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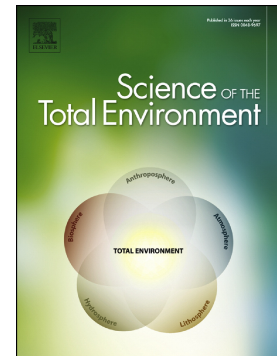
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**High denitrification potential but low nitrous oxide emission in a constructed wetland treating nitrate-polluted agricultural run-off**

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**Abstract**

Constructed wetlands (CW) can efficiently remove nitrogen from polluted agricultural run-off, however, a potential caveat is nitrous oxide ($N_2O$), a harmful greenhouse gas and stratospheric ozone depleter. During five sampling campaigns, we measured $N_2O$ fluxes from a 0.53 ha off-stream CW treating nitrate-rich water from the intensively fertilized watershed in Rampillon, France, using automated chambers with a quantum cascade laser system, and manual chambers. Sediment samples were analysed for potential $N_2$ flux using the He-$O_2$ incubation method.

Both inlet nitrate ($NO_3^-$) concentrations and $N_2O$ emission varied significantly between the seasons. In the Autumn and Winter inlet concentrations were about 11 mg $NO_3^-$-N $L^{-1}$, and < 6.5 mg $NO_3^-$-N $L^{-1}$ in the Spring and Summer. $N_2O$ emission was highest in the Autumn (mean±standard error: 9.7±0.2 µg N $m^{-2}$ $h^{-1}$) and lowest in the Summer (wet period: 0.2±0.3 µg N $m^{-2}$ $h^{-1}$). The CW was a very weak source of $N_2O$ emitting 0.32 kg $N_2O$-N $ha^{-1}$ $yr^{-1}$ and removing around 938 kg $NO_3^-$-N $ha^{-1}$ $yr^{-1}$, the ratio of $N_2O$-N emitted to $NO_3^-$-N removed was 0.033%. The automated and manual chambers gave similar results. From the potential $N_2O$ formation in the sediment, only 9% was emitted to the atmosphere, the average $N_2$ $N_2O$ ratio was high: 89:1 for $N_2$-$N_{potential}$ : $N_2O$-$N_{potential}$ and 1353:1 for $N_2$-$N_{potential}$ : $N_2O$-$N_{emitted}$. These results indicate complete denitrification. The focused principal component analysis showed strong positive correlation between the gaseous $N_2O$ fluxes and the following environmental factors: $NO_3^-$-N concentrations in inlet water, streamflow, and nitrate reduction rate. Water temperature, TOC and DOC in the water and hydraulic residence time showed negative correlations with $N_2O$ emissions.

Shallow off-stream CWs such as Rampillon may have good nitrate removal capacity with low $N_2O$ emissions.

**Keywords**: automated chambers; greenhouse gas; manual chambers; $N_2$; nitrate removal;

quantum cascade laser absorption spectrometer

# 1    Introduction

Nutrient export from anthropogenic activities in urban areas or agricultural fields harm water bodies through eutrophication and pollution of drinking water. Excess nutrients in freshwaters can be reduced efficiently by buffer zones and natural or constructed wetlands (CW; Verhoeven et al., 2006; Cheng et al., 2020). Nitrogen (N) is removed from the water via the following processes: (i) temporary direct plant uptake of inorganic N and sedimentation (Brix, 1997; Abe et al., 2014; Li et al., 2015); (ii) microbial N transformation into gaseous dinitrogen ($N_2$) via denitrification and anammox (Erler et al., 2008; Ligi et al., 2015; Wang et al., 2018; Ma et al., 2019), and (iii) nitrous oxide ($N_2O$) from denitrification (Erler et al., 2008; Jia et al., 2011; Batson et al., 2012), nitrification (Jia et al., 2011; Zhang et al., 2018) or DNRA (Jahangir et al., 2017; Wang et al., 2018). Although wetlands play a larger role in climate change mitigation through the carbon cycle, N removal may also contribute significantly through $N_2O$ emissions, especially at higher N loading rates. $N_2O$ is a greenhouse gas with a global warming potential about 153-310 times more powerful than carbon dioxide depending on the time horizon (IPCC, 2007), and it is 298 times larger in the 100 years timespan (IPCC, 2013). Moreover, $N_2O$ has been predicted to be the dominant stratospheric ozone-depleting substance in the 21st century (Ravishankara et al., 2009).

Long-term monitoring in wetlands demonstrated that nitrate ($NO_3^-$) removal efficiency depends on climate conditions, input load, carbon availability, residence time, and vegetation cover (Pulou et al., 2011, Tournebize et al., 2017). However, few wetland studies have evaluated the environmental risks associated with $NO_3^-$ removal and the balance between water quality improvement and $N_2O$ emissions (Freeman et al., 1997). For instance, increased anthropogenic N loads potentially strengthen the risk for $N_2O$ emissions in freshwater wetlands (Paludan & Blicher-Mathiesen, 1996; Wang et al., 2017; D'Acunha & Johnson, 2019) and salt marshes (Chmura et al., 2016). Most commonly, $N_2O$ is a product of incomplete processes of denitrification or nitrification (Butterbach-Bahl et al., 2013). Complete heterotrophic denitrification ends up with production of dinitrogen ($N_2$; Wilcock & Sorrell, 2008; Butterbach-Bahl et al., 2013). Therefore, analysis of environmental conditions regulating both production and consumption of $N_2O$ is crucial for evaluation of the net $N_2O$ emission in ecosystems.

