



**HAL**  
open science

## Study of the carbonization and activation of different lignocellulosic precursors for the elaboration of activated carbons with controlled porosity

Martin Pajot, Benoît Cagnon, Stéphanie de Persis, Jean-Paul J.-P. Charpentier, Marjorie Roulet, Alain Pineau, Sylvie Bonnamy

### ► To cite this version:

Martin Pajot, Benoît Cagnon, Stéphanie de Persis, Jean-Paul J.-P. Charpentier, Marjorie Roulet, et al.. Study of the carbonization and activation of different lignocellulosic precursors for the elaboration of activated carbons with controlled porosity. Exploring lignocellulosic biomass! Explorons la biomasse lignocellulosique!, Jun 2016, Reims, France. , 2016. hal-02799598

**HAL Id: hal-02799598**

**<https://hal.inrae.fr/hal-02799598>**

Submitted on 5 Jun 2020

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



# Study of the carbonization and activation of different lignocellulosic

## precursors for the elaboration of activated carbons with controlled porosity

Martin PAJOT<sup>1,2</sup>, Benoît CAGNON<sup>1</sup>, Stéphanie de PERSIS<sup>2</sup>, Jean-Paul CHARPENTIER<sup>3</sup>, Marjorie ROULET<sup>1</sup>, Alain PINEAU<sup>1</sup>, Sylvie BONNAMY<sup>1</sup>



<sup>1</sup>ICMN-Interfaces, Confinement, Matériaux et Nanostructures, CNRS, Orléans, France  
<sup>2</sup>ICARE-Institut de Combustion Aérodynamique Réactivité et Environnement, CNRS, Orléans France  
<sup>3</sup>INRA-Institut National de la Recherche Agronomique, plateforme GénoBois, Orléans, France



Valorization of Lignocellulosic Biomass



Energy

Solid Fuels



Materials

Activated Carbons

### Context

#### PYROLYSIS

gas, liquid

#### GASIFICATION

CO + H<sub>2</sub>

#### COMBUSTION

Heat, CO<sub>2</sub> + H<sub>2</sub>O

#### PHYSICAL ACTIVATION

Carbonization  
Ar or N<sub>2</sub>  
500-800°C

Activation  
H<sub>2</sub>O or CO<sub>2</sub> or O<sub>2</sub>  
up to 800°C

#### CHEMICAL ACTIVATION

Activating Agent + Carbonization  
H<sub>3</sub>PO<sub>4</sub> or HNO<sub>3</sub> + Ar or N<sub>2</sub>  
500-800°C

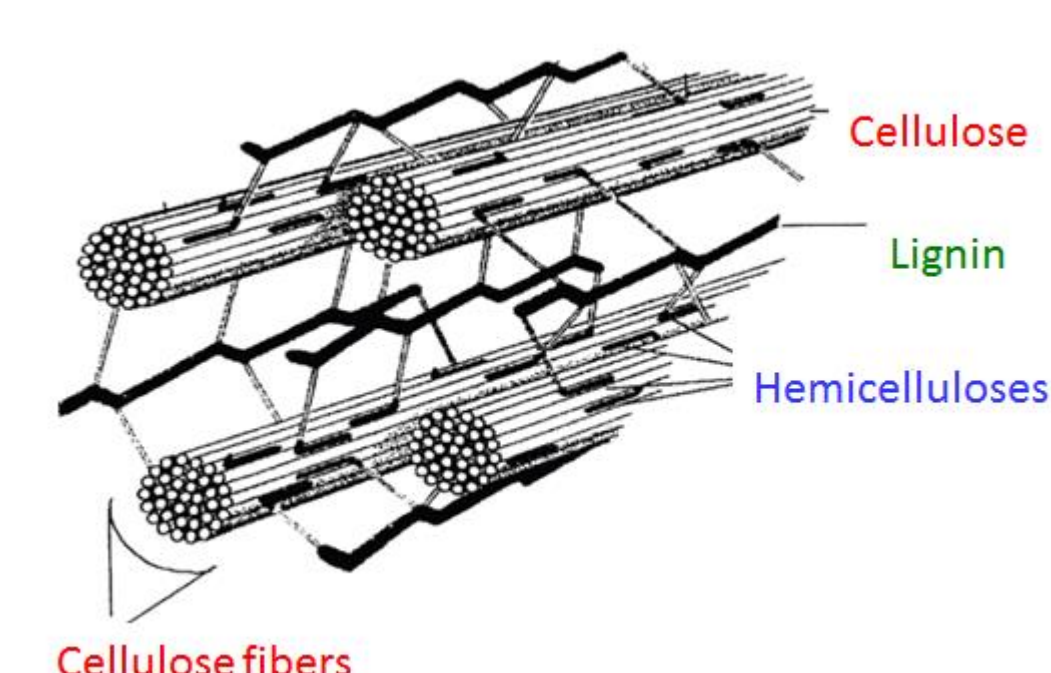
### OPTIMIZATION via the the study of the CARBONIZATION/PYROLYSIS

