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1 Sequential fed batch extractive fermentation for enhanced bioethanol production

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using recycled Spathaspora passalidarum and mixed sugar composition.

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ABSTRACT

The simultaneous ethanol production and removal during sequential cell recycle fed 7 batch fermentation provides a complementary route to produce this biofuel from sugar 8 mixtures, which may greatly improve yields and productivity from lignocellulosic 9 hydrolysates. Spathaspora passalidarum is a wild-type strain able to naturally convert 10 glucose, fructose, xylose and arabinose into ethanol. Therefore, the present work has 11 12 focused on 2G bioethanol production by S. passalidarum aiming at the consumption of 13 all sugars released after pre-treatment and enzymatic hydrolysis of sugarcane bagasse in a single fermentation step. The fermentation strategy with sequential cell recycle, fed-14 15 batch mode and ethanol removal in situ was performed on a hemicellulosic hydrolysate medium supplemented with molasses. This strategy gave improved fermentation 16 performance and enabled the co-fermention of all sugars under microaerobic conditions. 17 The maximum ethanol yield and productivity was 0.482 g.g⁻¹ and 9.5 g.L⁻¹.h⁻¹, 18 respectively, showing a process efficiency of 94.3%. The selective ethanol removal 19 enables the operation of the bioreactor at low levels of ethanol (20-30 g.L⁻¹), even with 20

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high sugar concentration inputs, accelerating the fermentation performance and avoiding inhibitory effects on yeast metabolism. Applying the cell recycle strategy, *S. passalidarum* was able to increase its robustness, as shown by a 10-fold increase in ethanol productivity, and it was also able to tolerate a high acetic acid concentration (4.5 g.L⁻¹) during long-term fermentations. These results demonstrate that the bioprocess strategy has a strong potential to improve bioethanol production of rich mixed sugar from lignocellulosic hydrolysates in a single fermentation step.

Keywords: *Spathaspora passalidarum*, cell recycle, extractive fermentation, product
recovery, hemicellulosic hydrolysate, molasses, biofuel.

30 **1. Introduction**

Considering today's global energy situation, the need for energy security and 31 32 environmental safety has intensified the demand for an alternative and eco-friendly energy source [1]. Bioethanol is one of the most promising alternatives for renewable 33 34 fuels, since this fuel may be produced from a wide range of renewable sources (rich in 35 fermentable sugars) [2]. According to the Paris Agreement, many countries and several 36 EU member states (around 195) have already proclaimed commitments to bioethanol programs as part of efforts to reduce dependence on fossil fuel, as well as reduce 37 greenhouse gas emissions [3]. 38

Industrial bioethanol production by Brazilian biorefineries is a well establish process, using sugarcane molasses, enabling the achievements of high yields (90-95%) and productivities (10-15 g.L⁻¹.h⁻¹), with a technology known as first generation bioethanol (1G) [4]. On the other hand, second generation bioethanol (2G) production still needs process improvements. Lignocellulosic biomass appears to be the most

44 abundant and promising agro-industrial raw material around the world, which may be 45 used to produce bioethanol and others bio-based products [2]. The biomass-based bioethanol industry in Brazil utilizes sugarcane bagasse and straw, which is considered 46 the cheapest waste raw material, derived from sugarcane processing [5]. On average, 47 48 one ton of sugarcane used in sugar and bioethanol production can generate a mixture of 49 around 50/50% bagasse (250 - 270 kg) and straw (more than 200 kg). Considering the 2018/2019 harvest, Brazil produced 620.8 million tons of sugarcane, resulting in 29 50 51 million tons of sugar, 33.1 billion liters of bioethanol and around 310-380 million tons 52 of bagasse [6]. In order to generate bioethanol from sugarcane bagasse (2G), the 53 biomass essentially needs to be degraded into its individual polymers: cellulose (42-54 48%), hemicellulose (19-25%) and lignin (20-42%), using an appropriate pre-treatment method [7]. In this way, the released cellulose and hemicellulose molecules are then 55 hydrolysed into soluble sugars (by chemical or enzymatic procedures). Finally, after 56 biomass pre-treatment and hydrolyses steps have been performed, this mixed soluble 57 sugar composition (composed essentially of hexoses and pentoses) is then converted 58 59 into bioethanol using microbial fermentation strategies [8].

Simultaneous co-fermentation of mixed sugar (xylose, glucose, arabinose and 60 cellobiose), mainly glucose and xylose (two main sugars released after bagasse 61 62 hydrolyses and pre-treatment), is still problematic for most microorganisms because the presence of glucose represses the utilization of the other saccharides [9]. For most 63 microorganisms, the consumption of glucose (or other readily metabolizable carbon 64 65 sources) represses the expression of genes encoding enzymes responsible for the metabolization of other carbon sources, a phenomenon known as catabolic repression 66 67 [10]. During consumption of mixed sugars, a lot of enzymes, especially those engulfed

68 in catabolic pathways, are subject to such repressive regulation, what represents one of the major bottlenecks for 2G bioethanol fermentation processes, preventing the 69 70 achievement of high yields and productivities at the industrial scale. In order to tackle this challenge, an important technique for overcoming catabolic repression in enzyme 71 72 biosynthesis is the use of fed batch cultures. Fed batch fermentations allows the 73 concentration of sugars in the fermenting medium to be kept at low levels, the cell 74 growth restricted, and the biosynthesis of specific enzymes depressed [11,12]. 75 Remarkably, the ascomycetous, beetle-associated yeast S. passlidarum, can co-ferment 76 a wide variety of sugars simultaneously, showing high ethanol yield [9,13-15]. This 77 strain exhibits rapid D-xylose consumption and the ability to ferment glucose, 78 arabinose, fructose and cellobiose, simultaneously, but at different sugar uptake rates, allowing the possibility to consume all mixed sugar released after bagasse pre-treatment 79 for 2G bioethanol production [15]. 80

One of the other possibilities to reduce capital costs for 2G bioethanol 81 production is process integration [16]. Until this date, there isn't any report in the 82 83 literature considering process integration for bioethanol production by S. passalidarum. However, integration of 1G and 2G bioethanol production may provide solutions for 84 several challenges in standalone 2G process, showing the opportunity to convert the 85 86 whole agricultural crops into ethanol, which consequently will increase ethanol yield per hectare of cultivated land disposable for feedstocks production, such as sugarcane 87 [17]. Considering this aspect, simultaneous ethanol production and removal during 88 microbial fermentation has been studied with the aim of developing more cost-effective 89 processes for bioethanol production [18,19]. The conceptual idea is the use of fed batch 90 extractive fermentation with cell recycle, which combines mixed sugar feeding at 91

