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Oligosaccharide binding and thermostability of two

related AA9 Lytic Polysaccharide Monooxygenases

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KEYWORDS.

Crystal structure, substrate specificity, structural biology, oligosaccharide, copper monooxygenase, lytic polysaccharide monooxygenases (LPMOs).

ABSTRACT.

Lytic polysaccharide monooxygenases (LPMOs) are copper-dependent enzymes which cleave polysaccharide substrates oxidatively. First discovered because of their action on recalcitrant crystalline substrates (chitin and cellulose) a number of LPMOs are now reported to act on soluble substrates including oligosaccharides. However, crystallographic complexes with oligosaccharides have only been reported for a single LPMO so far, an enzyme from the basidiomycete fungus Lentinus similis (LsAA9_A). Here we present a more detailed comparative study of LsAA9_A and an LPMO from the ascomycete fungus Collariella virescens (CvAA9_A) with which it shares 41.5% sequence identity. LsAA9_A is considerably more thermostable than CvAA9_A, and the structural basis for the difference has been investigated. We have compared the patterns of oligosaccharide cleavage and the patterns of binding in several new crystal structures explaining the basis for product preferences by the two enzymes. Obtaining structural information on complexes of LPMOs with carbohydrates has proven very difficult in general judging from the structures reported in the literature thus far and this can only partly be attributed to low affinity for small substrates. We have thus evaluated the use of differential scanning fluorimetry as a guide to obtaining complex structures. Furthermore, an analysis of crystal packing of LPMOs and glycoside hydrolases, corroborates the hypothesis that active site occlusion is a very significant problem for LPMO-substrate interaction analysis by crystallography, due to their relatively flat and extended substrate binding sites.

INTRODUCTION

Lytic polysaccharide monooxygenases (LPMOs) seem to play vital roles in the degradation of recalcitrant biomass. These copper containing enzymes cleave polysaccharides oxidatively rather than through the more wide-spread hydrolytic mechanism of glycoside hydrolases^{1,2}. The discovery of cellulose-active LPMOs has led to their incorporation in commercial enzyme cocktails (such as Cellic[®] CTec) for a more efficient enzymatic degradation of biomass in the biorefinery industry^{3,4}.

Since 2013 LPMOs have been classified in Auxiliary Activity enzyme families 9-11 and 13-16 (AA9-AA11 and AA13-AA16) in the Carbohydrate-Active enzyme database (CAZy)^{5–7}. The first LPMOs studied were of fungal or bacterial origin and demonstrated to be active on cellulose or chitin^{1,8}. Later, enzymes in the AA13 family were found to act on starch-derived substrates⁹. The most recently discovered LPMO families are AA14 showing activity on cellulose-associated xylan¹⁰, AA15 for which members are found in arthropod species (*e.g.* insects)^{7,11} and the fungal, cellulose active AA16⁷. Recently an AA10 of fern origin has also been discovered¹² as has an AA9 LPMO with preferential activity on xylan¹³.

Shared by all LPMOs is the Cu-coordinating histidine-brace motif (His-brace)¹ made up of an N-terminal His (often Nɛ-methylated in filamentous fungal species), and a second His later in the sequence. These occupy the ligand positions in the equatorial plane of the Cu and are often accompanied by a Tyr occupying the axial ligand position. In addition, exogenous ligands are often found in the axial- and/or equatorial coordination positions depending on the oxidation state of the Cu. LPMOs use a co-substrate (molecular oxygen or hydrogen peroxide) and an electron donor (such as small molecules, other oxidoreductases or lignin^{14–16}) in order to produce an oxygen species able to oxidize its substrate^{2,17}.

A large number of high-resolution LPMO structures have been determined (more than 38 different proteins), which have provided useful information for understanding the LPMO function, as reviewed recently *e.g.* in^{18–22}, but despite this the crystallographic information on LPMO substrate binding is scarce compared to other CAZymes. A pivotal structural finding came with the first crystallographic information on carbohydrate ligand binding by LPMOs from studies on an AA9 family enzyme from the basidiomycete fungus *Lentinus similis* (*Ls*AA9_A) in 2016²³. In that study enzyme kinetics, EPR spectroscopy and X-ray crystallography indicated that cellooligosaccharide binding leads to displacement of the axial exogenous ligand of the copper, a shortening of the Tyr-Cu(II) distance, and that the substrate is connected to the LPMO N-terminus via a hydrogen bonding network, leading to activation of the copper. In several of the *Ls*AA9_A crystal structures, a Cl⁻ ion in the equatorial position (from crystallization conditions) has been proposed as an oxygen species mimic²³.

In a follow up biochemical and structural study, the same $LsAA9_A$ enzyme and a homologue from the ascomycete fungus *Collariella virescens* ($CvAA9_A$) with which it shares 41.5% sequence identity, were further investigated. Both enzymes had activity on a range of soluble poly- and oligosaccharides²⁴, although with some notable differences in activity on xylan/xylo-oligosaccharides, with which $CvAA9_A$ exhibited a much weaker activity compared to $LsAA9_A$. Furthermore, crystallographic complexes with a range of oligosaccharides could be obtained for $LsAA9_A$. The crystal structure of $CvAA9_A$ in space group P1 was also determined.

However, a catalytically relevant complex structure could not be achieved at that time. This was likely due to a partial occlusion of the active site surface of *CvAA9_A* in the *P1* crystal form.

Here we have determined a new structure of $CvAA9_A$ in space group $P2_1$ with a less occluded active site surface. We present crystallographic complexes of both $LsAA9_A$ and $CvAA9_A$ with cello-oligosaccharides in different crystal forms, correlating the structures to the pattern of oligosaccharide binding and cleavage by the two enzymes. Generally, it has proven difficult to obtain complexes of LPMOs with their saccharide substrates. We suggest that this is because LPMOs tend to have flatter, generally more occluded substrate binding sites than glycoside hydrolases. Additionally, the difficulty in forming complexes is due to many LPMOs having low affinity for smaller substrates. Finally, we have evaluated the potential of native differential scanning fluorimetry (DSF) as a guide to crystallographic experiments of LPMOs with ligands. During this work it became apparent that $LsAA9_A$ is considerably more stable than $CvAA9_A$ and the structural basis of this additional stability was analyzed.

MATERIALS AND METHODS

Crystallization, X-ray data collection and refinement

Proteins for crystallization have been produced and deglycosylated as previously^{23,24}. Crystals of CvAA9_A and LsAA9_A were grown using the sitting-drop vapor diffusion method. Crystallization drops were set up in MRC two-well plates (Hampton Research) using an Oryx-8 crystallization robot (Douglas Instruments). During both screening and optimization the reservoir volume was kept at 100 µL, while the drops were 0.3 µL during screening and 0.4-0.5 µL for optimization. Drops of CvAA9 A were set up in a protein to reservoir ratio of 3:1 and 1:1 for both screening and optimization. 6.3 mg/ml CvAA9 A in 20 mM sodium acetate was incubated with a slight molar excess of Cu(II)-acetate for 1 hour at 4 °C before being dispensed into the MRC twowell plates. Crystallization conditions for CvAA9 A were established previously²⁴, but more open crystal forms (with an unoccluded active site) in different conditions (reservoir consisting of 0.1 M NaCl, 0.1 M Bis-Tris pH 6.5 and 1.5 M ammonium sulfate) were additionally identified in the INDEX screen (Hampton Research). These conditions were optimized by varying ammonium sulfate concentrations gradually from 1.0 M to 2.0 M. The final crystals resulting in the presented structures were grown from 1.4 M to 2.0 M ammonium sulfate (see Table 1). In these conditions CvAA9_A produces crystals with a plate-like morphology. They were harvested and flash frozen in liquid N₂ without added cryoprotectant.

*Cv*AA9_A crystals used for soaking were adjusted to pH 5.5 by moving the crystal to similar conditions to the ones they were grown in, but with a different pH, since most of the previous complexes were obtained at pH 5.5. Oligosaccharides (Megazyme) were dissolved in mother-liquor-like conditions. The saturated oligosaccharide solution was gently spun down to remove any residual material before being pipetted into the drop for a final concentration of between 0.1 M and 1.0 M. Crystals were soaked in oligosaccharides for 5-60 minutes before being harvested and flash frozen without any added cryoprotectant.

