



HAL
open science

Conformational features of C-glycosyl compounds : crystal structure and molecular modelling of methyl C-gentiobioside

A. Neuman, F. Longchambon, O. Abbes, H. Gillier-Pandraud, Sarah Perez, D.
Rouzaud, P. Sinay

► To cite this version:

A. Neuman, F. Longchambon, O. Abbes, H. Gillier-Pandraud, Sarah Perez, et al.. Conformational features of C-glycosyl compounds : crystal structure and molecular modelling of methyl C-gentiobioside. Carbohydrate Research, 1990, 195 (2), pp.187-197. hal-02709733

HAL Id: hal-02709733

<https://hal.inrae.fr/hal-02709733>

Submitted on 1 Jun 2020

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

CONFORMATIONAL FEATURES OF C-GLYCOSYL COMPOUNDS: CRYSTAL STRUCTURE AND MOLECULAR MODELLING OF "METHYL C-GENTIOBIOSIDE"

ALAIN NEUMAN, FRANÇOIS LONGCHAMBON, OUKACHA ABBES, HÉLÈNE GILLIER-PANDRAUD*,
*Laboratoire de Chimie Structurale Biomoléculaire, U.F.R. Santé, Médecine, Biologie Humaine,
Université de Paris-Nord, 74 rue Marcel Cachin F-93012 Bobigny (France)*

SERGE PÉREZ,
*Laboratoire de Physico-Chimie des Macromolécules, I.N.R.A., rue de la Géraudière, F-44072 Nantes
(France)*

DOMINIQUE ROUZAUD AND PIERRE SINAY
Laboratoire de Chimie, Ecole Normale Supérieure, 24 rue Lhomond, F-75005, Paris (France)
(Received March 3rd, 1989; accepted for publication, June 24th, 1989)

ABSTRACT

The crystal of "methyl C-gentiobioside" (methyl 8,12-anhydro-6,7-dideoxy-D-glycero-D-gulo- α -D-gluco-tridecapyranoside) ($C_{14}H_{26}O_{10}$) is triclinic, space group $P1$, with $a = 1.0181$ (6) nm, $b = 0.8093$ (5) nm, $c = 0.5066$ (4) nm, $\alpha = 96.03$ (5) $^\circ$, $\beta = 99.94$ (5) $^\circ$, $\gamma = 90.85$ (5) $^\circ$. The two D-glucose residues have the 4C_1 conformation. The orientation of the β -(1 \rightarrow 6) linkage is characterized by torsion angles $\Phi = 55.9^\circ$, $\Psi = 175.1^\circ$, and $\Omega = -63.9^\circ$. The orientation of the primary hydroxyl group at the non-reducing residue is *gauche-trans* ($\Omega' = -53.6^\circ$). There is no intramolecular hydrogen bond. Molecules are held together by a network of hydrogen bonds involving all of the hydroxyl groups. This crystal structure is the first experimental characterization of a "C-disaccharide". Unlike methyl gentiobioside, which has a high level of conformational flexibility, the "C-disaccharide" has a restricted flexibility. Each of the low-energy conformers *in vacuo* has a value of Φ centered about 60° , in agreement with the solid state conformation, and the exo-anomeric effect is no longer predominant.

INTRODUCTION

The biological importance of C-nucleosides has stimulated a search for routes of synthesis of C-glycosyl compounds¹. Disaccharide derivatives are now available in which the inter-residue oxygen atom is replaced by a methylene group. Such C-linked oligosaccharides may function as active-site-directed inhibitors of en-

* Author for correspondence.

zymes² and are used in the study of sugar-protein interactions³. Also, they provide a unique opportunity to evaluate the contribution of the exo-anomeric effect on the conformational preference of an oligosaccharide.

The first "C-disaccharide" synthesised was methyl 8,12-anhydro-6,7-dideoxy-D-glycero-D-gulo- α -D-gluco-tridecapyranside⁴ ("methyl C-gentiobioside") in which the bridging oxygen atom of the β -(1 \rightarrow 6) linkage was replaced by a methylene group. The present work describes the conformational features displayed by such a molecule in the solid state and a comparison with gentiobiose⁵.

EXPERIMENTAL

X-Ray investigation. — Crystals suitable for X-ray investigations were grown by cooling an ethanolic solution. Cell parameters were refined from diffractometer measurements of 25 well-centered, high-angle reflections. These and other pertinent crystal data are given in Table I. Intensity data were collected in the θ - 2θ scan mode, by using $\text{CuK}\alpha$ radiation, on a Philips PW 1100 diffractometer.

The intensities of reference reflections measured every 2 h did not decrease significantly during collection of the data. The data were corrected for Lorentz and polarization effects. No absorption correction was employed. Of the 3122 independent reflections measured, 1496 with $I/\sigma(I) > 2$ were used for structural analysis.

The structure was solved by direct methods using the DEVIN program⁶. The molecule of "methyl C-gentiobioside" displays a pseudo center of symmetry which influenced the statistics of the normalized structure factors, and perturbed the first steps of the structure solution.

