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# USING BIOCHAR FOR DECREASING THE MOBILITY OF METALS IN CONTAMINATED SOILS

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## INTRODUCTION

**Biochar\*** has emerged as a promising soil improver and carbon sink but its effects on **trace elements in soils** are still poorly known.

Literature<sup>[1]</sup> suggests **different interaction mechanisms** depending on element, biochar nature and environment.

\* (solid product from biomass pyrolysis used as soil amendment)

Cationic metals  $M^{2+}$  may be retained on the surface of biochar by various processes<sup>[1]</sup>:

- **Electrostatic interactions** between  $M^{2+}$  and biochar surface
- **Cationic exchange** between  $M^{2+}$  and major cations
- **$\pi$ -coordination** with electrons from C=C bonds of biochar
- **Surface complexation** with biochar functional groups
- **(Co)precipitation** with e.g. carbonate or phosphate

**Indirect effects of biochars on soil metal mobility** also exist, e.g. **an increased soil pH** affecting metal retention on soil particles<sup>[2]</sup>.

→ We compared here sorption studies with contaminated soil extractions in order to:

- Understand **how and where biochar can immobilize metals** at its surface
- Determine the **main mechanisms involved in metal immobilization in soils**
- Assess the **long-term influence** of biochar on **soil metal mobility**

## MATERIALS:

## METHODS:

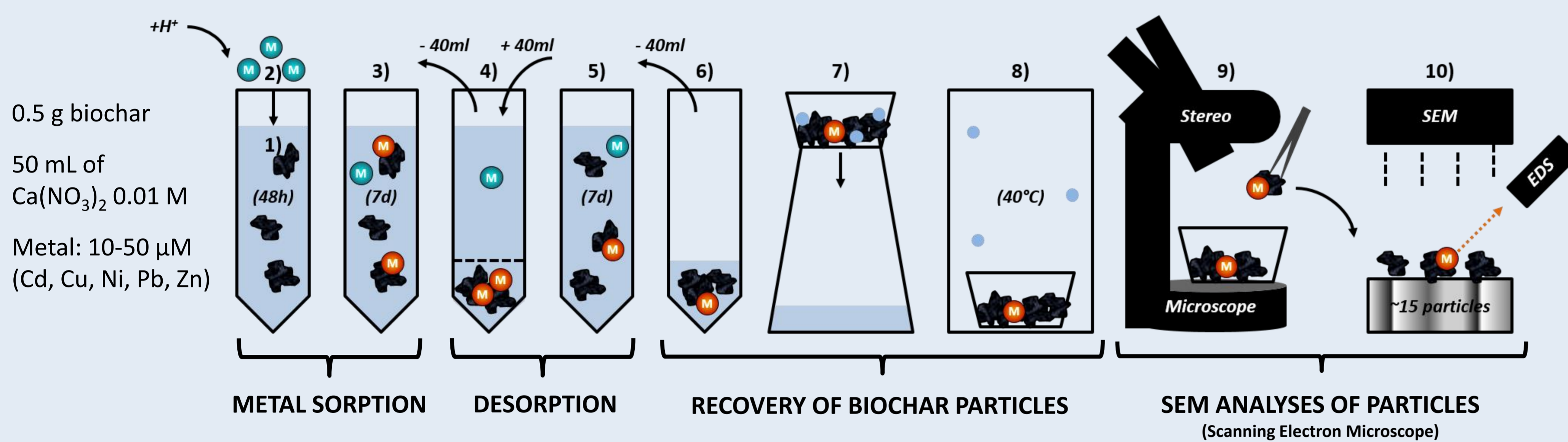
	Biochar	Soil A	Soil B	
<b>pH</b>	9.2	5.9	8.1	-
<b>Cd</b>	0.4 (0.3%)	17.6 (34%)	18.6 (1.3%)	mg kg <sup>-1</sup>
<b>Pb</b>	7.2 (-)	1120 (0.2%)	1080 (0.005%)	mg kg <sup>-1</sup>
<b>Zn</b>	112 (0.03%)	3170 (22%)	1380 (0.2%)	mg kg <sup>-1</sup>