Crystallization conditions of *Ls*AA9_A have also been established previously²³ as 3-4.4 M NaCl, 0.1 M citric acid pH 3.0-4.0. The crystal used for solving the *Ls*AA9_A-Cell₄ structure was

grown in 3.3 M NaCl, 0.1 M citric acid pH 3.5. The drops were of 0.5 μ l with a protein to reservoir to water ratio of 3:1:1. The crystal was pH adjusted and soaked using the same procedure as described for *CvAA9_A* above. The crystal was soaked for 10 minutes before cryo-cooling without added cryoprotectant.

All datasets were processed and scaled using XDS and XSCALE²⁵, with resolution cutoff chosen based on $CC_{1/2}$ close to 50%. Phases for *Cv*AA9_A structures were obtained from molecular replacement, using Molrep²⁶ and the already published *Cv*AA9_A structure (without substrate, space group *P*1, PDB 5NLT)²⁴. Phases for *Ls*AA9_A were obtained from the high resolution *Ls*AA9_A structure (PDB 5ACH)²³. Restrained refinement was performed in Refmac5²⁷ of the CCP4 suite²⁸, with manual rebuilding and validation performed in COOT²⁹. Ramachandran statistics and B-factors were calculated using Procheck (Wilson B-factor)³⁰ and Baverage (Average B-factors) of the CCP4 suite²⁸. Figures were prepared in PyMOL³¹.

Enzyme reactions and PACE analysis

Enzyme reaction conditions were as described by²⁴. Briefly, apo-*Cv*AA9_A and apo-*Ls*AA9_A were pre-incubated for 1 h at 4 °C with stoichiometric amounts of Cu(II)(NO₃)₂ immediately before enzyme reactions. AA9 enzyme reactions on cello-oligosaccharides were performed at 20 °C for 4 h in 10 µl reaction volume containing: 5 nmol oligosaccharide, 100 mM ammonium formate pH 6.0, 4 mM ascorbate and 5 pmol *Cv*AA9_A or *Ls*AA9_A. Identical conditions without enzyme were used as the control. Reactions were stopped by addition of three reaction volumes of 96% (v/v) ethanol before precipitation of the undigested substrates at 4 °C, and separation of the reaction products for further analysis.

High concentration gel polysaccharide analysis by carbohydrate electrophoresis (PACE) was performed as described by²⁴ with some alterations. Briefly, reaction products and oligosaccharide standards (Megazyme) were reductively aminated with 8-aminonaphthalene-1,3,6-trisulfonic acid (ANTS, Invitrogen) and separated by acrylamide gel electrophoresis. To ensure uniform electrophoresis conditions during PACE, small impurities were removed from the gel by prerunning at 1000 V for 1 h immediately before electrophoresis. Electrophoresis was carried out at 10 °C at 1000 V for 220 min using a 192 mM glycine, 25 mM Tris, pH 8.5, running buffer. Gel composition was as follows: resolving gel, 37.5 ml 40% (w/v) acrylamide, 12.5 ml 375 mM Tris-HCl buffer, pH 8.8, 100 μ l 10% ammonium persulfate, 50 μ l tetramethylethylenediamine (TEMED); stacking gel, 2 ml 40% (w/v) acrylamide, 2.5 ml 375 mM Tris-HCl buffer, pH 8.8, 5.4 ml water, 100 μ l 10% ammonium persulfate, 10 μ l TEMED. Gel visualization was performed with a G-box (Syngene) equipped with a short pass detection filter (500–600 nm) and long-wave UV tubes (365 nm emission).

Mass spectrometry

Mass spectrometry was performed using MALDI-TOF. A 0.1% (v/v) TFA solution and an acetonitrile solution were mixed in a 2:1 (v/v) ratio and sinapic acid was added to the solution until saturated. The mixture was shortly vortexed and then spun down. The supernatant (synaptic acid matrix solution) and protein sample in solution were sequentially pipetted onto to the target plate

(in a final 2:1 ratio) and left for the solvent to evaporate. The TOF experiments were performed in linear mode calibrated for 20-50 kDa molecules.

DSF measurements

The inflection point (T_i) of thermal unfolding was measured using differential scanning fluorimetry on a Tycho-NT6 (NanoTemper). The Tycho-NT6 measures the native fluorescence at 330 and 350 nm, the ratio of which changes as the enzyme unfolds. The measurement is performed from 35-95 °C over 3 minutes (these settings cannot be changed with this instrument). Changes in T_i were used to probe interactions between enzymes and substrates. *Cv*AA9_A and *Ls*AA9_A were incubated with a slight molar excess of Cu(II)-acetate for 1 hour at 4 °C before being mixed with oligosaccharides (Megazyme). The final enzyme concentration for the measurement was 0.2 mg/ml in 20 mM Na-acetate pH 5.5, with an oligosaccharide concentration of 10 mM, with the exception of Cell₃, Cell₄, Xyl₅ and Mal₅ which were also tested at 100 mM, and Xyl₅ at 1 M. The mixture was incubated for 5 min at RT before being transferred into a capillary and recorded on the instrument. T_i values are determined automatically by the instrument software based on the peak in the first derivative of the unfolding curve. All measurements were carried out at least in triplicates.

Sequence and structure analysis

CLUSTAL was used to determine the sequence identity between $LsAA9_A$ and $CvAA9_A^{32}$. AREAIMOL^{33,34} from the CCP4 suite²⁸ was used to analyze buried surface on ligand complexation, respectively. For the subsite analysis, the analysis was carried out with an intact glucose residue for each subsite.

Crystal contacts were analysed using NCONT within the CCP4 suite²⁸. For each LPMO (AA9-AA11 and AA13-AA15) and endo-acting GH5 enzyme one representative of isomorphous structures was chosen (the highest resolution structure without bound substrate). Structures inappropriate for analysis (such as mutants or with severely disordered active sites) were discarded resulting in almost 100 structures of proteins of both bacterial and eukaryotic origin. We searched for intermolecular contacts (*i.e.* between symmetry-related molecules and between different chains within the ASU) within 5 Å, 7.5 Å and 10 Å. The analysis were performed for active site residues (His-brace in LPMOs and catalytic dyad of Glu in GH5s) and aromatic residues near the minus subsites (conserved Tyr in AA9s equivalent of Tyr208 in *Cv*AA9_A) or plus subsites (Tyr or Trp in the L2 loop in AA10s; Tyr54 in *Sm*AA10A). In GH5s two Trp (Trp170 and Trp273 in *Ta*GH5_A) on each side of the active site (Glu133 and Glu240 in *Ta*GH5_A) are involved in substrate interactions near the plus and minus subsites, respectively.

For the analysis of possible thermostabilizing features, hydrogen bonds were identified using HBPLUS³⁵. Secondary structure was identified using DSSP from CCP4³⁶. Within PyMOL the solvent accessible surface area was calculated using the *get_area* command while *dot_solvent* was set to *on*, *dot_density* was set to 4 and water and ligands was ignored. Ion-pairs were estimated manually within PyMOL using side chain oxygen atoms from Asp/Glu residues as donor and any nitrogen atom from His/Arg/Lys as acceptor within a 4.0 Å distance from each other.

Molecular dynamics simulations were performed on the high resolution structure of *Ls*AA9_A (PDB-ID 5ACH), and on a modified model of *Cv*AA9_A (PDB-ID 5NLT). The *Cv*AA9_A structure (PDB-ID 5NLT) was submitted to Swiss-model³⁷ with *Ls*AA9_A (PDB-ID 5ACH) as a template, adding residues 224-227 to the C-terminal. Additionally, residues 228-236 were modeled manually in COOT²⁹ in a similar conformation as in the *Ls*AA9_A model. The modified *Cv*AA9_A model contributed to hydrogen bonding through the added residues, and conformed to favored torsion angles (only outlier being Glu227) which were validated with RAMPAGE³⁸. The initial models were stripped of water and ions (including the active-site Cu). Using VMD³⁹, the proteins were placed at the center of a cubic water box, with 15 nm to the edges in each direction, and the charge of the proteins was balanced with Na⁺ ions. The simulations were performed using NAMD-2.12⁴⁰ with the CHARMM36⁴¹ force field and run for 10 ns at 400 K in 2 fs time steps. The simulated data was analyzed using VMDs *time line* tool³⁹. Snapshots from the MD simulations were made using VMD³⁹.

