

Rapid assessment and prediction of the efficiency of two preservatives
against *S. aureus* in cosmetic products using High Content Screening
– Confocal Laser Scanning Microscopy

Running title: Assessment and prediction of preservative efficiency by HCS-CLSM

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Abstract

Most cosmetic products are susceptible to microbiological spoilage due to contaminations that
could happen during fabrication or by consumer's repetitive manipulation. The composition of

cosmetic products must guarantee efficient bacterial inactivation all along with the product shelf life, which is usually assessed by challenge-tests. A challenge-test consists in inoculating specific bacteria, i.e. *Staphylococcus aureus*, in the formula and then investigating the bacterial log reduction over time. The main limitation of this method is relative to the time-consuming protocol, where 30 days are needed to obtain results. In this study, we have proposed a rapid alternative method coupling High Content Screening - Confocal Laser Scanning Microscopy (HCS-CLSM), image analysis and modeling. It consists in acquiring real-time *S. aureus* inactivation kinetics on short-time periods (typically 4h) and in predicting the efficiency of preservatives on longer scale periods (up to 7 days). The action of two preservatives, chlorphenesin and benzyl alcohol, was evaluated against *S. aureus* at several concentrations in a cosmetic matrix. From these datasets, we compared two secondary models to determine the logarithm reduction time (Dc) for each preservative concentration. Afterwards, we used two primary inactivation models to predict log reductions for up to 7 days and we compared them to observed log reductions. The IQ model better fits datasets and the Q value gives information about the matrix level of interference.

Introduction

Each year around the world, official authorities in Europe (Rapid Alert System for Non-Food Products) or USA (US Consumer Product Safety Commission) notify many recalls for cosmetic products due to microbiological contamination [1-3]. Cosmetic formulas are complex and are susceptible to microbiological spoilage due to their composition, containing water and nutrients such as lipids, polysaccharides, proteins...[4]. Contamination of cosmetic products could happen during their fabrication but also by consumer's repetitive manipulations [5, 6]. The main pathogens frequently found in cosmetic formulas are *Pseudomonas aeruginosa*, *Escherichia coli*, *Burkholderia cepacia*, *Candida albicans*, *Klebsiella oxytoca*, *Enterobacter gergoviae*,

Serratia marcescens and *Staphylococcus aureus* [2, 7-9]. *S. aureus* has been found in various cosmetic products such as shaving cream, moisturizing cream, face care cream and depilatory cream [10-12]. It is a Gram-positive bacterium present on human skin and mucous membranes in 30% of the population [13]. Many *S. aureus* strains produce exfoliative toxins secreted on the skin that cause a wide range of clinical infections, including abscesses, furuncles or impetigo [14-17].

Each cosmetic product has a different level of microbiological risk according to the standard ISO 29621:2017, which depends on several parameters such as the formula composition (preservative, ethanol, A_w , pH) or the type of packaging (unidose, airless pump, pots) [6, 18]. Preservatives that can be used in cosmetic products are listed in Annex V of the European Regulation No. 1223/2009. Among them are listed chlorphenesin and benzyl alcohol, which have been tested in this study. Chlorphenesin or 3-(4-chlorophenoxy)-1,2-propanediol is an antifungal and antibacterial agent (active against both Gram-positive and Gram-negative bacteria). It can be used at a maximum concentration of 0.32% in rinse-off products and up to 0.30 % in leave-on products [19]. Benzyl alcohol can be used in various cosmetic formulations as a preservative, but also as a solvent, a fragrance or a viscosity-controlling agent. Its maximum in-use concentration is 1% [20].

The preservation efficiency of a given product is evaluated by proceeding to a challenge-test, as defined in the European standard EN ISO 11930:2019. During this procedure, specific microorganisms, including *S. aureus*, are inoculated in the product at a final concentration between 1.10^5 and 1.10^6 CFU.ml⁻¹ for bacteria, and 1.10^4 and 1.10^5 CFU.ml⁻¹ for molds or yeast. The microbial population is evaluated at defined time intervals by enumerating the survivors at 7, 14 and 28 days after inoculation. A preservative system is considered as efficient against bacteria if the formula composition leads to a bacterial logarithm reduction ≥ 3 seven days after inoculation and without the growth of bacteria after 14 and 28 days. Challenge-test, as described

in the European standard involves several steps including sampling, neutralization, serial dilutions, bacterial plating in duplicate, incubation time and colony counting [21]. The reliability of challenge-tests depends on several parameters such as the manipulation errors (pipetting and serial dilutions) [22], the type of plating method (spiral or pour plating), the level of bacterial enumeration [23], and on the ability of stressed microorganisms to recover and grow on agar plates [24]. It also relies on the efficiency of the neutralization step which consists of stopping the antimicrobial activity of preservatives by diluting the surviving population in a quenching solution [25]. The main limitation of the challenge-test procedure is relative to the time-consuming protocol (inoculation, sampling, counting) and to the duration of the whole test process (last sample analyzed on day 28).

Confocal Laser Scanning Microscopy (CLSM) allows *in-situ* 3-D visualization of microbial consortia thanks to various fluorescent markers. It is commonly used to investigate complex microbial spatial organizations such as biofilms [26], to analyze interactions between bacteria and oil droplets [27] or to evaluate bacterial distribution in food systems [28-30]. Moreover, CLSM was previously used to study the spatiotemporal action of biocide in biofilms [26, 31-34]. This method enables a real-time and *in situ* visualization of the bacterial inactivation kinetics after biocide addition. Typically, living cells are stained with an esterase viability marker, such as cFDA or calcein-AM, and after subsequent biocide addition, the fluorescence is lost due to the leakage of the fluorescent marker out of the cell when the cell membrane is permeabilized.

In this study, we used CLSM and image analysis for acquiring datasets of bacterial inactivation kinetics upon short periods in model cosmetic matrices containing various concentrations of preservatives and we accurately predicted the number of bacterial log reductions on longer periods, which are similar to challenge-test ones.

Materials and Methods

Chemicals and materials

Cetearyl glucoside and glyceryl stearate were purchased from SEPPIC (Puteaux, France), carbomer from Gattefossé (Lyon, France), glycerin from Oleon (Ertvelde, Belgium), cetearyl isononanoate from BASF France (Lyon, France), tocopheryl acetate from DSM (Heerlen, the Netherlands), tromethamine from Azelis (Heusden, Belgium), chlorphenesin and benzyl alcohol from Thor (Compiègne, France). Eugon LT 100 supplemented broth was purchased from Indicia production (Saint Genis l'Argentière, France).

Bacterial strain and culture conditions

The strain used in this study is *Staphylococcus aureus* CIP 4.83 recommended by the EN ISO 11930:2019 standard for cosmetic-product challenge tests. It was stored in cryovials at -80°C and resuscitated by two successive subcultures in tryptic soy broth (TSB, Biomérieux, Marcy-l'étoile, France) before each experiment. Cultures were grown at 30°C until the end of the exponential growth phase.

Preparation and characterization of the emulsified model matrix

The aqueous phase was first prepared with 0.25% carbomer in water and heated to 75°C before glycerin (moisturizer, 9%) was added. The oil phase is composed of 28.8% cetearyl isononanoate (emollient), 3.5% cetearyl glucoside (emulsifier), 0.2% tocopheryl acetate (antioxidant), and 2.5% glyceryl stearate (co-emulsifier). It was heated to 75-80°C before it was blended with the aqueous phase (20/80 o/w %) at 1,800 rpm using a rotor-stator homogenizer (Rayneri 33/300P, Group VMI) to obtain an emulsion. Benzyl alcohol and chlorphenesin at 7 different concentrations (respectively from 1.00 to 1.85 % and from 0.30 to 0.60 %) were

respectively pre-mixed with glycerin or water at 40°C. Tromethamine (base, 0.15%) was finally added. The viscosity was measured using a penetrometer (PNR10, PetroMesures). A specific cone was released in 300 g of matrix and the penetration depth measured (in mm \pm 0.1 mm) after 5 s. The penetrometry measured on each batch in triplicate is 33.06 mm \pm 1.33 mm. The pH, measured using pH-meter (SI Analytics, Lab 870) on each batch in triplicate, is 5.76 \pm 0.03.

