

Acetate oxidation Is the dominant methanogenic pathway from acetate in the absence of Methanosaetaceae

Dimitar Karakashev, Damien J. Batstone, Eric Trably, Irini Angelidaki

▶ To cite this version:

Dimitar Karakashev, Damien J. Batstone, Eric Trably, Irini Angelidaki. Acetate oxidation Is the dominant methanogenic pathway from acetate in the absence of Methanosaetaceae. Applied and Environmental Microbiology, 2006, 72 (7), pp.5138-5141. 10.1128/AEM.00489-06. hal-02666051

HAL Id: hal-02666051 https://hal.inrae.fr/hal-02666051

Submitted on 9 Aug 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1	Acetate oxidation is the dominant methanogenic pathway from acetate in the
2	absence of Methanosaetaceae.
3	
4	Acetate oxidation dominates without Methanosaetaceae
5	
6	Dimitar Karakashev, Damien J. Batstone, Eric Trably and Irini Angelidaki *
7	
8	Institute of Environment & Resources DTU
9	Technical University of Denmark
10	Building 113, DK-2800, Lyngby, Denmark
11	
12	* Corresponding author. Phone: (+45) 45251429, Fax: (+45) 45932850,
13	E-mail: ria@er.dtu.dk
14	

15

ABSTRACT

16 Oxidation of the acetate to hydrogen, and subsequent combination of hydrogen with 17 carbon-dioxide to methane, has largely been regarded as a niche mechanism at high 18 temperatures or under inhibitory conditions. In this study, 13 anaerobic reactors, and 19 sediment from a temperate anaerobic lake were surveyed for dominant methanogenic 20 population using fluorescent in situ hybridisation, and degree of acetate oxidation relative to aceticlastic conversion, using radiolabelled 2-[¹⁴C] acetate in batch incubations. When 21 22 Methanosaetaceae was not present, acetate oxidation was the dominant methanogenic 23 pathway. Aceticlastic conversion was only observed in presence of Methanosaetaceae.

24

25

26 Acetate is the main precursor for methane production during anaerobic digestion of 27 organic matter. Two mechanisms for methane formation from acetate have been 28 described. The first one is aceticlastic, carried out by Methanosarcinaceae or 29 Methanosaetaceae (2). Methanosarcinaceae generally have a higher acetate threshold, 30 but higher growth rate and yield compared to Methanosaetaceae (2). The second 31 mechanism encompasses a two-step reaction in which acetate is first oxidised to H_2/CO_2 32 and with these products subsequently converted to methane (15). This reaction is 33 performed by acetate-oxidising bacteria (often *Clostridium*) in a syntrophic association 34 with hydrogenotrophic methanogens (often *Methanomicrobiales* or *Methanobacteriales*) 35 (4, 10, 12).

36 Some important environmental factors influencing the rate of anaerobic aceticlastic 37 activity are temperature, organic acid concentrations and ammonia concentration (9). At 38 temperatures between 50°C and 65°C, acetate oxidation is favoured at low acetate 39 concentrations, while aceticlastic methanogenesis is favoured at high acetate 40 concentrations (15). The dominance of acetate oxidation at lower concentrations 41 increases with increased temperature. Syntrophic acetate oxidation is the main 42 mechanism for acetate degradation in the presence of inhibitors, particularly ammonium, 43 and volatile fatty acids (VFAs) (13). Syntrophic acetate oxidation has been reported for 44 natural anoxic environments in subtropical lake sediments at temperatures down to 15°C 45 (8).

It is relatively straightforward to detect acetate oxidation activity by measuring the production of ${}^{14}CH_4$ and ${}^{14}CO_2$ from acetate labeled in the methyl group (C-2). When aceticlastic methanogens degrade acetate, the labeled methyl group will form only 49 labeled methane (2). During syntrophic acetate oxidation, both carbon atoms of acetate 50 are converted to carbon-dioxide, and some of the carbon dioxide is subsequently reduced 51 to methane (13). Therefore, significant levels of labeled carbon dioxide from $[2-^{14}C]$ -52 acetate will only be formed during oxidation of acetate.