Following several cycles of full matrix, anisotropic, least-squares refinement for the non-hydrogen atoms, with the SHELX⁷ refinement section, the hydrogen atoms were located approximately on successive difference Fourier syntheses. They were submitted to complete refinement with isotropic thermal parameters.

The quantity minimized was $w(F_o - F_c)^2$, each reflection being assigned a weight $w = 1/\sigma^2(F)$ derived from $\sigma(I)$. A final electron-density map showed no

TABLE I

CRYSTAL DATA FOR "METHYL C-GENTIOBIOSIDE"

Formula	$\text{C}_{14}\text{H}_{26}\text{O}_{10}$
Mol. wt.	354
a 1.0181(6) nm;	α 96.03(5)°
b 0.8093(5) nm;	β 99.95(5)°
c 0.5066(4) nm;	γ 90.85(5)°
Space group	$P1$
$Z = 1$	
$V(\text{nm}^3)$	0.408
D_c	1.439 g.cm ⁻³

TABLE II

ATOMIC POSITIONS

Atom

C-1
C-2
C-3
C-4
C-5
C-6
C-7
C-16
O-1
O-2
O-3
O-4
O-5
C-1'
C-2'
C-3'
C-4'
C-5'
C-6'
O-2'
O-3'
O-4'
O-5'
O-6'

$$U_{eq} = 1/3 \sum_i \sum_j U_{ij}$$

significant residues. The final coordinates, a table of anisotropic, thermal

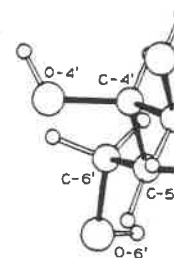


Fig. 1. Molecular structure.

*These data have B.B.A. Data Dep No. BBA/DD/421

TABLE II

ATOMIC POSITIONAL PARAMETERS (AND THEIR E.S.D. VALUES IN PARENTHESES)

Atom	x/a	y/b	z/c	U_{eq}^a
C-1	-0.4071(2)	-1.3220(3)	-0.2714(6)	0.031(3)
C-2	-0.5383(2)	-1.2573(3)	-0.2146(5)	0.030(3)
C-3	-0.5720(2)	-1.0985(3)	-0.3459(6)	0.029(2)
C-4	-0.4560(2)	-0.9723(3)	-0.2489(6)	0.031(3)
C-5	-0.3263(2)	-1.0466(3)	-0.3122(5)	0.028(2)
C-6	-0.2038(2)	-0.9327(3)	-0.2142(6)	0.029(2)
C-7	-0.3259(4)	-1.4908(4)	-0.6152(8)	0.098(5)
C-16	-0.1788(2)	-0.8751(3)	0.0883(5)	0.027(2)
O-1	-0.4204(2)	-1.3735(2)	-0.5452(4)	0.036(2)
O-2	-0.6409(2)	-1.3832(2)	-0.2923(4)	0.040(2)
O-3	-0.6913(2)	-1.0331(3)	-0.2671(4)	0.044(2)
O-4	-0.4735(2)	-0.8207(3)	-0.3607(6)	0.053(3)
O-5	-0.3027(1)	-1.1980(2)	-0.1815(4)	0.030(2)
C-1'	-0.0489(2)	-0.7730(3)	0.1859(5)	0.026(2)
C-2'	0.0765(2)	-0.8723(3)	0.1685(5)	0.025(2)
C-3'	0.2006(2)	-0.7599(3)	0.2367(5)	0.026(2)
C-4'	0.1874(2)	-0.6123(3)	0.0739(5)	0.025(2)
C-5'	0.0595(2)	-0.5235(3)	0.1130(5)	0.029(2)
C-6'	0.0346(3)	-0.3749(3)	-0.0422(7)	0.052(3)
O-2'	0.0873(2)	-0.9930(2)	0.3581(4)	0.038(2)
O-3'	0.3180(2)	-0.8497(2)	0.2131(4)	0.041(2)
O-4'	0.3001(2)	-0.5029(2)	0.1654(4)	0.039(2)
O-5'	-0.0521(1)	-0.6358(2)	0.0288(4)	0.031(2)
O-6'	-0.0490(2)	-0.2626(2)	0.0837(5)	0.049(2)

$$^a U_{eq} = 1/3 \sum_i \sum_j U_{ij} a_i^* \cdot a_j^* a_i \cdot a_j$$

significant residual density. The final R and R_w values were 0.036 and 0.032, respectively. The final atomic coordinates are listed in Table II. The hydrogen-atom coordinates, a tabulation of observed and calculated structure factors, and the final anisotropic, thermal parameters are available as supplementary materials*.