92 controlled levels, cell adaptation due to sequential recycling, fermentation processes and 93 ethanol extraction within a single bioreactor. This fermentation technology is an 94 attractive strategy due to the possibility of reducing operation costs, increasing conversion efficiencies, and reducing products and by-products inhibition. However, 95 this technology is still far from being suitable for industrial application. Despite all 96 97 these features, this study investigates the fermentation performance of a Spathaspora passalidarum strain, using fed batch extractive fermentations with cell recycle and 98 99 simultaneous ethanol removal in situ, under conditions that are relevant to the 2G bioethanol production process (high cell concentrations, mixed sugar composition, high 100 101 yields and process efficiency).

- 102 2. Materials and Methods
- 103 **2.1. Agro-industrial raw-materials**

104 2.1.1. Sugarcane molasses

Sugarcane molasses was obtained from Costa Pinto Sugar Mill (Piracicaba, SP, 105 106 Brazil), containing around 65% total reducing sugar (TRS). Because S. passalidarum is unable to produce ethanol from sucrose, the sugarcane molasses was hydrolysed 107 108 beforehand (to glucose and fructose) using a commercial Saccharomyces cerevisiae 109 baker's yeast strain. The sucrose hydrolysis was held in a rotatory shaker at 45 °C, at 200 rpm for 12 h, using 10% of inoculum size. This strategy achieved a fermenting 110 111 media with 100% of hexoses. The sugarcane molasses was used in the experiments as a strategy to dilute the hemicellulosic hydrolysate, reducing the effect of inhibitors, 112 mainly acetic acid, on yeast metabolism. Furthermore, it also works as a source of 113 114 nutrient supplementation as well as having a buffering effect, keeping the pH stable

115 during the fermentations. Sugarcane molasses has an acid pH, usually around 5, and its salt content (2 - 8%) has great buffering capacity, which contributes to stabilize pH 116 117 during fermentation [20]. The composition of hydrolysed molasses is presented on Table 1. 118

119

2.2.2. Sugarcane bagasse: hemicellulose hydrolysate

120 Diluted acid pre-treatment was carried out in order to release a high concentration of pentoses. Reactions were performed in a 350 L reactor (Pope Scientific 121 122 Inc, Saukville, WI) equipped with a stirrer and heated through an oil thermal jacket. 25 kg of dry sugarcane bagasse (Serrana Mill, São Paulo) were processed at 145°C, for 12 123 124 min, with H₂SO₄ diluted solution (0.5% w/v), and a final solid-to-liquid ratio of 1:10. At the end of this reaction time, solid-liquid separation was carried out through a filtration 125 step, in a Nutshe filter with 100 L capacity (Pope Scientific Inc). To alleviate the 126 127 inhibitory effects to microbes as well as to promote sugar concentration, a detoxification process (by evaporation) was carried out. The liquid fraction (hemicellulose 128 hydrolysate) was concentrated (from 5 to 19.5 °Brix) using an evaporation step (50 L.h⁻¹ 129 of hemicellulosic hydrolysate, 475 mbar, 80 °C) [15]. The hemicellulose hydrolysate 130 was stored at 4°C for later use in fermentation processes. Table 1 shown the 131 hemicellulosic hydrolysate composition, before and after the detoxification step been 132 133 perfomed.

- 134 **2.2.** Fermentation
- 2.2.1. Microorganism and propagation step 135

Spathaspora passalidarum NRRL Y-27907 was obtained from the ARS Culture 136 137 Collection (National Center of Agricultural Utilization Research, Peoria, IL) and was 138 used to ferment the mixed sugar composition (hexose and xylose). Stocks were kept at -80 °C in XYMP with 20% of glycerol. S. passalidarum cultures were streaked from -139 80°C stocks and transferred to XYPM media containing (g.L⁻¹): xylose (10); yeast 140 141 extract (5); malt extract (20); and monobasic sodium phosphate (2) and incubated in a 142 rotatory shaker for 24 h at 28°C and 150 rpm. A two-step propagation process was 143 standardized: the first step was used to obtain exponential cell growth with defined 144 media (XYMP), and the second one was for cell adaptation in a complex medium with a 145 carbon source from an agro-industrial raw material, in order to mimic an industrial process. These procedures were carried out to adapt yeast cells to an industrial medium. 146

After the first step had been performed, cells were centrifuged at 10.000 rpm 147 (19.762 xg) at 4 °C for 10 min in a Sorvall centrifuge, and suspended in an appropriate 148 149 volume of sterile water. The suspensions with microorganism were transferred to inoculum media containing (g.L⁻¹): yeast extract (3), malt extract (3), MgSO₄.7H₂O 150 (0.5), (NH₄)₂HPO₄ (2), KH₂PO₄ (4), xylose (20) from hemicellulosic hydrolysate and 151 glucose and fructose (20) from sugarcane molasses. The inoculum was kept in a 152 153 rotatory shaker for 24 h, 28°C and 150 rpm. After this time, cells were centrifuged again 154 (with the same conditions as described above) and re-suspended in an appropriate volume of sterile water to follow the next steps of propagation in the bioreactor. 155

156

2.2.2. Fed batch fermentation

157 A *S. passalidarum* suspension was used to perform fed batch fermentations in 158 the bioreactor. Cells were transferred to the bioreactor, (Bioflo III, 2 L, New Brunswick 159 Scientific Co., Inc., Edison, NJ) with 0.5 L of initial working volume (1.5 L). Agitation 160 was kept at 150 rpm, temperature at 28°C and aeration at 0.05 vvm. The growth media 161 had the following composition (g.L⁻¹): yeast extract (3); malt extract (3); MgSO₄.7H₂O 162 (0.5); (NH₄)₂HPO₄ (2) and KH₂PO₄ (4). The initial concentration of mixed sugar 163 composition inside the bioreactor vessel was zero and biomass was inoculated to give 164 an initial cell concentration of 10 g.L⁻¹.