RESULTS

A new crystal structure of CvAA9_A with unoccluded active site

The previously published native structure of $CvAA9_A$ (in space group P1 with 6 mol/ASU) showed a partially occluded substrate binding surface and thus we looked to determine the $CvAA9_A$ structure in a different crystal form. In the course of optimization of crystallization conditions, several parameters were varied in the composition of the reservoir solution (originally 1.6 M ammonium sulfate, 0.1 M NaCl, 0.1 M HEPES, pH 7.5 as described in²⁴). Analysis of X-ray diffraction data suggested that crystal forms were pH-dependent, with lower pHs favoring monoclinic (P2₁) or orthorhombic (P2₁2₁2) crystal forms with similar cell dimensions. This trend was however not completely consistent as in at least one occasion (see next section) crystals grown under identical conditions turned out both in space group P1 and P2₁2₁2.

With crystals grown in 0.1 M NaCl, 0.1 M Bis-Tris pH 6.5 and 1.5 M ammonium sulfate as reservoir, we managed to solve a $CvAA9_A$ structure in space group $P2_1$ with two molecules pr. ASU, to 2.1 Å resolution. Interestingly, this new $CvAA9_A$ structure exhibits a less occluded active site surface than the original crystals in P1. The His-brace motifs are equally free in both crystal forms, but in the P1 crystals a surface exposed Tyr208 equivalent to $LsAA9_A$ Tyr203 involved in oligosaccharide binding is occluded by crystal contacts. In the $P2_1$ structure crystal form also has a relatively unoccluded active site surface: the N-terminal His has its closest crystal contact at more than 13 Å away, and Tyr208 at more than 9 Å away.

Novel oligosaccharide complex structures of *Ls*AA9_A and *Cv*AA9_A

Since the new crystal forms showed a more accessible substrate-binding surface, they were used as starting points for renewed efforts to obtain oligosaccharide complexes. Crystals were grown with 0.1 M NaCl, 0.1 M Bis-Tris pH 6.5 and 1.4-2.0 M ammonium sulfate as reservoir. The crystals were transferred to drops with cello-oligosaccharides (Cell₄, Cell₅ or Cell₆) and soaked for up to 29 minutes. From these we managed to solve *CvAA9_A*-ligand complexes obtained with Cell₄,

Cell₅ (in $P2_1$ with four and two mol/ASU, respectively) and Cell₆ (in $P2_12_12$ with one mol/ASU) to 2.0-2.8 Å resolution.

In the $CvAA9_A$ -Cell₄ complex, Cell₄ is bound in subsites -3 to +1 in two of the four molecules of the ASU (Figure 1a). In the third molecule only three glucosyl units (subsites -3 to -1) could be modelled in the electron density. The fourth molecule of the ASU has the active site surface partially blocked by a symmetry related molecule around 4 Å from Tyr208 (likely involved with aromatic stacking interaction, similar to Tyr203 in LsAA9_A^{23,24}), and no electron density that could be interpreted as cello-oligosaccharide was present near the active site of this molecule. The CvAA9_A-Cell₅ structure was more difficult to interpret with electron density only accommodating three glucose units in one molecule of the ASU (-1 to -3 subsites) and a fourth disordered glucose unit with higher B-factors (+1 subsite) (Figure 1b). In the other molecule of the ASU two glucose units can be clearly seen at the -1 and -2 subsites (Figure S1), with the active site surface being partially occluded directly above Tyr208 from a symmetry related molecule down to a distance of 4.7 Å. In the CvAA9_A-Cell₆ complex structure, Cell₆ is bound in the -3 to +3 subsites. This structure exhibits the most accessible substrate surface of the CvAA9_A structures determined (Figure 1c). Unlike several of the published complex structures with LsAA9_A, no Cl⁻ ions could be identified by any anomalous signal, or could confidently be modeled into any difference density for the CvAA9_A structures. Soaking of CvAA9_A crystals with Cell₃ or Xyl₅ did not yield any complexes. Addition of Xyl₅ in a 1.0 M concentration dissolved the crystals for soaks longer than 10 min, or abolished diffraction for shorter soaks.



Figure 1. Complex structures of $CvAA9_A$ and $LsAA9_A$. (a-c) $CvAA9_A$ with Cell₄, Cell₅ and Cell₆ respectively. Only molecule A of $CvAA9_A$ -Cell₄ and -Cell₅ complexes are shown. Cell₅ could only be built with four glucosyl units as no electron density was available to adequately accommodate the remaining glucosyl unit. (d) $LsAA9_A$ with Cell₄. $2F_0$ - F_c electron density is shown as blue mesh at 1.0 σ contour level.

Previously, we have reported crystallographic complexes of $LsAA9_A$ with Cell₃, Cell₅ and Cell₆^{23,24}, which showed the substrates being bound from subsites -1 to +2, -3 to +2 and -4 to +2, respectively. For completeness, we here also determined the structure of $LsAA9_A$ in complex with Cell₄ (Figure 1d). In this Cell₄ complex the subsites -2 to +2 are occupied. The crystal for the $LsAA9_A$ -Cell₄ complex structure (determined in $P4_132$) was formed as described in²³. For details on crystallization conditions, data collection- and refinement statistics see Table 1.