53 The diversity of environments in which syntrophic acetate oxidation has been found 54 indicates it may also be important for commercial gas production in biogas reactors, 55 digesting wastewater sludge and manure. Aceticlastic activity has generally been 56 considered to be the dominant pathway, with either Methanosarcinaceae or 57 Methanosaetaceae dominating (9, 15). If a second pathway such as acetate oxidation 58 dominates, it is necessary to re-evaluate reactor operation and optimisation, currently 59 based on maintaining *Methanosaetaceae* populations. The objective of this work was to 60 assess the degree of acetate oxidation relative to aceticlastic conversion in a wide range 61 of industrial anaerobic digesters, fed either with manure or wastewater sludge. A low-62 temperature environmental sample was also evaluated.

Sampling. Thirteen Danish full-scale anaerobic digesters were sampled (Table 3) as
described in (5). organic waste from abattoirs or food industries. An anaerobic sediment
sample was collected from a lake situated in Orholm (Sollerod municipality, Denmark) at
0.2 m depth with a gravity corer (6).

Analysis of the samples. The samples were analyzed for volatile fatty acids (VFAs) and ammonia by standard methods (1). Microbial ecology was evaluated using fluorescent *in-situ* hybridisation FISH with established probes, and method as previously reported (5). Methanogenic populations not identified by FISH were assessed using polymerase chain reaction coupled to temperature gradient gel electrophoresis (PCR- TGGE). Sequence data for identified microbes have been submitted to the Genbank
database under the accession numbers DQ409324 to DQ409326.

74 Medium. Basal anaerobic (BA) medium was used for acetic oxidation batch tests as 75 described previously (5). The medium was dispensed anaerobically under a N_2/CO_2 (80) %: 20 %) headspace in 100 ml incubation bottles, amended with labeled 2-[¹⁴C]-sodium 76 77 acetate and non-labeled sodium acetate. The medium was reduced with Na₂S.9H₂O and 78 supplemented aseptically with a sterilely filtered anaerobic vitamin solution as described 79 previously. After inoculation with raw sample the bottles were closed hermetically and 80 incubated until methane production ceased. This was treated as the end of the test, and 81 analysis followed.

Radioisotope analyses. The liquid and headspace was sparged with approx 2L N₂, through a 5M NaOH trap to collect the ¹⁴CO₂. The ¹⁴CH₄ collected after trapping was combusted to ¹⁴CO₂ in a tube furnace at 800°C. ¹⁴CO₂ generated in this furnace was then trapped in Carbosorb-E (carbon dioxide absorber for liquid scintillation counting, Packard Bioscience Company, USA). Radioactivity measurements of liquid samples were performed on a liquid scintillation counter (Tri-Carb 1600, Perkin Elmer, England).

Simulation of Methane Production Rates. A simple kinetic batch model, based on Monod kinetics with zero order lag, for conversion of acetate to methane was implemented in Aquasim 2.1d (11). The maximum acetate removal rate and lag phase were estimated by fitting measured cumulative methane to modeled cumulative methane. The Secant method, with an objective function of residual sum of squares was used to fit the data. 94 An overview of the results from the acetate oxidation survey experiment is given in95 Table 1.

96 Rates of methane production and acetate removal. Methane production rates varied 97 considerably, with fast samples (such as Lundtofte and Hillerød) stopping methane 98 production in 3 days, and slow samples (e.g., Nysted and Vegger) requiring more than 10 99 days. The anaerobic lake sediment sample (Orholm) had a lag phase of 31.5 ± 0.8 days. 100 Acetate removal rates also varied within a factor of approximately 10 (Table 1). These rates were higher (>4 mM.day⁻¹) in cultures with low acetate oxidation degree in 101 102 comparison with cultures with high acetate oxidation degree (the acetate utilisation rates lower than 4 mM.day⁻¹). Our rates compare with acetate removal rates in pure culture for 103 104 mesophilic (12) and a thermophilic acetate oxidising cultures (4).