**1 biochar:** produced at ~450°C from woody biomass, sieved to <2mm, untreated.

**2 soils:** contaminated by smelters activity, with similar properties except soil pH

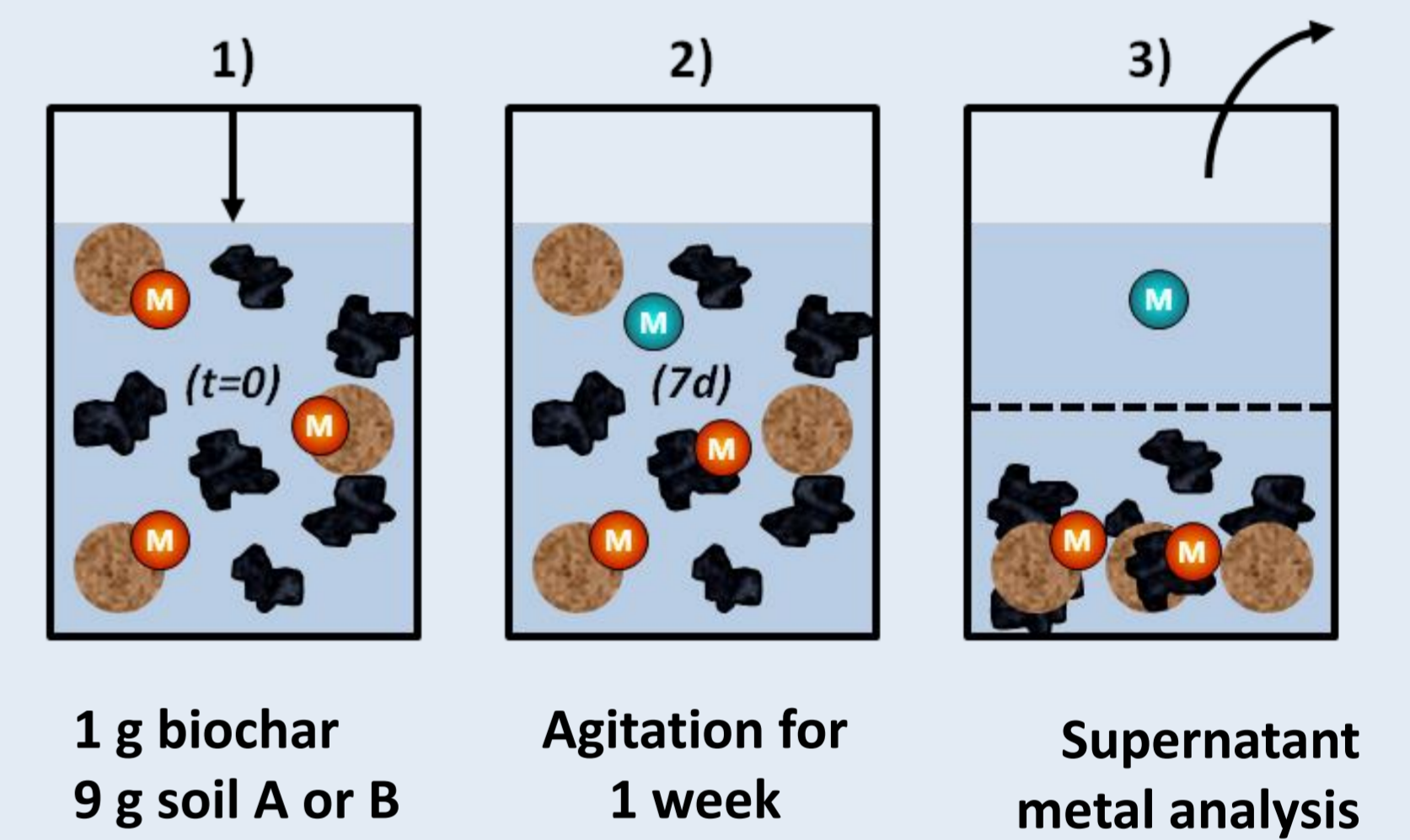
TOTAL CONTENT (CaCl<sub>2</sub> exchangeable fraction)