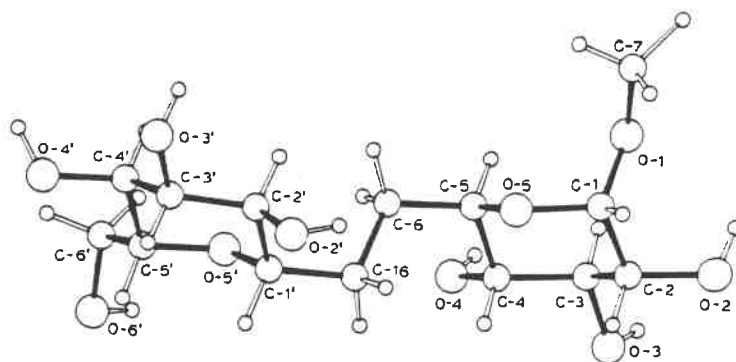


Fig. 1. Molecular conformation (ORTEP¹¹) of "methyl C-gentiobioside" as found in the crystal structure.

*These data have been deposited with, and may be obtained from Elsevier Science Publishers B.V., B.B.A. Data Deposition, P.O. Box 1527, Amsterdam, The Netherlands. Reference should be made to No. BBA/DD/421/Carbohydr. Res., 195 (1990) 187-197.

Conformational analysis. — The numbering of the atoms, shown in Fig. 1, proceeds from the non-reducing end (primed atoms, ring II) to the reducing end (unprimed atoms, ring I). The conformational analysis to determine the likely orientation about the inter-residue junction was conducted using the program PFOS⁸. The potential energy was calculated by including the partitioned contributions arising from the van der Waals, torsional, and hydrogen-bond contributions. The van der Waals interactions were evaluated by using 6–12 potential functions with the parameters proposed by Scott and Scheraga⁹. For “methyl *C*-gentiobioside”, a three-fold, sinusoidal potential was used for rotation about torsion angles Φ and Ψ , with a barrier of 3.5 kcal.mol⁻¹ for the eclipsed orientation. As for gentiobiose, a barrier of 1.0 kcal.mol⁻¹ was assigned to the eclipsed orientation around Ψ . For rotation about the glycosidic angle Φ , the intramolecular mechanism responsible for the exo-anomeric effect was taken into account by using the potential function proposed by Tvaroska¹⁰; a sinusoidal, three-fold, eclipsed, rotational barrier of 1.0 kcal.mol⁻¹ was also included. The hydrogen-bond energy was computed by an empirical expression proposed by Pérez and Vergelati¹¹. In these calculations, the hydroxylic hydrogen atoms were not considered, and the geometry of each residue was considered to be rigid. No electrostatic interaction was taken into account.

RESULTS AND DISCUSSION

A representation (ORTEP¹²) of the molecule of “methyl *C*-gentiobioside” is shown in Fig. 1. The bond distances and angles are given in Tables III and IV, respectively.

Molecular features. — The bond distances and angles observed in this crystal structure compare well with those for gentiobiose⁵, and conform to the tabulated values of carbohydrates¹³. The expected ⁴C₁ chair conformation was found for the

TABLE III

BOND DISTANCES^a (NM) AND THEIR E.S.D. VALUES

C-1-C-2	0.1505(4)	C-1'-C-2'	0.1530(3)
C-1-O-1	0.1389(3)	C-16-C-1'	0.1527(3)
C-1-O-5	0.1432(3)	C-1'-O-5'	0.1430(3)
C-2-C-3	0.1525(4)	C-2'-C-3'	0.1514(3)
C-2-O-2	0.1424(3)	C-2'-O-2'	0.1432(3)
C-3-C-4	0.1529(4)	C-3'-C-4'	0.1516(3)
C-3-O-3	0.1434(3)	C-3'-O-3'	0.1424(3)
C-4-C-5	0.1528(3)	C-4'-C-5'	0.1529(3)
C-4-O-4	0.1406(3)	C-4'-O-4'	0.1420(3)
C-5-C-6	0.1520(4)	C-5'-C-6'	0.1506(4)
C-5-O-5	0.1456(3)	C-5'-O-5'	0.1421(3)
C-6-C-16	0.1531(4)		
C-7-O-1	0.1424(4)		

^aMean bond lengths: C-H, 0.110; O-H, 0.091 nm.

TABLE IV

BOND ANGLES

C-2-C-1-O-1
C-2-C-1-O-5
O-5-C-1-O-1
C-1-C-2-C-3
C-1-C-2-O-2
C-3-C-2-O-2
C-2-C-3-C-4
C-2-C-3-O-3
C-4-C-3-O-3
C-3-C-4-C-5
C-3-C-4-O-4
C-5-C-4-O-4
C-4-C-5-C-6
C-4-C-5-O-5
C-6-C-5-O-5
C-5-C-6-C-16
C-1-O-5-C-5
C-6-C-16-C-1'
C-1-O-1-C-7

pyranose re
given in Ta
“methyl *C*-
extreme var
similar conc
can be draw
“methyl *C*-
The *C*
reflects the
The o

TABLE V

TORSION ANGLE

Ring I

O-5-C-1-C-2-O-1
C-1-C-2-C-3-C-4
C-2-C-3-C-4-C-5
C-3-C-4-C-5-C-6
C-4-C-5-C-6-C-16
C-5-O-5-C-1-O-5

Mean absolute

^aE.s.d., 0.2°.