	CvAA9_A	CvAA9_A :Cell ₄	CvAA9_A :Cell ₅	CvAA9_A :Cell ₆	LsAA9_A :Cell ₄		
Beamline	ID30A-3 (ESRF)	ID30A-3 (ESRF)	ID30B (ESRF)	ID30B (ESRF)	P11 (PETRA III)		
Wavelength [Å]	0.96	0.96	0.90	0.90	1.03		
Space group	$P2_{1}$	$P2_1$	$P2_1$	$P2_{1}2_{1}2$	P4132		
Cell parameters							
(a, b, c)[Å]	51.12, 39.39,	39.79, 120.8,	52.14, 39.70,	39.55, 124.37,	125.67		
	119.93	136.36	120.65	51.67			
$(\alpha,\beta,\gamma)[^{\circ}]$	90.0, 93.77, 90.0	90.0, 92.09, 90.0	90.0, 93.491, 90.0	90.0, 90.0, 90.0	90.0, 90.0, 90.0		
Resolution [Å]	50-2.12 (2.18-	50-2.0 (2.05-2.0)	50-2.80 (2.87-	50-2.20 (2.26-	50.0-1.9 (1.95-		
	2.12)		2.80)	2.20)	1.9)		
Completeness [%]	94.5 (91.5)	98.4 (97.2)	98.2 (99.1)	96.9 (99.8)	100.0 (100.0)		
R _{meas} [%]	12.4 (88.0)	23.2 (115.4)	26.0 (115.4)	27.9 (136.0)	7.7 (183.0)		
I/ σ(I)	8.70 (1.58)	5.54 (1.44)	4.23 (1.08)	5.03 (1.28)	20.25 (1.28)		
$CC_{1/2}$ [%]	99.5 (50.8)	98.9 (54.4)	96.9 (54.7)	98.2 (62.0)	100.0 (54.7)		
Unique reflections	26050 (1853)	85419 (6184)	23359 (1783)	13174 (977)	50564 (3763)		
Observed	104767 (7161)	389031 (25746)	68143 (5290)	68277 (5185)	468655 (34891)		
reflections		()			, , ,		
Redundancy	4.02 (3.86)	4.55 (4.16)	2.91 (2.96)	5.41 (5.61)	9.26 (9.27)		
No. mol/ASU	2	4	2	1	1		
Rwork [%]	1973	21.46	19 78	19.40	21.90		
$\mathbf{R}_{\text{Free}}[\%]$	24 94	26.21	26.09	26.36	26.11		
Ramachandran[%]8	21.91	20.21	20.09	20.50	20.11		
Preferred	94.6	95.2	92.1	93 7	92.2		
Allowed	5 /	18	7 2	63	73		
Outlier	0.0	4.0	0.7	0.0	0.4		
Avr. B factors $[Å^2]$	0.0	0.0	0.7	0.0	0.4		
Drotein	15 50	11.01	13.01	17 77	25.00		
Ligand / jon	78 47	20.74	34.17	50.17	25.90		
Water	30.11	29.74	24.17 28.42	31.30	45.26		
Wilson B	<i>JJJJJJJJJJJJJ</i>	20.46	20.42 53.62	10.33	45,20		
footor [Å2]	41.10	29.40	55.02	40.55	44.79		
No. of stams							
NO. OI atoms	2502	7126	2506	1765	1016		
	5505	1150	5500	1705	1810		
Ligand / ion	15	144	08	0/	40 107		
DMCD	241	1401	82	1/8	197		
	0.014	0.014	0.012	0.014	0.016		
Bond lengths [A]	0.014	0.014	0.013	0.014	0.016		
Bond Angles [*]	1.50	1.48	1.57	1.69	1.70		
Crystallization	0.1 M NaCl	0.1 M NaCl	0.1 M NaCl	0.1 M NaCl	3.3 M NaCl		
conditions	0.1 M HEPES	0.1M Bis-Tris	0.1 M Bis-Tris	0.1 M Bis-Tris	0.1 M citric acid		
	pH 6.5	pH 6.5	pH 6.5	pH 6.5	рН 3.5		
	1.5 M (NH ₄) ₂ SO ₄	1.8 M (NH4)2SO4	2.0 M (NH ₄) ₂ SO ₄	1.4 M (NH4)2SO4			
Drop volume	0.4 µl	0.4 μl	0.4 μl	0.4 μl	0.5		
Ratio§§	1:1:0	3:1:0	1:1:0	3:1:0	3:1:1		
Soak concentration	-	1.0 M	0.1-0.5 M	0.1-0.5 M	1.0 M		
Soak duration	-	12	29	28	10		
[min]							
PDB-ID	6YDF	6YDC	6YDD	6YDE	6YDG		
Highest resolution shell shown in parenthesis.							
§ Ramachandran preferred/allowed/outlier regions calculated by Rampage ³⁸							

Table 1. C	Crystallogi	aphic data	and refineme	nt statistics.
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§§ Protein to reservoir to water volume ratio

The modes for cello-oligosaccharides binding to $LsAA9_A$ and $CvAA9_A$ in the crystal structures from this and previous studies^{23,24} are summarized in Figure 2. Table S1 shows the values of torsion angles found in the oligosaccharides in the complexes described in this paper, and indicate a conformation close to cellulose.



Figure 2. Schematic of cello-oligosaccharide binding in *Ls*AA9_A and *Cv*AA9_A. The active site surface is schematically shown as subsites -4 to +3 (semicircles). Filled circles indicate glucosyl units. For each substrate the predominant products identified in the PACE analysis (here, indicated by # and in previous work as indicated in parentheses and with reference number), and the binding mode found in the corresponding crystal structures (here, indicated by # and/or in previous work as indicated by the reference number) have been illustrated. Binding modes consistent with the product patterns are shown. For Cell₃ and Cell₄ the binding modes in the crystal can fully account for the solution products. Reaction products on Cell₆-2AB (Cell₅ analogue) determined by mass spectrometry^{23,24} are also shown and suggest a main binding mode. The crystal structure of *Cv*AA9_A with Cell₅ has the two molecules pr. ASU shown with only four glucosyl units in molecule A, and two units in molecule B. The products of *Cv*AA9_A with Cell₅ digestion agrees with the crystal structure assuming the missing (presumably disordered) glucosyl unit of molecule A is found in subsite +2.

The protein-carbohydrate interactions at individual subsites were analyzed in terms of hydrogen bonds and buried accessible area on the protein using the *Ls*AA9_A and *Cv*AA9_A complexes with Cell₆ in Table S2. Total interactions in complexes with different oligosaccharides are summarized in Table S3. Figures S2 and S3 compare complex structures of *Ls*AA9_A and *Cv*AA9_A with Cell₄ and Cell₆. These figures highlight that residues interacting with the substrates are mostly the same with some notable differences at subsite -1 (Ala75 in *Ls*AA9_A and Asp76 in *Cv*AA9_A) and subsite +2 (Asn28/His66/Asn67 in *Ls*AA9_A and Thr28/Arg67/Val68 in *Cv*AA9_A).

LsAA9_A and CvAA9_A activity on cello-oligosaccharides

We have previously reported some of the patterns of cello-oligosaccharide degradation for $LsAA9_A$ and $CvAA9_A^{24}$. Here, we have further investigated digestion of cello-oligosaccharides (Cell₃, Cell₄, Cell₅ and Cell₆) by the two LPMOs, and analyzed the reaction products by Polysaccharide Analysis using Carbohydrate Electrophoresis (PACE). Since these two enzymes are known to oxidize cello-oligosaccharides at the C4 position, all products (oxidized and non-

oxidized) are visible by PACE analysis (Figure 3), and by the relative migration of the products it is possible to investigate the site of cleavage. Under the reaction conditions tested, *Cv*AA9_A hardly degraded Cell₃, while *Ls*AA9_A readily degraded Cell₃ yielding considerable amounts of products migrating as both cellobiose (likely the C4-oxidized 4-ketoaldose form, based on Cell₃ binding in *Ls*AA9_A-Cell₃ structure, PDB-ID: 5ACF) and glucose.



Figure 3. PACE gel showing the digestion products of $CvAA9_A$ and $LsAA9_A$ on Cell₆, Cell₅, Cell₄ and Cell₃ oligosaccharides. Enzymatic reactions with the indicated enzyme, 1.0 stochiometric amounts of Cu and excess reducing agent were run for 4 hours. Reaction products were aminated and separated using acrylamide gel electrophoresis. Oxidized and non-oxidized reaction products with the same degree of polymerization are expected to migrate similarly.

The pattern of degradation for Cell₄ is very different, with *Ls*AA9_A producing predominantly cellobiose, while *Cv*AA9_A produces mostly Cell₃, indicating different binding modes. With Cell₅ the products are consistent with what has been identified earlier using mass spectrometry^{23,24}, where Cell6-2AB was used as a Cell₅ analogue and produce Cell₃ and Cell₂ for both *Ls*AA9_A and *Cv*AA9_A. *Ls*AA9_A is then expected to degrade Cell₃ further to Cell₂ and glucose, which does not appear to be the case for *Cv*AA9_A. With Cell₆ as substrate *Ls*AA9_A produces mostly cellobiose and glucose, while *Cv*AA9_A produced mostly Cell₃ and cellobiose. In this case it is difficult to relate this pattern to a particular binding mode, since the initial products are likely to be cleaved further²⁴, and the binding modes in crystal may be affected by crystal contacts. Product profiles from this and previous work are summarized in Figure 2 together with the crystal binding modes.

Probing oligosaccharide binding in solution by differential scanning calorimetry (DSF)

Given the difficulties in obtaining crystallographic complexes with LPMOs, we were curious to see how DSF experiments in solutions correlate with the crystallographic results obtained for *Ls*AA9_A and *Cv*AA9_A. Thermal curves shifts on addition of ligands (even relatively poor

ligands) can indicate binding in solution, as already shown for LPMOs⁴², and guide crystallographic complex formation studies. In our case, we used native DSF, based on the differences in intrinsic fluorescence of the enzymes before and after thermal unfolding, in particular following the intrinsic fluorescence ratio (350 nm / 330 nm) and determining the inflection point in the thermal unfolding transition (T_i). For Cu-loaded *Ls*AA9_A the thermal unfolding is easily followed by this method, as the difference in the fluorescence ratio is pronounced between the folded and unfolded state. T_i was thus easily determined as 71.8 ± 1.7 °C (T_i values listed in Table 2 and visualized in Figure S4).