$N_2O$ is highly soluble in water (Haynes, 2015). Therefore, in addition to the primary losses of the $N_2O$ gas produced from nitrification and denitrification in the soil, the dissolution and eventual degassing of $N_2O$ in leaching and drainage water from agricultural fields is also a

significant loss pathway (Dowdell et al., 1979; Roper et al 2013). $N_2O$ fluxes from water bodies (degassing) in agricultural areas are considered as indirect $N_2O$ emission. They are calculated as the proportion of the N leached from soils to the N applied as fertilizers. By default, this proportion is considered as 30% of the amount of applied N in fertilizers; Kasper et al., 2019). The partitioned emission factors for $N_2O$ degassing from leached water, rivers and estuaries are 0.015, 0.0075, and 0.0025, respectively (IPCC, 2000, IPCC, 2002). However, recent publications propose degassing factors for the surface waters up to one order of magnitude lower (Sawamoto et al., 2005; Maavara et al., 2019). In a larger riverine system of complex land-use pattern, the indirect fluxes were estimated to account only for 3.7% of the total greenhouse gas emissions from the basin, while $N_2O$ emissions accounted for 4.4% of all indirect fluxes (Marescaux et al., 2018). Beaulieu et al. (2011) found a high average emission rate (37 µg $N_2O$-N $m^{-2}$ $h^{-1}$) from US rivers, while Xiao et al. (2019) reported 1.5 times higher emissions from rivers and ditches of an agricultural watershed in Eastern China. Furthermore, emissions from shallow lakes vary remarkably, from −12 through 48 (Soja et al., 2014) to 969.2 µg $N_2O$-N $m^{-2}$ $h^{-1}$ (Liu et al., 2014) showing lower values in their littoral zone. Likewise, larger and cleaner lakes emit significantly less $N_2O$ (Chen et al., 2011). In several studies, $N_2O$ emissions were calculated from dissolved $N_2O$, as a water-atmosphere equilibrium in stream water (Garnier et al., 2009; Audet et al., 2017). Large concentrations of dissolved $N_2O$ are shown in nitrate-contaminated groundwater (Vilain et al., 2012; Jurado et al., 2017) and eutrophic ponds (Cao et al., 2016) Although wetlands can be sinks for dissolved $N_2O$, they show low concentrations when compared to the inflowing groundwater or streamwater (Blicher-Mathiesen & Hoffmann, 1999; Hinshaw & Dahlgren, 2016; Wang et al., 2017), while low pH (< 0.6) in rivers may decrease dissolved $N_2O$ concentration and increase degassing (Audet et al., 2020).

The proportion of denitrified $NO_3^-$ that is converted to $N_2O$ rather than $N_2$ – the $N_2O$ yield – is usually applied to characterize the denitrification potential of aquatic ecosystems (Beaulieu et al., 2011). Likewise, the ratio of $N_2 : N_2O$ in sediments is an important parameter of nitrogen cycle in water bodies, but only few ^{15}N isotope studies considering the $N_2 : N_2O$ ratio in wetland sediments are known from early 1990s (Lindau & DeLaune, 1991; Lindau et al., 1991).

For free water surface (FWS) CWs (Erler et al., 2008; Batson et al., 2012) as well as for wastewater stabilization ponds (Glaz et al., 2017) only limited information on $N_2O$ investigations is available.

The main objectives of this study are: (1) to analyse $N_2O$ emission dynamics in relation to

water quality and environmental factors and (2) to estimate the denitrification potential in an off-stream constructed wetland treating nitrate-polluted water from an intensively managed agricultural catchment. To meet the first objective, we applied a new methodology of semi-continuous measurement using a quantum cascade laser absorption spectroscopy (QCLAS) system, which allows high temporal frequency measurement of $N_2O$ fluxes. For the second purpose, a novel He-$O_2$ incubation method of intact soil cores was implemented (Butterbach-Bahl et al., 2002). The manual chamber technique is a common method for the measurement of $N_2O$ fluxes in CWs (Mander et al., 2014) and no previous attempt using automated chambers with $N_2O$ analysers is documented. The latter is quite a common technique used in various other ecosystems for decades (Hargreaves et al., 1996; Yao et al., 2009). High-frequency laser systems are widely used to analyse $N_2O$ fluxes in terrestrial ecosystems, mostly for eddy covariance studies (Eugster et al., 2007, Cowan et al., 2014; Huang et al., 2014; Merbold et al., 2014). However, these systems are still scarcely used in wetlands (Savage et al., 2014).