105 Anaerobic acetate conversion pathways and environmental conditions. In all cases, 106 populations dominated by Methanosaetaceae had minimal acetate oxidation degree $({}^{14}CO_2 / {}^{14}CH_4 < 0.1)$, while populations dominated by other methanogenic Archaea, and 107 without *Methanosaetaceae* had a high degree of acetate oxidation (${}^{14}CO_2 / {}^{14}CH_4 > 1$) 108 109 (Table 1). Results obtained clearly showed a strong correlation between the absence of 110 Methanosaetaceae and involvement of acetate oxidation pathway. Other factors (e.g., 111 source and inoculum temperature) had no influence. Acetate cleavage has been generally 112 regarded as a bimodal system, dominated by *Methanosarcinaceae* at high acetate 113 concentrations, and Methanosaetaceae at low acetate concentrations (2, 14). From the 114 data presented here, we propose instead a different bimodal system in mixed cultures, 115 with aceticlastic methanogenesis in the presence of *Methanosaetaceae*, and acetate 116 oxidation in its absence. The absence of this methanogenic phylogenetic group has been 117 previously investigated in the systems analysed here, and was linked to the presence of 118 high ammonia and VFA levels (5). Most probably the high ammonia concentrations 119 inhibit the aceticlastic methanogens much more than the hydrogenotrophic methanogens, 120 and methane formation is mainly by hydrogen-utilising methanogens. This is supported 121 from previous studies (3) indicating that acetate-utilising methanogens are more sensitive 122 to ammonia than are hydrogenotrophic methanogens. The high degree of acetate 123 oxidation in digested manure at high ammonia and VFA levels is also in agreement with 124 other results (13). However a large potential for syntrophic acetate oxidation was also 125 observed at low acetate concentrations (in the Orholm sample). It is likely that inhibition 126 or other factors prevent growth of Methanosaetaceae, and allow dominance by acetate 127 oxidation by default.

128 The bimodality of the system is also highlighted in Figure 1, which shows two distinct 129 groups, with hydrogen-utilising methanogens (Methanobacteriales, Methanococcales, 130 Methanomicrobiales, possibly Methanosarcinaceae, uncultured archae (Hashøj and 131 Lemvig) and unidentified archae (Studsgard) in syntrophic cooperation with acetate 132 oxidisers at low maximum acetate removal rates, and the strict aceticlastic methanogen 133 at maximum acetate removal rates. The presence of Methanosaetaceae 134 Methanosarcinaceae as a hydrogen utilising syntrophic partner in the acetate oxidising 135 cultures is not surprising. In contrast to the *Methanosaetaceae*, which is a strict aceticlast, 136 most Methanosarcinaceae species are mixotrophic, utilising not only acetate but also 137 hydrogen and carbon dioxide, methanol and methylamines (2). In addition, 138 Methanosarcinaceae are capable themselves of acetate oxidation (7), and could be therefore be mediating the entire process of acetate oxidation, and subsequentmethanogenesis, rather than acting as an acetate sink via aceticlastis.

141 Methanogenic populations. The FISH observations showed that dominant 142 methanogenic population of wastewater sludge samples was consistently 143 Methanosaetaceae, as previously reported (5), while manure samples were 144 phylogenetically more diverse. In every case, dominance of specific groups as observed 145 by FISH was clear, and they constituted more than 90% of the archaeal population, as 146 described previously (5)). Diversity in subdominant methanogens was limited, except in 147 the Orholm (sediment sample), where archaea belonging to Methanosaetaceae, 148 Methanomicrobiales, and Methanococcales were observed. Methanogenic population 149 changes were observed during growth on acetate in the incubations. For the 150 Methanosaetaceae dominated samples, the only changes observed during incubation 151 were elimination of subdominant populations. In the other samples, there was a shift to 152 known hydrogen consumers (Methanobacteriales, Methanomicrobiales, or 153 Methanococcales) or uncultured archae (samples M2 and M6).

154 Methanogenic communities in several samples (M2 and M5 before incubation and M2 155 and M6 after incubation) were not identified by FISH. This was due to the limitations of 156 visual *in situ* hybridisation. FISH is very convenient for the rapid analysis of a large 157 number of environmental samples but is limited when carried beyond the limits of 158 oligonucleotide probes. ARC915 is an effective general probe, and order-level probes 159 have been used in a wide range of systems, however in complicated systems such as 160 manure, they might fail to detect all methanogens. Therefore, unidentified methanogens 161 were phylogenetically characterised by PCR-TGGE. Samples not identified by FISH