Bacterial staining and matrix inoculation

Bacterial cells were harvested by centrifugation at 1,575 g for 10 min and washed twice in 150 mM NaCl. The bacterial suspension was calibrated to 1.10^{10} to 1.10^{11} CFU.ml⁻¹ in 150 mM NaCl to observe at least 10-100 bacteria per CLSM image (290.6 x 290.6 x 1.6 μm^3). 300 μl of bacterial suspension were labeled with 13 μl calcein-AM (53.55 μM in DMSO, Invitrogen by Thermofisher Scientific), incubated in the dark for 1h30 at 37°C and inoculated in 30 g of model cosmetic matrix which was vortexed for 30 s. The average of the bacterial concentration in the matrix is 1.10^8 to 1.10^9 bacteria/g. Calcein-AM is a viability marker that penetrates passively into a cell where it is cleaved by cytoplasmic esterases and leads to green fluorescence. Each experiment was performed respectively two or three times from independent cultures for benzyl alcohol and chlorphenesin.

Enumeration of the bacterial population by drop-plate method

For each enumeration, 1 g of inoculated matrix was dispersed in 9 ml of neutralization solution (Eugon LT 100 supplemented broth). After 30 minutes, the bacterial population is enumerated by serial dilution in 150 mM NaCl on tryptone soya agar (TSA, Biomérieux) using the drop-plate method [35]. Plates were incubated at 30°C for 24 to 48 h before counting. Bacterial enumeration is processed every twenty minutes for four hours after inoculation and then at least

once every day until seven days. Each enumeration was performed at least in duplicate.

Acquisition of bacterial inactivation curves by High Content

Screening - Confocal Laser Scanning Microscopy (HCS-CLSM)

The evolution of bacterial population was acquired upon a short time (typically 4h) for 7 different concentrations in duplicate for benzyl alcohol and in triplicate for chlorphenesin. To obtain one inactivation curve, the inoculated matrix containing a specific concentration of a preservative was dropped into several wells of polystyrene 96-well microtiter plates (Greiner Bio-One, France) and CLSM acquisition was achieved in each well at a specific time to avoid photobleaching. Thanks to the HCS-CLSM, the stage was programmed to move automatically to the next well every 15 min during 4h or every hour during 13h for low concentrations.

Image acquisition was performed using a Leica SP8 AOBS Confocal Laser Scanning Microscope (Leica Microsystems, France) at the MIMA2 imaging platform (<https://doi.org/10.15454/1.5572348210007727E12>). Calcein-AM is excited at 488 nm and the emitted fluorescence collected in the range 498 to 560 nm. Images size were 290.6 x 290.6 x 1.6 μm^3 (512 x 512 pixels) and were acquired at 600 Hz using a 40x air objective (N.A. = 0.85) and a hybrid detector. The HCS-CLSM control software was programmed to take a mosaic of 10 x 10 images per well, corresponding to a volume of 1.3×10^{-5} ml. The number of bacteria by mosaic was counted by binarizing each image using the MaxEntropy algorithm in an automatic macro executed in ImageJ software (National Institutes of Health, USA) [36]. The obtained number of bacteria per ml was converted per g according to the matrix density (1.15 g/ml). In our experimental conditions, we consider that our threshold value is at 1 bacteria per image or 100 bacteria per mosaic, which corresponds to 6.10^6 bacteria/g.

Primary model for bacterial inactivation on short times

The log-linear model of Bigelow *et al.* [37], described in equation 1, was used to fit each CLSM inactivation curve acquired on short-times.

$$\log_{10}(N) = \log_{10}(N_0) - \frac{t}{Dc} \quad (1)$$

where N_0 is the initial bacterial population, N is the bacterial population at the sampling time, Dc is the decimal reduction time and t is the time (min).

The GinaFit freeware add-in for Microsoft Excel was used to fit each curve [39] and to obtain the Dc value to which we applied a correction factor to take into account to the correlation between CLSM enumeration and plate enumeration. Hence, we obtained a dataset of Dc , each of them corresponding to a specific concentration of one preservative.

Secondary model for estimation of the Dc -value according to the concentration

From obtained Dc datasets, the Dc values were fitted according to concentration using a semi-log approach, derived from Mafart *et al.* (2001) [40], and expressed in equation 2.

$$\log_{10}(Dc) = \log_{10}(Dc^*) - \left(\frac{C-C^*}{z_c}\right)^n \quad (2)$$

where Dc is the decimal reduction time for the concentration C , Dc^* is the decimal reduction time for the reference concentration C^* , z_c is the increase of concentration which leads to a ten-fold reduction of the decimal reduction, n is a shape parameter which can be set to 1 (model #1, linear model) or 2 (model #2, second-degree model). Dc^* and z_c were the estimated parameters.

The model parameters were fitted with `nls` R function according to the minimization of the residual sum of square errors (RSS). Confidence intervals of fitted parameters were assessed by bootstrap using `nlsBoot` function from `nlsMicrobio` R package [41]. The two models were compared according to the Bayesian information criterion (BIC) (equation 3). The lower the BIC, the better the model fits the dataset.

$$BIC = p \cdot \ln\left(\frac{RSS}{p}\right) + k \cdot \ln(p) \quad (3)$$

Where p is the number of experimental points and k the number of parameters of the model

Prediction of the log-reduction of the bacterial population over a period of several days

To predict the log reduction of the bacterial population over several days, we first predict D_c with the secondary model #2 (equation 2) at some tested concentrations of preservative.

Afterwards, two different models were used to predict the inactivation of the bacterial population as a function of time: the log-linear model (equation 1) and IQ model (equation 4).

The intrinsic quenching model (Lambert et al., 2000) [38] was constructed with the hypothesis that the disinfection concentration decreases during the test period and can be described by the equation 4.

$$\log_{10}(N) = \log_{10}(N_0) - \frac{(1 - e^{-Q \cdot t})}{Q \cdot D_c} \quad (4)$$

where N_0 is the initial bacterial population, N is the bacterial population at the sampling time, D_c is the decimal reduction time, t is the time (min) and Q is the quenching coefficient. Q was the estimated parameter.

The logarithm reduction of the bacterial population that should be obtained after a defined time, from 1 to 7 days was predicted. To optimize and validate the model, a dataset of log-reductions of the bacterial population was acquired by plate enumeration on the corresponding periods (1 to 7 days) for 7 concentrations of each preservative, as described before. Predicted and observed log reductions were compared.