Table 2. Points of thermal inflection (T_i) for *Ls*AA9_A in the presence of oligosaccharides. Inflection temperatures extracted from DSF unfolding curves of *Ls*AA9_A. T_i measurements are listed in °C. Measurements were performed using 0.2 mg/ml LPMO in 20 mM Na-acetate pH 5.5. Values highlighted in bold font are calculated to be significantly different from the LPMO (without substrate) measurement, with a 95% confidence in Student's t-test. In the case of cellooligosaccharides producing a biphasic unfolding curve, only the first T_i is listed.

	LsAA9_A	
	69 1 + 0 1	
	09.1 ± 0.1	
	71.8 ± 1.7	
Calla	72.4 ± 1.7	
Cell3	72.4 ± 1.7	
	72.5 ± 1.0	
Cell ₅	75.2 ± 0.9	
Cell ₆	76.0 ± 0.2	
Mal ₅	71.9 ± 1.0	
Man ₆	73.0 ± 0.3	
G4G3G4G	72.0 ± 0.3	
Xyl ₅	72.8 ± 0.5	
XXXG	78.7 ± 0.1	
Chi ₅	72.0 ± 0.1	
Cell ₃	72.2 ± 0.37	
Cell ₄	73.5 ± 0.1	
Xvl5	73.5 ± 0.1	
Mal ₅	72.4 ± 0.36	
Xyl ₅	76.0 ± 0.57	
	Cell ₃ Cell ₄ Cell ₅ Cell ₆ Mal ₅ Man ₆ G4G3G4G Xyl ₅ XXXG Chi ₅ Cell ₃ Cell ₄ Xyl ₅ Mal ₅ Xyl ₅	LsAA9_A 69.1 ± 0.1 71.8 ± 1.7 Cell3 72.4 ± 1.7 Cell4 72.3 ± 1.6 Cell5 75.2 ± 0.9 Cell6 76.0 ± 0.2 Mal5 71.9 ± 1.0 Man6 73.0 ± 0.3 G4G3G4G 72.0 ± 0.3 Xyl5 72.8 ± 0.5 XXXG 78.7 ± 0.1 Chis 72.0 ± 0.1 Cell4 73.5 ± 0.1 Xyl5 72.4 ± 0.36 Xyl5 72.4 ± 0.36

The addition of 10 mM Cell₅, Cell₆ or the xyloglucan oligosaccharide XXXG gives a clear thermal shift towards higher T_i , while the other oligosaccharides tested at this concentration do not (Figure 4). However, Cell₄ gives a biphasic unfolding curve when added at a concentration of 10 mM, indicating that Cell₄ likely binds. None of the other oligosaccharides significantly affected the T_i or shape of the curve (see Figure S5 and S6).



Figure 4. Unfolding curves of *Ls*AA9_A with selected oligosaccharides. Unfolding curves are shown for Cu-loaded *Ls*AA9_A with 10 mM of chitopentaose (Chi₅, red, negative control), cellohexaose (Cell₆, magenta), and Xyloglucan oligosaccharide (XXXG, green). Transparent color area indicate standard deviation. Dashed lines indicate the inflection point (T_i). *Ls*AA9_A alone has similar T_i to *Ls*AA9_A with Chi₅, while with Cell₆ or XXXG the T_i increases.

Binding of Cell₃ and Xyl₅ (xylopentaose) has previously been observed in crystal structures with LsAA9_A, where the oligosaccharide concentrations used have been higher than 150 mM, and as high as 0.9 M for Xyl5^{23,24}. For Cell₃, Cell₄ and Xyl5 we chose to repeat the LsAA9_A thermal unfolding experiments with oligosaccharide concentrations closer to the ones used for complex formation (see Figure S7 and S8). At 100 mM concentrations Cell₄ still exhibit a biphasic curve with the largest inflection indicating a T_i of 87.7 °C. Using Cell₃ at 100 mM concentration the curve is biphasic as well indicating that a population of the enzyme is shifted to higher T_i, however the shift is much less pronounced than when using Cell₄. Using Xyl₅ or Mal₅ (maltopentaose, an α -1,4-linked oligosaccharide of glucose, which is not expected to be cleaved by AA9 LPMOs²⁴) at 100 mM concentrations the curves are unchanged in appearance and inflection points are consistent with the ones observed without added oligosaccharide. Since Xyl5 is bound in a crystal structure with LsAA9_A where the oligosaccharide concentration was 0.9 M²⁴ we wanted to approach this concentration for the DSF experiment as well. With 1.0 M of Xyl₅ the T_i increased to 76.0 \pm 0.57 °C, and the fluorescence ratio increased throughout the measurement (see Figure S7C and S8C). Similar observations were however made using 1-4 M of xylose (Figure S9), indicating that at these high concentrations the thermal shifts may be due to unspecific effects.

A similar study was done with $CvAA9_A$, but the melting curve is shallower than for $LsAA9_A$, and this gives challenges in precise determination of T_i which is however clearly much lower than for $LsAA9_A$, 57.7 ± 0.9 °C in the presence of copper. Small changes in folding curves are seen for some cello-oligosaccharides, but generally the results are less clear (Table S4 and Figure S10 to S13).

Analysis of AA9 family crystal packing and comparison to a glycosyl hydrolase family

We have previously noted a flat and extended active site surface in LPMOs compared to the deeper substrate binding clefts in glycoside hydrolases such as GH5 (Figure S14). Thus, we hypothesized that the aromatic platforms involved in LPMO-substrate interaction would be more frequently involved in crystal contacts compared to substrate-interacting aromatic residues in GH5s, providing a rationale as to why so few LPMO-complex structures have been obtained.

Therefore, we investigated crystal contacts in AA9 and GH5 structures involving key functional residues: the His-brace and aromatic platforms of LPMOs, and the catalytic dyad and substrate interacting Trp residues in GH5s. Indeed, we found that within a 5 Å distance a higher number of intermolecular contacts were observed for LPMOs compared to GH5s. In LPMO structures, an average of 6, 13, 33 and 37 contacts pr. chain were found for the N-terminal His, second His, the conserved AA9 Tyr and AA10 L2 Tyr/Trp, respectively. In contrast, in GH5 structures 8 contacts pr. chain were found on average for the Trp near the negative subsites (Trp273 in *Ta*GH5_A, PDB 1H1N), while the value was less than one for both the Trp near the positive subsite and for the active site Glu catalytic dyad (Acid/Base and Nucleophile) (Figure 5). A similar trend was observed when inspecting the number of contacts found within 7.5 Å and 10 Å, showing that LPMO structures have significantly more crystal contacts involving functional residues compared to GH5 structures (Figure S14d).





Structural basis of thermostability

In order to identify possible reasons for the significantly higher thermostability of *Ls*AA9_A compared to *Cv*AA9_A a quantification of structural features often correlated with thermostability was carried out, including analysis of Pro/Gly ratio, potential hydrogen bonds and ion pairs, secondary structure and accessible surface composition (Table S5). Although there is a clear difference in terms of *Cv*AA9_A having a more charged surface the other differences seem small and unable to explain the large difference in stability.

One clear difference is the construct length used for structure determination (252 amino acids for $CvAA9_A$), however the number visible in the structure is only 224. In order to clarify if this is due to disorder in the structure or degradation during production, storage or crystallization we performed MALDI-TOF MS of a $CvAA9_A$ sample in solution (data not shown). The determined weight of 25974.15 Da corresponds to 236 residues suggesting indeed that degradation has occurred and that the preparations of $CvAA9_A$ and $LsAA9_A$ comprise polypeptide chains of similar length (235 residues for $LsAA9_A$). Despite the similar length of the two enzymes, more residues are visible in the $LsAA9_A$ structure (Figure S15). The C-terminus of $LsAA9_A$ is able to form an additional β -strand which elongates one of the β -sheets of the $LsAA9_A$ β -sandwich. At the end of this additional β -strand residue IIe226 is buried in a small hydrophobic pocket which act as an anchor point for the extended C-terminus (Figure S15A). The equivalent position of IIe226 in $LsAA9_A$ is Cys230 in $CvAA9_A$ (Figure S15C), which is not able to make similar interactions. Furthermore, the hydrophobic pocket is already filled by Trp129 in $CvAA9_A$. Thus, the much lower thermostability of the $CvAA9_A$ construct studied could be due to a more disordered C-terminus.