TABLE IV

BOND ANGLES (°) AND THEIR E.S.D. VALUES

C-2-C-1-O-1	108.6(2)	C-2'-C-1'-C-16	113.8(2)
C-2-C-1-O-5	110.5(2)	C-2'-C-1'-O-5'	110.5(2)
O-5-C-1-O-1	112.3(2)	O-5'-C-1'-C-16	107.5(2)
C-1-C-2-C-3	111.2(2)	C-1'-C-2'-C-3'	111.0(2)
C-1-C-2-O-2	110.5(2)	C-1'-C-2'-O-2'	109.3(2)
C-3-C-2-O-2	112.6(2)	C-3'-C-2'-O-2'	108.1(2)
C-2-C-3-C-4	108.3(2)	C-2'-C-3'-C-4'	111.3(2)
C-2-C-3-O-3	109.3(2)	C-2'-C-3'-O-3'	111.7(2)
C-4-C-3-O-3	109.7(2)	C-4'-C-3'-O-3'	112.7(2)
C-3-C-4-C-5	109.0(2)	C-3'-C-4'-C-5'	108.9(2)
C-3-C-4-O-4	113.9(2)	C-3'-C-4'-O-4'	108.8(2)
C-5-C-4-O-4	108.8(2)	C-5'-C-4'-O-4'	110.2(2)
C-4-C-5-C-6	114.0(2)	C-4'-C-5'-C-6'	113.9(2)
C-4-C-5-O-5	109.2(2)	C-4'-C-5'-O-5'	109.6(2)
C-6-C-5-O-5	106.9(2)	C-6'-C-5'-O-5'	107.6(2)
C-5-C-6-C-16	114.7(2)	C-5'-C-6'-O-6'	109.9(2)
C-1-O-5-C-5	112.8(2)	C-1'-O-5'-C-5'	113.2(2)
C-6-C-16-C-1'	113.6(2)		
C-1-O-1-C-7	114.8(2)		

pyranose residues, and the torsion angles about the skeletal bonds of the rings are given in Table V. The mean values of the intracyclic torsion angles in ring I of "methyl C-gentiobioside" and gentiobiose are 57.8° and 57.6°, respectively. The extreme variations are 55.4°–60.5° for the former and 49.7°–64.7° for the latter. A similar conclusion about the rigidity of the residues in "methyl C-gentiobioside" can be drawn; for ring II, the mean absolute values are 56.5° (51.2°–63.2°) in "methyl C-gentiobioside", and 55.4° (49.2°–66.4°) in gentiobiose.

The O-1 Me group is in a *gauche* orientation with respect to O-5, which reflects the influence of the exo-anomeric effect¹⁴.

The orientation of the HO-6 is usually defined as *gauche-gauche*, *gauche-*

TABLE V

TORSION ANGLES (°)^a

Ring I		Ring II	
O-5-C-1-C-2-C-3	56.9	O-5'-C-1'-C-2'-C-3'	52.3
C-1-C-2-C-3-C-4	-55.4	C-1'-C-2'-C-3'-C-4'	-51.2
C-2-C-3-C-4-C-5	56.2	C-2'-C-3'-C-4'-C-5'	53.9
C-3-C-4-C-5-O-5	-58.4	C-3'-C-4'-C-5'-O-5'	-58.8
C-4-C-5-O-5-C-1	60.5	C-4'-C-5'-O-5'-C-1'	63.2
C-5-O-5-C-1-C-2	-59.7	C-5'-O-5'-C-1'-C-2'	-59.6
Mean absolute values:	57.9		56.5

^aE.s.d., 0.2°.

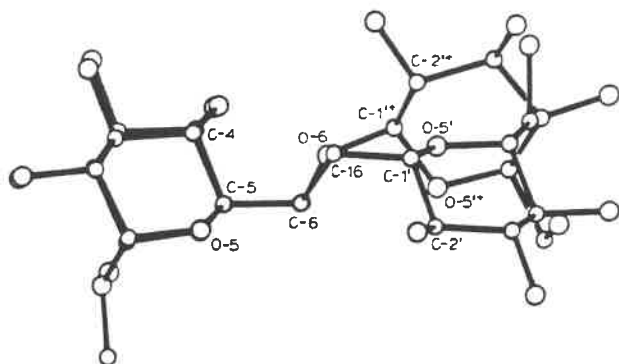


Fig. 2. Comparison between the solid state conformation of "methyl C-gentiobioside" and gentiobiose.

trans, or *trans-gauche*¹⁵. In this terminology, the torsion angle (O-5-C-5-C-6-O-6) is stated first, and the torsion angle (C-4-C-5-C-6-O-6) second. For the *gluco* configuration, the orientations observed in the solid state are usually almost equally distributed between *gauche-gauche* and *gauche-trans*¹⁴. In the present molecule, the orientation of HO-6' *i.e.*, about C-5'-C-6', is in a *gauche-gauche* conformation (-53.6°). A *gauche-trans* conformation (66.1°) was found for gentiobiose.