#### COMPLEX

for a *lignocellulosic precursor* due to:  
-differences in reactivity between hemicellulose, cellulose and lignin  
-the competition of the reactions accompanying their decomposition

#### KNOWLEDGE OF THE RAW MATERIAL MANDATORY:

Elementary analysis, biochemical composition, FTIR, XRD, Thermal decompositions by thermogravimetry (TGA-DTA)



## Thermal behavior of Raw Material and lignocellulosic compounds

TGA-DTA carried out on all the lignocellulosic precursors and on the three compounds (H., C., L.) => estimation of their respective contribution in the final mass of the solid phase (chars and activated carbons)

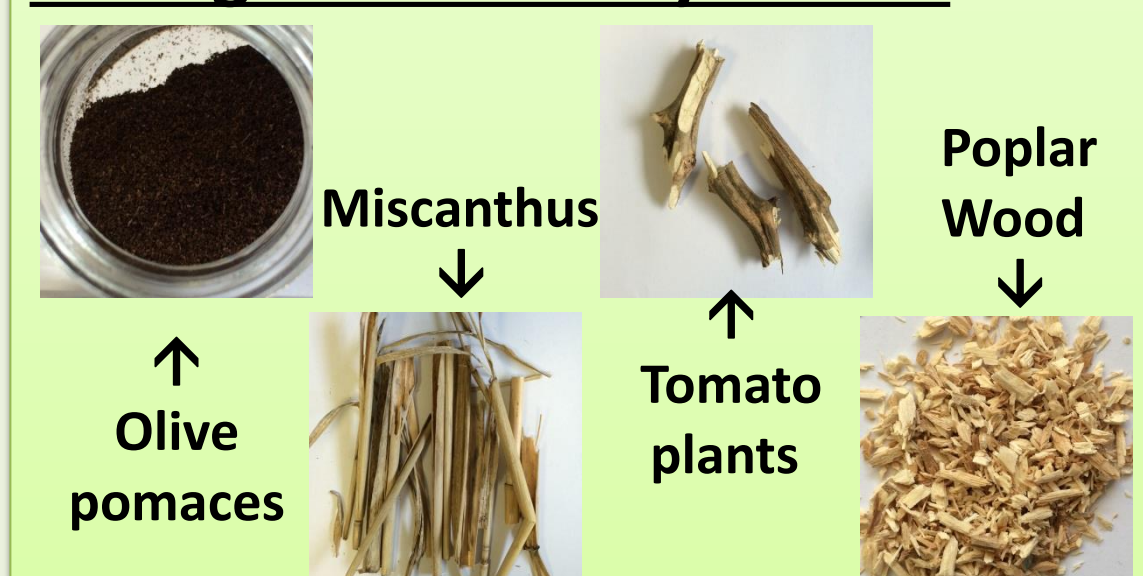
### Lignocellulosic compounds (commercial)

Hemicellulose (H.)