To investigate this hypothesis further, the missing residues of $CvAA9_A$ (225-236) were modeled through a combination of Swiss-model³⁷ (residues 224-227) and manual modeling (residue 228-236) using $LsAA9_A$ (PDB 5ACH) as a template (Figure S15C). 10 ns molecular dynamics simulations were performed for both $LsAA9_A$ and $CvAA9_A$ at 400 K (Figure S16 and S17). The root mean square fluctuations (RMSF) over the whole simulation indicates that the extended C-terminal of $CvAA9_A$ is more disordered than the equivalent segment of $LsAA9_A$. Over the course of the simulation the β -strand modelled in $CvAA9_A$ (residues 226-230) becomes untethered, unlike the equivalent amino acids in $LsAA9_A$ (residues 222-226). Additionally, the surface L7 and LC loops ($LsAA9_A$ naming) are destabilized in $CvAA9_A$ as seen in the larger RMSF for residues 150-166/196-216, compared to the equivalent regions 145-161/195-212 in $LsAA9_A$ (see Figure S16 and S17).

DISCUSSION

As indicated now by a number of other studies, the clear difference in both binding pattern and cleavage pattern of *e.g.* Cell₄ by *Cv*AA9_A and *Ls*AA9_A highlight that LPMOs primarily cleave through binding to substrate and bringing of substrate and reactive oxygen species precisely together, and not merely by generation of reactive species. Binding of substrate is a prerequisite for efficient use of LPMOs – such precise and appropriate binding of substrate is in fact necessary for preventing untimely inactivation of LPMOs⁴³⁻⁴⁵. In this respect we have previously reported that binding of *Ls*AA9_A to xylo-oligosaccharides may be less 'appropriate' than for cello-oligosaccharides²⁴, in the sense that the binding mode is different, preventing a similar geometry of interaction with the oxygen species, and probably resulting in diminished protection from oxidative damage.

From this study we can extract some general features of oligosaccharide binding to LPMOs, with on average 97 Å² accessible area buried for an intact glucose unit and 1.5 H-bonds per subsite. Generally cello-oligosaccharides bind in a cellulose-like conformation. The analysis of interactions reflects well substrate preference in cleavage studies, for example we see both lower buried accessible area and fewer hydrogen bonds with Xyl₅ than Cell₅ for *Ls*AA9_A. The

symmetrical cleavage of Cell₄ by *Ls*AA9_A compared to the asymmetrical cleavage by *Cv*AA9_A correlates well with extra hydrogen bonds interactions of *Ls*AA9_A at +2 and *Cv*AA9_A's extra interactions at -3.

Generally speaking, there is for $LsAA9_A$ an overall good correlation between the degraded substrates, oligosaccharides that could bind in the crystals, and oligosaccharides that gave a thermal shift or otherwise clear changes in the DSF analysis. For very soluble substrates in the DSF analysis it is possible to use very high concentrations, as in crystal soaks, so that even weakly binding fragments, such as xylo-oligosaccharides, can be identified. However, at molar concentrations non-specific interactions will also be reflected in the thermal shifts, as was found in measurements with >1 M xylose (Figure S9). DSF (in that case using a fluorescence dye) was first utilized in LPMO research for an in depth study of $NcAA9_C^{42}$ where the effect of several additives on stability was investigated, including soluble polymeric substrates, but not oligosaccharides. In two recent publications, we have attempted to use native DSF with an LPMO and an LPMO-like protein to probe oligosaccharide binding, but with somewhat inconclusive results^{46,47}. Here we establish that DSF, native and presumably also using fluorescent dyes, can be useful to pre-screen for useful binding fragments for crystallographic studies. However, as shown for $CvAA9_A$, this is mostly useful where a clear inflection point can be shown in the melting curve.

One exception to the correlation between crystallographic results and DSF is with xyloglucan oligosaccharides. As previously reported²⁴, no crystallographic complexes could be obtained, though xyloglucan is degraded by *Ls*AA9_A and other AA9 LPMOs acting on soluble substrates, where direct interactions with the substrate have been clearly shown by NMR⁴⁸. Thermal shift analysis confirm that XXXG is bound by *Ls*AA9_A, thus as discussed before, this oligosaccharide is probably too big to fit between crystal contacts in the crystal form used in this and previous studies.

The synergistic binding of chloride ions and oligosaccharides has been previously proposed²³, and chloride ions are often found in complex structures with $LsAA9_A$. In the presented complex structures of $CvAA9_A$ no chloride ions could be identified. This is likely due to the much lower Cl⁻ concentration in conditions favoring $CvAA9_A$ crystal growth (100 mM compared to 3-4 M with $LsAA9_A$).

In general, complex formation with LPMOs in crystals, just like for other enzymes, requires an accessible active site surface. Because LPMOs substrate binding sites are 'flatter' than for most other enzymes, we suggest that this is particularly problematic for crystallographic studies of LPMOs. Indeed, for the enzyme sets investigated here, LPMOs active sites in crystals are more often occluded by symmetry-related molecules compared to GH5 hydrolases.

In the course of this investigation we observed that $CvAA9_A$ is much less thermostable than $LsAA9_A$. Thermostability is an important feature for industrial enzymes⁴⁹ including AA9 family LPMOs, and has previously been investigated for $AfAA9_B^{50}$, which had unfolding temperature in similar range as $LsAA9_A$. Protein engineering of $AfAA9_B$ generated variants with a 6-7 °C increase in melting temperature and improved industrial performance⁵⁰, probably due to a combination of improved electrostatic and packing interactions. These variants were

inspired by the structure of $TaAA9_A$, which is unique in terms of its exceptional thermostable nature, and a T_i of up to 89.5 °C has recently been demonstrated using DSF⁵¹. The engineered region of $AfAA9_B$ is already similar to the equivalent regions in both the structures of $CvAA9_A$ and $LsAA9_A$, so for these two enzymes further improvement of thermostability cannot be expected by applying the same $TaAA9_A$ inspired mutations.

In contrast, most other AA9 LPMOs where this has been investigated had lower melting temperatures^{42,52}. In this case, comparison of the structures of *Ls*AA9_A and *Cv*AA9_A pointed to improved interactions of the C-terminal portion of *Ls*AA9_A with the rest of the structure, compared to *Cv*AA9_A, may be the main sources of additional stability. Molecular dynamics simulations at 400 K, while not necessarily biologically relevant per se, illustrate clearly the more flexible nature of the C-terminus in *Cv*AA9_A compared to *Ls*AA9_A already indicated by proteolysis during crystallization and lack of electron density for several C-terminal residues. In the hyperstable *Ta*AA9_A, the very C-terminal residue Gly228 corresponds to *Ls*AA9_A Gly220, preceding the flexible C-terminus. Our findings are thus is in line with previous work^{52,53}, highlighting the importance of C-terminal and linker regions in AA9 for their stability.

In summary, our further work on $CvAA9_A$ and $LsAA9_A$, two AA9 LPMOs sharing 41.5% sequence identity, provides new details on similarities and differences. Despite their sequence and structural similarity, $LsAA9_A$ is considerably more thermostable. We proposed that the lower thermostability of $CvAA9_A$ is caused by a more flexible C-terminus, which is also prone to proteolysis. We further investigated interactions with oligosaccharide substrates and found native DSF to be a promising technique to guide crystallographic complex structure determination of LPMOs in the future, which however is likely to continue to be hampered by the involvement of the shallow active site of LPMOs in crystal contacts. We show that $CvAA9_A$, only the second LPMO for which crystallographic information on substrate binding has been obtained, binds oligosaccharides similarly to $LsAA9_A$ overall, but with subtle differences mediated particularly by structural differences at the +2 subsites. These differences are reflected in an asymmetric pattern of Cell₄ degradation by $CvAA9_A$, while $LsAA9_A$ produces only DP2 products.