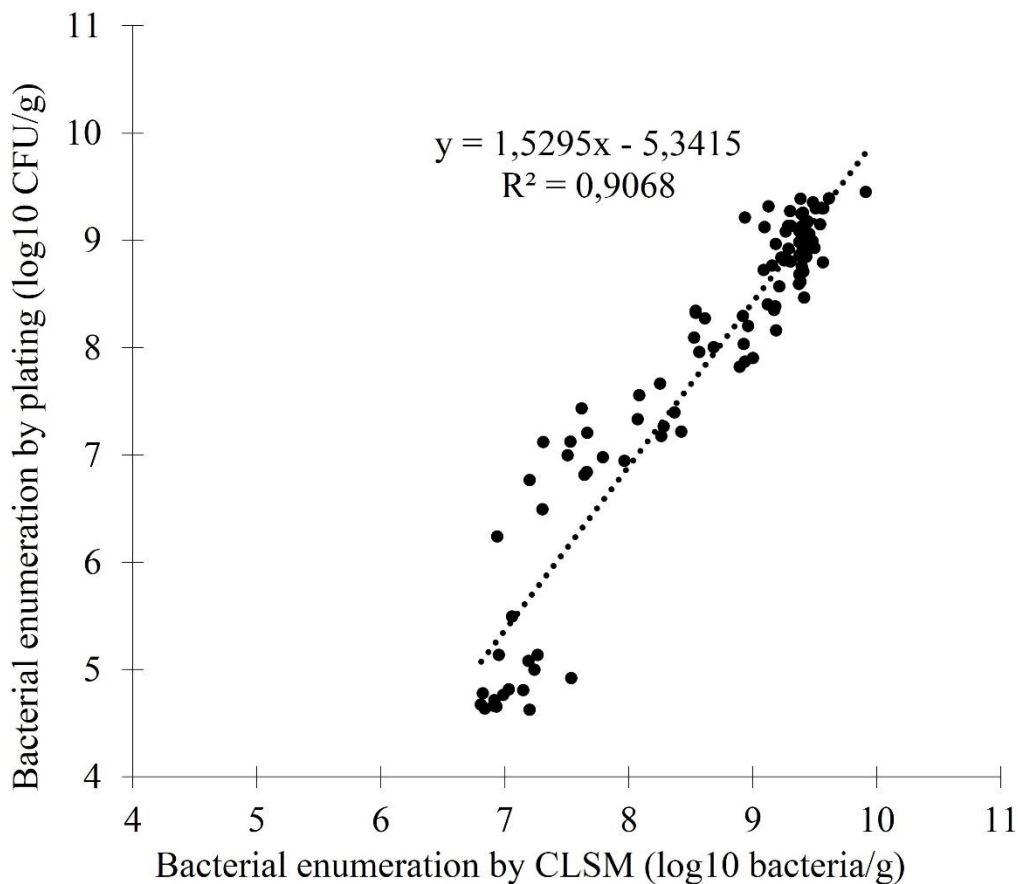
The model parameters were fitted with nls R function according to the minimization of the residual sum of square errors (RSS). Confidence intervals of fitted parameters were assessed by bootstrap using nlsBoot function from nlsMicrobio R package [41].

220

221 Results

222 Correlation between enumeration by CLSM and plate counting

223 Model cosmetic matrices were formulated with different concentrations of chlorphenesin or
224 benzyl alcohol. Bacterial enumeration of *S. aureus* was achieved at several contact times
225 (between 10 min and 4h) by both plate counting (log CFU/g) and CLSM enumeration (log
226 bacteria/g). Fig 1 gives the relationship between both techniques. The relationship between both
227 techniques is linear ($y = 1.530x - 5.342$; $R^2 = 0.907$) for a level of population over the detection
228 threshold of the technique (6.10^6 bacteria/g).



229

230 **Fig 1. Correlation between bacterial enumeration by plating (log₁₀ CFU/g) and bacterial**
231 **enumeration by CLSM imaging (log₁₀ bacteria/g).**

Inactivation of *S. aureus* according to the concentration of the preservative

Fig 2 shows the kinetics of bacterial reduction obtained by CLSM during four hours for seven different concentrations of chlorphenesin (Fig 2A) or benzyl alcohol (Fig 2B). According to the correlation between enumeration by CLSM and plate counting (Fig 1), we only took into account data in the range of population above $6 \cdot 10^6$ (maximum $2.5 \log_{10}$ reductions). The higher the concentration of preservative the higher the slope of inactivation and the lower the Dc. For chlorphenesin, 0.3% is the smallest concentration for which Dc is measurable ($17.89 \text{ h} \pm 1.12$) on a CLSM kinetics (maximum 17h). For the range between 0.40 and 0.50%, Dc varies between $10.05 \text{ h} \pm 0.44$ and $3.55 \text{ h} \pm 1.04$. For the range between 0.55 to 0.60%, Dc varies between $1.48 \text{ h} \pm 0.10$ and $0.45 \text{ h} \pm 0.07$. For benzyl alcohol, Dc for the smallest concentration 1% is $28.09 \text{ h} \pm 7.50$. From 1.5% Dc increasingly decreases to reach $1.07 \text{ h} \pm 0.05$ at 1.85 %. To obtain similar log reductions of *S. aureus*, the concentrations of benzyl alcohol should be higher than those of chlorphenesin. For example, we obtained one log reduction in $0.45 \text{ h} \pm 0.07$ with 0.6% chlorphenesin whereas $1.07 \text{ h} \pm 0.05$ is necessary with 1.85% benzyl alcohol. Fig 2C illustrates the loss of fluorescence of *S. aureus* in a model matrix with 0.3% and 0.6% chlorphenesin over time. With 0.3% chlorphenesin, the number of fluorescent bacteria decreases very slowly over time. At 4 h, the slight decrease of fluorescent bacteria number corresponds to a bacterial reduction of about 0.2 log bacteria/g (Fig 2A). In contrast, with 0.6% chlorphenesin, the fluorescent bacteria number decreased rapidly in 1h which corresponds to a reduction of $5 \cdot 10^2$ bacteria/g (Fig 2A). After 2h, no bacteria were visible anymore.