In the "methyl C-gentiobioside" molecule, a pseudo-inversion center is located between C-6 and C-16 (mean coordinates = 0.1817, -0.9249 , -0.0597).

The relative orientation of two residues connected by a (1 \rightarrow 6) linkage is given by the torsional angles: Φ O-5'-C-1'-C-16-C-6, Ψ C-1'-C-16-C-6-C-5, and Ω C-16-C-6-C-5-O-5. For both "methyl C-gentiobioside" and gentiobiose, the conformations about Ω are similar [$\Omega = -64.0$ (2) $^\circ$ and -61.5 (5) $^\circ$, respectively], and correspond to a *gauche-gauche* type orientation. Similarly, the conformation about Ψ is almost perfectly *trans* [175.1 (2) $^\circ$] in "methyl C-gentiobioside", which is 29° away from the conformation (-156.3°) for gentiobiose. A drastic difference, see Fig. 2, of $\sim 120^\circ$ is found for Φ for the linkage conformation of the two molecules (55.9° in the C-glycoside, *versus* -58.3° in the O-glycoside).

Hydrogen bonding and packing. — The crystal structure of "methyl C-gentiobioside" involves an extensive network of hydrogen bonds, described in Table VI, a projection of which is shown in Fig. 3. All the hydroxyl oxygen atoms participate in hydrogen bonds (76.5% as both donor and acceptor, and 12.4% as donor only) and they follow the pattern proposed by Avenel¹⁶. Moreover, the molecules adopt an optimum mode of packing. All the non-polar groups (CH₃, CCH₂C) interact and create cylindrical hydrophobic channels which are embedded in a hydrophilic matrix where all the hydrogen bonds are located. As seen in Fig. 3, hydrophobic and hydrophilic zones are of equal importance. The crystal structure is dense, each "methyl C-gentiobioside" molecule being surrounded by 10 neighbors.

Molecular modelling. — In order to rationalize the observed solid-state conformation, and to compare it with that of gentiobiose, a set of iso-energy maps was computed and plotted. These maps correspond to intramolecular-energy variations

O-2 ...

O-2 ...

\

H-O-2 ...

O-3 ...

O-3 ...

\

H-O-3 ...

O-4-H-O-4

O-5 ...

O-5 ...

O-2' ...

O-2' ...

\

H-O-2' ...

O-3' ...

O-3' ...

\

H-O-3 ...

O-4' ...

O-4' ...

\

H-O-4' ...

O-6' ...

O-6' ...

\

H-O-6' ...

as a funct
non-eclips
trans-gauc
gentiobios
The data :
energy sur
able. Ther
over 180° :
of the low-
indicated i
occurring,
occurs for
similar to t

TABLE VI

HYDROGEN BONDING (DISTANCE, NM)

O-2 . . .	H-O-4'	0.188	
O-2 . . .	O-4'	0.276	-1, -1, 0
\			
H-O-2 . . .	O-4'	0.187	-1, -1, -1
O-3 . . .	H-O-3'	0.184	
O-3 . . .	O-3'	0.270	-1, 0, 0
\			
H-O-3 . . .	O-2'	0.173	-1, 0, -1
O-4-H-O-4 . . .	O-3'	0.198	-1, 0, -1
O-5 . . .	H-O-6'	0.190	
O-5 . . .	O-6'	0.278	0, -1, 0
O-2' . . .	H-O-3	0.173	
O-2' . . .	O-3	0.273	-1, 0, 1
\			
H-O-2' . . .	O-6'	0.188	0, -1, 0
O-3' . . .	H-O-4	0.198	
O-3' . . .	O-4	0.274	1, 0, 0
\			
H-O-3 . . .	O-3	0.184	1, 0, 0
O-4' . . .	H-O-2	0.187	
O-4' . . .	O-2	0.276	1, 1, 1
\			
H-O-4' . . .	O-2	0.187	1, 1, 0
O-6' . . .	H-O-2'	0.188	
O-6' . . .	O-2'	0.269	0, 1, 0
\			
H-O-6' . . .	O-5	0.190	0, 1, 0

as a function of Φ and Ψ , with Ω being assigned successively one of the three non-eclipsing conformations corresponding to *gauche-gauche*, *gauche-trans*, and *trans-gauche*. Calculations were performed on "methyl C-gentiobioside" and gentiobiose, with the ring geometry found in the corresponding crystal structures. The data are summarized in Fig. 4. For "methyl C-gentiobioside", the potential energy surfaces computed for each of the stable conformations of Ω are comparable. There is only one low-energy domain centered about $\Phi = 60^\circ$, which spans over 180° along Ψ , with a well-defined three-fold symmetry. The precise locations of the low-energy conformers, along with the main contributors to the energy, are indicated in Table VII; the corresponding inter-residue hydrogen bonds, when occurring, are also listed. The calculated conformation having the lowest energy occurs for $\Phi = 55^\circ$, $\Psi = 180^\circ$, $\Omega = -60^\circ$, and $E = -4.57$ kcal.mol⁻¹, which is similar to that in the crystal [$\Phi = 55.8^\circ$, $\Psi = 175.1^\circ$ (or -184.9°), $\Omega = -63.9^\circ$, E