The fermentation feed medium with mixed sugar composition was composed of 165 hemicellulosic hydrolysate diluted in sugarcane molasses, resulting in about 50.0 g.L⁻¹ 166 of pentoses (xylose and arabinose) and around 50.0 g.L⁻¹ of hexoses (fructose and 167 glucose). In order to avoid nutrient limitation, the feed medium was also supplemented 168 169 with a nutrient solution at concentration levels as described above. The feed rate was ranged from 30 to 60 mL.h⁻¹ to evaluate fermentation kinetics, and the feed was carried 170 171 out until the maximum reactor working volume was achieved. These values were defined based on our previous study, regarding the use of mixed sugar composition by 172 S. passalidarum during batch runs [15]. These feed rates were based on xylose 173 174 consumption rate by this strain, because this sugar is consumed slowly compared to glucose, when mixed sugar concentration is target. So, in order to avoid xylose 175 176 accumulation in fermentation medium, we set up these different flow rates to identify 177 the best fermentation performance during fed batch runs. The fermentation was extended until total sugar consumption. Samples were taken periodically for 178 measurements of cell, ethanol and sugar concentrations. The best fed batch fermentation 179 180 performance as evaluated in this step was used for the subsequent steps.

181

2.2.3. Fed batch fermentation with ethanol removal and cell recycling

In order to reduce ethanol toxicity in the fermenting medium, as well as to improve yeast activity, fed batch experiments with simultaneous ethanol removal and repeated recycle of cells were performed. For this, the cells obtained in the propagation step were inoculated (1 L) in a proportion of 1/3 (v/v) of the system work volume (3 L)
and supplemented with nutrients, similarly to previous steps. Agitation was kept at 150
rpm, temperature at 28°C, and aeration at 0.1 vvm. The reaction system is represented
in Fig. 1.

189 The composition of the fermentation feed medium was similar to that described in Table 2. The feed rate was 57.2 mL.h⁻¹, based on S. passalidarum kinetic 190 performance during fed batch fermentations as evaluated previously. The reactor 191 192 feeding was carried out until the system working volume was reached, and fermentation 193 was run until sugar depletion. At the end of fermentation, the yeast cells were recycled to a subsequent fed batch fermentation step. The cell recycle was performed using a 194 micro-filter, coupled to a bioreactor vessel, so that cells could be completely retained 195 and accumulated in the system, and the fermented broth was drained from the 196 197 bioreactor. In this way, cell recycling was promoted by continuous pumping through the micro-filter (tangential filtration) and the cell free liquid effluent (permeate) was 198 removed by a peristaltic pump coupled to the output of the filtering system (see Fig. 1). 199 200 The yeast cells were concentrated until achieving the initial inoculum condition (1/3 of 201 the working volume system). The micro-filtration system was composed of a crossflow micro-filtration type, which was designed by Millipore Corporation (Ceraflow model: 202 pore size 0.22 µm, filtration area of 0.0372 m², length of 22.8 cm). It consisted of an 203 204 external inorganic tubular filter made of alumina (high purity) and mounted inside a stainless-steel housing. 205

Ethanol removal was performed intermittently, every 6 h, aiming at a constant and low ethanol concentration inside the bioreactor vessel (lower than 40 g.L⁻¹, below toxic concentration for the yeast). Each ethanol extraction cycle was about 1 h. The

209 fermentation system was totally controlled and monitored on line, through the use of a 210 supervisory system, using the software LabVIEW 10.0. To promote ethanol removal, a 211 flash tank was coupled to the bioreactor vessel (see Fig. 1). The flash tank operated under vacuum conditions (100-150 mmHg). The extraction of ethanol from the 212 fermented medium is possible due to the volatility difference between the molecules. 213 214 So, an enriched vapour mixture of ethanol and water evaporates and passes through a 215 condenser (5°C) and the condensed alcoholic solution was collected in another vessel. 216 After each ethanol removal cycle, the fermenting medium was sent back to the bioreactor. A more detailed description regarding the fermentative ethanol extraction 217 218 system can be found in Farias et al. [15].

219 **2.3. Sterilization**

In order to minimize variations in medium composition, fermentation medium was cold sterilized separately using a sterile system of cellulose ester membrane, 0.2 mm of pore diameter (model Minikap HF Filter MK2M-512-V6S, Sprectum Laboratories, Inc., Fl, USA) and aseptically added to the system. The sterilization of the entire vacuum extractive fermentation prototype (Fig. 1) was performed in an autoclave at 121 °C for 30 min.

226

6 **2.4. Analytical methodology**

The optical cell density was determined using a spectrophotometer detector at 600 nm. Yeast concentrations were measured by gravimetrical analysis in triplicate, and 2 mL of each sample was centrifuged (13.000 rpm, 4 °C for 10 min), re-suspended twice in distilled water in order to remove soluble components and dried in an oven at 70 °C.

232 The cell free supernatant was used after dilution to quantify sugars, alcohols, 233 glycerol and organic acids using a HPLC apparatus (Varian Inc. Scientific Instruments, 234 Palo Alto, CA), equipped with a refractive index (RI) and an Aminex HPX-87H column (Bio-Rad, 300 x 7.8 mm), at 35 °C, with eluent flow rate 0.6 mL.min⁻¹ (degassed and 235 236 ultrapure water, pH adjusted to 2.6 with H₂SO₄). Furfural and hydroxymethylfurfural were separated by a Nova-Pack C18 column (Water Co., Milford, MA), with an effluent 237 flow rate of 0.8 mL.min⁻¹ (acetonitrile: water / 1:8) and detected by UV at 276 nm. The 238 column was placed in an oven at 30°C. The system was equipped with guard cartridges 239 to avoid contaminant interference in the analysis. 240

Sucrose and monomeric sugars (arabinose, fructose, glucose and xylose) were analyzed using a HPAEC-PAD DX-500 system (Dionex, USA) equipped with a CarboPac PA-1 column (0.4x25 cm; Dionex, USA). Elution was performed at 1 mL.min⁻¹ and involved an isocratic step of 1mM NaOH for 8 min followed by a linear gradient of 1-5 mM NaOH for 8 min and an isocratic step of 5 mM NaOH for 14 min. The column was cleaned between runs for 5 min with 150 mM NaOH and equilibrated for 5 min with 1 mM NaOH.