### 1) Metal sorption isotherms and analysis of metal-loaded biochars



### 2) Soil extraction

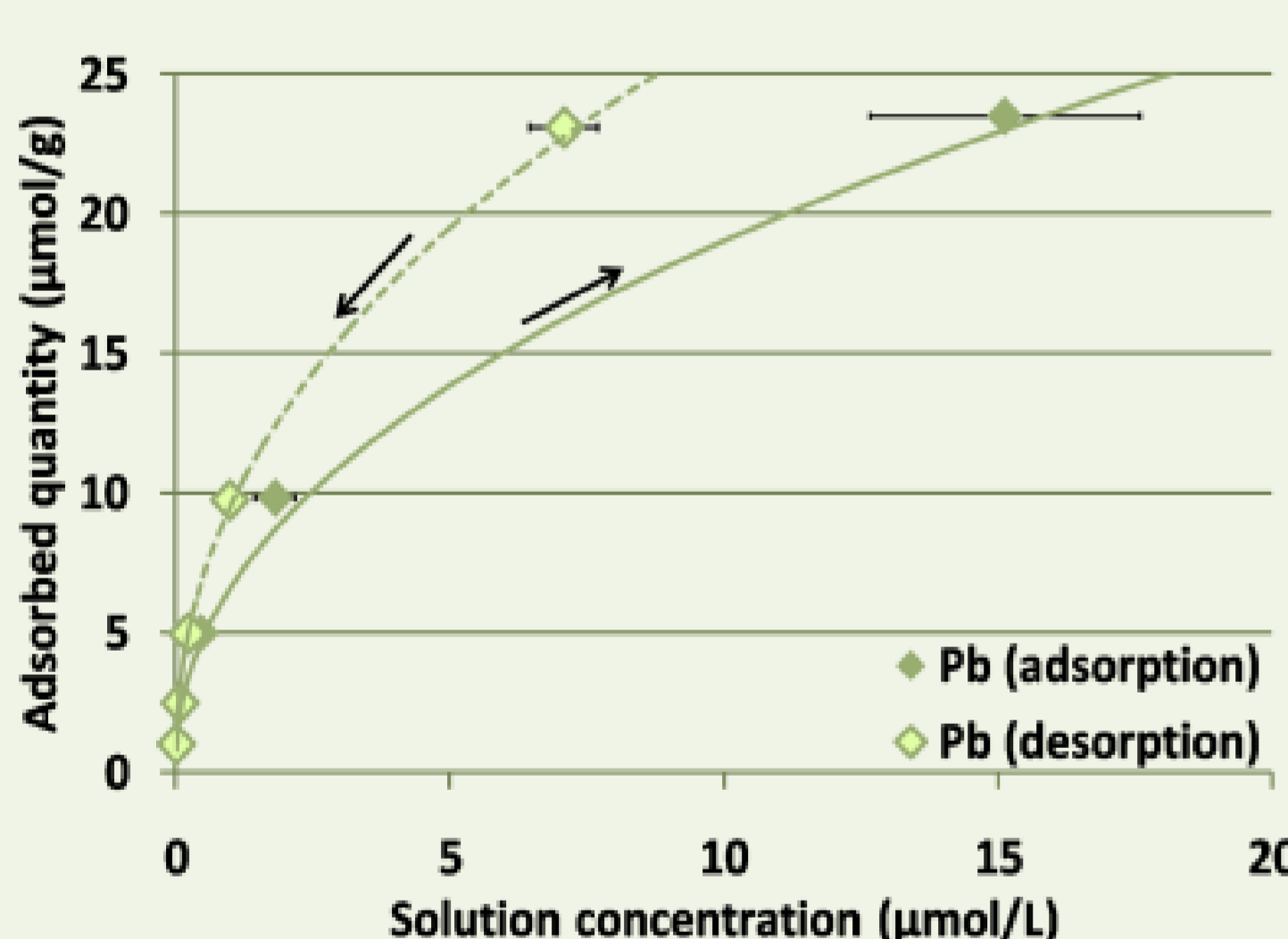
→ With 0% or 10% (w/w) biochar amendments



## RESULTS & DISCUSSION

▪ Metals are **immobilized at the surface of biochar**, following the order: Pb>Cu>Cd>Zn>Ni<sup>[3]</sup>

