TS)] / 1.13 (2),

¹¹⁴ where 1.11 is the ratio of the molecular weights of glucose to glucan (180/162) and 1.13 is the ratio of the molecular weights of xylose ¹¹⁵ and arabinose to xylan (150/132).

116 Metabolites and byproducts contents

¹¹⁷ Volatile Fatty Acids (VFA) composition, *i.e.* acetic (C2), propionic (C3), butyric and iso-butyric (C4 andiC4), valeric and iso-valeric (C5 ¹¹⁸ and iC5) and caproic (C6) acids were determined in the liquid phase using a gas chromatograph (GC-3900, Varian) equipped with a ¹¹⁹ flame ionization detector. The concentrations of non-VFA metabolic byproducts such as lactate and ethanol and other byproducts ¹²⁰ (furfural, 5-hydroxylmethylfurfural) were measured by HPLC analysis and refractometric detection (Waters R410). The components ¹²¹ were separated in an Aminex HPX-87H column (Biorad) maintained at 35°C. The eluent corresponded to 0.005 M H₂SO₄ and flow-rate ¹²² was 0.4 mL.min⁻¹. Total phenols in the liquid fraction of alkaline pretreated samples were determined using a microtube test ¹²³ (Spectroquant, Merck) followed by a 4-aminoantipyrine colorimetric measurement after a two-hundred dilution.

124 Accessible surface area and porosity

¹²⁵ Nitrogen adsorption–desorption was performed on a micromeritics ASAP 2010 volumetric apparatus at 77 K. Before adsorption ¹²⁶ measurements, samples (0.5–0.3 g) were outgassed at 373 K under vacuum during 72 h. Specific surface area and porosity of the samples ¹²⁷ were evaluated using the BET method.¹⁸

128 Biochemical Hydrogen Potential (BHP) tests

¹²⁹ The H₂ production experiments were carried out in batch in 600 mL glass bottles. An anaerobically digested sludge was pretreated by ¹³⁰ heat shock (90°C, 15 min) and then used as inoculum. One milliliter of the inoculum (final concentration around 250 mg-COD L⁻¹) was ¹³¹ added into a culture medium (final working volume of 200 mL) containing 100 mM of 2-(N-morpholino) ethanesulfonic acid (MES) ¹³² buffer, 5 g VS.L⁻¹ of untreated or pretreated sunflower stalks. The initial pH was adjusted to 5.5 using HCl 1 M. In a first series of ¹³³ experiments, BHP tests were performed on thermoalkaline pretreated samples and on the separated liquid and solid fractions to determine ¹³⁴ the part of hydrogen originating from both fractions. In a second series of experiments, BHP tests were performed only on the solid ¹³⁵ residues after alkaline pretreatment with or with no enzymatic hydrolysis. All BHP tests were performed in duplicates. Once prepared, ¹³⁶ the flasks were bubbled with nitrogen to obtain anaerobic conditions, and closed with air impermeable red butyl rubber septum-type ¹³⁷ stoppers. Bottles were then incubated at 35°C. Two milliliters of the mixed cultures were periodically collected and centrifuged (13,000 ¹³⁸ rpm, 10 min). The supernatants were stored at 4°C for further metabolites analysis.

139 DNA extraction, PCR amplification and CE-SSCP fingerprinting

¹⁴⁰ Two milliliters of the cultures were collected after 40 h and 92 h of incubation from all batch experiments. Microbial cells were then ¹⁴¹ centrifuged at 12,100 g for 15 min. Genomic DNA was extracted and purified using the Wizard Genomic DNA Purification kit ¹⁴² (Promega). DNA purity and quantity in the extracts were measured by spectrophotometry (Infinite NanoQuant M200, Tecan). 16S rRNA ¹⁴³ genes were amplified for further fingerprinting analysis using Capillary Electrophoresis Single Strand Conformation Polymorphism (CE-¹⁴⁴ SSCP) as described elsewhere.¹⁹ Fingerprint normalization and comparison were performed using the StatFingerprint library ²⁰ of the R ¹⁴⁵ software version 2.10.1. ²¹