248 **2.5. Kinetic performance**

In order to access the fermentation performance of *S. passalidarum* using a mixed sugar composition, the kinetic parameters were calculated based on the fermentation strategy adopted. The ethanol yield (Y_{ps}) was calculated based upon the ratio of maximum product titer and consumed sugars, as described by Eq. 1 and Eq. 3, for fed batch and fed batch extractive fermentation, respectively. The consumption of mixed sugar was based on the total sugar concentration (glucose, xylose, fructose and

arabinose) at the beginning, minus the residual sugar remaining at the end of the fermentation process. Ethanol productivity (Q_p) was calculated considering the ratio of ethanol produced and fermentation time, as described by Eq. 2 and Eq. 4. The list of symbol is provided as supplementary material.

259 In fed-batch fermentation:

$$260 Y_{ps} = \frac{P_f - P_i}{S_a - S_f} (1)$$

$$261 \qquad Q_p = \frac{P_f - P_i}{t_f} \tag{2}$$

262 In fed-batch extractive fermentation:

263
$$Y_{ps} = \frac{\Delta P_{FBE}}{\Delta S_{BAE}}$$
(3)

264
$$Q_p = \frac{\Delta P_{FBE}}{t_f}$$
(4)

265 Where:

266
$$\Delta P_{FBE} = \left[\left(P_f - P_i \right) V_r \right]_{bioreactor} + \left[\left(P_{extracted} \right) V_{cond} \right]_{flash}$$
(5)

267
$$\Delta S_{FBE} = \left[\left(S_f - S_i \right) V_r \right]_{bioreactor} + \left[F_a * S_a * (t_a) \right]_{feeding}$$
(6)

Ethanol efficiency (Eq. 7) was calculated based on ethanol yield (Y_{ps}) described by Eq. (1) and Eq. (3) and compared to the maximum theoretical ethanol yield ($Y_{ps}^{max} =$ 0.511 g. g⁻¹). This value corresponds to the stoichiometric conversion of xylose and glucose into ethanol by *S. passalidarum*. The values reported in the tables are the average of two independent experiments.

273
$$E(\%) = \frac{Y_{ps}}{Y_{ps}^{max}}$$
 (7)

The substrate consumption rate (r_s) and specific consumption rate (q_s) for each individual sugar were calculated using Eq. (8) and Eq. (9), respectively:

$$r_s = \frac{F_a}{V_r} (S_a - S) - \frac{dS}{dt}$$
(8)

$$q_S = r_S \frac{1}{X} \tag{9}$$

278 3. Results and Discussion

279 **3.1. Raw material compositions**

280 Table 1 shows the composition of hemicellulose hydrolysate and molasses used 281 as a carbon source during fermentation experiments. According to Table 1, after the 282 diluted acid pre-treatment and detoxification steps (by evaporation) been performed, an 283 enriched pentose hydrolysate was obtained. The evaporation step enables to concentrate 284 4.5 times the hemicellulosic hydrolysate in terms of xylose concentration (from 21 – 95.0 g.L⁻¹). A diluted acid pre-treatment was chosen because it shows high specificity 285 for pentose removal and allows for better consumption of sugars, when ethanol is 286 287 concerned, due to low generation of furfural and HMF. During bagasse pre-treatment, short-chain aliphatic acids can be formed, including formic acid, acetic acid and 288 levulinic acid [21]. Acetic acid is formed mainly from the hydrolysis of acetyl groups of 289 290 hemicellulose, while formic acid and levulinic acid are formed from 5hydroxymethylfurfural (HMF) degradation [21]. The current work presents low 291 concentrations of formic acid (0.282 g.L⁻¹) and levulinic acid (0.007 g.L⁻¹). On the other 292 hand, significant concentrations of acetic acid was detected (5.5 g.L⁻¹). The acetic acid 293

in fermentation media is responsible for the reduction of specific growth rate, sugar consumption uptake, cell yield and it prolongs lag phase, decreasing the overall fermentation performance [15]. Crude lignocellulosic hydrolysates generally contain acetic acid concentrations around 1 - 8 g.L⁻¹ [23,24]. Considering this aspect, the concentration used in this work (hemicellulosic hydrolysate diluted with molasses) implies that the suggested fermenting media are of practical relevance for lignocellulose-based bioethanol production processes.

301 According to previous works, S. passalidarum can consume sucrose, although 302 this sugar is used for cell growth instead of producing ethanol. On the other hand, this 303 yeast is able to convert hexoses into ethanol. In this way, the molasses used to supplement the fermenting medium was hydrolysed in order to improve product 304 305 accumulation from mixed sugars [15]. The enzymatic hydrolysis of molasses enabled 306 100% of sucrose conversion into glucose and fructose. Finally, in order to prepare a mixed fermenting medium composed of pentoses and hexoses, a mixture of 307 308 hemicellulosic hydrolysate and hydrolysed molasses (50:50) was carried out. This 309 strategy enabled a low acetic acid concentration, as well as to supplement the fermentation medium with nutrients from molasses. The final composition of the 310 fermentation culture medium used in all experiments is shown in Table 2. 311

312 **3.2.** *S. passalidarum*: fed batch fermentations

Figure 2 shows the fermentation results of *S. passalidarum* growing at different flow rates (from 30 mL.h⁻¹ until 60 mL.h⁻¹). The dashed line identifies the fermentation time when the feeding of fresh culture medium was interrupted. The feeding was stopped when the maximum working volume of the bioreactor was achieved, and this 317 time varied according to the flow rate. It can be observed in Figure 2 that glucose was 318 quickly and completely consumed in all cases, even with changes in flow rate. In fact, 319 all sugars were consumed simultaneously, but the substrate uptake rate of each 320 individual sugar was different. This simultaneous sugar consumption occurs probably because the glucose content was always kept low during fermentation, and the diauxic 321 322 phenomenon was not verified. At higher flow rates, it was also possible to verify that 323 xylose and fructose started to accumulate in the bioreactor, and the higher the flow rate 324 the higher the residual sugar concentration.