**Biochar Pb sorption isotherms fitted by Freundlich model**



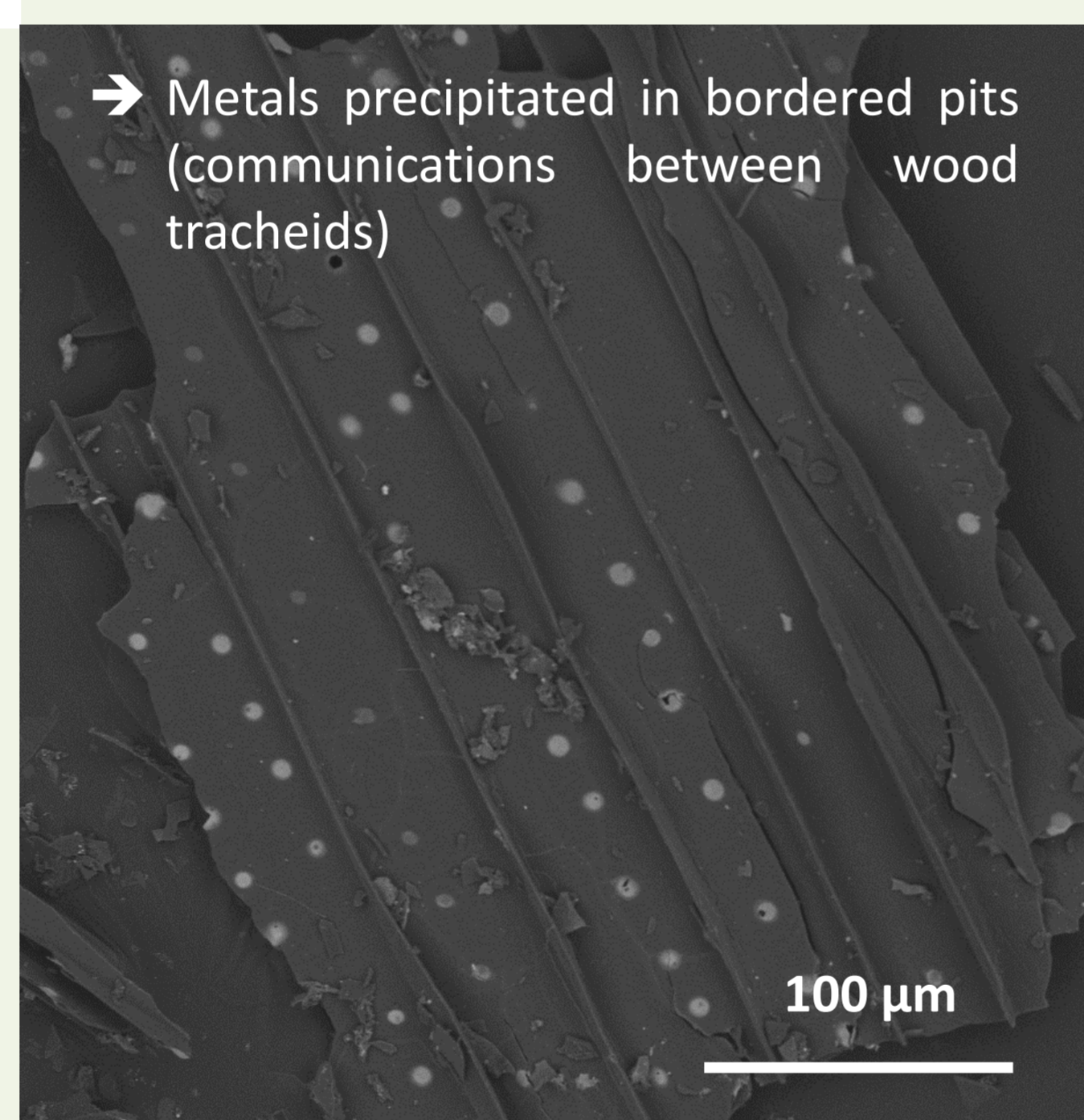
▪ Isotherms hysteresis:  
→ **sorption is partially irreversible**

▪ No significant evolution of Na<sup>+</sup>, K<sup>+</sup> or Mg<sup>2+</sup> with increasing sorbed metal:  
→ **cation exchange is limited**