146 Identification of microbial communities by pyrosequencing

¹⁴⁷ Comparison of fingerprints led to the selection of one genomic DNA representative for each experimental condition. Identification of the ¹⁴⁸ main dominant bacterial populations was performed by pyrosequencing of the V4-V5 regions of the 16S rRNA gene (Molecular ¹⁴⁹ Research Laboratory, TX, USA). An average of 5161 sequences per sample was obtained, and sequence data derived from the ¹⁵⁰ sequencing process was processed using a proprietary analysis pipeline (Molecular Research Laboratory, TX, USA). Sequences were ¹⁵¹ first clustered at 3% of divergence and then taxonomically classified using BLASTn against a curated GreenGenes database.

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153 Results and discussion:

154 Effect of thermo-alkaline and two-stage alkaline-enzymatic pretreatments on biochemical and structural features

To investigate the biochemical changes induced by the thermo-alkaline pretreatment, variations in proteins, uronic acids, hemicelluloses, 156 cellulose and Klason lignin contents were estimated (Figure 1). Uronic acids which are constituents of hemicelluloses and pectins were 157 removed at 64% after alkali treatment at 55°C, and were totally removed at 170°C, as a consequence of the cleavage of the lignin-158 hemicellulose linkages through uronic acids, so-called lignin–carbohydrate complexes (LCC). LCC remain the main obstacle for the 159 utilization of lignocellulosic biomass.⁴ Their cleavage makes holocelluloses more accessible for enzymatic hydrolysis and microbial 160 fermentation. Thermo-alkaline pretreatment at 170°C was found efficient to solubilize proteins since 54% were removed. Significant 161 hemicelluloses removals of 26% and 58% were observed after thermo-alkaline pretreatment at 55°C and 170°C, respectively. In contrast, ¹⁶² it was shown less than 8% of cellulose removal after thermo-alkaline pretreatment at both 55°C and 170°C. Finally, high lignin removals ¹⁶³ of 22% and 36% were observed after thermo-alkaline pretreatment at 55°C and 170°C, respectively. Such results are in accordance with ¹⁶⁴ previous studies reporting that alkaline pretreatment is efficient to delignify and remove partly lignin and hemicelluloses whereas ¹⁶⁵ cellulose is preserved.^{9,10} As a result of such concomitant and similar removal of lignin and holocelluloses, no significant variation of the ¹⁶⁶ lignin content ranging from 32.2 to 33.5 g/100g remaining VS was observed between alkaline pretreated solids and untreated sunflower ¹⁶⁷ stalks. In contrast, a relative increase of 24% and 60.5% of the cellulose content compared to raw sunflower stalks was observed after ¹⁶⁸ alkaline pretreatment at 55°C and 170°C, respectively. This result shows that cellulose was less removed compared to the other ¹⁶⁹ compounds during thermo-alkaline pretreatments.

¹⁷⁰ Table 1 shows the structural features such as accessible surface area (SA) and pore volume (Vp) of the untreated and alkaline-pretreated ¹⁷¹ solid fractions of the sunflower stalks. Low increase of SA and Vp was observed after thermo-alkaline pretreatment at 55°C. SA ¹⁷² increased from 1.55 m² g⁻¹ TS (sunflower stalks) to 1.59 m² g⁻¹ TS and Vp increased from 0.083 cm³ g⁻¹ TS (sunflower stalks) to 0.106 ¹⁷³ cm³ g⁻¹ TS. Thermo-alkaline pretreatment at 170°C was found more efficient in increasing SA and Vp with final values of 2.55 m² g⁻¹ TS ¹⁷⁴ and 0.235 cm³ g⁻¹ TS, respectively. Similarly, Gharpuray et al. (1983) observed an increase of the accessible surface area from 0.64 to 1.7 ¹⁷⁵ m² g⁻¹ TS by pretreating wheat straw at 100°C with 10% NaOH (w/w) during 30 min.²²