162	(e.g.,	M5, Studsgard before inoculation), were found by PCR-TGGE to be far outside							
163	know	known phylogenetic groupings for methanogens. It is likely that these microbes are still							
164	methanogens, since bacterial methanogenesis is unknown. These unknown microbial								
165	groups are interesting scientifically, and deserve further investigation.								
166	Thanks are due to Lene Kirstejn Jensen, Birthe Ebert and Hector Garcia for the technical assistance with								
167	the experiments. This work was supported by Danish Government Scholarship and Danish Research								
168	Progra	mme (EFP 05).							
169		REFERENCES							
170									
171	1.	American Public Health Association 1985. Standard methods for the							
172		examination of waste and wastewater. APHA AWWA WPCF, Washington, D.C.							
173	2.	Ferry, J. 1993. Fermentation of Acetate, p. 305-334. In J. G. Ferry (ed.),							
174		Methanogenesis. Ecology, Physiology, Biochemistry and Genetics. Chapman and							
175		Hall, New York.							
176	3.	Garcia, J. L., B. K. C. Patel, and B. Ollivier. 2000. Taxonomic, Phylogenetic,							
177		and Ecological Diversity of Methanogenic Archae. Anaerobe 6.							
178	4.	Hattori, S., Y. Kamagata, S. Hanada, and H. Shoun. 2000. Thermacetogenium							
179		phaeum gen.nov., sp.nov., a strictly anaerobic, thermophilic, syntrophic acetate-							
180		oxidizing bacterium. Int.J. Syst. Evol. Microb. 50.							
181	5.	Karakashev, D., D. J. Batstone, and I. Angelidaki. 2005. Influence of							
182		environmental conditions on methanogenic compositions in anaerobic biogas							
183		reactors. Appl. Env. Microb 71:331-338.							

184	6.	Kelts, K. U., K. Briegel, K. Ghilardi, and H. K. K. 1986. The limnogeology-
185		ETH coring system. Schweitz, Z. Hydrol. 1986.
186	7.	Lovley, D. R., and J. G. Ferry. 1985. Production and consumption of H2 during
187		growth of Methanosarcina spp. on acetate. Appl. Environ. Microbiol. 49:247-249.
188	8.	Nusslein, B., KJ. Chin, W. Eckert, and R. Conrad. 2001. Evidence for
189		anaerobic syntrophic acetate oxidation during methane production in the
190		profundal sediment of subtropical Lake Kinneret (Israel). Env. Microbiol. 3:460-
191		470.
192	9.	Pavlostathis, S. G., and E. Giraldo-Gomez. 1991. Kinetics of anaerobic
193		treatment: A critical review. Crit. Rev. Environ. Control 21:411-490.
194	10.	Petersen, S., and B. Ahring. 1991. Acetate oxidation in a thermophilic anaerobic
195		sewage - sludge digestor: the importance of non-aceticlastic methanogenesis from
196		acetate. FEMS Microbiol. Ecol. 86:149-158.
197	11.	Reichert, P. 1994. AQUASIM - A tool for simulation and data analysis of aquatic
198		systems Wat. Sci. Technol. 30.
199	12.	Schnurer, A., B. Svensson, and B. Schink. 1997. Enzyme activities in energetics
200		of acetate metabolism by the mesophilic syntrophically acetate-oxidizing
201		anaerobe Clostridium ultunense. FEMS Microbiol. Lett. 154.
202	13.	Schnurer, A. G., G. Zellner, and B. Svensson. 1999. Mesophilic syntrophic
203		acetate oxidation during methane formation in biogas reactors. FEMS
204		Microb.Ecol. 29: 249-261.
205	14.	Speece, R. E. 1996. Anaerobic Biotechnology for Industrial Wastewaters. Archae
206		Press, Nashville, TN.

- 207 15. Zinder, S., and M. Koch. 1984. Non-aceticlastic methanogenesis from acetate:
 208 Acetate oxidation by a thermophilic syntrophic coculture. Arch. Microbiol.
- **138:**263-272.