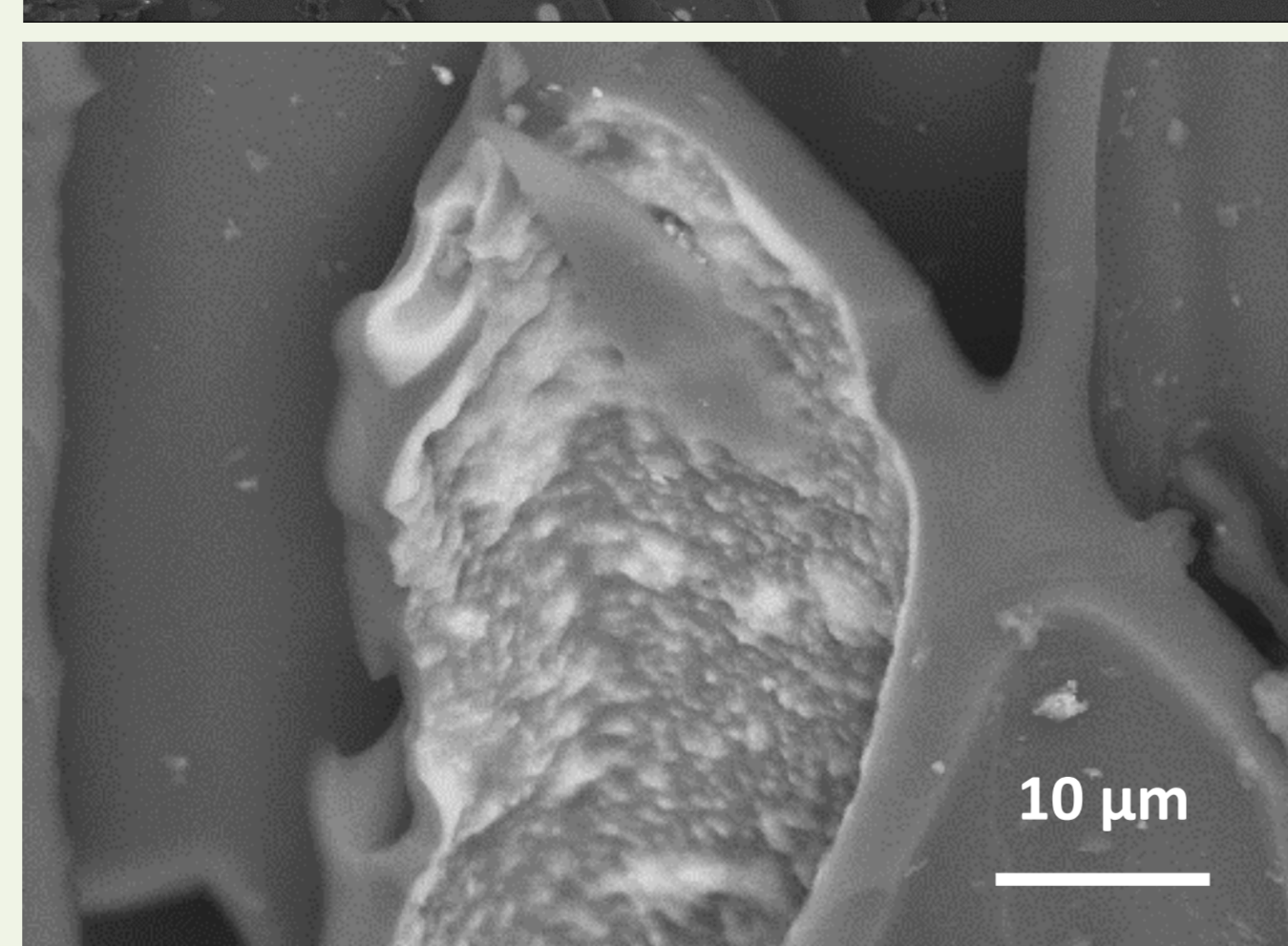
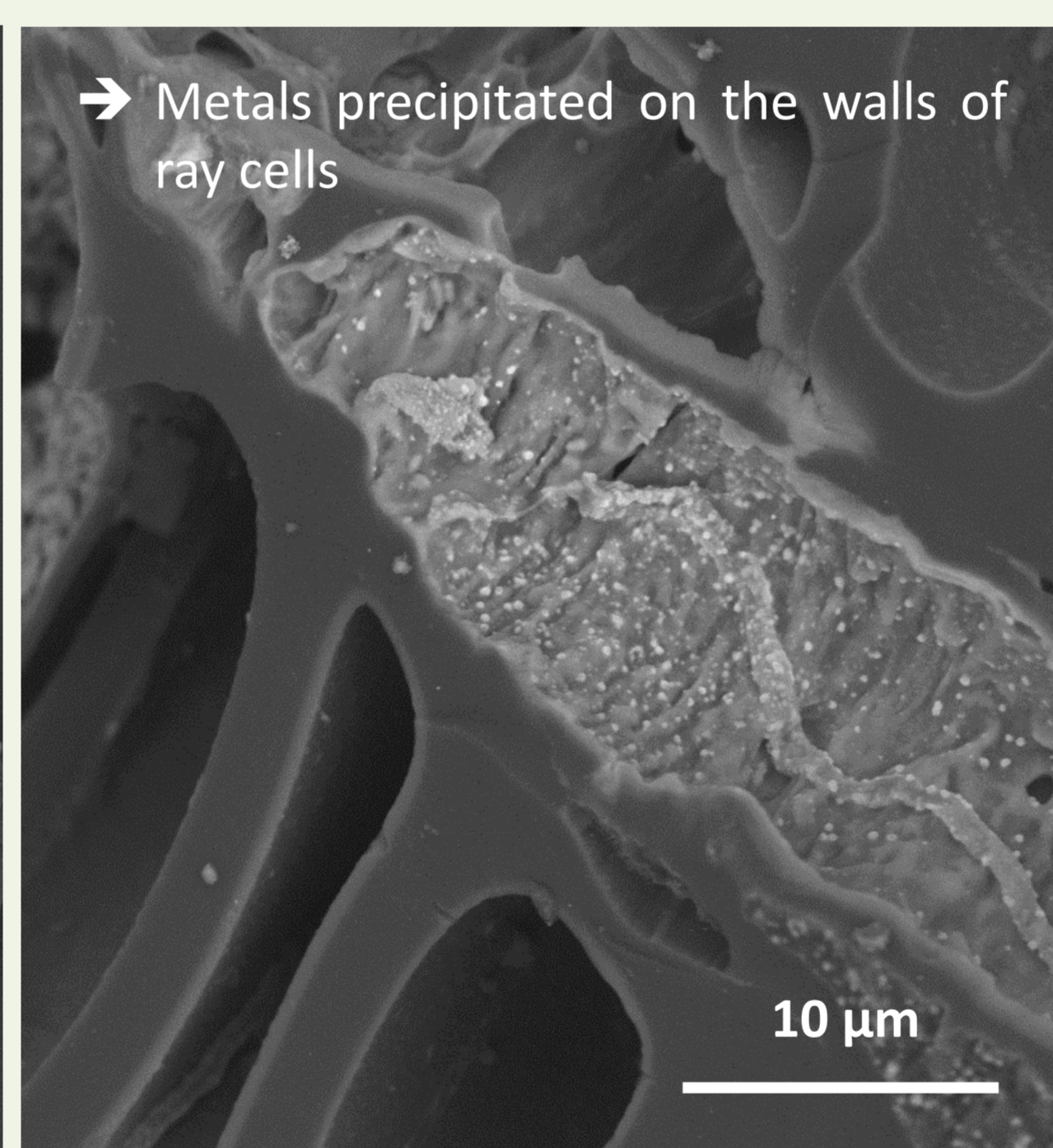
▪ pH and carbonates ↓ when sorbed Pb ↑:  
→ **precipitation of metals with carbonates**