176 In addition, enzymatic hydrolysis of the raw sunflower stalks and the residual solids obtained after alkaline pretreatment was further 177 investigated by estimating the solubilization of cellulose and hemicelluloses to glucose and xylose which are the two main monomeric 178 sugars found in holocelluloses (Figure 1). A low enzymatic hydrolysis of hemicelluloses (12%) and cellulose (35%) was observed on raw 179 sunflower stalks. After thermo-alkaline pretreatments, the enzymatic hydrolysis yields increased for both cellulose and hemicelluloses. 180 Indeed, after alkaline pretreatment at 55°C and enzymatic hydrolysis, an increase of 45% cellulose and 56% hemicelluloses 181 solubilization was observed. After alkaline pretreatment at 170°C and enzymatic hydrolysis, a higher increase of 51% and 82% was 182 observed for cellulose and hemicelluloses, respectively. Similar results were previously reported by combining alkaline-enzymatic 183 pretreatment on spruce and where enzymatic hydrolysis of both cellulose and hemicelluloses was substantially enhanced.²³ Several 184 factors may explain the increase of enzymatic hydrolysis after thermo-alkaline pretreatment. First, the increase of accessible surface area 185 and pore volume, as previously observed after thermo-alkaline pretreatment, can enhance the efficiency of enzymatic hydrolysis. 186 Consistently, good correlations between pore volume, accessible surface area and enzymatic digestibility of lignocellulosic residues were 187 previously reported.²⁴ Gharpuray et al. (1983) showed that the specific surface area can impact biomass digestibility: an increase of the 188 accessible surface area resulted in higher hydrolysis yields.²² However, the low increase of both accessible surface area and pore volumes 189 between raw sunflower stalks and sunflower stalks alkaline-pretreated at 55°C suggested that probably other factors were favorable for 190 enzymatic hydrolysis. Among them, removal of pectins (polymers of uronic acids) after thermo-alkaline pretreatment might explain the 191 increase of enzymatic hydrolysis efficiency. As described elsewhere, removal of pectins in hemp was shown to increase enzymatic 192 hydrolysis by 26%.²⁵ Even though thermo-alkaline pretreatment did not change the lignin content in the solid residues, a physical 193 redistribution of the lignin probably occurred with a cleavage of the LCC, that play a key role for enzyme accessibility and further 194 biomass digestibility.

196 Biohydrogen performances

197 Effect of thermo-alkaline and two-stage alkaline-enzymatic pretreatments on biohydrogen yields

¹⁹⁸ Gas product analysis of Biological Hydrogen Potential (BHP) tests showed that only H_2 and CO_2 were produced during dark ¹⁹⁹ fermentation, with no detectable CH_4 in the biogas. This suggests that heat shock treatment (90°C, 15 min) of the inoculum was efficient ²⁰⁰ to suppress methanogens from anaerobic sludge. A low hydrogen yield of 2.3 ± 0.9 mL $H_2.g^{-1}$ VS was observed for the raw sunflower ²⁰¹ stalks (Table 1). A small increase in hydrogen yields to 4.4 ± 2.6 mL $H_2.g^{-1}$ VS and 20.6 ± 5.6 mL $H_2.g^{-1}$ VS were observed after thermo-²⁰² alkaline pretreatments at 55°C and 170°C, respectively. Similar results were obtained by pretreating grass with 0.5% NaOH (w/v) boiled ²⁰³ for 30 min with an increase of the hydrogen yield from 4.4 mL $H_2.g^{-1}$ dry grass for the untreated sample to 19.3 mL $H_2.g^{-1}$ dry grass after ²⁰⁴ treatment.¹¹ Alkaline pretreatment (0.4% NaOH, 24 h, room temperature) of sweet sorghum stalks led also to an increase of hydrogen ²⁰⁵ potentials (127 mL $H_2.g^{-1}$ VS compared to 52 mL $H_2.g^{-1}$ VS for untreated).²⁶ The enhancement of hydrogen yields coincided with an ²⁰⁶ increase in soluble sugars after alkali pretreatment compared to raw sweet sorghum stalks with a sugar concentration of 2.23 and 0.86 ²⁰⁷ g.L⁻¹, respectively.²⁶