Sample Id	Reactor name	Feed type	Temp. (type) ¹	Incubation period (days)	Dominant Methanogen ³ (nondominant)		Level (mean \pm SD ²) of:				
					Before incubation	After incubation	VFA (gHAc L ⁻¹)	Ammonia (gN L ⁻¹)	Maximim acetate removal rate (mM day ⁻¹)	¹⁴ CO ₂ / ¹⁴ CH ₄	Recov of ¹⁴ C
M1	Nysted	manure	38°C (M)	9	MS (NO)	MG (MS)	2.7 ± 0.11^2	5.6±0.12	$0.5^4 \pm 0.1$	2.9 ± 0.18^{2}	96 ± .
M2	Hashøj	manure	37°C (M)	13	Uncultured archae DQ409324 (MC)	Uncultured archae DQ409325 (NO)	1.9 ± 0.07	4 ± 0.1	1.8 ± 0.1	2.75 ± 0.11	97.5 ±
M3	Lemvig	manure	52.5°C (T)	8	MS (NO)	MB, MC (MS)	0.6 ± 0.01	2.6 ± 0.08	5.3 ± 0.6	2.45 ± 0.15	92.5
M4	Fangel	manure	37°C (M)	10	MB (MC)	MB (MC)	2.33 ± 0.12	4.5 ± 0.10	1.5 ± 0.1	2.2 ± 0.14	97.4
M5	Studsgard	manure	52°C (T)	10	UnIDd filaments (NO)	MG (MS)	0.22±0.003	2.20 ± 0.09	2.7	2.11 ± 0.16	99 ±
M6	Vester Hjermitslev	manure	37°C (M)	16	MS (NO)	Uncultured archae DQ409326 (MS)	1.81 ± 0.06	4.4±0.11	0.7 ± 0.03	1.9 ± 0.18	92.2 ±
M7	Vegger	manure	55°C (T)	8	MS (NO)	MS (NO)	0.77 ± 0.04	2.34±0.07	3.5 ± 0.4	1.61 ± 0.11	94 ±
LS1	Orholm	lake sediment	4°C (P)	49	MB (MX, MG, MC)	MC (NO)	0.01	0.03	$1.4^{4} \pm 0.2$	1.4 ± 0.09	96.2 ±
M8	Sinding	manure	55°C (T)	8	MS (NO)	MS (NO)	0.41 ± 0.01	2.5 ± 0.06	3.4	1.1 ± 0.06	101 :
S 1	Hillerød	WW sludge	55°C(T)	5	MX (MS)	MX (NO)	0.13 ± 0.001	1.44 ± 0.05	10 ± 0.5	0.11 ± 0.007	$97 \pm$
S 2	Lundtofte	WW sludge	37°C (M)	10	MX (NO)	MX (NO)	0.02	1.00 ± 0.04	9.9 ± 0.6	0.08 ± 0.003	98.1
S 3	Fakse	WW sludge	37°C (M)	6	MX (NO)	MX (NO)	0.04	1.50 ± 0.03	7.1	0.065 ± 0.004	91 ±
S4	Sydkyst	WW sludge	37°C (M)	8	MX (NO)	MX (NO)	0.13	0.50 ± 0.001	$4.8^4 \pm 0.4$	0.04 ± 0.002	92 ±
S5	Helsingør	WW sludge	37°C (M)	6	MX (MB, MG)	MX (NO)	0.06	0.84 ± 0.03	6.1 ± 0.7	0.025 ± 0.001	95.4 ±

210 Table 1: Results from acetate oxidation survey

211 ^{1.} P: Psychrophilic ($<20^{\circ}$ C), M: mesophilic (35° C- 40° C), T: thermophilic ($>50^{\circ}$ C); ². SD based on triplicate analysis; ³. MS,

212 Methanosarcinaceae; MX, Methanosaetaceae; MB, Methanobacteriales; MG, Methanomicrobiales; MC, Methanococcales; NO, not

213 observed; UnIDd, unidentified (both by FISH and TGGE). The term "dominant methanogens" was used in the sense of more than 90 % of

the total number of methanogenic cells (Archae responding to ARC915). The term "nondominant methanogens" was used in the sense of 1

to 10 % of the total number of methanogenic cells. Cells in 20 fields were countered. ⁴. Long lag phase observed before acetate removal



216

217 Figure 1. Distribution of the dominant methanogens observed in the samples as a function of the acetate oxidation degree versus acetate utilisation rates. Error bars indicate standard 218 219 deviations. indicates Methanosaetaceae while other abbreviations indicate MX 220 Methanosarcinaca (MS), Methanobacteriales (MB), *Methanomicrobiales* (MG), Methanococcales (MC), and Unidentified (UnIDd). 221 222 223 224

225