▪ SEM observations confirm that **metals (Pb, Cd, Zn) concentrate in carbonate phases:**

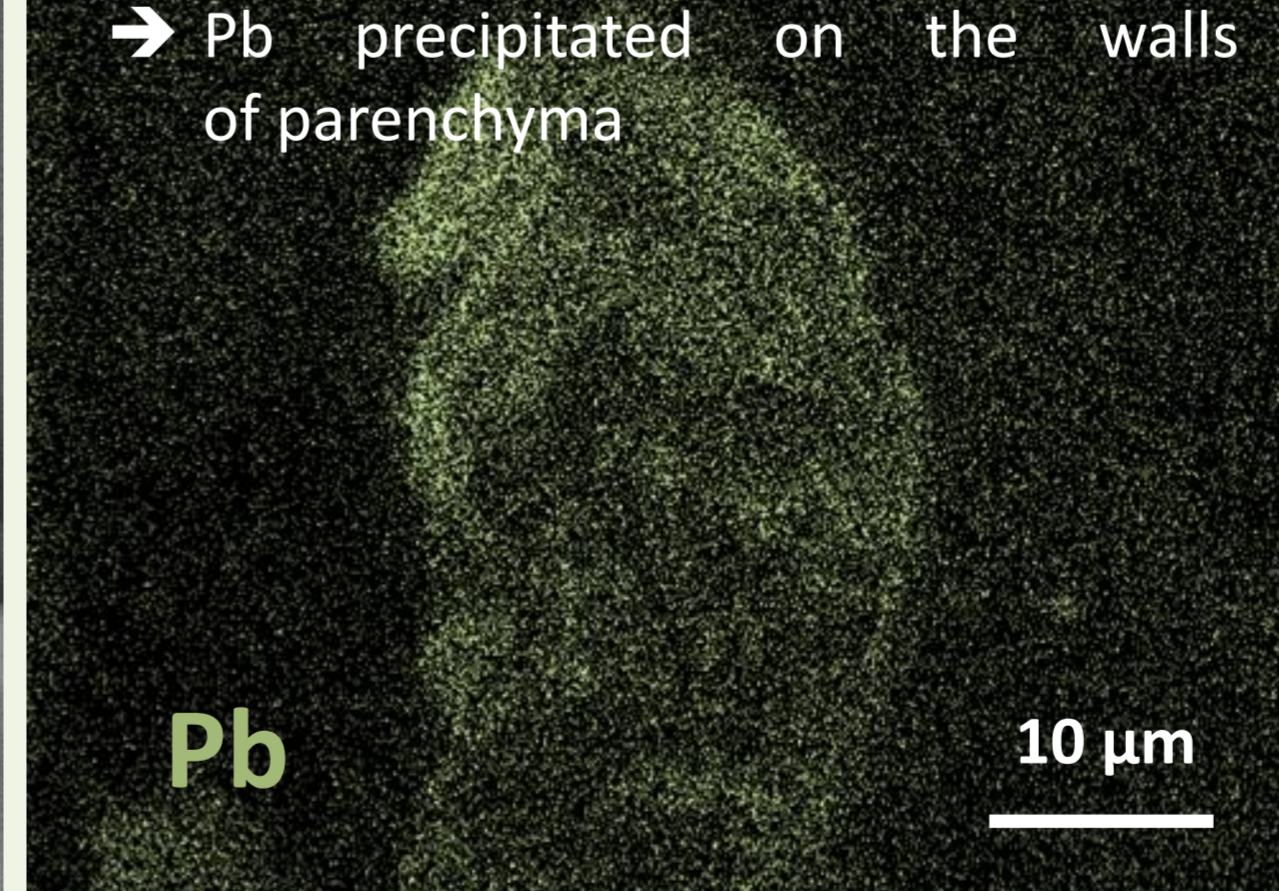
→ Metals precipitated in bordered pits (communications between wood tracheids)



→ Metals precipitated on the walls of ray cells



→ Pb precipitated on the walls of parenchyma



▪ The presence of biochar in both soils leads to<sup>[3]</sup>:

→ **Soil A: pH ↑**; metal extractability strongly ↓  
→ **Soil B: no pH change**; Cd extractability ↓ but no significant effects for Zn or Pb

**pH and equilibrium concentration of metal measured in soil extracts**

C <sub>eq</sub> (μmol/L)		Zn	Cd	Pb	pH (-)
		Control	1072	3.56	0.62
Soil A	Biochar	369	1.73	0.16	6.9
Soil B	Control	6.3	0.19	0.04	7.2
	Biochar	4.1	0.13	0.03	7.4

▪ The amount of immobilized metals in soils is different from the maximal amounts of metals sorbed on biochar assessed from sorption isotherms.