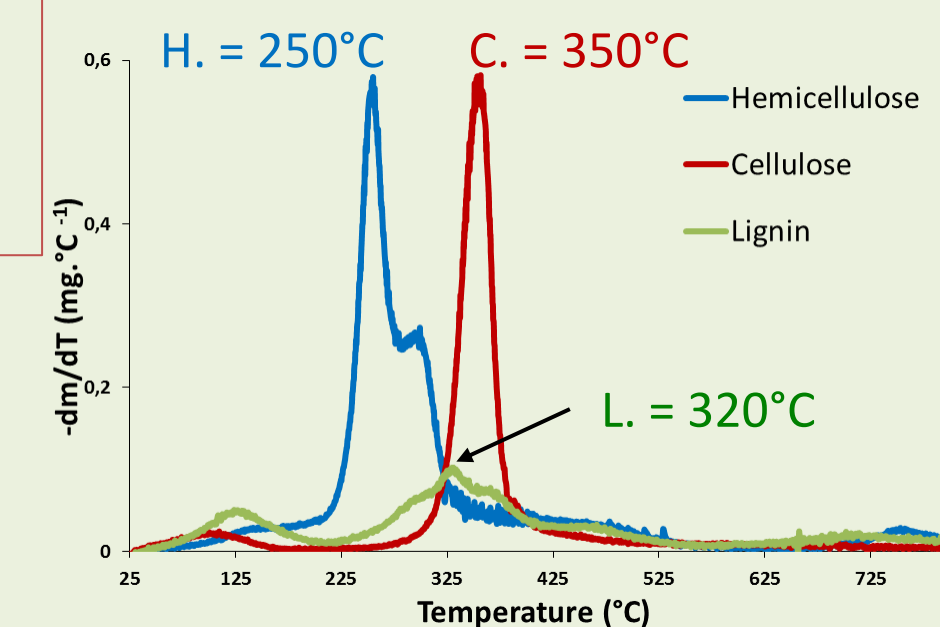
Cellulose (C.)

Lignin (L.)

### Raw material from agricultural and agro-alimentary wastes



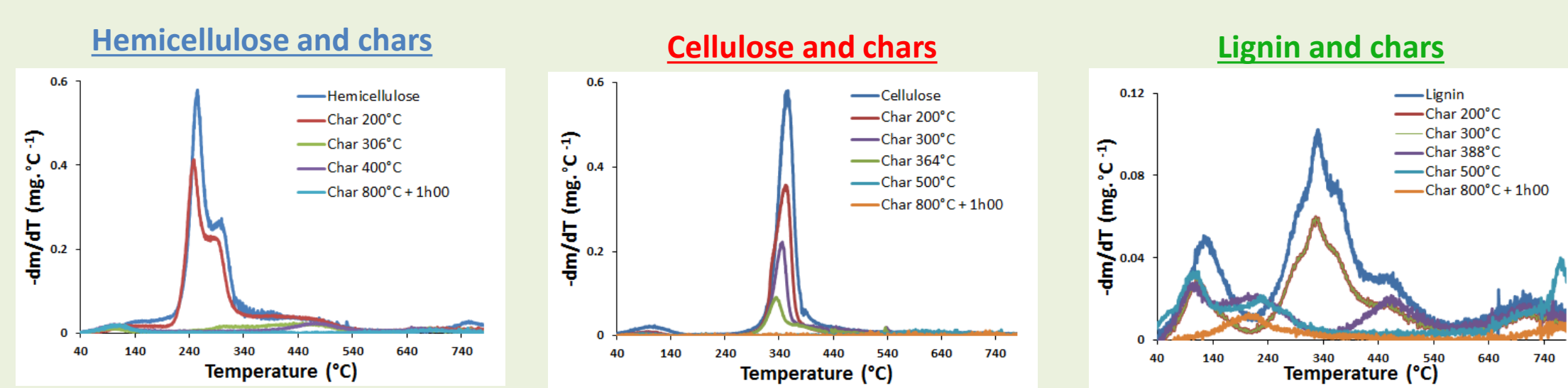
Experimental conditions:  
T<sub>final</sub>: 800°C-1h  
Heating rate: 20°C.min<sup>-1</sup>  
Vector gas: argon  
Flow: 160 mL.min<sup>-1</sup>



#### Carbonization of the pure compounds

Lignin decomposes over a wide temperature range whereas hemicellulose and cellulose have characteristic peaks