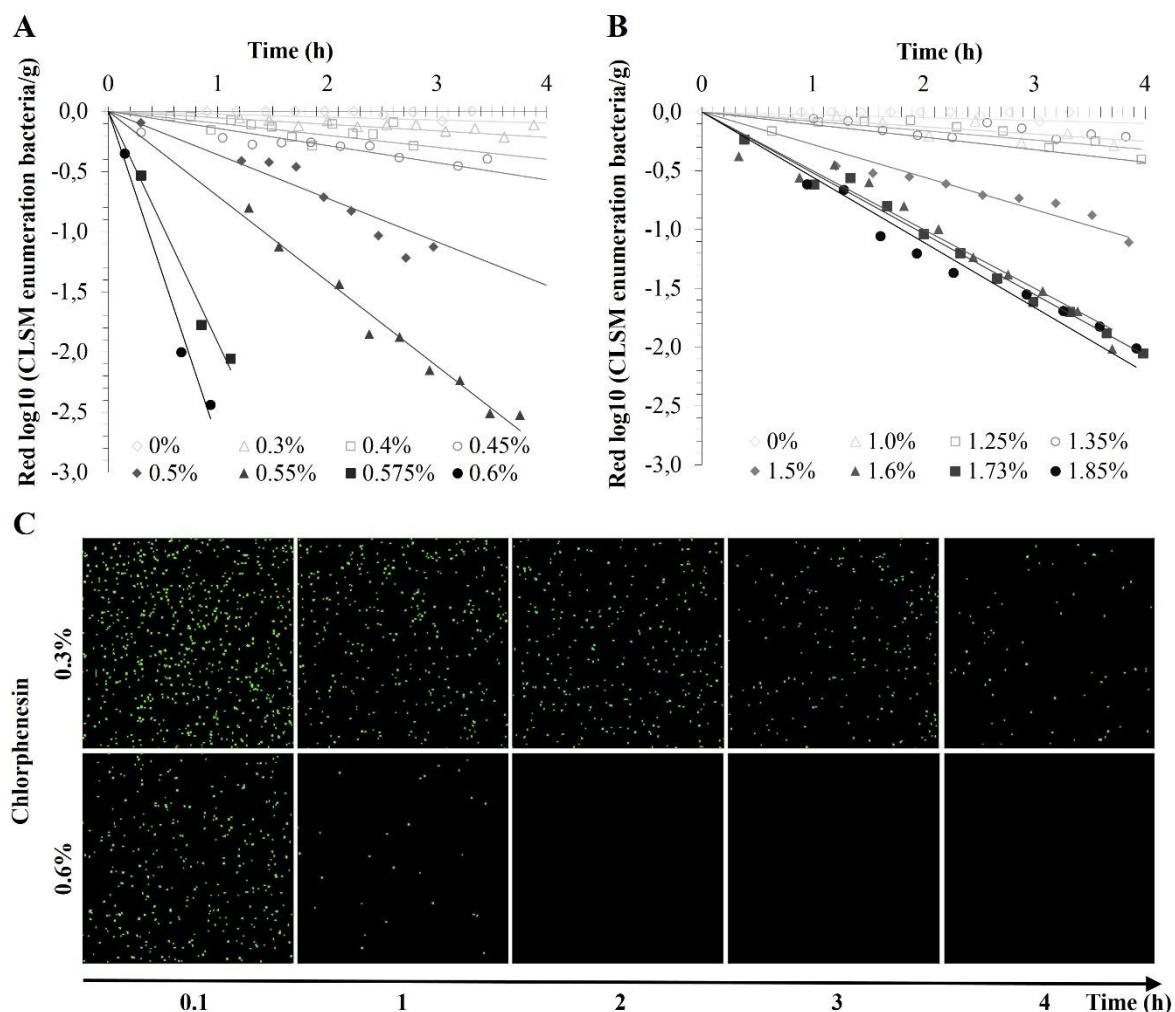


Fig 2. *S. aureus* inactivation kinetics obtained by HCS-CLSM in cosmetic model matrices with several concentrations of chlorphenesin (A) and benzyl alcohol (B). Example of the loss of bacterial fluorescence assessed by HCS-CLSM over time for two concentrations of chlorphenesin (C).

Estimation of Dc value according to the preservative concentration

Semi-log models were used to fit datasets of Dc values upon the preservative concentration. The shape parameter n was set at 1 in model #1 (linear-model, Fig 3 A and C) and set at 2 in model#2 (second-degree model, Fig 3 B and D). Model parameters of the two models, Dc* and z_c , are given in Table 1 together with the RSS and BIC for both preservatives. Second-degree

model allows the lowest BIC for both preservatives, meaning that the shape parameter is significant. Accordingly to the BIC, model #1 does not fit well and was not used for the following prediction.

Table 1. Estimated parameters (and their 95% CI intervals) and performance criteria of both secondary models.

	Chlorphenesin		Benzyl alcohol	
	<i>model#1</i>	<i>model#2</i>	<i>model#1</i>	<i>model#2</i>
n	1	2	1	2
Number of data	21	21	14	14
C^*	0.25	0.25	0.95	0.95
$\log(Dc^*)$	1.96 [1.76-2.16]	1.54 [1.47 – 1.62]	1.98 [1.70 – 2.22]	1.65 [1.44 – 1.78]
z_c	0.18 [0.15-0.21]	0.27 [0.26 – 0.28]	0.51 [0.41 – 0.65]	0.71 [0.64 – 0.79]
RSS	0.76	0.20	0.72	0.60
BIC	-63.49	-91.48	-36.22	-38.86

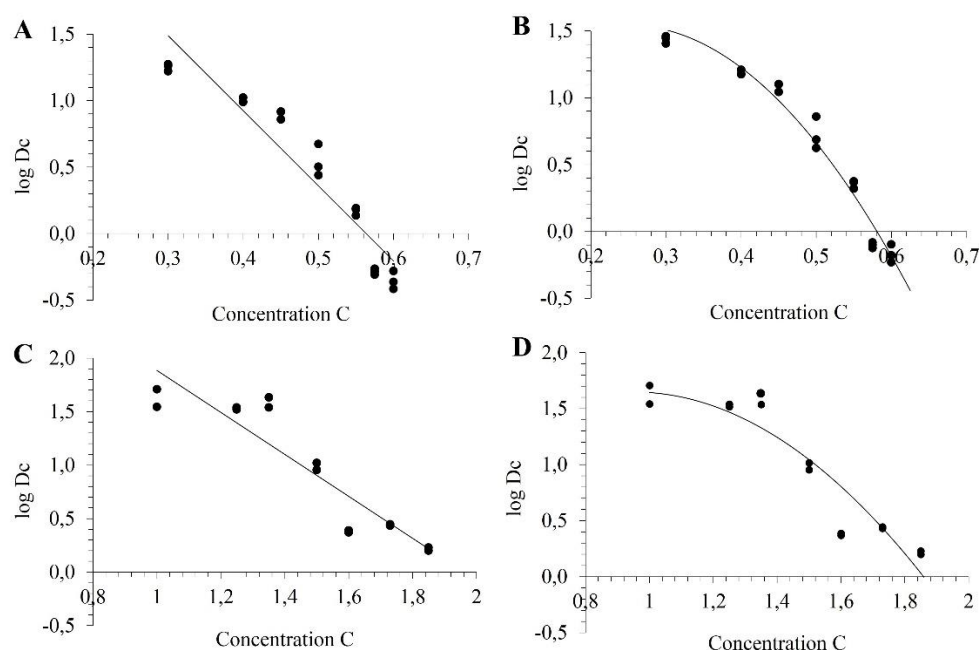


Fig 3. Relation between the Dc value and the concentration of chlorphenesin (A, B) and benzyl alcohol (C, D) by fitting of model#1 (A and C) and model#2 (B and D).

274

275 **Prediction of bacterial log reduction on long periods**

276 Dc values at specific concentrations were first estimated from model #2. The logarithm
277 reduction of the bacterial population was then calculated for specific times (from 1 to 7 days)
278 using the Bigelow linear-model (equation 1) or the IQ model (equation 4). Fig 4 presents the
279 relationships between predicted and experimental bacterial reductions. For both preservatives,
280 the best combination is obtained when using IQ model for log-reduction estimation. For
281 chlorphenesin (Fig 4A), the IQ model prediction for log-reduction datasets is far better than the
282 linear model. The Q coefficient could be optimized at 0.0141 (CI 95% 0.0124 – 0.0156) and
283 the slope of regression curve is 1.12 ($R^2=0.906$). Linear model is less relevant with a slope
284 around 0.55 and lower R^2 . For benzyl alcohol (Fig 4 B), predictions with both models are less
285 different than for chlorphenesin. The Q coefficient is optimized at 0.0043 (CI 95% 0.0022 –
286 0.0076) and the slope of the regression curve with the IQ model is 0.93 ($R^2=0.796$). For both
287 preservatives, one could note that the prediction is relevant only for maximum 5 log-reductions
288 because of the initial level of contamination and the experimental protocol used to obtain the
289 observed datasets. Fig 5 shows the prediction of the evolution of bacterial enumerations over
290 seven days for 4 tested concentrations of chlorphenesin (Fig 5A) and benzyl alcohol (Fig 5B)
291 with Bigelow linear-model (dotted lines) or IQ model (plain lines). These curves could be
292 generated for any concentrations in the range of 0.3 to 0.6 % of chlorphenesin and 1 to 1.9 %
293 for alcohol benzyl from model #2 and IQ model with respective Q to 0.014 for chlorphenesin
294 and 0.003 for benzyl alcohol.