ACCESSION CODES

Atomic coordinates and structure factors for the reported crystal structures were deposited in the Protein Data Bank under accession codes: 6YDC (*CvAA9_A-Cell*₄), 6YDD (*CvAA9_A-Cell*₅), 6YDE (*CvAA9_A-Cell*₆), 6YDF (*CvAA9_A* (without substrate), 6YDG (*LsAA9_A-Cell*₄). UniprotKB Protein accession codes are *CvAA9_A*: A0A223GEC9; *LsAA9_A*: A0A0S2GKZ1.

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Conflict of interest

The authors declare that they have no conflicts of interest with the contents of this article.

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ABBREVIATIONS

AA, Auxiliary Activity; AfAA9_B, isoform B of Auxiliary Activity family 9 LPMO from Aspergillus fumigatus; ASU, Asymmetric Unit; CAZy, Carbohydrate Active enzyme; Cell₃, cellotriose oligosaccharide; Cell₄, cellotetraose oligosaccharide; Cell₅, cellopentaose oligosaccharide; Cell₆, cellohexaose oligosaccharide; Chi₅, chitopentaose oligosaccharide; CvAA9_A, isoform A of Auxiliary Activity family 9 LPMO from Collariella virecens; DSF, native Differential Scanning Fluorimetry; G4G3G4G, Glu-1,4-Glu-1,3-Glu-1,4-Glu; GH5, Glycoside Hydrolase family 5; LPMO, Lytic Polysaccharide Monooxygenase; LsAA9 A, isoform A of Auxiliary Activity family 9 LPMO from Lentinus similis; Mal₅, maltopentaose oligosaccharide; Man₆, mannohexaose oligosaccharide; NcAA9_A, isoform A of Auxiliary Activity family 9 LPMO from Neurospora crassa; NCS, Non-Crystallographic Symmetry; PACE, polysaccharide analysis by carbohydrate electrophoresis; RMSF, root mean square fluctuations; TaAA9_A, isoform A of Auxiliary Activity family 9 LPMO from Thermoascus aurantiacus; TEMED, tetramethylethylenediamine; T_i, point of thermal inflection; XXXG, xyloglucan oligosaccharide, cellotetraose backbone with xylose decorations on three consecutive glucose units from the non-reducing end, see nomenclature⁵⁴; Xyl₅, xylopentaose oligosaccharide.

SUPPORTING INFORMATION

Tables S1 – S5

Figures S1 – S17

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Supporting Information:

Oligosaccharide binding and thermostability of two

related AA9 Lytic Polysaccharide Monooxygenases

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SUPPLEMENTARY TABLES

		CvAA9_A	CvAA9_A	CvAA9_A	LsAA9_A	
Glycoslaic torsion angles		Cell ₄	Cell ₅	Cell ₆	Cell ₄	
Subsite	Φ	-	-	-74.56		
+2/+3 (°)	Ψ	-	-	128.19		
Subsite	Φ	-	-	-85.48	-91.65	
+1/+2 (°)	Ψ	-	-	87.27	88.43	
Subsite	Φ	(A) -82.40 (B) -97.39	-97.14	-101.22	-90.68	
-1/+1 (°)	Ψ	(A) 94.37	99.99	105.39	96.99	
		(A) -75.79				
	Φ	(B) -70.03	-82.32 (-87.35)	-74.50	-78.75	
Subsite		(C) -66.37				
-1/-2 (°)	174	(A) 96.16	100.83	08.24	101.20	
	Ψ	(B) 105.57 (C) 103.77	(98.75)	98.34	101.20	
		(A) -78.86				
	Φ	(B) -88.05	-76.64	-88.54	-	
Subsite		(C) -89.51	_			
-2/-3 (°)		(A) 103.27				
	Ψ	(B) 98.28	101.51	99.37	-	
		(C) 93.93				
Values from model (°)	φ		-98	8.75		
	Ψ	90.91				
Definitions	φ		O ₅ ·-C ₁	·-O ₄ -C ₄		
	Ψ		C1'-O	4-C4-C3		

Table S1. Torsion angles in bound cellooligosaccharides of presented complexes. Model values fromcellulose microfibril in http://polysac3db.cermav.cnrs.fr¹.

Subsite	-4	-3	-2	-1	+1	+2	+3
CvAA9_A H-bonds*	NA	1	2	1	(2)	2(1)	1(1)
		Asp155	Glu153, Arg164	Asp76, Ser78		Thr28, Arg67	Arg3 (+crystal contacts)
LsAA9_A H-bonds*	0	0	3	1	(1)	3(1)	NA
	(+ crystal contacts)		Glu148, Arg159	Ser77		Asn28, His66, Asn67	
CvAA9_A buried area §	NA	84.2 Å ²	114.9 Å ²	103.6 Å ²	103.5 Å ²	99.0 Å ²	101.1 Å ²
LsAA9_A buried area §	66.8 Å ²	72.0 Å ²	118.7 Å ²	109.1 Å ²	106.9 Å ²	87.2 Å ²	NA

* number of direct H-bonds between ligand and protein within 3.4 Å cutoff, identified in PyMOL through the ligand site preset. Numbers in parenthesis are the number of H-bonds between ligand and protein mediated by a single water molecule. Residues given are the ones interacting directly with the ligand. Interactions arising from crystal contacts are not considered.

§ Area of buried accessible surface on complexation. The calculation is carried out at each subsite with an intact glucosyl unit (thus the sum of all subsites is larger than the surface area buried by the oligosaccharide)

Table S2. Analysis of individual subsites interactions based on Cell₆ complexes with $LsAA9_A$ (PDB 5ACI, ²) and $CvAA9_A$ (this manuscript).

Ligand	Cell ₃	Cell ₄	Cell5	Cell ₆	Xyl5
LsAA9_A					
PDB code	5ACF	6YDG	5NLS	5ACI	5NLN
Buried area (Å ²)	222	286	376	386	313
Total number of H- bonds	8	9	10	9	7
CvAA9_A					
PDB code	-	6YDC	6YDD*	6YDE	-
Buried area in Å ²		271/298#		421	
Total number of H- bonds		8		11	

Table S3. Analysis of total buried surface for different complexes. # Buried protein surface is given for chains A and B (where Cell₄ binding is clearly visible) * not calculated since only 2 and 4 glucosyl units are clearly bound to different protein chains

		CvAA9_A	
LPMO		57.7 ± 0.9	
	Cell ₃	58.4 ± 0.8	
	Cell ₄	59.1 ± 2.2	
	Cells	59.5 ± 1.2	
	Cell ₆	59.4 ± 0.2	
10 10	Mal ₅	58.4 ± 0.1	
+10 mM	Man ₆	58.4 ± 0.2	
	G4G3G4G	57.6 ± 0.6	
	Xyl5	58.6 ± 0.4	
	XXXG	58.0 ± 0.1	
	Chi5	58.1 ± 0.5	
	Cell ₃	59.2 ± 0.1	
+100 mM	Cell ₄	61.4 ± 0.2	
	Xyl5	58.6 ± 0.2	
	Mal ₅	58.2 ± 0.4	

Table S4. Points of thermal inflection (T_i) for $CvAA9_A$ in the absence and presence of oligosaccharides. Inflection temperatures extracted from DSF unfolding curves $CvAA9_A$. T_i measurements are listed in °C. Measurements were performed using 0.2 mg/ml LPMO in 20 mM Naacetate pH 5.5.