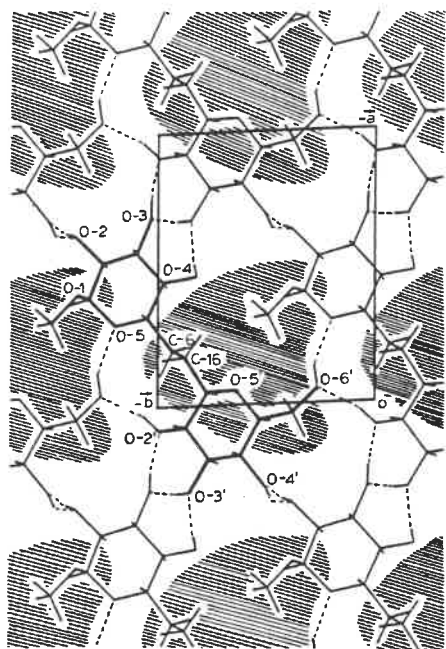


Fig. 3. Projected view of the packing of "methyl C-gentiobioside" in the crystallographic (a, b) plane. Hatched zones indicate the hydrophobic regions.

$= -4.46 \text{ kcal.mol}^{-1}$]. Such good agreement indicates the suitability of the set of potential energy functions and associated parameters for the conformational analysis of C-glycosyl compounds.

Iso-energy surfaces were calculated for the molecule of gentiobiose for each of the stable conformations of Ω (-60° , 60° , 180°) and they are represented in Fig. 4d-f. The low-energy domains encompass $\sim 50\%$ of the total (Φ , Ψ) surface. Many local energy minima are found (Table VIII). The conformation found for gentiobiose in the solid state ($\Phi = -58.3^\circ$, $\Psi = -156.3^\circ$, $\Omega = -61.5^\circ$, $E = -1.77 \text{ kcal.mol}^{-1}$) belongs to the low-energy domain where the deepest minimum is found ($\Phi = -75^\circ$, $\Psi = -175^\circ$, $\Omega = -60^\circ$, $E = -2.22 \text{ kcal.mol}^{-1}$). Almost irrespective of the Ω values, two distinct regions of the (Φ , Ψ) map define the low-energy domains. The low-energy area centered around $\Phi = -80^\circ$, and spanning from $\Psi = 60^\circ$ to $\Psi = -60^\circ$, contains most of the low-energy conformers. The second more restricted domain corresponds to conformations having Φ close to 60° , the Ψ values spanning from $\Psi = 90^\circ$ to $\Psi = -90^\circ$. All the local minima belonging to this domain have energy values significantly higher than those of the other domain.

The conformation of "methyl C-gentiobioside" in solution, determined by high-resolution n.m.r. spectroscopy¹⁷, is described by $\Phi = -85^\circ$, $\Psi = 180^\circ$, and $\Omega = 85^\circ$ (these values have been recalculated from those reported, because the authors made use of a different convention¹⁸). Such a conformation does not corres-

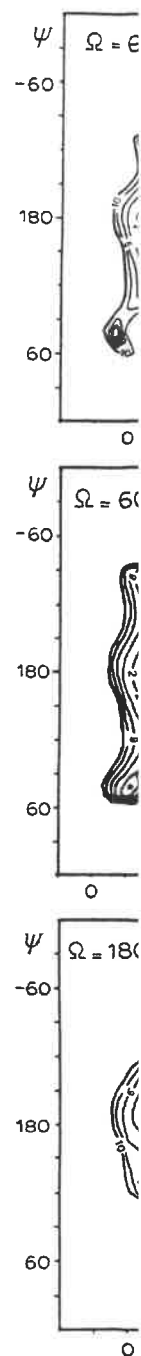


Fig. 4. Iso-energy surfaces for gentiobiose at different Ω values. Relative iso-energy surfaces of the minima.