Considering the results from previous studies we hypothesize that: (1) the studied CW is a major source of gaseous $N_2O$ emissions and (2) the denitrification potential of the sediments is high. To check these hypotheses, we aimed to (i) better understand $N_2O$ emission dynamics at diurnal and seasonal time scales, (ii) analyse the dependence of $N_2O$ flux on various environmental conditions, (iii) analyse the ratio of gaseous to dissolved $N_2O$ fluxes, and (iv) determine the $N_2$:$N_2O$ ratio in the CW. Several continuous monitoring campaigns were carried out at key seasons in an experimental comprehensively studied CW within an agricultural watershed.

## 2 Material And Methods

### 2.1 Site description

2.1.1 Context of the study

The Champigny aquifer (limestone of the Eocene Epoch) is a major groundwater resource for Paris Area providing drinking water for more than 1.5 million inhabitants living in 1700 km². About 60% of the annual water reload of this karstic aquifer is directly infiltrated from surface water through natural sinkholes, increasing the vulnerability of the groundwater to diffuse source of pollution. We investigated a 355-ha watershed of Rampillon in the historic Brie region of Île-de-France (Paris Region) (Seine-et-Marne, (03°03'47" E, 48°32'19" N, 70 km south-east of Paris, France).

2.1.2 Rampillon constructed wetland

The Rampillon CW in the Brie plateau was implemented in 2010 to collect drainage and run-

off waters from an agricultural catchment and meet local environmental and health issues. The 5300 m² constructed wetland consists of ecologically engineered sub-basins separated by bunds to enhance water time retention and mitigation processes (Fig. 1). A first sedimentation sub-basin (deep inflow zone) is about 1 m deep and 300 $m^3$ in volume. A middle zone of 5000 m² has a maximum 0.5 m depth with a volume max of 1200 $m^3$. A 1000 $m^3$ with 0.8 m deep outflow sub-zone was implemented upstream from the outlet. In 2015, about 40% of the CW was covered by sedges (*Carex* spp.), common reed (*Phragmites australis*), broadleaf cattail (*Typha latifolia*), bulrushes (*Juncus* spp.) and various algae.

Five different campaigns were organized to cover the seasons of a full hydrological year. The beginning of the campaign in May 2014, with already high temperatures and low flow conditions, was classified as Spring. The end of May 2014 and October 2015 were classified respectively, as wet or dry Summer (Summer W or D, respectively), respectively. Summer D had no water flow and some parts of the CW sediments were dried up. Autumn was represented by November 2014, characterized by low temperature and intensive flow. Winter was represented by the March 2015 campaign, corresponding to low temperature, intensive water flow and the highest nitrate concentrations. The duration of a campaign was 3-6 days (Supplementary Table S1).

An automatic weather station (Campbell Scientific Inc., Logan, UT, USA) measured rainfall, solar radiation, air temperature and humidity at the wetland location.

## 2.2 Water quality measurements

Continuous water flow and quality monitoring system installed at the inlet and outlet of the CW measured flow turbidity, dissolved organic carbon (DOC), total organic carbon (TOC) and $NO_3^-$ concentrations at hourly time step using a Doppler sensor (Sigma 950 Flow Meter; Hach Company, Loveland, CO, USA) and multiparameter spectrophotometer (spectro::lyser UV-vis; s::can Messtechnik GmbH, Vienna, Austria). Hydraulic retention time (HRT) and nitrate retention ratio (RR-N) were calculated based on the flow dynamics and $NO_3^-$ concentrations on inlet and outlet.

During the five campaigns, temperature, pH, electric conductivity (EC), dissolved oxygen concentration and its saturation were measured using portable devices (Multiparameter portable meter MultiLine® Multi 3630 IDS; and YSI ProODO Optical Dissolved Oxygen Meter; both by Xylem Analytics, Beverly, MA, USA). In each compartment of the CW, water samples for the analysis of daily nutrient concentration was sampled. Anions ($SO_4^{2-}$, $Cl^-$) and cations ($NH_4^+$, $K^+$, $Na^+$, $Ca^{2+}$, $Mg^{2+}$) from all water samples (automatic and manual) were analysed at the lab of the UR HYCAR, French National Institute for Agriculture, Food,

and Environment (INRAE), Paris, using ion chromatography (DIONEX DX-120) with a pre-column Dionex IONPAC1 AG9-HC and a IONPAC1 AS9-HS column (Thermo Fisher Scientific Inc., Waltham, MA, USA). A 9-mM carbonate buffer solution was used as the mobile phase. Water temperature and water level were measured using a pressure transducer model Madofil by IRIS (IRIS Instruments, Orléans, France).

Water samples for dissolved $N_2O$ analysis were collected at 5 to 8 different points of the wetland, about 2 to 5 times per campaigns between 10h and 12h of the daytime. Samples were stored in a 100 ml penicillin flask, poisoned with $HgCl_2$ (2%) and stoppered with a rubber septum excluding any headspace. In the laboratory of UMR METIS, Sorbonne University, Paris, dissolved $N_2O$ was determined with a gas chromatograph Perichrom ST 2000 (Perichrom, Saulx-les-Chartreux, France) combined with an electron capture detector (GC-ECD) (Garnier et al., 2009; 2010).