LPMO	LsAA9_A	CvAA9_A
Sequence length	235	236*
Sequence modelled in structure	1-235	1-224 (1-227 in some chains in
-		5NLT)
Charged residues [%]		
(Arg, Asp, Glu, His, MeHis, Lys)	16.6	25.0
Polar residues [%]		
(Asn, Cys, Gln, Gly, Ser, Thr, Tyr)	38.7	33.3
Hydrophobic residues [%]		
(Ala, Ile, Leu, Met, Phe, Pro, Trp, Val)	44.7	41.7
Aromatic residues [%]		
(His, MeHis, Phe, Trp, Tyr)	12.8	13.1
Number of Pro / Gly residues	20 / 18	20 / 20
Accessible surface area [Å ²]		
Total	9890	10217
Charged residues	2381 (24%)	4423 (43%)
Polar residues	4375 (44%)	2776 (27%)
Hydrophobic residues	3134 (32%)	3018 (30%)
Secondary structure content [%]		
a-helices	4.7	6.7
β-sheet	29.4	29.0
3 ₁₀ -helices	2.6	0
No. of hydrogen bonds §		
Main chain – Main chain	112 (0.47)	103 (0.43)
Main chain – Side chain	69 (0.29)	54 (0.22)
Side chain – Side chain	56 (0.23)	38 (0.16)
Residues in possible ion-pairs (4 Å cutoff)		
His/Lys/Arg	11	12
Asp/Glu	13	15
PDB ID	5ACH	6YDF

Table S5. Structural features of $LsAA9_A$ published in ² and $CvAA9_A$ (this manuscript).

* Sequence length determined by mass spectrometry. Construct length is 252 residues, thus proteolysis must have occurred during crystallization.

§ Number in parenthesis is average number of hydrogen bonds pr residue.

SUPPLEMENTARY FIGURES



Figure S1. Active site surface of molecule B in *Ls*AA9_A-Cell₅. Only two glucosyl units could be modeled into the available electron density (final $2F_0$ - F_c is shown as blue mesh at 1.0 σ contour level).



Figure S2. Superposition made using chain A of $CvAA9_A-Cell_4$ and $LsAA9_A-Cell_4$. The superposition was made using 1371 atoms with the PyMOL command *super*. (a) $LsAA9_A-Cell_4$ showing cellotetraose bound in subsites -2 to +2. (b) $CvAA9_A-Cell_4$ showing cellotetraose bound in subsites -3 to +1. Residues and residue names are shown in green and black respectively for $LsAA9_A$, while they are shown in pink and magenta for $CvAA9_A$.



Figure S3. Superposition made using chain A of $CvAA9_A-Cell_6$ and $LsAA9_A-Cell_6$. The superposition was made using 1389 atoms with the PyMOL command *super*. (a) $LsAA9_A-Cell_6$ showing cellohexaose bound in subsites -4 to +2. (b) $CvAA9_A-Cell_6$ showing cellohexaose bound in subsites -4 to +2. (b) $CvAA9_A-Cell_6$ showing cellohexaose bound in subsites -3 to +3. Residues and residue names are shown in green and black respectively for $LsAA9_A$, while they are shown in pink and magenta for $CvAA9_A$.



Figure S4. Inflection temperatures (T_i) of Cu-loaded *Ls*AA9_A and *Cv*AA9_A in solution without/with 10 mM oligosaccharides. Graph is based on values presented in Table 2 and Table S4. Cellopentaose (Cell₅), Cellohexaose (Cell₆) and xyloglucan oligosaccharide (X₃Glc₄) significantly increases the T_i for *Ls*AA9_A, while shifts in T_is are less conclusive for *Cv*AA9_A. Error bars indicate standard deviation.



Figure S5. Unfolding curves of *LsAA9*_A in solution with 10 mM oligosaccharides. Panels (A-J) with individual oligosaccharides as labeled. Transparent colored areas represent the standard deviation.



Figure S6. First derivative of unfolding curves of *LsAA9*_A in solution with 10 mM oligosaccharides. Panels (A-J) with individual oligosaccharides as labeled. Transparent colored areas represent the standard deviation.



Figure S7. Unfolding curves of *Ls*AA9_A in solution with 10 mM, 100 mM or 1 M oligosaccharides as indicated in the legends. *Ls*AA9_A with Cell₃ or Cell₄ result in biphasic unfolding curves. With 10 mM or 100 mM Xyl₅ the unfolding curves for *Ls*AA9_A appear unchanged from the unfolding curve of the native enzyme. At 1 M Xyl₅ the unfolding curve of *Ls*AA9_A has shifted towards a higher T_i . Transparent colored areas represent the standard deviation.



Figure S8. First derivatives of unfolding curves of *Ls*AA9_A in solution with 10 mM, 100 mM or 1 M oligosaccharides as indicated in the legends.



Figure S9. Unfolding curve of Cu-loaded *LsAA9* A in solution with xylose in increasing concentrations. At concentrations up to 0.1 M xylose the Ti is not significantly changed, but slight differences are found in the shape of the unfolding curves. At concentrations of 1-4 M xylose there is a clear thermal shift, indicative of non-specific interactions having an effect on the thermal unfolding at these concentrations.



Figure S10. Unfolding curves of *CvAA9*_A in solution with 10 mM oligosaccharides. Panels (A-J) with individual oligosaccharides as labeled. Transparent colored areas represent the standard deviation.



Figure S11. First derivative of unfolding curves of *CvAA9_A* in solution with 10 mM oligosaccharides. Panels (A-J) with individual oligosaccharides as labeled. Transparent colored areas represent the standard deviation.



Figure S12. Unfolding curves of *CvAA9*_A in solution with 10 mM or 100 mM oligosaccharides as indicated in the legends. Transparent colored areas represent the standard deviation.



Figure S13. First derivative of unfolding curves of *CvAA9*_A in solution with 10 mM or 100 mM oligosaccharides as indicated in the legends. Transparent colored areas represent the standard deviation.



Figure S14. Crystal contacts in LPMOs and GH5. (a) $LsAA9_A$ (PDB 5ACH) active site surface with the conserved Tyr (in AA9s) near the minus subsites (red). (b) $SmAA10_A$ (PDB 2BEM) active site surface with extended L2 loop harboring the Tyr/Trp position (orange). For both (a) and (b) the His-brace residues are in cyan (c) $TaGH5_A$ (PDB 1H1N) active site surface showing a cleft around the catalytic Glu dyad (magenta) and the two substrate interacting Trp (yellow). (d) average number of crystal contacts pr. chain observed within 5 Å, 7.5 Å and 10 Å of the above mentioned residues.



Figure S15. C-terminus of *Ls*AA9_A (PDB 5ACH), *Cv*AA9_A (*P*2₁, PDB 6YDF) and *Cv*AA9_A (extended C-terminal used in molecular dynamics simulations). (A) For *Ls*AA9_A residue numbers 216-235 of the C-terminal are highlighted in green. Indicated with an arrow is the buried Ile226. (B) C-terminal residues 220-224 are highlighted in magenta. (C) *Cv*AA9_A with an extended C-terminal modeled using *Ls*AA9_A (PDB 5ACH) as a template. Highlighted in blue is residue number 220-236. The equivalent position to Ile226 in (A) is Cys230 show in (C). The elongated C-terminus of *Cv*AA9_A was modeled to compare C-terminal flexibility between *Cv*AA9_A and *Ls*AA9_A in molecular dynamics simulations.



Figure S16. Molecular dynamics simulation of *Ls*AA9_A at 400 K. (A) RMSF of *Ls*AA9_A over the 10 ns simulation. (B) is the displacement of each amino acid C α shown as a heat map over the course of the simulation. (C) panel displays modeled residues 220-235 over the course of the simulation. The C-terminal segment from residue 227-235 exhibits large fluctuations in coordinates, while the segment 220-226 largely retains the initial conformation, as found from panels (A) and (B). Arrows indicate the L7 and LC loops.



Figure S17. Molecular dynamics simulation of $CvAA9_A 400$ K. (A) RMSF of $CvAA9_A$ over the 10 ns simulation. (B) is the displacement of each amino acid C α shown as a heat map over the course of the simulation. (C) panel displays modeled residues 220-236 over the course of the simulation. The modeled C-terminal part of $CvAA9_A$ exhibits large fluctuations, as found from panels (A) and (B). Arrows indicate the L7 and LC loops.

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