325 Table 3 shows the fermentation performance of S. passalidarum during fed 326 batch runs using mixed sugar composition. According to Table 3, the time required for the total consumption of sugars increased with a higher flow rate. During the fed batch 327 performed at flow rate of 30 mL.h⁻¹ (Fig. 2a), all sugars were depleted at around 34 h. 328 At this condition, the highest ethanol productivity $(2.4 \text{ g.L}^{-1}.\text{h}^{-1})$ was achieved. On the 329 other hand, in order to avoid sugar starvation (and consequently reduced ethanol 330 accumulation), it is desirable to keep the sugar concentration inside bioreactor at around 331 10 g.L⁻¹ [11]. It is important to emphasize that if sugar concentrations are kept next to 332 333 zero during fed batch fermentations, the yeast starts to assimilate ethanol as a carbon 334 source, consequently reducing the process yield [15].

The maximum ethanol titer (30.3 g.L⁻¹) was achieved at flow rate of 34.2 mL.h⁻¹, with ethanol yield of 0.470 g.g⁻¹ and productivity of 2.04 g.L⁻¹.h⁻¹. On the other hand, the highest ethanol yield (0.501) and efficiency (98.0 %) was achieved for fed batch fermentation performed at a flow rate of 37.2 mL.min⁻¹. This experiment reached the highest process yield and efficiency, and demanded 90 h of fermentation. However, there were still small amounts of residual sugars (17 g.L⁻¹) at the end of the process, which were responsible for a drop in ethanol productivity $(1.92 \text{ g.L}^{-1}.\text{h}^{-1})$ and titer (26.3 g.L⁻¹). At higher flow rates the fermentation performance was reduced. The ethanol titer, yield and productivity were also reduced.

Regarding by-products, acetic acid, glycerol and xylitol were detected during fed 344 345 batches runs. Glycerol production was very low and reached a maximum concentration around 0.3 g.L⁻¹. The maximum acetic acid and xylitol concentrations at the end of 346 experiments are shown on Table 3. As mentioned previously, acetic acid is produced 347 348 during the pre-treatment and accumulated in the bioreactor according to the flow rate, reaching maximum concentration levels of about 2.3 g.L⁻¹, as shown on Fig. 2. Xylitol 349 accumulation changed accordingly to the flow rate and reached levels around 2.6 g.L⁻¹. 350 According to Table 3, it can be noted that the higher the flow rate the higher the xylitol 351 352 accumulation.

353 The sugar consumption rates (r_s) and specific sugar consumption rates (q_s) for 354 each individual substrate during fed batches fermentations are shown in Figure 3. Virtually, all sugars were consumed simultaneously, but at different uptake rates. It is 355 356 also possible to verify that xylose was consumed at the highest consumption rate, followed by glucose, fructose and arabinose. As the flow rate increases (from 30 up to 357 60 mL.h^{-1}), higher rate values were observed. The highest xylose consumption rate (rs) 358 was of the order of 3.4 g.L⁻¹.h⁻¹ achieved at a flow rate of 60 mL.h⁻¹. However, higher 359 360 flow rates resulted in higher residual sugar concentrations, reducing the overall 361 fermentation performance. For glucose, a low substrate consumption rate was observed compared to xylose, with values around 2 g.L⁻¹.h⁻¹. Therefore, this fermentation strategy 362 363 led to a xylose uptake rate higher than that of glucose, indicating that S. passalidarum 364 directed its metabolism to the xylose consumption while maintaining glucose, fructose

365 and arabinose at low rates. This behaviour is probably due to a higher expression of 366 specific enzymes necessary to promote the xylose uptake - xylose reductase (XR) and 367 xylitol dehydrogenase (XDH). Considering the xylose assimilation pathway reported to be express for S. passalidarum, the xylose is firstly reduced to xylitol due the activity of 368 enzyme XR, and then this molecule is oxidized to D-xylulose promoted by activity of 369 370 XDH. The activity of XDH is exclusively dependent of coenzyme NAD⁺. However, it 371 has been reported that S. passalidarum has two different active genes that encode XR 372 (genes XYL1.1 and XYL1.2); the first one is strictly dependent on NADH, while the second one is capable to use both coenzymes (NADH or NAPH), but it presents a 373 374 higher affinity for NADH [25]. Regarding these aspects, the excessive utilization of 375 NAPDH leads to insufficient regeneration of NAD⁺, producing a redox imbalance and, 376 consequently the xylose consumption will be interrupted, leading to a xylitol 377 accumulation instead of ethanol [26]. Adding to this, it is well established that a low concentration of glucose in the fermenting media promotes higher xylose conversion 378 into ethanol [27]. Nevertheless, the sequential utilization of these sugars extends 379 380 fermentations times, resulting in an incomplete substrate consumption. These 381 behaviours are due to the product accumulation, which achieves inhibitory levels before sugars are depleted, and the complete consumption of sugar is delayed [28]. 382

The specific sugar consumption rate exhibited similar trends. The maximum xylose specific consumption rate was 0.180 $g_{consumed xylose}.g_{dry cell}^{-1}.h^{-1}$ at a flow rate of 30 mL.h⁻¹ and 0.356 $g_{consumed xylose}.g_{dry cell}^{-1}.h^{-1}$ at flow rate of 60 mL.h⁻¹. Despite the fact that sugar consumption rates increased with increasing flow rates, from the fed batch performed at 37.2 mL.h⁻¹, the sugar content began to accumulate in the bioreactor, hindering the accumulation of ethanol. The glucose specific consumption rate varied

from 0.08 to 0.258 g_{consumed} glucose.g_{dry} cell⁻¹.h⁻¹, proving that S. passalidarum directed its 389 390 metabolism to xylose consumption when a mixed sugar substrate was used. For 391 fructose and arabinose specific consumption rates, a lower conversion was also 392 observed. The use of mixed sugar in fed batches runs showed better results at an intermediate flow rate, when no residual sugar was detected. Therefore, the results 393 394 suggest that the combined mixed sugar composition favoured the simultaneous 395 utilization of xylose, glucose, fructose and arabinose. The positive and expressive effect 396 of fed batches runs at high mixed sugar concentrations, is a relevant point in driving further advances in 2G bioethanol production from sugarcane bagasse and molasses. 397

Despite acetic acid concentration being 2.8 g.L⁻¹ in the feed medium, *S. passalidarum* was able to grow, to produce ethanol and efficiently consume all mixed sugars. *S. passalidarum* was even able to degrade acetic acid, as reported in a previous study using mixed sugar in mono and co-cultures batch runs [15]. However, the mechanisms involved in this degradation have not been completely elucidated [29].