## 2.3 $N_2O$ measurements

### 2.3.1 Automated chamber measurements

A system of 12 automated floating chambers was connected via the multiplexer to a quantum cascade laser system (Aerodyne QCLAS $N_2O$-$H_2O$-$CH_4$ analyser; Aerodyne Research Inc, Billerica, MA, USA) for a continuous high frequency (1-2 Hz) monitoring of $N_2O$ fluxes. This instrument uses tunable infrared laser differential absorption spectroscopy (TILDAS) enabling an on-line analysis of atmospheric trace gases with high sensitivity.
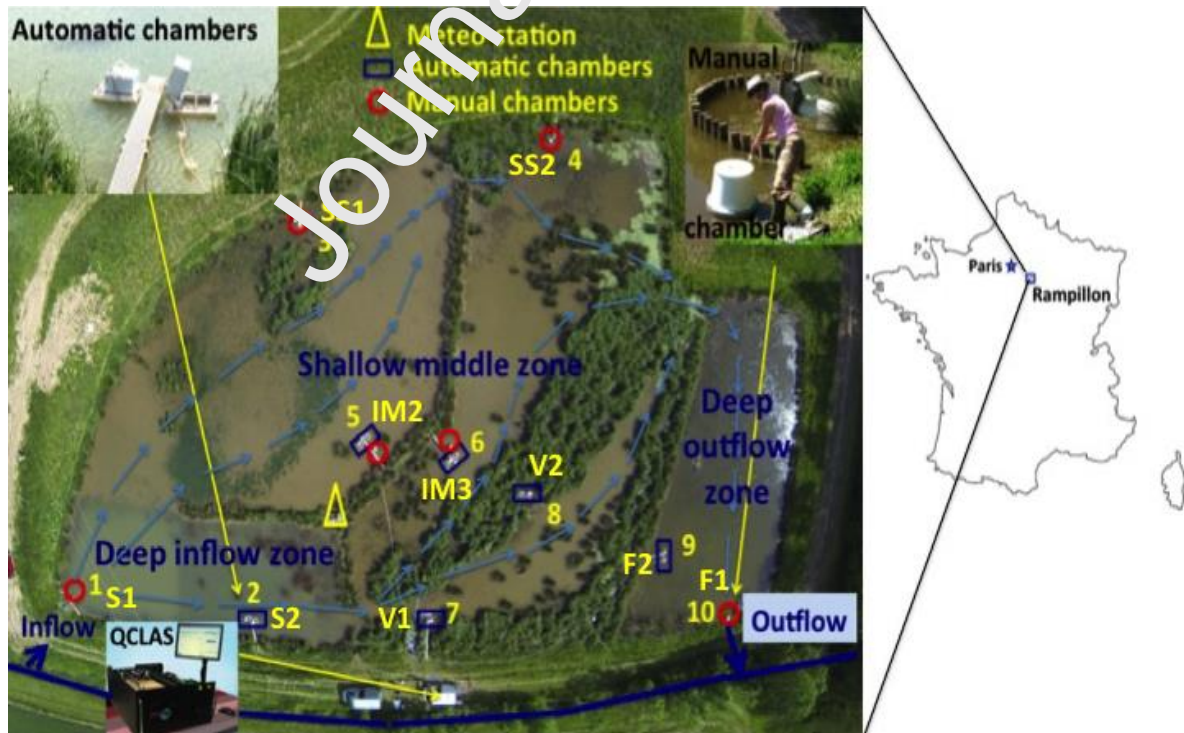


**Figure 1.** Location of Rampillon constructed wetland in France and set-up of chambers for

gas measurements (1-10) and sediment sampling points: deep water - S1, S2 (inflow) and F1, F2 (outflow); shallow and stagnant water – SS1, SS2, IM2, IM3, V2 and V2.

The laser frequency is 1271 cm$^{-1}$ for $N_2O$, the system is thermoelectrically cooled (Thermocube) to 32$^{\circ}$C. For a complete description of the QCLAS instrument, see Nelson et al. (2004). The QCLAS analyser was connected through pumps and PTFE pipes to the multiplexer linked to 12 chambers located in 6 pairs on the 30-35 m distance from the gas analyser installed in a van (Fig. 1). Each rectangular PVC chamber covered a 0.16 m² wetland surface, containing a volume of 0.032 m³. To avoid stratification of gas inside the chamber, air with a 1.8 L min$^{-1}$ constant flow rate was circulated within a closed loop between the chamber and gas analyser unit during the measurements by a diaphragm pump. The air sample was taken from the top of the chamber headspace and pumped back by distributing it to each side of the chamber. For the measurements, the chambers were closed automatically for a duration of 6 minutes each. Flushing time of the whole system with ambient air between measurement periods was 1 minute. Thus, there were approximately 18 measurements per chamber per day. Each automated chamber was installed on a floating frame of extruded polystyrene foam. The frame was anchored to ensure stability in opening and closing and support the equilibrium of the water column.

The air temperature in the chamber was measured and used to convert the concentrations from ppm[v] to mg m$^{-3}$ according to the ideal gas law before flux calculation (Collier et al., 2014). Gas fluxes were calculated for a selected time window (150 seconds), after discarding initial 90 seconds to exclude the initial stabilisation period, using the linear model which is based on the assumption of a linear relationship between concentrations inside the chamber headspace and time (Livingston & Hutchison, 1995).