208 Biohydrogen potentials of separated liquid and solid fractions obtained after thermo-alkaline pretreatment at 55°C and 170°C were 209 further investigated to determine from which fraction the biohydrogen was produced (Table 1). At 55°C, biohydrogen production from $_{210}$ the liquid and solid fractions was similar with 2.6 and 2 mL H₂.g⁻¹ VS, respectively. In contrast, at 170°C, biohydrogen yield of the liquid ²¹¹ fraction was twice higher than solid fraction with 11.4 and 5.1 mL H₂.g⁻¹ VS, respectively. To analyze the performances of biohydrogen 212 production on these different fractions, biohydrogen yields were expressed in terms of initial sugar contents (glucose, xylose and 213 arabinose) according to the amount of holocelluloses present in the different fractions (total, solid and liquid) as shown in Table 1. 214 Concerning the liquid fractions, hydrogen yields of 0.344 and 0.865 mol H2.mol⁻¹ hexose were obtained after 55°C and 170°C 215 pretreatments, respectively. Nevertheless, the hydrogen yields from the liquid phase remained lower than the values commonly reported ²¹⁶ in the literature with simple sugars and mixed cultures, *i.e.* 2 mol H₂.mol⁻¹ glucose and 1.92–2.25 mol H₂.mol⁻¹ xylose. ^{19, 27} Two 217 hypotheses can be formulated to explain these results. The first one is that soluble sugars released in the liquid phase are present in the 218 form of oligomeric and monomeric sugars. Recently, Quéméneur et al. (2011) showed the impact of the low carbohydrate structure ²¹⁹ where di- and tri-saccharides affected significantly hydrogen production, and longer chains of oligomers lowered the hydrogen yields.²⁸ 220 The second hypothesis is the generation of byproducts during thermo-alkaline pretreatments such as furfural, 5-HMF and phenol 221 compounds. Even though no byproducts of sugar degradation (furfural and 5-HMF) were detected in the liquid phase after thermo- $_{222}$ alkaline pretreatments, phenols coming from lignin degradation accumulated in the liquid phase at a low concentration of 25.7 mg L⁻¹ 223 and 42.7 mg L⁻¹ at 55°C and 170°C, respectively (data not shown), and phenolic compounds were previously shown to negatively impact ²²⁴ biohydrogen production from xylose using mixed cultures ^{15,19,29}.

²²⁵ Concerning the solid fractions, low hydrogen yields of 0.043 and 0.081 mol H_2 .mol⁻¹ hexose were obtained, respectively, with raw ²²⁶ sunflower stalks and after alkaline pretreatment at 55°C. The increase of the hydrogen yields to 0.380 mol H_2 .mol⁻¹ hexose after thermo²²⁷ alkaline pretreatment at 170°C on the solid fraction was likely due to the increase of the accessible surface area as shown in Table 1. ²²⁸ However, these hydrogen yields remained low likely because of the nature of the sugars present in the solid fractions which are in their ²²⁹ polymeric and not monomeric form. Since low hydrogen potentials were obtained whatever the solids, enzymatic hydrolysis was ²³⁰ performed on raw sunflower stalks as well as on the solid fractions after alkaline pretreatment of the straws at 55°C and 170°C to ²³¹ enhance the solubilization of holocelluloses into monomeric sugars.