295

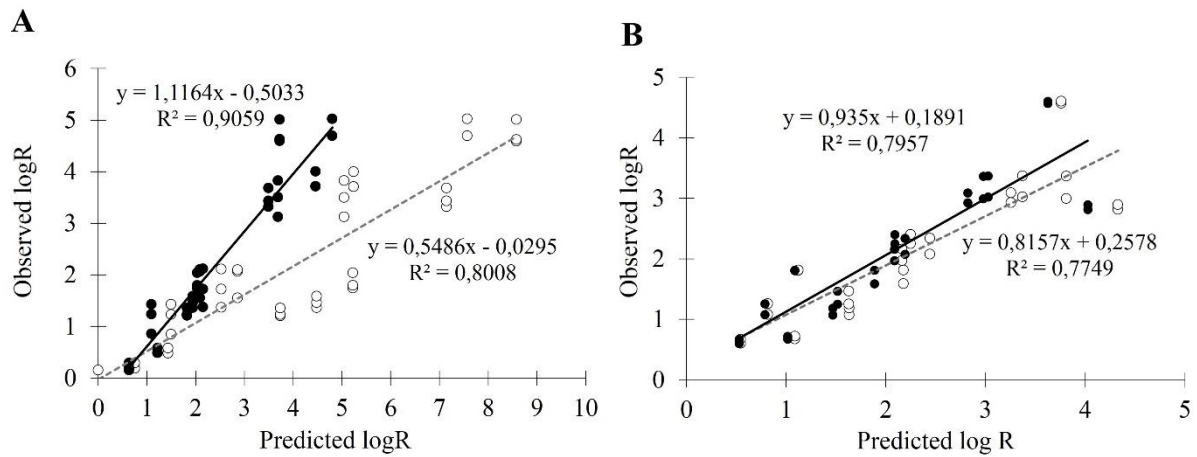


Fig 4. Correlation between the observed bacterial log-reductions and the predicted ones using Bigelow linear-model (white dots) or IQ model (black dots) for chlorphenesin (A) and benzyl alcohol (B).

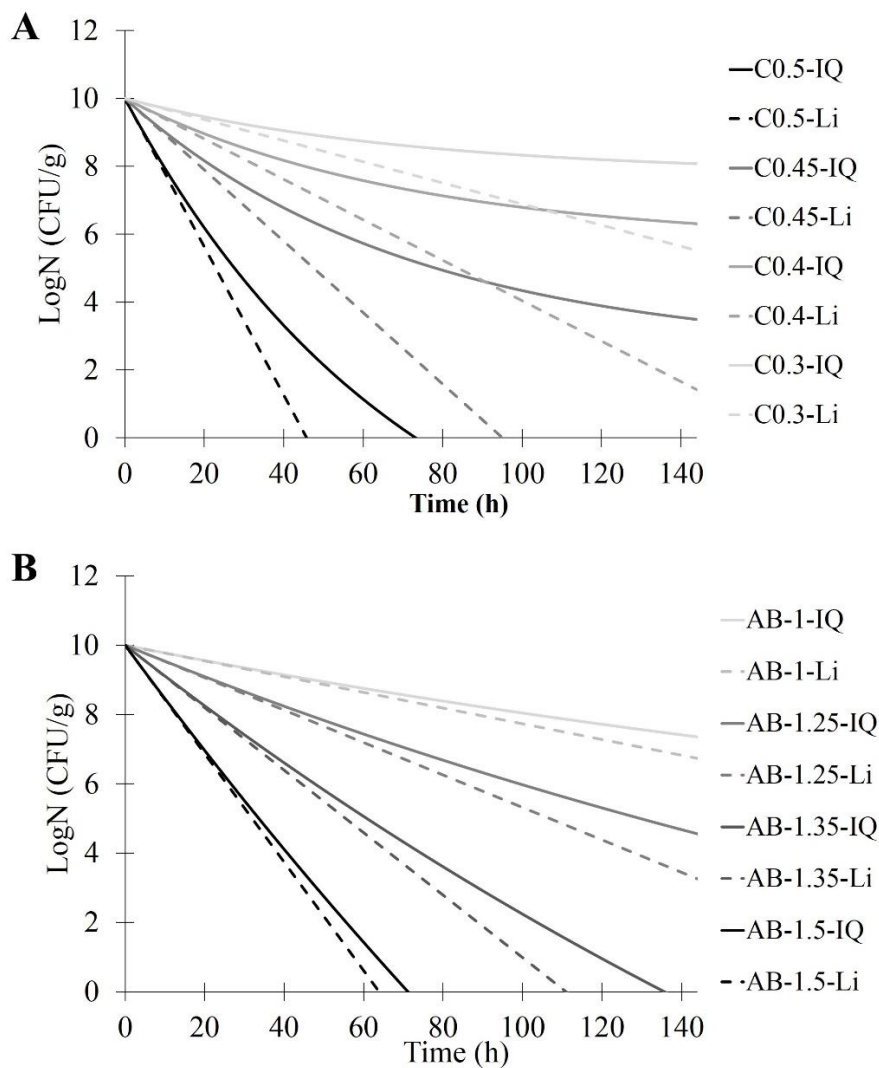


Fig 5. Illustration of the possible prediction of the evolution of the bacterial population over seven days for four concentrations of chlorphenesin (A) or benzyl alcohol (B) with Bigelow linear-model (dotted lines) or IQ model (plain lines).

Discussion

Challenge-tests are necessary to assess the efficiency of preservation in cosmetic products. Nevertheless, the procedure of the challenge-test is time-consuming due to the numerous enumerations by plate-counting necessary and to the results that are available only 48h after the last assessment point (day 28). By consequence, the challenge-test method lacks reactivity and flexibility for optimizing the preservation of a formula. In this study, we propose a new alternative method allowing the prediction of the log reduction of a bacterial population in long-term preservation by acquiring data on short-time periods. This method relies on the acquisition of CLSM kinetics of bacterial inactivation in the presence of several concentrations of preservatives during short times: acquisition during 4h is generally enough to evaluate D_c but 13h could be necessary for very low concentrations. Bacteria are first stained with a viability fluorescent marker, calcein-AM. This marker is widely used to assess bacterial viability by CLSM or by flow cytometry [31, 42]. Its precursor diffuses passively into the cytoplasm, where it is cleaved by intracellular esterases into green-fluorescent calcein [43]. This non-permeant fluorescent dye is released out of the cell when the membrane is permeabilized (dead cell). We have first shown that the enumeration obtained by CSLM and dedicated image analysis can be correlated to bacterial plate-counting during the action (10 min to 4h) of the preservative (chlorphenesin or benzyl alcohol at specific concentrations). Our detection threshold by CLSM imaging is 6.10^6 bacteria/g which is lower than the one obtained by CLSM by Auty *et al.* [29]

(1.10^8 bacteria/ml). This is probably due to the observed surface which was enlarged to a mosaic of 100 CLSM images. Here, bacterial CLSM enumeration is always higher (between 0.5 and 1 log) than the enumeration by plate counting. Auty *et al.* [29] also compared enumeration by CLSM and plate counting before they assess the viability of human probiotic strains in dairy products. They used Live/Dead BacLight marker and also underlined an overestimation of the CLSM enumeration of about 1 log. They suggested that this might be due to the bacterial clumping on plates. Indeed, the accuracy of enumeration by plate counting is usually estimated in the range of 0.3 to 0.7 log [23, 44]. However, we can notice that the difference between both techniques increases when enumeration decreases. Lower enumerations correspond to bacterial populations that remain alive after action of the preservative. Among this persistent population, a high fraction of bacteria is under stress which could explain why this fraction could not have the ability to recover and grow on agar plates while it is still stained by the viability marker by CLSM [24].

We used CLSM enumeration technique to follow the action of two preservatives at different concentrations in model cosmetic matrices. The bacterial inactivation kinetics was assessed by acquiring the calcein-AM loss of fluorescence over a few hours. These acquisitions were only possible thanks to the HCS-module of the CLSM. The automated high content screening (HCS) system is an emerging software solution that allows a CLSM to acquire automatically high content images for analysis of numerous samples, thanks to an automatically xyz-positioning in multiple wells as a function of time [45]. Automatic movements from well to well allow to acquire images for the same sample over time while avoiding photobleaching by enlightening each well only once. Moreover, we can also investigate several preservative concentrations over the same time lapse.

The CLSM method used during this study is very well suited to evaluate the efficiency of preservatives that cause membrane permeabilization. Chlorphenesin is a phenol ether with a

349 chlorine atom and it belongs to the class of organo-halogen organic compounds. Phenols disrupt
350 the cytoplasmic membrane and induce leakage of potassium ions of the cytosol. Their
351 halogenation is known to improve their antibacterial activity [46]. Benzyl alcohol is an organic
352 aromatic alcohol. Alcohols are known to damage cell membranes and denature bacterial
353 proteins that are essential to the cell metabolism which leads to the cell lysis [46].