2.3.2   Manual chamber measurements

Additional manual chamber measurements (Hutchinson & Livingston, 1993) were conducted to cover the whole wetland. Six of them were located in remote places where automated measurements were not possible to install. Four manual chambers were set up close to the automated chamber allowing a comparison between the short-term (1 hour) and continuous (several days) monitoring results (Fig. 1). Sampling was carried out twice a day (in the morning and in the afternoon). Gas samplers (opaque static PVC chambers of truncated-cone shape, height 50 cm, Ø 50 cm, volume 65 L, painted white to avoid heating during application; similar to those used in soil flux analyses; Mander et al., 2005) were installed on hollow-foam swimming-pool noodles and anchored for stability. Gas samples were taken

right after the closure of the chambers (0 moment) and after 20 min, 40 min and 60 min using pre-evacuated (0.3 mbar) 50-mL gas bottles (Hansen et al., 2013) to determine emission rates. Gas concentration in the samples was analysed using the Shimadzu GC-2014 gas chromatography system (Shimadzu Corporation, Kyoto, Japan) equipped with ECD detector modified by Loftfield et al (1997). The emission rate of trace gas was calculated based on the linear increase of gas concentrations in time, corrected with the area and volume of the chamber (Soosaar et al., 2011).

## 2.4. Sediment sampling and analysis

Sediment samples were taken before and after each campaign from two depths (0-5 cm and 10-15 cm) in 12 locations (Fig. 1) and analysed in dried sediment for $pH_{KCl}$ and content of organic matter (OM), total carbon (TC) and total nitrogen (TN) using a varioMAX CNS elemental analyser, dissolved nitrogen (DN), TOC, DOC and dissolved inorganic carbon (DIC) using Vario TOC/TNb detector (Elementar Analysentechnik GmbH, Langensebold, Germany), and $NO_3^-$-N , $NH_4^+$-N, total P, Ca and Mg according to standard methods (APHA-AWWA-WEF, 2005) at the laboratories of the University of Tartu and the Estonian University of Life Sciences. Ten intact sediment samples with metallic cylinders were taken from the 0-10 cm upper layer for further analysis of potential $N_2$ and $N_2O$ fluxes in the gas chromatography lab of the Department of Geography, University of Tartu, Estonia. Helium atmosphere soil incubation technique was used to measure potential $N_2$ and $N_2O$ fluxes from intact soil cores in laboratory (Butterbach-Bahl et al. 2002). The cylinders with the intact soil cores were placed into special gas-tight incubation vessels located in a climate chamber. Gases were removed by flushing with an artificial gas mixture (21.0% $O_2$, 358 ppm $CO_2$, 0.313 ppm $N_2O$, 1.67 ppm $CH_4$, 5.97 ppm $N_2$ and rest He). The new atmosphere equilibrium by continuously flushing the vessel headspace with the artificial gas mixture at 20 mL per minute was established after 12-24 h. The flushing time depended on the soil moisture. The temperature was kept similar to the field conditions during incubation. The gas-chromatograph (Shimadzu GC-2014) equipped with a thermal conductivity detector was used to measure $N_2$ and $N_2O$ concentration in the mixture of emitted gases accumulated in the headspace (0, 40, 80, and 120 minutes) of the cylinder after 2 h of closure. The gas concentration in the chambers increased in a near-linear way so that linear regression was applied for calculation of the fluxes. For all gas analyses only flux measurements leading to a $R^2$ of 0.81 or greater (p<0.01) were used.

## 2.5. Statistical analysis of data

The data set was analysed by two complementary statistical methods: principal component

analysis (PCA) and focused PCA (FPCA). Differences in PCA between the sampling campaigns and subsites were evaluated using PERMANOVA with 9999 permutations. Pairwise comparisons were corrected with the Bonferroni method. Focused PCA allows determining the main factors explaining the variability of $N_2O$ and $N_2$ emissions. Correlations between the dependent variable and the other variables are represented faithfully in FPCA, contrary to PCA. Randomesque missing values for both PCA and FPCA were imputed using the random forest technique. One-way ANOVA with the Tukey HSD post hoc test was applied to evaluate the significance of the differences between seasons in dissolved $N_2O$ concentrations and ratios ($N_2$-$N_{potential}$ : $N_2O$-$N_{potential}$ and $N_2$-$N_{potential}$ : $N_2O$-$N_{emitted}$). In addition, the t-test was used to evaluate the significance of the differences between ratios of $N_2$-$N_{potential}$ : $N_2O$-$N_{potential}$ and $N_2$-$N_{potential}$ : $N_2O$-$N_{emitted}$ according to each season. Exploratory data analyses were performed with the *missRange* (Mayer, 2019), *ade4* (Dray & Dufour, 2007), *vegan* (Oksanen et al., 2019) and *psy* packages (Falissard, 1999) of the R software.