403 The data from this work may represent a good prospect for improved ethanol production from lignocellulosic sugars. On the other hand, the ethanol produced during 404 405 fermentation can be inhibitory leading to drops in the yeast performance. In order to 406 promote a significant effect on ethanol production, improvements in ethanol-fermenting 407 microorganisms as well as bioreactor design and fermentations techniques have been 408 developed. In this sense, the selective ethanol production and removal using sequential 409 cell recycles and mixed sugar composition were carried out to investigate the 410 improvements in process yields and productivities.

411 3.3. Sequential fed batch extractive fermentation: a high cell density of 412 Spasthaspora passilidarum with bioethanol removal in situ

413 The results achieved during fed batch extractive fermentation are shown in Figure 4, regarding the investigation of the capability of *S. passalidarum* to tolerate five 414 415 repeated cell recycles with intermittent bioethanol removal in situ. The dashed lines define the end of each fed batch run, when the cell recycle for the next stage was 416 performed. The fermentation was carried out for around 400 h at a flow rate of 57.2 417 418 mL.h⁻¹, considering all five sequential steps. Bioethanol removal was carried out with a flash tank, every 6 h, during the entire process through the application of vacuum 419 420 conditions at around 150 mmHg. Each bioethanol removal cycle was about 1 h, and with this strategy it was possible to maintain a low bioethanol concentration inside the 421 bioreactor, around 20-30 g.L⁻¹ (below toxic concentrations for *S. passalidarum* strain). 422 Regarding the bioethanol condensed in the flash tank, it was possible to achieve a 423 maximum ethanolic concentration of 25°GL, reached at the end of the second cell 424 425 recycle.

The cell concentration was 56.92 g.L⁻¹ at the beginning of the process and 426 approximately 36.5 g.L⁻¹ at the end of the first fed batch fermentation (Fig. 4). Until the 427 third sequential cell recycle, the cell concentration showed similar performance. At the 428 beginning of the fourth cell recycle, the fermentation achieved the highest cell 429 concentration of 105.1 g.L⁻¹ and around 33.8 g.L⁻¹at the end of this step. This behaviour 430 431 is in agreement with the concept of sequential fed batch cell recycling, widely known as Melle-Boinot [4]. The strategy of this process was to start fermentation at high cell 432 density $(60 - 90 \text{ g.L}^{-1}, \text{ dry mass})$, which was used as inoculum for each fermentation 433 recycling, corresponding to approximately 1/3 of total bioreactor working volume. After 434

that, the fresh culture medium was gradually fed into bioreactor until reaching the total working volume of the system. At this point, yeast concentrations were supposed to achieve approximately 30-40 g.L⁻¹ at the end of each fermentation step [4,11].

The major advantage of applying cell recycling is that it is possible to provide 438 439 high cell concentrations, minimizing the consumption of sugars for yeast growth, what 440 consequently reduce the fermentation time and others costs related to cell propagation steps, and consequently will increase the overall fermentation performance [11]. The 441 442 sequential cell recycling allowed the starting of each subsequent fermentation with high 443 cell density, and with the additional strategy of ethanol removal it was possible to achieve impressive values for ethanol productivity. Accordingly in Table 4, under these 444 conditions, the maximum ethanol productivity using sugar mixture reaches 9.5 g.L⁻¹.h⁻¹, 445 showing an ethanol yield of 0.482 g.g⁻¹ and a process efficiency of 94.3%, achieved 446 during second cell recycle of fed batch extractive fermentation. This ethanol 447 productivity was around 10-fold higher than conventional processes, which is the 448 449 highest 2G ethanol productivity reported to date. The ethanol productivity declined 450 significantly after the forth cell recycle (200 h), but cell growth continued. The 451 adaptation of S. passalidarum strains during sequential cell recycle also allowed the 452 simultaneous consumption of all sugars, under microaerobic conditions. On the other 453 hand, it is well known that sugar uptake rate depends on the initial cell concentration and the nutrient supplementation. The glucose in the medium was promptly metabolized 454 455 (100%), even after prolonged fermentation. Regarding xylose content, the time required 456 for complete consumption increased at each cell recycle. Reduced values were observed 457 after the third cell recycle, delaying the fermentation performance. The main reason for 458 this behaviour is probably the accumulation of acetic acid, which is due to the ethanol removal. It may be possible that the increased concentration of this inhibitor compoundled to the deficient xylose bioconversion.

461 Despite of the increasing values of residual xylose over time, at the end of each cell recycling, both ethanol yield and productivity increased compared to the first cycle. 462 463 These increasing values reinforce the two main ideas proposed by this work: i) the 464 benefits of cellular recycling during sequential fed batch fermentation, which provided energy-saving proprieties (cells converted all carbon content into ethanol instead of 465 466 shifting for the metabolism of cell growth); and ii) the intermittently ethanol removal, 467 what will avoid inhibition of cells metabolism by the ethanol accumulated. Considering these aspects, a physiological adaptive effect of cell recycling was still observed cycle-468 469 to-cycle, implying that ethanol removal was essential to improve yeast metabolism and 470 that the adaptation of cells after third-cycle was necessary for the cell population to 471 thrive at high acetic acid concentration.

472 Figure 5 shows the sugar consumption rates (r_s) and specific sugar consumption rates (q_S) for each individual substrate. It is possible to see the interference of glucose 473 474 on the xylose consumption and it is also possible to observe the difference between mixed sugars consumption profiles. At the first fed batch fermentation, xylose was 475 consumed at a higher substrate consumption rate (2.208 g.L⁻¹.h⁻¹), followed by glucose 476 (1.752 g.L⁻¹.h⁻¹), fructose (1.112 g.L⁻¹.h⁻¹) and arabinose (0.115 g.L⁻¹.h⁻¹), respectively. 477 478 Interestingly, the xylose consumption rate decreased from the second cell recycle (1.795 g.L⁻¹.h⁻¹) and became similar to glucose (1.789 g.L⁻¹.h⁻¹). Considering this scenario, it 479 can be noted that the cell population of S. passalidarum was able to adapt itself to the 480 481 mixed sugar composition, showing similar consumption rates for glucose and xylose 482 after sequential cell recycles. Remarkably, concerning fructose and arabinose contents,

the substrate uptake rate became higher during the sequential recycles, showing that the
strategies of ethanol removal and cell recycling improved the overall fermentation
performance.