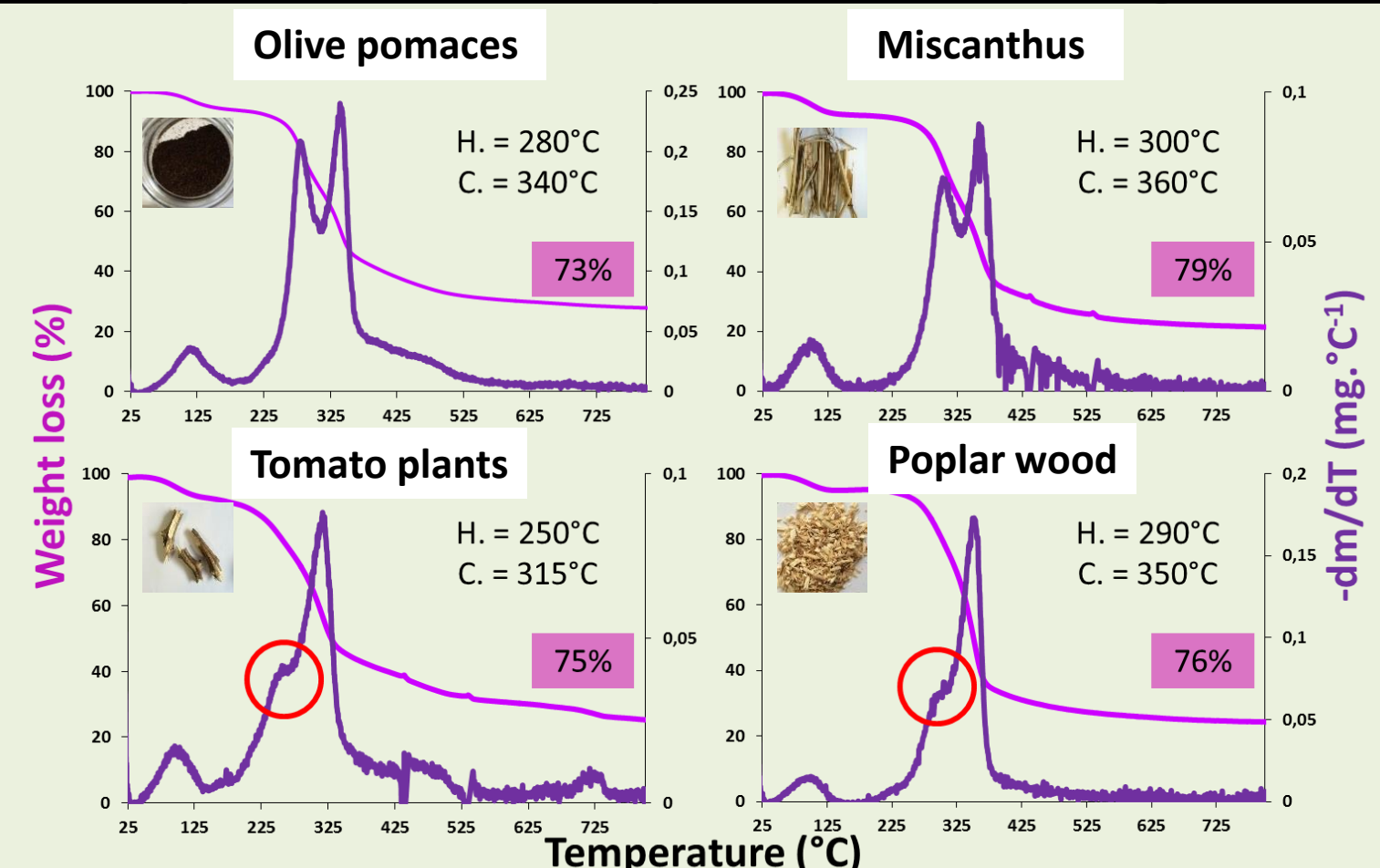
Thermal evolution followed by charring in an oven => Carbonization was done in several steps as a function of the previous thermograms



The thermal decomposition of Lignin seems to be more complex than Hemicellulose and Cellulose.