## 3 Results

### 3.1 Water regime and quality at inlet and outlet

During the campaigns, hydrological regimes were relatively stable due to low rainfall and low maximum flows in response to high rainfall events: from no flow during the summer campaign to 5.9 and 14.8 L s$^{-1}$ for the Autumn and Winter campaigns respectively (Fig. 2, Table 1). Discharge during the campaigns was representative for 82% of a full hydrological year while the discharge above 15 L s$^{-1}$ represented the rest of time. Water depth was above full storage in CW during the Autumn, Winter and Spring but low during the Summer W and Summer D campaigns. Inlet flow was zero during the Summer W and D events, low (2.5 L s$^{-1}$), and Spring, and high during Autumn and Winter (29.4 and 26.8 L s$^{-1}$, respectively).
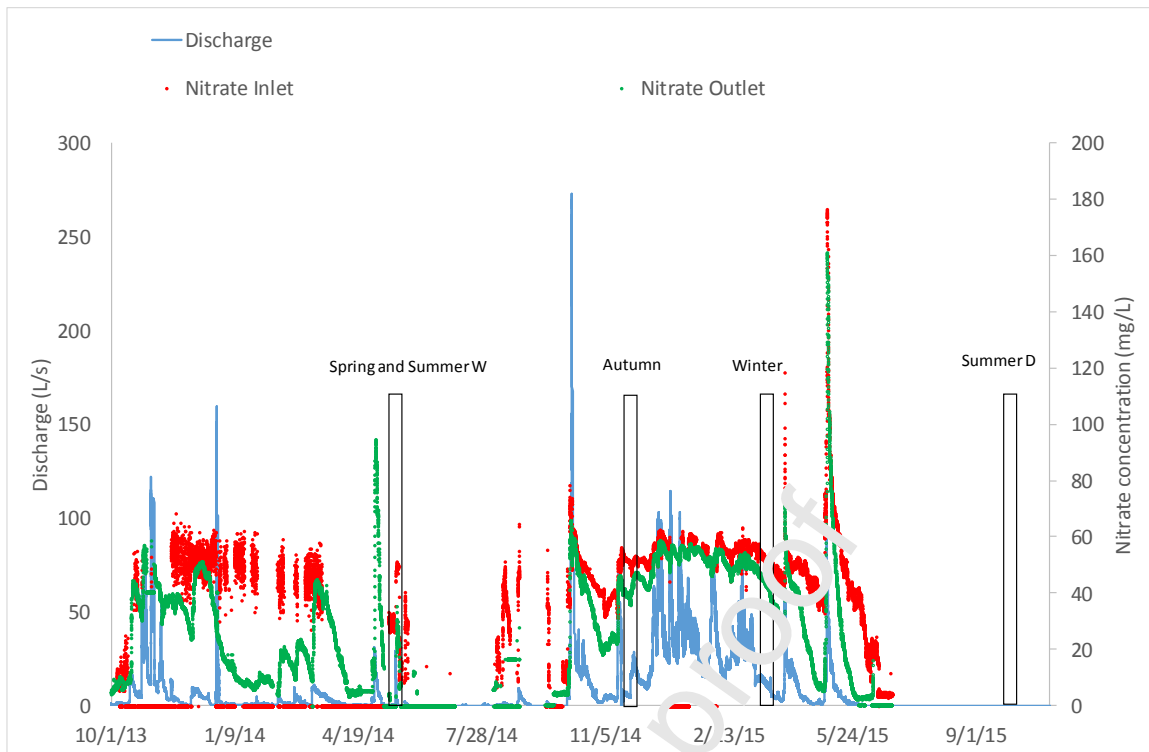
**Figure 2.** Discharge and nitrate concentrations at the inlet and outlet of the Rampillon CW during the period 2013-2015. Black rectangles indicate five campaigns: Spring and Summer W in May 2014, Autumn in November 2014, Winter in March 2015, and Summer D in October 2015.

The corresponding theoretical hydraulic residence time ranged from short (2 or 4 days) during Autumn and Winter to long (216 days) during Spring and could not be calculated for the no-flow period of Summer W and D.

The dynamics of nitrate concentrations during the campaigns was representative for 75% of the full drainage season. In general, the CW reduced $NO_3^-$ concentrations from inlet water (Fig. 2). Nitrate concentrations were higher during Autumn and Winter (about 11 mg $NO_3^-$-N $L^{-1}$) and lower during Spring and Summer (< 6.5 mg $NO_3^-$-N $L^{-1}$). The average nitrate removal rate in Autumn, Winter and Spring varied from 0.08 to 0.64 mg $NO_3^-$-N $m^{-2}$ $d^{-1}$. The negative removal value (−0.13 mg $NO_3^-$-N $m^{-2}$ $d^{-1}$) in Summer W was explained by the water management strategy: the inlet was closed, but the outlet gate kept open (Table S1). $NO_3^-$ concentrations in the water at different points within the CW were similar to the outlet measurements for all campaigns. The other water quality variables showed limited changes, e.g. pH only varied between 7.4 and 7.6, and no difference was detected for $NH_4$ neither at inlet/outlet nor in the ponding water. Dissolved oxygen exhibited the highest diurnal and seasonal variation from 79% to oversaturation 143% in Winter, including a high standard

variation coefficient (36% during Spring and Summer W). DOC and TOC measurements showed a stable value throughout the campaigns, nevertheless there was a big difference between the inlet and outlet values: 22.2 and 5.0 DOC mg L$^{-1}$ and 46.8 and 43.6 TOC mg L$^{-1}$ respectively.