354 Chlorphenesin seems to be more effective than benzyl alcohol against *S. aureus* in the model
355 cosmetic matrix. We observed that obtaining the same logarithm reduction needs lower
356 concentrations of chlorphenesin than benzyl alcohol. According to the literature, the partition
357 coefficient (logP) can be a parameter influencing bacterial inactivation [47]. The higher the
358 logP the higher the antibacterial activity. Chlorphenesin could have a better ability than benzyl
359 alcohol to intercalate into the bacterial membrane of *S. aureus* because its logP is higher (1.713)
360 than the one of benzyl alcohol (1.100) [48, 49].

361 In this study, we were able to predict the number of log reduction at any time for one
362 preservative at any concentration in a specific range from inactivation datasets obtained over
363 short-term times. We fitted the datasets with two models describing the effect of the
364 concentration on the log reduction time. These models derived from Mafart models [40] can
365 take several forms by setting the shape parameter at 1 (linear model#1) or 2 (second-degree
366 model#2). Mafart *et al.* (2001) compared these two first semi-log models for describing the
367 effect of pH on the heat resistance of spores (reduction time D_T) and showed that second-degree
368 model presents a better safety than the linear one. From our side, we used the BIC calculation
369 to choose the most relevant model while adjusting the minimum number of parameters. BICs
370 of models#2 are better than model#1 for both chlorphenesin and benzyl alcohol (Table 1). This
371 indicates that the preservative concentration and the contact time do not have a similar impact
372 on the reduction time. As noticed by Mafart *et al.* (2001) for the effect of pH on the resistance
373 of spores, we can hypothesize that the relationship between D_c and the preservative

concentration is more complex than that of the effect of temperature on heat resistance. Hence, the linear model was discarded from the following prediction.

The next step was to predict, from the D_c values calculated with model#2, the log reduction of the bacterial population on longer times (up to seven days) using two primary models, the Bigelow linear model and the IQ model. The Q coefficient is the characteristic parameter of the IQ model which indicates the level of quenching of the preservative in the matrix (Lambert et al 2000). Below 0.005, which appears to be the case of benzyl alcohol, the level of quenching is very low and the inactivation curves are quite similar to linear log-survivor curves. On the contrary, the Q coefficient for chlorphenesin is 0.014 which indicates a quenching of the preservative in the matrix. The level of quenching increases over time as it is demonstrated by the comparison between the predictions of linear-model and IQ models (Fig 5). As it is not similar for both preservatives, we can hypothesize that it is influenced by the interactions between the antimicrobial and the matrix. As the model cosmetic matrix used here is an emulsion, we can hypothesize that chlorphenesin which has a higher log P (1.713) than benzyl alcohol (1.100) could progressively partition into the hydrophobic droplets, thus losing its preservative efficiency. Pernin et al. (2019) studied the antimicrobial activity of two natural phenolic compounds, ferulic acid and eugenol, against *Listeria monocytogenes* in a model oil-in-water emulsion. They showed that eugenol, which has the highest logP, loses its antibacterial efficacy in emulsified systems, in contrast of ferulic acid. The authors suggest that once in the emulsion, the more hydrophobic antimicrobial agent would preferentially partition in the lipid droplets and thus the remaining concentration in the aqueous phase would not be able to inhibit microorganisms [50]. Polarity, antimicrobial charge, and environmental conditions such as temperature, ionic strength, and pH can also play a major role in the effectiveness of an antimicrobial [51]. Electrostatic and hydrophobic interactions between antimicrobials and the matrix constituents, such as lipids, proteins and charged polysaccharides, could interfere with

the antimicrobial activity [51]. For example, the addition of bovine meat proteins decreases the antimicrobial activity of phenolic compounds [52, 53]. Some gelling agents, such as hydroxypropylmethylcellulose, may be associated with the loss of effectiveness of preservatives [54]. Emulsifiers could also participate in the reduction of antimicrobial activity by sequestering antimicrobial molecules in micelles [50, 55, 56].

Nevertheless, from the estimations of D_c with model#2 and then of the log-survivors from IQ model, we propose here a method of prediction of the efficiency of two preservatives. The log-reduction of *S. aureus* population could be estimated at any concentration and after any time in a period of a few days for both tested preservatives.

This prediction is matrix- and preservative- dependent. The Q parameter is a characteristic of the interactions between them. This method should be challenged for many other couples of preservatives and matrices before it can be used for industrial prediction purposes. Moreover, some other microorganisms should be tested besides *S. aureus*, i.e. environmental strains isolated from contaminated cosmetic products. Calcein-AM is relevant for many bacteria including some Gram negative ones such as *Salmonella* [57]. However, it doesn't work for some species including *Escherichia coli* [58]. Indeed, some strains, such as *Pseudomonas aeruginosa*, have efflux pumps that release the fluorescence outside the alive bacteria and prevent cell visualization [59]. To limit these pump interferences, it was suggested to add sodium azide in the staining solution [59], as used for the observation of biofilms [32]. Unfortunately, we cannot add this molecule in cosmetic matrices because it could modify the structure and composition of the formula. Hence, other impermeant fluorescent dyes should be evaluated.

Conclusions

In this paper, we propose a rapid HCS-CLSM method associated with modeling to predict the

preservative efficacy in a cosmetic matrix. This method could provide a quick evaluation of preservative efficiency and save a lot of time by replacing many microbiological analyses. It could be beneficially used for screening preservatives or for optimizing the formulation of a cosmetic product. Nevertheless this model has to be challenged in the future and adapted for several bacterial species, preservatives and matrices.

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References

1. Agence nationale de sécurité du médicament et des produits de santé. [30 january 2020]. Available from: www.ansm.sante.fr/S-informer/Informations-de-securite-Retraits-de-lots-et-de-produits.
2. Lundov MD, Zachariae C. Recalls of microbiologically contaminated cosmetics in EU from 2005 to May 2008. *International journal of cosmetic science*. 2008;30(6):471-4.
3. SGS. [30 january 2020]. Available from: www.sgs.com/en/consumer-goods-retail/product-recalls?p=30&d=4294967287+4294967286&id=16FF624085B0&dc=http&lb=
4. Herrera AG. Microbiological analysis of cosmetics. *Methods in molecular biology* (Clifton, NJ). 2004;268:293-5. Epub 2004/05/25. doi: 10.1385/1-59259-766-1:293. PubMed

447 PMID: 15156038.

448 5. Feuilleley MO, N. Évaluation des produits cosmétiques - La sécurité. Chartres, 2018.

449 6. Halla N, Fernandes IP, Heleno SA, Costa P, Boucherit-Otmani Z, Boucherit K, et al.
450 Cosmetics preservation: a review on present strategies. *Molecules*. 2018;23(7):1571. doi:
451 10.3390/molecules23071571. PubMed PMID: 29958439.

452 7. Lundov MD, Moesby L, Zachariae C, Johansen JD. Contamination versus preservation
453 of cosmetics: a review on legislation, usage, infections, and contact allergy. *Contact Dermatitis*.
454 2009;60(2):70-8. doi: 10.1111/j.1600-0536.2008.01501.x.

455 8. Hugbo P, Onyekweli A, Igwe I. Microbial contamination and preservative capacity of
456 some brands of cosmetic creams. *Tropical Journal of Pharmaceutical Research*. 2005;2. doi:
457 10.4314/tjpr.v2i2.14604.