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After enzymatic pretreatment of raw sunflower stalks, a maximal cumulative hydrogen production of 30.4 (\pm 3.8) mL H₂.g⁻¹ VS was reached which was 13-fold higher than untreated sunflower stalks. Coupling alkaline-pretreatment with enzymatic hydrolysis led to a significant increase of the hydrogen yields compared to enzymatic hydrolysis alone with 59.9 (\pm 2.6) mL H₂.g⁻¹ VS and 80.9 (\pm 3.0) mL H₂.g⁻¹ VS at 55°C and 170°C, respectively. Such increase of hydrogen production was concomitant to the increase of enzymatic hydrolysis through the release of soluble monomer sugars (ie glucose and xylose). Hydrogen yields of 0.562, 0.959 and 1.158 mol H₂.mol⁻¹ hexose were obtained after enzymatic hydrolysis of raw sunflower stalks and alkaline pretreated sunflower stalks at 55°C and 170°C, respectively. Although a substantial increase was observed, these results remained lower than the average values of 2 mol H₂.mol⁻¹ glucose and 1.92–2.25 mol H₂.mol⁻¹ xylose commonly reported in the literature when using mixed cultures.^{19,27} These results suggest that a part of holocelluloses was still not degraded during the dark fermentation process. Interestingly, if hydrogen potentials were expressed by considering only the initial content in soluble sugars monomers (ie xylose and glucose) released after enzymatic pretreatment, hydrogen yields ranged from 1.97 to 2.14 mol H₂.mol⁻¹ soluble sugars, in strict accordance with data previously published. Thus, hydrogen was mainly produced from soluble sugar monomers released during enzymatic hydrolysis which is also in accordance with the literature.^{6,26}

²⁴⁶ Furthermore, once the maximum cumulative hydrogen production was reached, microbial metabolites were also analyzed. The ²⁴⁷ fermentation metabolites produced with or without enzymatic hydrolysis of sunflower stalks and for the three conditions (raw, solid ²⁴⁸ residue from alkaline treatment at 55°C and 170°C) were only acetate and butyrate (data not shown). This suggests that H₂ was produced ²⁴⁹ only by the well described metabolic fermentative H₂-producing pathways of acetate and butyrate. Butyrate as well as acetate production ²⁵⁰ are indeed typical of a dark fermentation process when using substrates rich in carbohydrates.³⁰

251 General remarks on biohydrogen potentials

To assess the advantage of combining thermo-alkaline pretreatment with enzymatic hydrolysis, results were expressed according to the 253 initial VS, as shown in Figure 2. In the case of thermo-alkaline pretreated stalks, the hydrogen produced from the liquid fraction after 254 thermo alkaline pretreatment and the hydrogen produced from the solids alkaline-pretreated at 55°C and 170°C and with or without 255 enzymatic hydrolysis were considered separately. Moreover, a theoretical hydrogen potential (100 L H₂.kg⁻¹ initial VS) was calculated by 256 considering the content of holocelluloses in sunflower stalks and a hydrogen production of 2 mol H₂.mol⁻¹ eq glucose as commonly 257 observed with mixed cultures.¹⁶

²⁵⁸ As mentioned previously, a hydrogen yield of 30.4 mL H₂·g⁻¹ initial VS was obtained after enzymatic pretreatment of sunflower stalks,

which was 13-fold higher than raw sunflower stalks (2.3 \pm 0.9 mL H₂.g⁻¹ initial VS). When considering combined thermo-alkaline pretreatment at 55°C and 170°C with enzymatic hydrolysis, hydrogen potentials were respectively 21 to 26-fold higher than with untreated stalks. In both cases, the main part of hydrogen was coming from the enzymatic-alkaline-pretreated solid fraction which corresponded respectively to 81% and 95% of total biohydrogen production. When considering the initial VS content, coupling thermo-alkaline pretreatment at 170°C with enzymatic hydrolysis was not substantially beneficial compared to 55°C since hydrogen yields increased only from 49 to 59.5 mL H₂.g⁻¹ initial VS. Overall, hydrogen yields obtained after enzymatic hydrolysis with or without coupling with thermosalkaline pretreatments remained always lower than the highest hydrogen yields that can be expected if holocelluloses would have been fully converted to H₂ during the dark fermentation process. In addition, and to make the process economically viable, the effluents of the H₂ form H₂ fermentation process should also be considered for further valorisation as either hydrogen by photo-fermentation or methane by anaerobic digestion.^{4,31}

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270 Bacterial community analysis