Concerning by-products accumulation, acetic acid, glycerol and xylitol were 486 detected during sequential fed batch runs. Figure 6 shows the profile of by-products 487 accumulation during fermentation time. The glycerol concentration remained low all 488 along the sequential cycles performed $(0.14 - 0.75 \text{ g.L}^{-1})$. Regarding xylitol 489 490 accumulation, it is possible to note that at the end of the third fed batch (around 200 h), xylitol started to be accumulated. S. passalidarum produced 4.8 g.L⁻¹ of xylitol in the 491 last fed batch (fifth recycle), however this accumulation may probably be avoided by 492 controlling the oxygen levels during the fermentation. Precisely controlled micro-493 494 aeration is crucial during xylose fermentation by wild-type strains, in order to regenerate the NAD⁺ cofactor required by xylitol dehydrogenase enzyme (XDH) on 495 xylose metabolism [30]. Considering this aspect, if an insufficient oxygen level is 496 provided during fermentation runs, it will lead to an insufficient amount of NAD⁺ 497 498 regenerated and an increased xylitol accumulation, blocking the xylose metabolism as described previously [31]. 499

The accumulation of acetic acid during fermentation is related to two main factors. Firstly, this inhibitor is current in hemicellulosic hydrolysate and the final concentration of the culture medium is of the order of 2.89 g.L⁻¹. Adding to this, during the ethanol removal (by evaporation), there is also the concentration of acetic acid inside the bioreactor. Although acetic acid is also a volatile substance, the concentration of this inhibitor kept increasing during ethanol removal, probably because the conditions of vacuum x temperature adopted were favourable to the extraction of the

507 ethanol-water mixture. Besides the fact that selective ethanol removal during
508 fermentation decreases the alcohols toxicity, the presence of inhibitors generated during
509 the pre-treatment increased, which affects the microbial growth and fermentation
510 performance.

511 It is possible to note that after the third fed batch cell recycle, the acetic acid concentration is about 4.5 g.L⁻¹. Despite the high concentration of this compound, there 512 is still a fraction of highly resistant cells able to resume growth after prolonged 513 514 exposure to this inhibitor, and probably this seems to be related to a superior phenotype 515 for acetic acid tolerance compared to the strains in the first fed batch cycle. The low 516 reproducibility during sequential fed batch with cell recycle led us to hypothesize that 517 for the S. passalidarum strain, only a small portion of cells are capable of keeping their 518 metabolic activities. Accordingly, any small variation in this cell population along the 519 sequential cell recycles, due to the higher concentration of acetic acid, would significantly decrease the xylose uptake rate, ethanol yield, productivity and 520 521 consequently the overall fermentation performance. The fact that only a subpopulation 522 of cells resume growth after dealing with prolonged stress conditions, can be attributed 523 to cell-to-cell heterogeneity, or even to mutations that have naturally increased in these 524 cells [32]. The current study shows that acetic acid resistance affects the performance of 525 microorganism in industrial fermentation. Once the high acetic acid tolerance of S. 526 passalidarum is accepted as a desirable phenotype in industrial 2G bioethanol, it is 527 possible to screen these portions of yeasts capable of growth after repeated cell recycling and investigate their genetics in order to construct more robust strains. Yeast 528 529 screening could be used to guide further studies related to strain-specific differences that

are genetically determined as well as to be used as a valuable starting point to identifygenetic targets for yeast improvements.

532 There is not any other report in the literature considering the use of sequential fed batch extractive fermentation with cell recycle and ethanol removal. Nevertheless, 533 534 the benefits of sequential fed batch fermentation using S. passalidarum was also reported by other researchers. Nakanihi et al. [13] investigated the performance of fed 535 batch cell recycle by the same strain using a fermentation medium composed by 536 537 sugarcane bagasse hydrolysate, however, unlike this study, with a higher concentration of hexoses instead pentoses (Glucose: 42.9 g.L⁻¹; Xylose: 14.9 g.L⁻¹). The authors 538 achieved improvements in ethanol productivity and process efficiency, from the first to 539 the fourth cell recycle (0.38 - 0.81 g.L⁻¹.h⁻¹ and 62 - 80%, respectively). Recently, 540 Neitzel et al. (2020) [33] also reported the development of sequential fed batch cell 541 542 recycle fermentation by S. passalidarum, using a synthetic fermentation medium with a similar composition applied in our work (Xylose: 63 g.L⁻¹; Glucose: 27 g.L⁻¹), but once 543 544 again without the extractive fermentation and sugarcane biomass. As a result, 545 researchers reported improvements in xylose consumption (from 84.2 to 99.5%), ethanol efficiency (from 68.9 to 90.8%) and productivity (from 1.34 to 1.79 g.L⁻¹.h⁻¹) 546 and they linked the better fermentation performance mainly due the physiological 547 548 adaption through cellular recycling. Anyway, with the application of ethanol removal techniques together with sequential fed bath cell recycle we successfully achieved the 549 highest productivity (9.5 g.L⁻¹.h⁻¹) reported in the literature, proving that the elimination 550 of inhibition promoted by product accumulation was crucial in order to achieved higher 551 fermentation performance. 552

553 The results from this study showed that high yields and productivities were 554 achieved during sequential cell recycle with ethanol removal in situ, suggesting that this 555 method can be a suitable strategy to improve the process. Additionally, what is believed 556 to be the first report has been given, regarding an organism capable of fermenting a wide range of sugars (xylose, arabinose and fructose) in the presence of glucose. This 557 ability makes S. passalidarum a potentially useful yeast strain for 2G ethanol 558 559 production, from mixed sugars released by lignocellulosic biomass, in a single reactor. 560 Furthermore, S. passalidarum is an interesting microorganism for unravelling the 561 regulatory mechanisms involving bioconversion of lignocellulosic materials by yeasts 562 and it is also possible to access this knowledge as a source of new D-xylose metabolism 563 genes for robust recombinant industrial strains of Saccharomyces cerevisiae. The results presented may establish a baseline for further improvements when mixed sugar 564 565 fermentation is the target.