458 9. Halla N, Fernandes IP, Heleno SA, Costa P, Boucherit-Otmani Z, Boucherit K, et al.
459 Cosmetics Preservation: A Review on Present Strategies. *Molecules*. 2018;23(7). Epub
460 2018/07/01. doi: 10.3390/molecules23071571. PubMed PMID: 29958439; PubMed Central
461 PMCID: PMC6099538.

462 10. Birteksoz Tan AS, Tuysuz M, Otuk G. Investigation of preservative efficacy and
463 microbiological content of some cosmetics found on the market. *Pakistan journal of*
464 *pharmaceutical sciences*. 2013;26(1):153-7. Epub 2012/12/25. PubMed PMID: 23261741.

465 11. Tran TT, Hurley FJ, Shurbaji M, Koopman LB. Adequacy of cosmetic preservation:
466 chemical analysis, microbial challenge and in-use testing. *International Journal of Cosmetic*
467 *Science*. 1994;16(2):61-76. doi: 10.1111/j.1467-2494.1994.tb00084.x.

468 12. Campana R, Scesa C, Patrone V, Vittoria E, Baffone W. Microbiological study of
469 cosmetic products during their use by consumers: health risk and efficacy of preservative
470 systems. *Letters in Applied Microbiology*. 2006;43(3):301-6. doi: 10.1111/j.1472-
471 765X.2006.01952.x.

- 472 13. Ryu S, Song P, Seo C, Cheong H, Park Y. Colonization and infection of the skin by
473 *S. aureus*: immune system evasion and the response to cationic antimicrobial peptides.
474 International Journal of Molecular Sciences. 2014;15(5):8753. PubMed PMID:
475 doi:10.3390/ijms15058753.
- 476 14. Findley K, Grice EA. The skin microbiome: a focus on pathogens and their association
477 with skin disease. PLoS Pathog. 2014;10(10):e1004436-e. doi: 10.1371/journal.ppat.1004436.
478 PubMed PMID: 25393405.
- 479 15. Lowy FD. *Staphylococcus aureus* infections. New England Journal of Medicine.
480 1998;339(8):520-32. doi: 10.1056/nejm199808203390806. PubMed PMID: 9709046.
- 481 16. Tong SYC, Davis JS, Eichenberger E, Holland TL, Fowler VG. *Staphylococcus aureus*
482 infections: epidemiology, pathophysiology, clinical manifestations, and management. Clinical
483 Microbiology Reviews. 2015;28(3):603.
- 484 17. Bukowski M, Wladyka B, Dubin G. Exfoliative Toxins of *Staphylococcus aureus*.
485 Toxins. 2010;2(5):1148-65. PubMed PMID: doi:10.3390/toxins2051148.
- 486 18. Berthele H, Sella O, Lavarde M, Mielcarek C, Pense-Lheritier AM, Pirnay S.
487 Determination of the influence of factors (ethanol, pH and a(w)) on the preservation of
488 cosmetics using experimental design. Int J Cosmet Sci. 2014;36(1):54-61. Epub 2013/10/15.
489 doi: 10.1111/ics.12094. PubMed PMID: 24117694.
- 490 19. Johnson W, Bergfeld W, Belsito D, Hill R, Klaassen C, Liebler D, et al. Safety
491 assessment of chlorphenesin as used in cosmetics. International journal of toxicology.
492 2014;33:5S-15S. doi: 10.1177/1091581814526893.
- 493 20. Final report on the safety assessment of benzyl alcohol, benzoic acid, and sodium
494 benzoate. International Journal of Toxicology. 2001;20(3_suppl):23-50. doi:
495 10.1080/10915810152630729.
- 496 21. Russell AD. Challenge testing: principles and practice. International Journal of

- 497 Cosmetic Science. 2003;25(3):147-53. doi: 10.1046/j.1467-2494.2003.00179.x.
- 498 22. Hedges AJ. Estimating the precision of serial dilutions and viable bacterial counts.
499 International Journal of Food Microbiology. 2002;76(3):207-14. doi:
500 [https://doi.org/10.1016/S0168-1605\(02\)00022-3](https://doi.org/10.1016/S0168-1605(02)00022-3).
- 501 23. Augustin J-C, Carlier V. Lessons from the organization of a proficiency testing program
502 in food microbiology by interlaboratory comparison: Analytical methods in use, impact of
503 methods on bacterial counts and measurement uncertainty of bacterial counts. Food
504 Microbiology. 2006;23(1):1-38. doi: <https://doi.org/10.1016/j.fm.2005.01.010>.
- 505 24. Stephens PJ, Mackey BM. Recovery of stressed microorganisms. In: Corry JEL, Curtis
506 GDW, Baird RM, editors. Progress in Industrial Microbiology. 37: Elsevier; 2003. p. 25-48.
- 507 25. Johnston MD, Lambert RJW, Hanlon GW, Denyer SP. A rapid method for assessing the
508 suitability of quenching agents for individual biocides as well as combinations. Journal of
509 Applied Microbiology. 2002;92(4):784-9. doi: 10.1046/j.1365-2672.2002.01584.x.
- 510 26. Bridier A, Dubois-Brissonnet F, Boubetra A, Thomas V, Briandet R. The biofilm
511 architecture of sixty opportunistic pathogens deciphered using a high throughput CLSM
512 method. Journal of Microbiological Methods. 2010;82(1):64-70. doi:
513 <https://doi.org/10.1016/j.mimet.2010.04.006>.
- 514 27. Dorobantu LS, Yeung AKC, Foght JM, Gray MR. Stabilization of oil-water emulsions
515 by hydrophobic bacteria. Appl Environ Microbiol. 2004;70(10):6333-6. doi:
516 10.1128/AEM.70.10.6333-6336.2004. PubMed PMID: 15466587.
- 517 28. Doherty SB, Gee VL, Ross RP, Stanton C, Fitzgerald GF, Brodkorb A. Efficacy of whey
518 protein gel networks as potential viability-enhancing scaffolds for cell immobilization of
519 *Lactobacillus rhamnosus* GG. J Microbiol Methods. 2010;80(3):231-41. Epub 2010/01/05. doi:
520 10.1016/j.mimet.2009.12.009. PubMed PMID: 20045713.
- 521 29. Auty MA, Gardiner GE, McBrearty SJ, O'Sullivan EO, Mulvihill DM, Collins JK, et al.

522 Direct in situ viability assessment of bacteria in probiotic dairy products using viability staining
 523 in conjunction with confocal scanning laser microscopy. Appl Environ Microbiol.
 524 2001;67(1):420-5. doi: 10.1128/AEM.67.1.420-425.2001. PubMed PMID: 11133474.

525 30. Jeanson S, Chadoeuf J, Madec MN, Aly S, Floury J, Brocklehurst TF, et al. Spatial
 526 distribution of bacterial colonies in a model cheese. Appl Environ Microbiol. 2011;77(4):1493-
 527 500. Epub 2010/12/21. doi: 10.1128/aem.02233-10. PubMed PMID: 21169438; PubMed
 528 Central PMCID: PMCPmc3067236.

529 31. Davison WM, Pitts B, Stewart PS. Spatial and temporal patterns of biocide action
 530 against *Staphylococcus epidermidis* biofilms. Antimicrob Agents Chemother.
 531 2010;54(7):2920-7. Epub 05/10. doi: 10.1128/AAC.01734-09. PubMed PMID: 20457816.

532 32. Bridier A, Dubois-Brissonnet F, Greub G, Thomas V, Briandet R. Dynamics of the
 533 action of biocides in *Pseudomonas aeruginosa* biofilms. Antimicrob Agents Chemother.
 534 2011;55(6):2648-54. Epub 2011/03/23. doi: 10.1128/aac.01760-10. PubMed PMID: 21422224;
 535 PubMed Central PMCID: PMCPmc3101418.