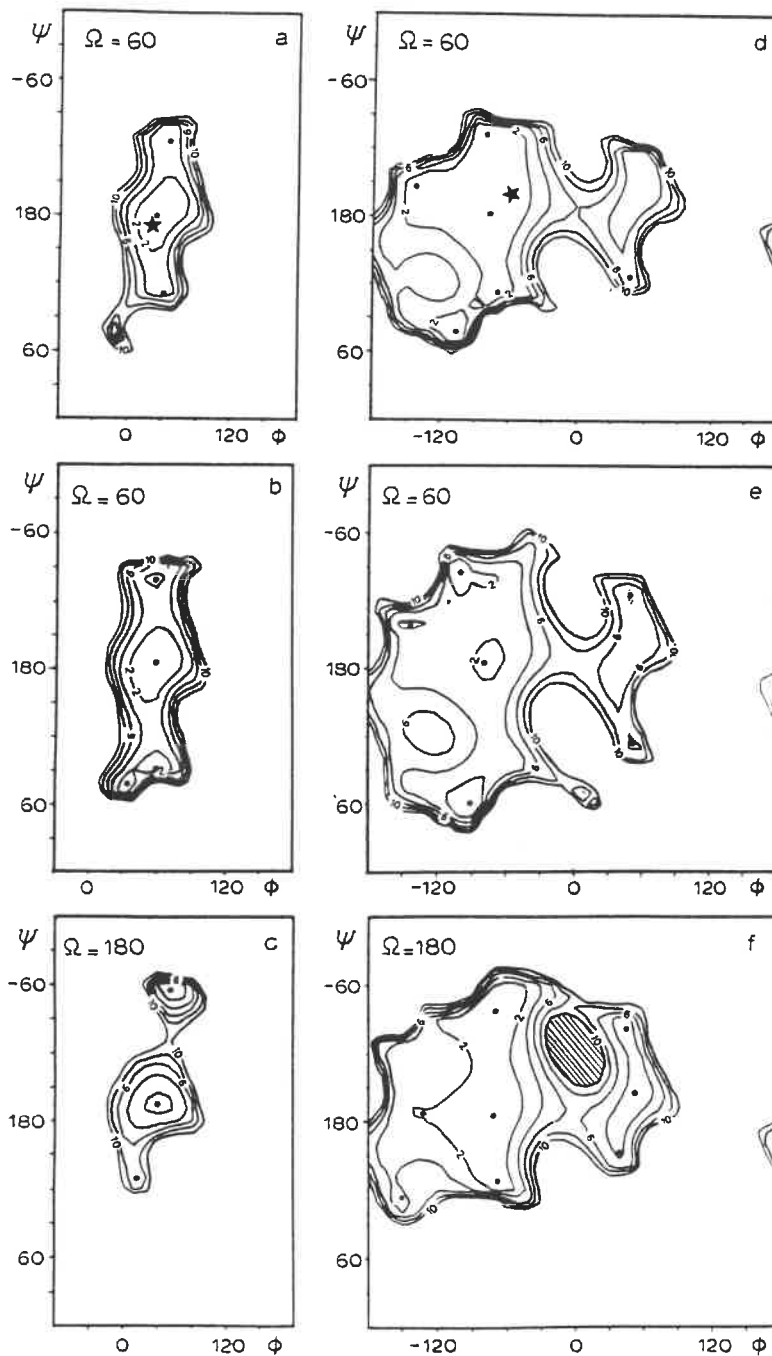


Fig. 4. Iso-energy maps (Φ , Ψ) about the (1 \rightarrow 6) linkage for "methyl C-gentiobioside" (a, b, c) and for gentiobiose (d, e, f). In each case, three orientations for the torsion angle Ω were taken into account. Relative iso-energy contours are drawn at intervals of 2 kcal/mol; the symbol * indicates the occurrence of the minima observed in the crystal structures.

graphic (a, b) plane.

ility of the set of conformational

tiobiose for each presented in Fig. Ψ) surface. Many found for gentio- 5° , $E = -1.77$ minimum is found most irrespective; the low-energy spanning from Ψ The second more 60° , the Ψ values ng to this domain main.

, determined by $\Psi = 180^\circ$, and Ω ed, because the does not corres-

TABLE VII

LOW-ENERGY CONFORMERS OF "METHYL C-GENTIOBIOSIDE"

$\Omega(^{\circ})$	$\Phi(^{\circ}), \Psi(^{\circ})$	$E_{nb} + E_{tor}$ (kcal.mol)	Hydrogen bond (nm)
-60	55, 180	-4.57	
	60, 110	-1.64	O-2' ... O-5 0.288
	65, -115	-2.7	O-5' ... O-4 0.274
	25, 65	-1.52	
60	60, -175	-4.26	
	60, 90	-3.46	
	60, -100	-2.89	
	35, 75	-3.72	
180	60, -165	-3.72	
	40, 130	1.63	O-2' ... O-4 0.279
	70, -65	-5.15	O-5' ... O-4 0.262

TABLE VIII

LOW-ENERGY CONFORMERS OF GENTIOBIOSE

$\Omega(^{\circ})$	$\Phi(^{\circ}), \Psi(^{\circ})$	$E_{nb} + E_{tor}$ (kcal.mol)	Hydrogen bond (nm)
-60	-80, -105	-2.48	
	-75, -175	-2.22	
	-105, 75	-2.57	
	-70, 110	-1.16	O-6' ... O-1 0.287
	-140, -150	-1.08	O-2' ... O-4 0.279
	50, -125	-0.23	O-5' ... O-4 0.277
			O-6' ... O-4 0.276
	50, 125	-0.42	O-2' ... O-5 0.278
60	-100, -95	-1.61	O-2' ... O-5 0.281
	-75, -175	-1.95	
	-90, 60	-3.15	
	-140, -140	-0.71	
	50, -115	0.08	
	55, 110	1.79	
180	-70, -80	-2.65	O-2' ... O-4 0.275
	-75, -175	-2.19	
	-70, 130	-1.54	O-5' ... O-4 0.296
	-135, 190	-1.70	
	45, -100	-2.12	
	55, -160	-1.25	
	-150, 110	0.81	O-5' ... O-4 0.276
	40, 150	1.01	O-2' ... O-4 0.285

pond to any magnitude of

The p
saccharide"
of the avai
analogue. T
120° differ
tions indica
saccharide"
interaction c
a bridging c
C-16-C-6 b
biose and its