²⁷¹ By investigating the metabolic routes of the dark fermentation process, a diauxic behavior of the mixed culture was observed on the ²⁷² consumption of soluble sugars (ie glucose and xylose) released after enzymatic hydrolysis. Glucose was first consumed and, once glucose ²⁷³ was completely exhausted, xylose was utilized by the microorganisms (Fig. 3). Such diauxic effect has been well described with pure ²⁷⁴ cultures, but to our knowledge, no study reported such effect with complex mixed cultures. Diauxic growth usually occurs when bacterial ²⁷⁵ cells of an individual species are exposed to multiple carbon sources leading to a catabolic repressive effect that does not permit ²⁷⁶ simultaneous consumption of all sugars, *i.e.* sugars are sequentially consumed resulting in two successive exponential growth phases that ²⁷⁷ are separated by an intermediate lag phase. Glucose is a preferred substrate for many microorganisms, and is responsible of a common ²⁷⁸ catabolic repression which is commonly called the "glucose effect".³² In mixed culture, several bacterial populations with different ²⁷⁹ simultaneously.³³ Therefore, a diauxic behaviour could result from a diauxic growth but also from a shift of highly specific microbial ²⁸⁰ simultaneously.³³ Therefore, a diauxic behaviour could result from a diauxic growth but also from a shift of highly specific microbial ²⁸¹ populations. To support the hypothesis of a diauxic catabolic repression effect, a bacterial analysis of the H₂-producing bacteria present in ²⁸² mixed culture was performed at two experimental times: 40 h and 92 h, corresponding to the end of glucose and xylose consumption, ²⁸⁴ respectively (red arrows in Figure 3). The number of sequences analyzed per sample ranged from 3980 to 6366 and the number of ²⁸⁴ identified species varied between 23 and 34 according to the sample. The bacterial species with a relative abundance above 1% at 40 h and ²⁸⁵ 92 h in the dark H₂ fermentation batch are summarized in Table 2.

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²⁸⁷ For all samples, the bacterial community was dominated by members of the *Clostridium* genus with an average of 92% of the total ²⁸⁸ abundance. The composition of the bacterial communities in this study was similar to the ones reported in previous studies carried out in ²⁸⁹ dark fermentation systems.^{19,34} For instance, in granular sludge of H₂-producing bioreactor using sucrose as substrate, 69.1% of the clones ²⁹⁰ were affiliated to four *Clostridium* species.³² Similarly, a proportion of *Clostridium* species of 85.8% was observed during dark ²⁹¹ fermentation of xylose and after heat shock pretreatment (90°C, 10 min) of the anaerobic mixed culture inoculum.¹⁹

292 During the glucose phase consumption, a specific enrichment of *Clostridium butyricum* amongst the other *Clostridium* species was 293 observed with proportions varying from 80.3 % to 92.9%. In contrast, during xylose consumption, even though Clostridium butyricum 294 remained the main dominant species, other *Clostridium* species clearly outcompeted. More particularly, and except for *Clostridium* 295 bifermentans, the relative abundance of Clostridium butyricum decreased between 40 h and 92 h, while the abundance of minority species 296 such as Clostridium roseum, Clostridium aciditolerans, Clostridium saccharoperbutylacetonicum and Clostridium tunisiense increased 297 after 40 h. Consequently, the diauxic effect observed on glucose/xylose consumption was not only related to a diauxic growth of 298 Clostridium butyricum but also to the development of highly competitive Clostridium species during the change of sugar degradation 299 pathway. This result suggests that the diauxic effect could be common to the whole *Clostridium* genus. Diauxic effects were previously 300 reported on ethanol production from glucose/xylose mixture using pure cultures, ie. Saccharomyces cerevisiae and E. coli ^{35, 36}. So far, and 301 to our knowledge, such diauxic effect has not been so clearly reported with fermentative mixed cultures. In general, diauxic effect can be 302 avoided by the use of mixed cultures instead of pure cultures since the presence of various bacterial populations enhances the possibilities 303 to convert concomitantly a large range of substrates due to microbial diversity, metabolic flexibility, and the possibility of specific ³⁰⁴ individual ecological niches.³⁵ In this study, the heat shock pretreatment initially applied on anaerobic sludge led not only to the complete 305 suppression of Archaea methanogens, but also to a strong selection of Clostridium species with a low final number of bacteria species 306 ranging from 23 to 34 only. This observation is in accordance with Baghchehsaraee et al. (2008) who showed that heat pretreatment (80°C ₃₀₇ or 95°C) of inocula led to a substantial decrease of bacterial diversity.³⁷ Comparatively, much higher numbers of bacteria ranging from ³⁰⁸ 114 to 164 were reported on three granular and non-granular mesophilic cultures which were not heat treated.³⁸