#### Thermal decomposition of the compounds inside the vegetal matrix

Identification of the characteristic peaks for lignocellulosic compounds



- Different thermal behaviors as a function of the raw material.
- Natural deconvolution of H. and C. for olive pomaces and miscanthus and L. is hidden.
- H. peak is less pronounced for tomato plants and poplar wood.



Analysis of the obtained data by means of a kinetic model required to better understand their thermal decomposition  
Optimization and modelling of the carbonization step required (whatever the activation method) as a function of the raw material in order to control final textural properties of the activated carbon.

## Is it possible to produce activated carbons from tomato plants?

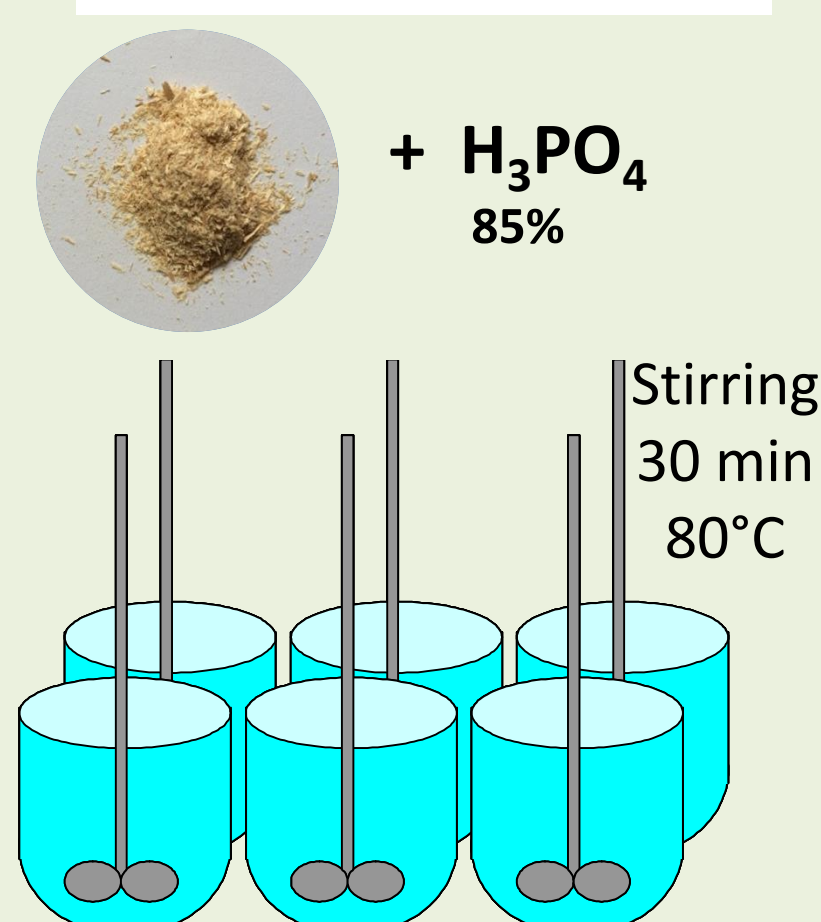
Raw material

Activated Carbon?