566 **4. Conclusions**

The simultaneous consumption of mixed sugar by S. passalidarum increases the 567 568 bioethanol yield and productivity during sequential cell recycling with simultaneous ethanol removal. Ethanol productivity reached values around 9.5 g.L⁻¹.h⁻¹, with mixed 569 sugar concentration of 100 g.L⁻¹ in a feed medium composed of hemicellulosic 570 hydrolysate supplemented with molasses. The selective ethanol removal was 571 572 responsible for a reduced ethanol concentration inside the bioreactor (20-30 g.L⁻¹), 573 achieving high concentrations, condensed from the flash tank, with a titer around 192.4 g.L⁻¹. The long-term adaptation of S. passalidarum cells through sequential 574 575 fermentation in hemicellulosic hydrolysate media was successfully established.

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Figures Caption

Fig. 1. Schematic illustration of the fed batch extractive fermentation with sequential cell recycling. This fermentation scheme was designed to obtain a high cell density of *Spathaspora passalidarum* in cell recycle fed batch fermentations and to promote cell adaptation in mixed sugar rich-media, with ethanol removal in situ.

Fig. 2. Fermentation performances of *Spathaspora passalidarum* during fed batch runs in a mixed sugar fermenting medium composed by 50% hemicellulosic hydrolysate enriched with 50% molasses. Symbols: (▲) Xylose, (□) Glucose, (■) Fructose, (◊) Arabinose, (○) Ethanol, (●) Acetic acid. The dashed lines identifies the fermentation time when feeding of fresh medium was interrupted.

Figure 3. Sugar consumption rate (r_s) and specific consumption rate (q_s) by *Spathaspora passalidarum* for each individual sugar from the mixture of hemicellulosic hydrolysate and sugarcane molasses in fed batch fermentations according to the flow rate.

Fig. 4. Time course for *Spathaspora passalidarum* during five sequential cellular recycle through fed batch extractive fermentation runs with selective bioethanol removal in situ. Symbols: (\blacktriangle) Xylose, (\square) Glucose, (\blacksquare) Fructose, (\diamondsuit) Arabinose, (\bigcirc) Ethanol inside bioreactor, (X) Ethanol extracted, (\bullet) Acetic acid, (\triangle) DCW. The dashed lines identifies the end of each fed batch run, when the cells were recycled to the next fermentation step.

Fig. 5. Sugar consumption rate (r_s) and specific consumption rate (q_s) by *Spathaspora passalidarum* for each individual sugar, during five sequential fed batch steps of the extractive fermentations with cell recycling.

Fig. 6. By-product accumulation profile during five sequential cellular recycle fed batch extractive fermentation by *Spathaspora passalidarum* using mixed sugar composition as substrate.

Figure 1.



Figure 2





(e)

Figure 3.



Figure 4



Figure 5



Figure 6



Time (h)

E	Before Detoxificat	Hemice tion (BD)	ellulosic hydrolysate / After Detoxification	n (Evaporatio	on) (AD)	
	Sugar (g.L ⁻¹)				.L ⁻¹)	
	BD	AD		BD	AD	
Xylose	21.0	95.0	Acetic acid	3.46	5.83	
Glucose	2.0	15.2	Formic acid	0.31	0.28	
Arabinose	3.8	8.5	Levulinic acid	0.02	0.01	
Cellobiose	0.4	2.1	HMF	0.11	0.07	
			Furfural	0.25	0.05	
		Sug	arcane Molasses			
Sugar (g.L ⁻¹)		In natu	ra	Hydrolysed		
Sucrose			393.59			
Fructose			143.41		322.16	
Glucose			79.68		287.15	
ART			637.40		609.31	

Table 1. Hemicellulosic hydrolysate and sugarcane molasses used in medium composition.

Components	Concentration (g.L ⁻¹)
Xylose	47.3
Glucose	25.9
Fructose	22.8
Arabinose	5.2
Acetic acid	2.8
Final sugar concentration	101.2

Table 2. Final fermentation composition used in fresh medium.

Table 3.Fermentations parameters for *S. passalidarum* in fed batches runs using mixedsugar fermenting medium composed of hemicellulosic hydrolysate and sugarcane molasses (50:50).

Flow rate (mL.h ⁻¹)	Y _{ps} (g.g ⁻¹)	Q _{pmax} (g.L ⁻¹ .h ⁻¹)	Titer (g.L ⁻¹)	Sresidual (g.L ⁻¹)	Xy (g.L ⁻¹)	AA (g.L ⁻¹)	Time (h)	E (%)
30.0	0.430	2.40	24.2	0.0	0.0	2.0	34	84.1
34.2	0.470	2.04	30.3	2.7^{a}	1.7	1.8	70	92.0
37.2	0.501	1.92	26.3	17.0 ^b	2.0	2.3	90	98.0
45.0	0.493	1.99	23.9	12.9 ^c	2.4	2.1	100	96.5
60.0	0.473	1.97	17.4	29.0 ^d	2.6	2.0	100	92.6

Residual sugars during fermentation: ^aArabinose: 2.1; Xylose: 1.3. ^bArabinose: 3.0; Fructose: 12.4; Xylose: 1.6. ^cArabinose: 1.9; Fructose: 11.0;. ^dArabinose: 2.7; Fructose: 7.2; Xylose: 19.1.

Fermentation parameters for sequential cell recycle by S. passalidarum Table 4. extractive fed batch runs using mixed sugar fermenting medium composed of hemicellulosic hydrolysate and sugarcane molasses (50:50).

Fed batch	Y _{ps} (g.g ⁻¹)	Qp _{max} (g.h ⁻¹)	Ethanol (Bioreactor) (g.L ⁻¹)	Ethanol (Condenser) (g.L ⁻¹)	Sresidual (g.L ⁻¹)	Xy (g.L ⁻¹)	Acetic Acid (g.L ⁻¹)	E (%)
1	0.383	2.2	30.9	168.8	0.0	0.0	1.9	75.0
2	0.482	9.5	31.4	192.4	3.8 ^a	0.0	2.2	94.3
3	0.427	8.8	30.7	180.7	18.1 ^b	2.8	2.7	83.6
4	0.424	4.3	28.2	136.7	17.3 ^c	4.5	3.9	82.9
5	0.468	7.0	26.7	128.7	22.8 ^d	4.8	4.5	91.6

Residual sugars during fermentation:

^aArabinose: 2.5; Xylose: 1.3. ^bArabinose: 4.3; Fructose: 1.9; Xylose: 11.9. ^cArabinose: 2.9; Fructose: 1.2; Xylose: 13.2. ^dArabinose: 1.7; Fructose: 0.7; Xylose: 20.4.