536 33. Karampoula F, Giaouris E, Deschamps J, Doulgeraki AI, Nychas G-JE, Dubois-
 537 Brissonnet F. Hydrosol of *Thymra capitata* is a highly efficient biocide against *Salmonella*
 538 *enterica* serovar Typhimurium biofilms. Appl Environ Microbiol. 2016;82(17):5309-19. doi:
 539 10.1128/AEM.01351-16. PubMed PMID: 27342550.

540 34. Takenaka S, Trivedi HM, Corbin A, Pitts B, Stewart PS. Direct visualization of spatial
 541 and temporal patterns of antimicrobial action within model oral biofilms. Appl Environ
 542 Microbiol. 2008;74(6):1869-75.

543 35. Chen C-Y, Nace GW, Irwin PL. A 6×6 drop plate method for simultaneous colony
 544 counting and MPN enumeration of *Campylobacter jejuni*, *Listeria monocytogenes*, and
 545 *Escherichia coli*. Journal of Microbiological Methods. 2003;55(2):475-9. doi:
 546 [https://doi.org/10.1016/S0167-7012\(03\)00194-5](https://doi.org/10.1016/S0167-7012(03)00194-5).

547 36. Schneider CA, Rasband WS, Eliceiri KW. NIH image to imageJ: 25 years of image
548 analysis. *Nature Methods*. 2012;9:671. doi: 10.1038/nmeth.2089.

549 37. Bigelow WD, Esty JR. The thermal death point in relation to time of typical
550 thermophilic organisms. *The Journal of Infectious Diseases*. 1920;27(6):602-17.

551 38. Lambert R, Johnston M. Disinfection kinetics: a new hypothesis and model for the
552 tailing of log survivor/time curves. *Journal of Applied Microbiology*. 2000;88(5):907-13.

553 39. Geeraerd AH, Valdramidis VP, Van Impe JF. GInaFit, a freeware tool to assess non-
554 log-linear microbial survivor curves. *International Journal of Food Microbiology*.
555 2005;102(1):95-105. doi: <https://doi.org/10.1016/j.ijfoodmicro.2004.11.038>.

556 40. Mafart P, Couvert O, Leguérinel I. Effect of pH on the heat resistance of spores:
557 Comparison of two models. *International Journal of Food Microbiology*. 2001;63(1):51-6. doi:
558 [https://doi.org/10.1016/S0168-1605\(00\)00397-4](https://doi.org/10.1016/S0168-1605(00)00397-4).

559 41. Baty F, Ritz C, Charles S, Brutsche M, Flandrois J-P, Delignette-Muller M. A toolbox
560 for nonlinear regression in R : the package nlstools. *Journal of Statistical Software*. 2015;66:1-
561 21. doi: 10.18637/jss.v066.i05.

562 42. Chitarra LG, Breeuwer P, Abee T, Bulk RW. The use of fluorescent probes to assess
563 viability of the plant pathogenic bacterium *Clavibacter michiganensis* subsp. *michiganensis* by
564 flow cytometry. *Fitopatologia Brasileira*. 2006;31:349-56.

565 43. Kaneshiro ES, Wyder MA, Wu Y-P, Cushion MT. Reliability of calcein acetoxy methyl
566 ester and ethidium homodimer or propidium iodide for viability assessment of microbes.
567 *Journal of Microbiological Methods*. 1993;17(1):1-16. doi: [https://doi.org/10.1016/S0167-](https://doi.org/10.1016/S0167-7012(93)80010-4)
568 [7012\(93\)80010-4](https://doi.org/10.1016/S0167-7012(93)80010-4).

569 44. Afssa. Avis de l'agence française de sécurité sanitaire des aliments concernant les
570 références applicables aux denrées alimentaires en tant que critères indicateurs d'hygiène des
571 procédés 2007.

- 572 45. Canette A, Deschamps J, Briandet R. High content screening confocal laser microscopy
573 (HCS-CLM) to characterize biofilm 4D structural dynamic of foodborne pathogens. In: Bridier
574 A, editor. Foodborne Bacterial Pathogens: Methods and Protocols. New York, NY: Springer
575 New York; 2019. p. 171-82.
- 576 46. Al-Adham I, Haddadin R, Collier P. Types of Microbicidal and Microbistatic Agents.
577 Russell, Hugo & Ayliffe's 2013. p. 5-70.
- 578 47. Pernin A, Dubois-Brissonnet F, Roux S, Masson M, Bosc V, Maillard M-N. Phenolic
579 compounds can delay the oxidation of polyunsaturated fatty acids and the growth of *Listeria*
580 *monocytogenes*: structure-activity relationships. Journal of the Science of Food and
581 Agriculture. 2018;98(14):5401-8. doi: 10.1002/jsfa.9082.
- 582 48. The Good Scents Compagny information system. Chlorphenesin [30 january 2020].
583 Available from: <http://www.thegoodscentscopy.com/data/rw1364911.html>.
- 584 49. The Good Scents Compagny information system. Benzyl alcohol [30 january 2020].
585 Available from: <http://www.thegoodscentscopy.com/data/rw1001652.html>.
- 586 50. Pernin A, Bosc V, Maillard M-N, Dubois-Brissonnet F. Ferulic acid and eugenol have
587 different abilities to maintain their inhibitory activity against *Listeria monocytogenes* in
588 emulsified systems. Front Microbiol. 2019;10:137-. doi: 10.3389/fmicb.2019.00137. PubMed
589 PMID: 30787916.
- 590 51. Weiss J, Loeffler M, Terjung N. The antimicrobial paradox: why preservatives lose
591 activity in foods. Current Opinion in Food Science. 2015;4:69-75. doi:
592 <https://doi.org/10.1016/j.cofs.2015.05.008>.
- 593 52. Bouarab-Chibane L, Forquet V, Clement Y, Lanteri P, Bordes C, Bouajila J, et al. Effect
594 of interactions of plant phenolics with bovine meat proteins on their antibacterial activity. Food
595 Control. 2018;90:189-98. doi: <https://doi.org/10.1016/j.foodcont.2018.03.006>.
- 596 53. Juven BJ, Kanner J, Schved F, Weisslowicz H. Factors that interact with the

antibacterial action of thyme essential oil and its active constituents. The Journal of applied bacteriology. 1994;76(6):626-31. Epub 1994/06/01. doi: 10.1111/j.1365-2672.1994.tb01661.x. PubMed PMID: 8027009.

54. Scholtyssek R. Protection of cosmetics and toiletries. In: Paulus W, editor. Directory of Microbicides for the Protection of Materials: A Handbook. Dordrecht: Springer Netherlands; 2005. p. 263-6.

55. Kirk–Othmer. Chemical Technology of Cosmetics. New Jersey: Wiley; 2013. 832 p.

56. Shimamoto T, Mima H. A model for the evaluation and prediction of preservative activity in oil-in-water emulsions. Chemical and Pharmaceutical Bulletin. 1979;27(11):2743-

57. Rawlinson LAB, Ryan SM, Mantovani G, Syrett JA, Haddleton DM, Brayden DJ. Antibacterial effects of Poly(2-(dimethylamino ethyl)methacrylate) against selected Gram-positive and Gram-negative bacteria. Biomacromolecules. 2010;11(2):443-53. doi: 10.1021/bm901166y. PubMed PMID: WOS:000274215000018.

58. Comas-Riu J, Vives-Rego J. Use of calcein and SYTO-13 to assess cell cycle phases and osmotic shock effects on *E. coli* and *Staphylococcus aureus* by flow cytometry. Journal of Microbiological Methods. 1999;34(3):215-21. doi: 10.1016/s0167-7012(98)00091-8. PubMed PMID: WOS:000078249500006.

59. Joux F, Lebaron P. Use of fluorescent probes to assess physiological functions of bacteria at single-cell level. Microbes and Infection. 2000;2(12):1523-35. doi: [https://doi.org/10.1016/S1286-4579\(00\)01307-1](https://doi.org/10.1016/S1286-4579(00)01307-1).