REFERENCE

- 1 S. HANNESS
- 2 M. L. SHU
- 3 L. A. REED
- 4 D. ROUZAU
- 5 D. C. ROH
650-654.
- 6 C. RICHE, *J*
- 7 G. SHELDRI
1976.
- 8 S. PÉREZ, *E*
- 9 R. A. SCOT
- 10 I. TVAROSK
- 11 S. PÉREZ AN
- 12 C. K. JOHN
- 13 F. LONGCH
LEHMAN, AN
- 14 F. LONGCHA
- 15 R. H. MAR
- 16 D. AVENEL
- 17 P. G. GOEK
- 18 D. Y. GAGI
430.

pond to any of the stable states calculated for the molecule in vacuum. The reported magnitude of $\Phi = -85^\circ$ lies far from the only stable values centered about $\Phi = 60^\circ$.

The present work provides the first experimental characterization of a "C-disaccharide" molecule. The molecular modelling study indicates a drastic decrease of the available conformational space, on going from gentiobiose to the C-analogue. The striking difference occurs about the Φ torsion angle, since there is a 120° difference between the magnitude of this angle in both molecules. The calculations indicate that the lack of influence of the exo-anomeric effect in the "C-disaccharide" is not responsible for such a shift. Indeed, it results simply from steric interaction of the substituents at the inter-residue linkage. Also, the substitution of a bridging oxygen atom by a CH_2 group favours a *trans* conformation about the C-16-C-6 bond. The conformations about the exocyclic C-6-C-5 bond of gentiobiose and its C-analogue are similar.

REFERENCES

- 1 S. HANNESSIAN AND A. G. PERNET, *Adv. Carbohydr. Chem. Biochem.*, 33 (1976) 111-188.
- 2 M. L. SHUMAN, S. D. SHIYAN, AND A. Y. KHORLIN, *Carbohydr. Res.*, 33 (1974) 229-235; D. CERRETTI, *ibid.*, 94 (1981) c10-c13.
- 3 L. A. REED, I. ITO, S. MASAMUNE, AND K. B. SHARPLESS, *J. Am. Chem. Soc.*, 104 (1982) 6468-6470.
- 4 D. ROUZAUD AND P. SINAY, *J. Chem. Soc., Chem. Commun.*, (1983) 1353-1354.
- 5 D. C. ROHRER, A. SARKO, T. L. BLUHM, AND Y. N. LEE, *Acta Crystallogr., Sect. B*, 36 (1980) 650-654.
- 6 C. RICHE, *Acta Crystallogr., Sect. A*, 29 (1973) 133-137.
- 7 G. SHELDRICK, *SHELX, A Program for Crystal Structure Determination*, University of Cambridge, 1976.
- 8 S. PÉREZ, D.Sc. Thesis, University of Grenoble, 1978.
- 9 R. A. SCOTT AND H. A. SCHERAGA, *J. Chem. Phys.*, 42 (1965) 2209-2215.
- 10 I. TVAROSKA, *Carbohydr. Res.*, 125 (1984) 1951-1960.
- 11 S. PÉREZ AND C. VERGELATI, *Polym. Bull.*, 17 (1987) 141-148.
- 12 C. K. JOHNSON, *ORTEP*, Report ORNL-3794, Oak Ridge National Laboratory, Tennessee, 1965.
- 13 F. LONGCHAMBON, H. GILLIER-PANDRAUD, R. WIEST, B. REES, A. MITSCHLER, R. FELD, M. LEHMAN, AND P. BECKER, *Acta Crystallogr., Sect. B*, 41 (1985) 47-56.
- 14 F. LONGCHAMBON, D.Sc. Thesis, Univ. Paris-Nord, 1983.
- 15 R. H. MARCHESSAULT AND S. PÉREZ, *Biopolymers*, 18 (1979) 2369-2374.
- 16 D. AVENEL, D.Sc. Thesis, Univ. Paris-Nord, 1986.
- 17 P. G. GOEKJIAN, T. C. WU, H. Y. KANG, AND Y. KISHI, *J. Org. Chem.*, 52 (1987) 4823-4825.
- 18 D. Y. GAGNAIRE, R. NARDIN, F. R. TARAVEL, AND M. R. VIGNON, *Nouv. J. Chim.*, 1 (1987) 423-430.

 bond (nm)

 O-5 0.288
 O-4 0.274

 O-4 0.279
 O-4 0.262

 bond (nm)

 O-1 0.287
 O-4 0.279
 O-4 0.277
 O-4 0.276
 O-5 0.278

 O-5 0.281

 O-4 0.275

 O-4 0.296

 O-4 0.276
 O-4 0.285