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310 Acknowledgements

³¹¹ The authors are grateful to ADEME, the French Environment and Energy Management Agency, for financial support in the form of F. ³¹² Monlau's PhD grant.

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TABLES and FIGURES CAPTIONS

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⁴²⁵ Table 1 Accessible surface area (SA), Volume pores (Vp) Sugars contents in equivalent hexose (g eq hexose g⁻¹ VS) and biohydrogen potentials originating 426 from total, liquid and solid fractions of raw sunflower stalks (SS) and alkaline-pretreated sunflower stalks at 55°C(SS 55°C) and 170°C (SS 170°C) with or without enzymatic hydrolysis 427

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- 429
- Table 2 Repartition of the bacterial population in % of order, genus and species during dark fermentation process at 40 h and 92 h after enzymatic hydrolysis of raw sunflower stalks (SS), alkaline pretreated sunflower stalks at 55°C (SS, 55°C) and at 170°C (SS, 170°C). 430
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- Figure 1 Scheme of pretreatment strategies used and biochemical composition of raw sunflower stalks (SS) expressed in g/100g initial VS and of the solid 432 fractions after thermo-alkaline pretreatments at 55°C (SS 55°C) and at 170°C (SS 170°C) expressed in g/100g initial VS and in g/100g remaining VS. 433 434 Values correspond to means of two replicates of independent values \pm standard deviation.

⁴³⁵ Figure 2 Impact of pretreatments on the origin of the biohydrogen production (ml H₂, g⁻¹ initial VS) for raw sunflower stalks and solid residue of alkaline $_{436}$ pretreated sunflower stalks at 55°C and 170°C with or without enzymatic hydrolysis. Values correspond to means of two replicates of independent values \pm standard deviations (error bars). 437

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Figure 3 Evolution of sugars monomers consumption (ie glucose and xylose) released from enzymatic hydrolysis during H₂ fermentation in mixed culture 439 (a) from raw sunflower stalks (b) from solid residue of sunflower stalks pretreated at 55°C, 24 h, 4% NaOH (c) from solid residue of sunflower stalks 440 pretreated at 170°C, 1 h, 4% NaOH. Values correspond to means of two replicates of independent values ± standard deviations (error bars). 441

		Structural features		Sugars content	Biohydrogen potentials	
		$SA (m^2.g^{-1})$	$Vp (cm^3.g^{-1})$	(g eq hexose.	mL $H_2.g^{-1}$ VS	mol H ₂ .mol ⁻¹
Conditions	Fractions	remaining TS)	remaining TS)	$100g^{-1}$ VS)		eq hexose
SS	Total	1.55	0.083	43.5	2.3 ± 0.9	0.043
SS (55°C)	Total			43.5	4.4 ± 2.6	0.081
	Liquid			6.1 (14%) ^a	$2.6 \pm 0.2 (57\%)^{b}$	0.344
	Solid	1.59	0.106	37.4 (86%) ^a	$2 \pm 0 (43\%)^{b}$	0.043
SS (170°C)	Total			43.5	20.6 ± 5.6	0.380
	Liquid			10.6 (24.3%) ^a	$11.4 \pm 0.3 \ (69\%)^{b}$	0.865
	Solid	2.55	0.235	32.9 (75.7%) ^a	$5.1 \pm 0.4 (31\%)^{b}$	0.125
SS + Enzymes	Total			43.5	30.4 ± 3.8	0.562
SS (55°C) +						
Enzymes	Total			50.2	$59.9 \pm 2.6^{\circ}$	0.959
SS (170°C) +						
Enzymes	Total			56.1	$80.9\pm3.0^{\rm c}$	1.158