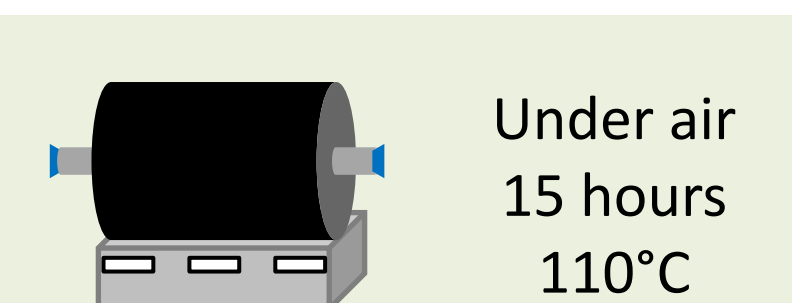
#### 1 Grinding



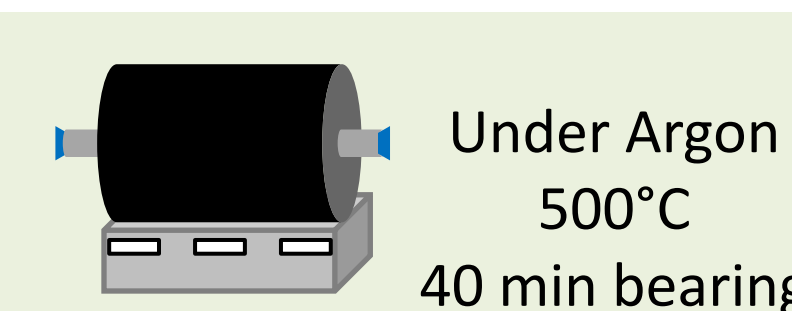
#### 2 Blending



#### 3 Drying



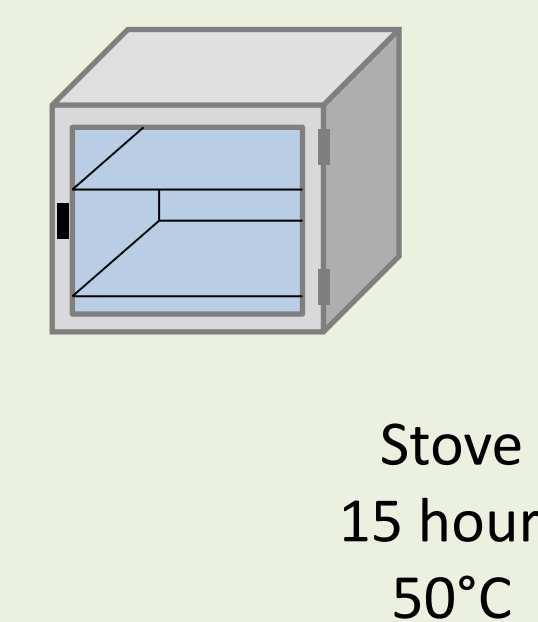