**Immobilized quantities of metal in the presence of biochar, predicted from sorption isotherms and actually observed**

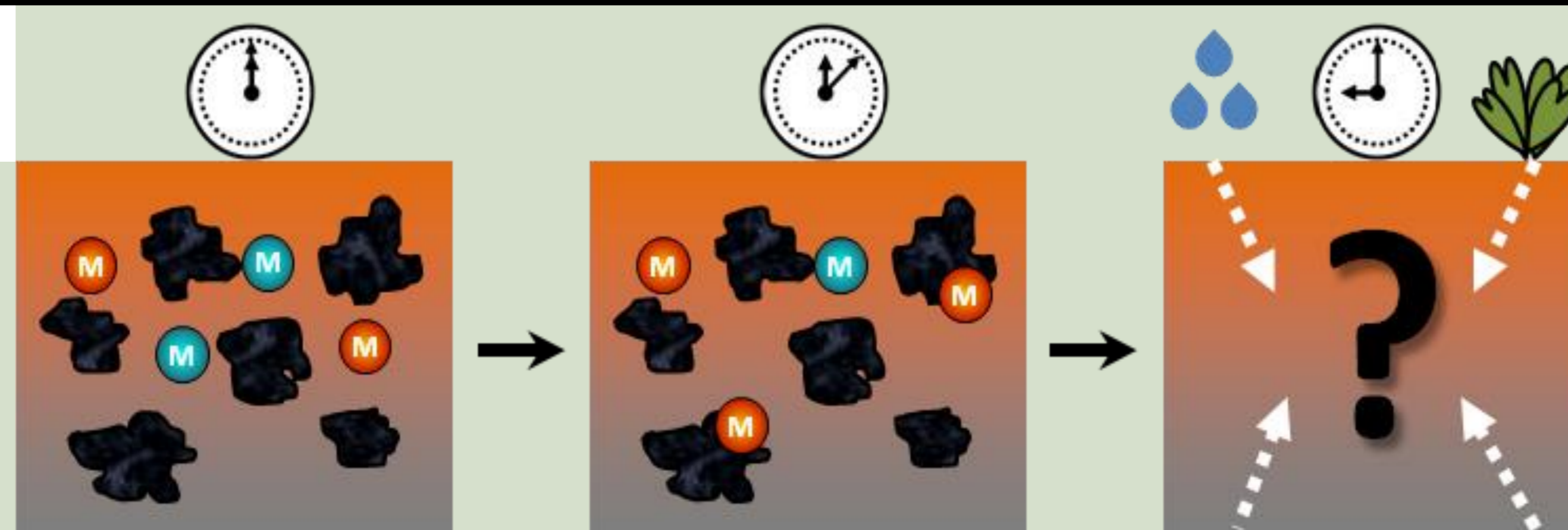
Q <sub>ads</sub> (μmol/g)		Zn	Cd	Pb
		Predicted	0.83	0.27
Soil A	Observed	56	0.15	0.04
Soil B	Predicted	0.25	0.030	0.494
	Observed	0.16	0.004	0.000

→ **Indirect effect of biochar by increased soil pH** explains the effects on soil metals mobility better than sorption on biochar

## CONCLUSIONS

▪ Biochar amendments in soils can decrease soil metal mobility by:

- 1) **Direct effects** by sorption at biochar's surface, e.g. by **precipitation of metals with carbonates**
- 2) **Indirect effects** on soil metal retention, by **soil pH increase**.



## PERSPECTIVES

- **Different feedstocks and pyrolysis conditions** could affect biochars ability to sorb metals (e.g. with various pH and carbonate content)
- **Soil biota and plants responses** to biochar need also to be considered in order to **predict long-term biochar effects** on metal mobility and availability.

[1] Beesley et al. (2011.) A review of biochars potential role in the remediation, revegetation and restoration of contaminated soils. *Environmental Pollution*

[2] Houben et al. (2013) Mobility, bioavailability and pH-dependent leaching of cadmium, zinc and lead in a contaminated soil amended with biochar. *Chemosphere*

[3] Rees et al. (2014) Short-term effects of biochar on soil heavy metal mobility are controlled by intra-particle diffusion and soil pH increase. *European Journal of Soil Science*