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 Table 1 Accessible surface area (SA), Volume pores (Vp) Sugars contents in equivalent hexose (g eq hexose.g⁻¹ VS) and biohydrogen potentials originating from total, liquid and solid fractions of raw sunflower

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 stalks (SS) and alkaline-pretreated sunflower stalks at 55°C(SS 55°C) and 170°C (SS 170°C) with or without enzymatic hydrolysis

^a Percentage of total glucose originating from the liquid and solid fractions. In the liquid fraction, sugar content in equivalent glucose was calculated according the removal of holocelluloses in the initial total fraction.

^b Percentage of biohydrogen potentials originating from the liquid and solid fractions ^c Biohydrogen potentials are expressed in ml H₂.g⁻¹ remaining VS

Conditions SS SS SS (55°C) SS (55°C) SS (170°C) SS (170°C) 40 h 92 h 40 h 92 h 40 h 92 h Number of sequencing 6366 4715 4791 6172 3980 4940 28 23 34 25 30 Number of species 26 % **ORDER** / Genus species^a **CLOSTRIDIALES** 91 92 95 96 92 85 Clostridium butyricum 89.0 80.3 Clostridium perfringens 0.0 0.0 0.0 0.0 0.1 Clostridium roseum 0.7 0.7 0.4 Clostridium aciditolerans 0.9 1.5 0.5 1.5 1.5 Clostridium saccharobutylicum 0.0 0.0 0.0 0.1 0.0 2.7 Clostridium tertium 0.2 0.8 0.8 1.9 0.6 0.4 Clostridium lundensis 0.0 1.7 0.0 0.4 0.4 0.0 Clostridium peptidivorans 0.2 0.6 0.1 1.1 0.4 0.2 Clostridium saccharoperbutylacetonicum 0.3 3.3 0.3 1.1 0.5 2.6 Clostridium tunisiense 1.0 0.0 3.5 0.1 0.1 0.3 Clostridium bifermentans 0.0 0.0 1.0 0.1 0.9 0.3 Clostridium thiosulfatireducens 0.0 3.2 0.0 0.0 0.0 0.2 Clostridium botulinum 0.0 1.7 0.0 0.0 0.0 0.4 Clostridium paraputrificum 0.0 1.0 0.0 0.0 0.0 0.1 BACILLALES 8 7 3 5 13 4 Rummeliibacillus pycnus 0.0 0.0 0.6 3.0 0.0 0.0 0.0 0.0 0.0 1.0 Sporolactobacillus racemilacticus 0.0 1.1 Bacillus ginsengihumi 2.0 3.0 0.6 1 1 1 3 2 1 Others

Table 2 Repartition of the bacterial population in % of order, genus and species during dark fermentation process at 40 h and 92 h after enzymatic hydrolysis of raw sunflower stalks (SS), alkaline pretreated sunflower stalks at 55°C (SS, 55°C) and at 170°C (SS, 170°C).

⁴⁵⁴ ^{*a*} Among the species observed, only species with a relative abundance above 1% are represented in the table

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Figure 1 Scheme of pretreatment strategies used and biochemical composition of raw sunflower stalks (SS) expressed in g/100g initial VS and of the solid
 fractions after thermo-alkaline pretreatments at 55°C (SS 55°C) and at 170°C (SS 170°C) expressed in g/100g initial VS and in g/100g remaining VS.
 Values correspond to means of two replicates of independent values ± standard deviation.



Figure 2 Impact of pretreatments on the origin of the biohydrogen production (ml H_2 .g⁻¹ initial VS) for raw sunflower stalks and solid residue of alkaline pretreated sunflower stalks at 55°C and 170°C with or without enzymatic hydrolysis. Values correspond to means of two replicates of independent values ± standard deviations (error bars).



Figure 3 Evolution of consumption of monomeric sugars (ie glucose and xylose) released from enzymatic hydrolysis during H₂ fermentation in mixed
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