#### 4 Carbonization



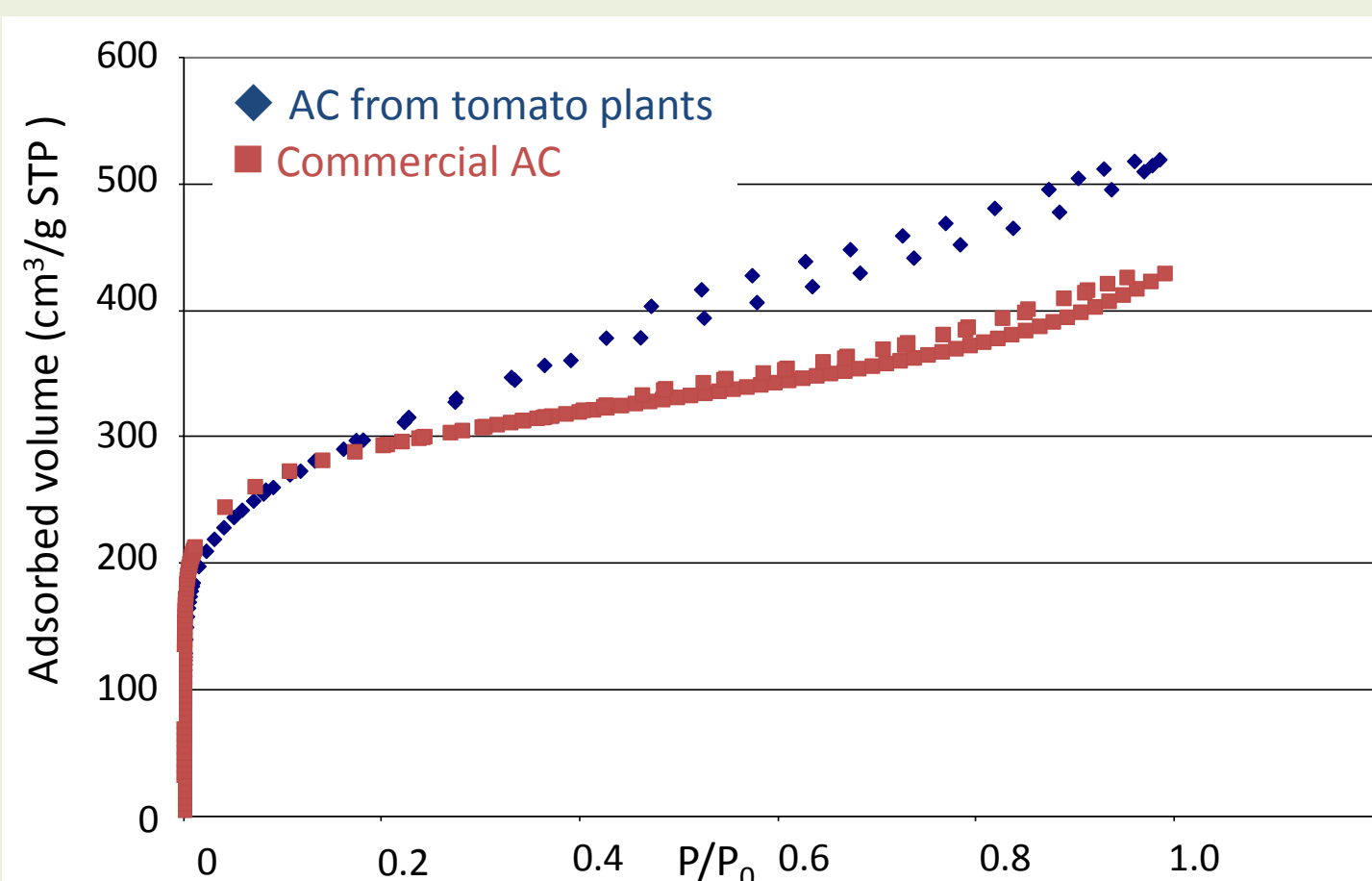
#### 5 Rinsing/Filtering

Product obtained + HCl 37%  
Stirring 20 min  
Filtering with distilled water until obtention of a neutral solution (pH = 7)