## 3.2    Sediment characteristics

Based on the texture (clayed silt soil, with 28%, 62% and [10]% clay, silt and sand respectively), the sediment of this CW is similar to the typical Luvisols of the Paris Region. Physical-chemical characteristics of sediments showed some spatial and seasonal variability. The following ones demonstrated relatively high stability: (i) pH (mean value varied from 7.35 to 7.62), (ii) TN content (0.13 to 0.16%), (iii) TC/TN ratio (14.1 to 19.0), and (iv) OM content (2.3 to 3.1%). Also, the sediment is characterized by a relatively high and stable concentration of Ca and Mg (average values varied 2642-3830 and 67-86 mg kg$^{-1}$, respectively).

The contents of $NH_4^+$, $NO_3^-$ and COD, as well as the TC/TN ratio showed a relatively large range of temporal-seasonal variations (see Table S1). $NH_4^+$ contents were three-fold lower in the shallow part to compare with the deeper zones close to inlet and outlet, and likewise, the seasonal variability was relatively high (14-39 mg kg$^{-1}$). The $NH_4^+$ contents were much higher than $NO_3^-$ content (0.26 mg kg$^{-1}$), likely due to denitrification in the anoxic area. The seasonal variability of sediment $NO_3^-$ contents ranged between low content in Autumn and Winter (0.03 mg kg$^{-1}$), intermediate in Spring (0.17 mg kg$^{-1}$) and high in Summer D (0.88 mg kg$^{-1}$). DOC content varied from 59 mg kg$^{-1}$ and 133 mg kg$^{-1}$ in Summer D and Spring, respectively, linked to the development of vegetation (e.g., C source for the root zone). Nevertheless, the TC/TN ratio of the sediment was about 10, considered as not limiting for the denitrifying processes although carbon biodegradability was not assessed.

## 3.3    Gaseous $N_2O$ emissions

### 3.3.1    Seasonal and diurnal patterns of $N_2O$ emissions

Annual average $N_2O$ emission from the Rampillon CW calculated either from automated or manual chambers was relatively low: 3.9 or 3.6 µg $N_2O$-N m$^{-2}$ h$^{-1}$, respectively (Fig. 3).
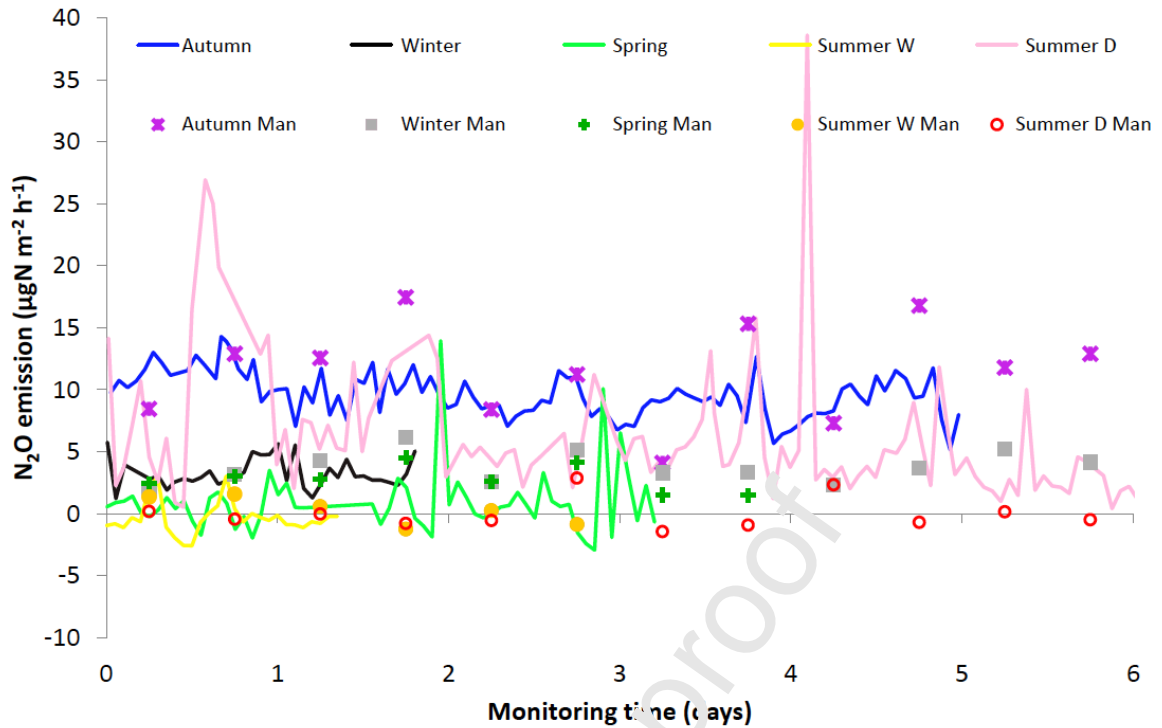
**Figure 3.** The daily evolution of $N_2O$ emissions in Rampillon constructed wetland during the five seasonal campaigns using automated (line) and manual (point, Man in legend) chamber techniques.