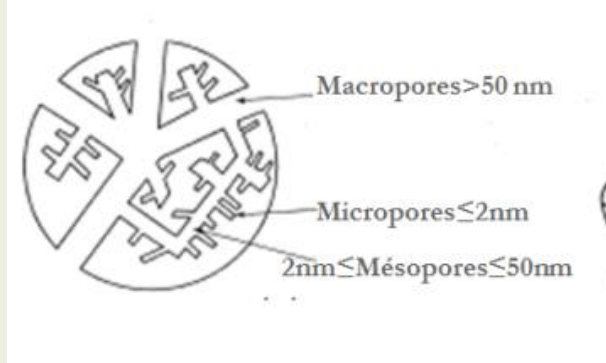
#### 6 Drying



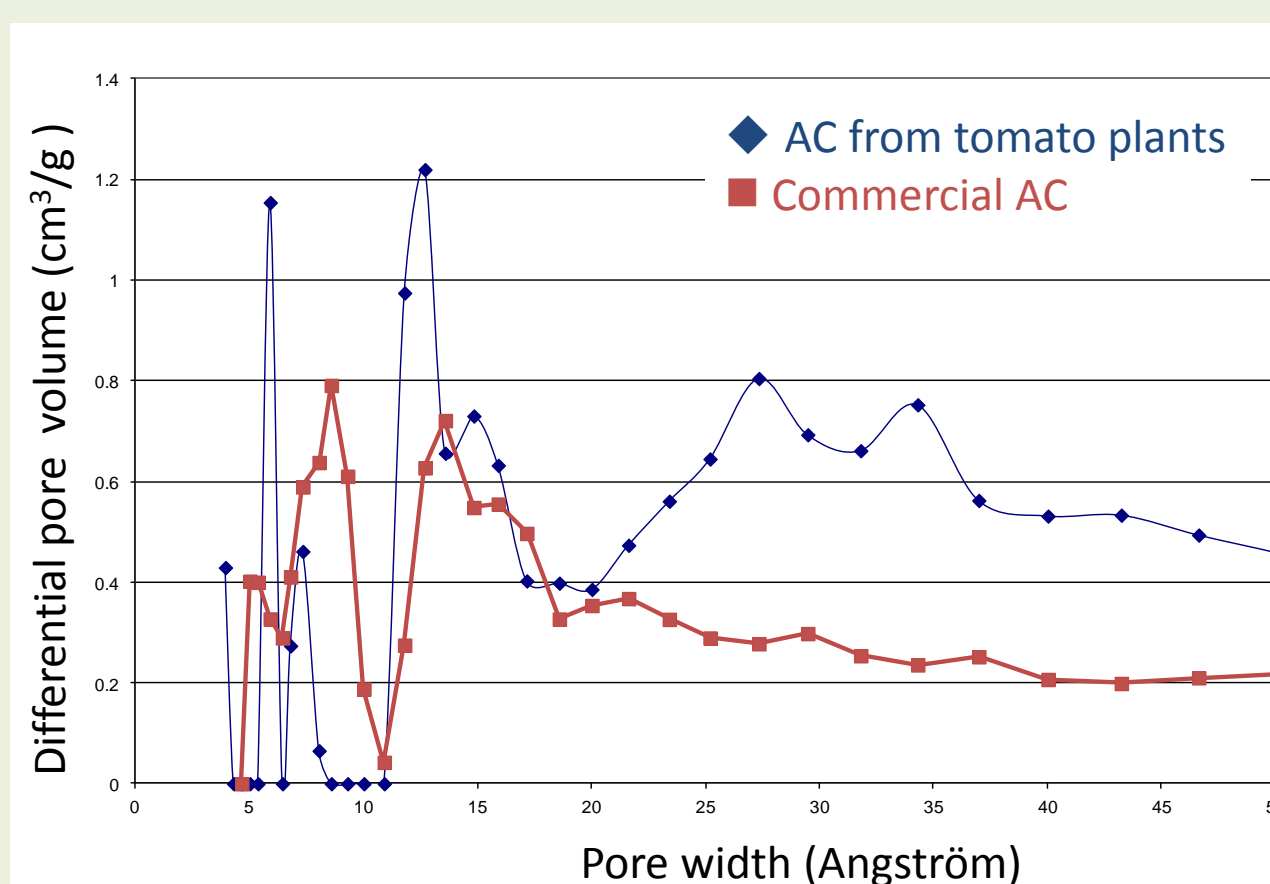
### Microporosity characterized by nitrogen adsorption at 77 K and comparison with a commercial AC



The isotherm shape indicates presence of micropores and mesopores



Confirmation by the distribution size of micropores by DFT method



#### Textural properties obtained from N<sub>2</sub> 77K adsorption isotherm

	W <sub>0</sub> (cm <sup>3</sup> /g)	L <sub>01</sub> (Å)	S <sub>BET</sub> (m <sup>2</sup> /g)	S <sub>ext</sub> (m <sup>2</sup> /g)	S <sub>micro</sub> (m <sup>2</sup> /g)	S <sub>total</sub> (m <sup>2</sup> /g)
Tomato AC	0.38	22.1	1090	404	344	748
Commercial AC	0.39	12.7	1077	256	614	870

W<sub>0</sub>: specific microporous volume; L<sub>01</sub>: mean microporous size; S<sub>BET</sub>: BET surface; S<sub>ext</sub>: external surface; S<sub>micro</sub>: microporous surface } S<sub>total</sub>: total surface



Very good result obtained without any optimization: AC from tomato plants has textural properties closed to commercial AC used for water treatment

## Conclusion

First tests of AC elaboration: very good result (without optimization) with tomato plants: textural properties fitted for water treatment

activation optimization required to control porosity for a given application

Perspective: development of a predictive method to estimate the mass and the textural properties of the final AC

based on the study of the raw material: elementary analyses, biochemical composition, FTIR, XRD, Calorimetry, TGA curves, modelling

For activated carbons => control of the porosity => for which application?

For solid fuels => High Heating Value (HHV), modelling of pyrolysis and secondary gas-phase reactions