The range of individual values was −8.5 to 138 µg $N_2O$-N $m^{-2}$ $h^{-1}$. The seasonal variation of $N_2O$ emissions was moderate (Table 1). Autumn's and Winter's emissions showed relatively stable values similar between the automated and manual chambers: Autumn: 9.2 vs 11.4 and Winter: 2.6 vs 3.8 µg $N_2O$-N $m^{-2}$ $h^{-1}$ respectively. Spring values between the automated and manual chambers were also in the same range (mean values 0.9 vs 1.6 µg $N_2O$-N $m^{-2}$ $h^{-1}$, respectively). Summer W emissions were very low (0.1 and 1.8 µg $N_2O$-N $m^{-2}$ $h^{-1}$ in automated and manual chambers, respectively) showing negative values (−2.59 and −1.26 µg $N_2O$-N $m^{-2}$ $h^{-1}$) at the end of the second day for both techniques. As expected from the differences in chamber positioning (see method section), during the Summer D period, average $N_2O$ fluxes measured with automated chambers were three magnitudes higher than those determined by manual chambers: 5.4 and −0.01 µg $N_2O$-N $m^{-2}$ $h^{-1}$, correspondingly. Summer D data from automated chambers showed the largest variability: individual values varied from −0.9 to 85 µg $N_2O$-N $m^{-2}$ $h^{-1}$.

The graph of cumulative fluxes of gaseous $N_2O$ demonstrates a rather linear increase during all the seasons and over all the chambers (Fig. 4). Although the diurnal variability of $N_2O$ fluxes was large (40%), cumulative emissions measured with high-frequency technique or

calculated based on two-day data with the manual method, were linear for the six consecutive days of the experiment. The comparison between manual and automated techniques shows congruent values in four out of the five campaigns (Fig. 4; Supplementary Figures S1 and S2). Without the values from the exceptional Summer D period the correlation between the data gathered using the two techniques was significant ($R^2 = 0.53$; $p < 0.01$; Fig. S2).
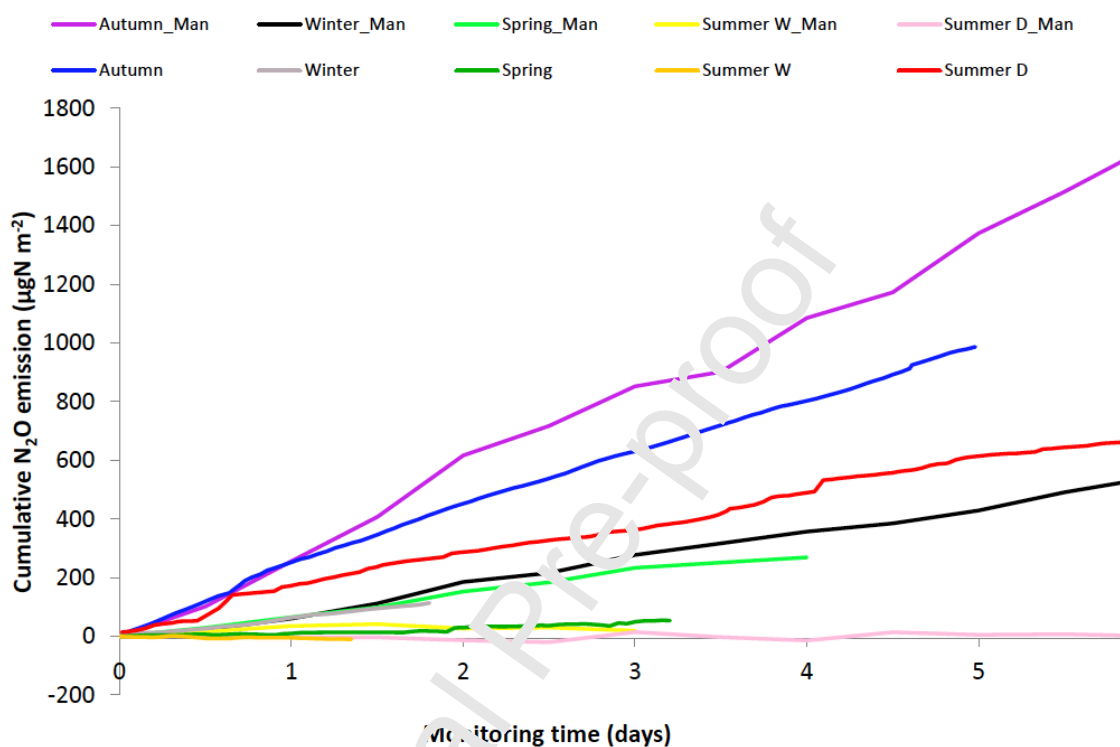


**Figure 4.** Cumulative $N_2O$ emissions measured by the automated and manual chamber techniques calculated across the five campaigns. Man – manual chambers.

The high-frequency measurements enabled us to evaluate the diurnal variability of $N_2O$ emissions. All campaigns exhibited a high variability of emission values within a day: from 38% in Autumn to 176% in Summer W (Fig. S1). The maximum peak emissions often occurred at the maximum solar radiation around 14h for Autumn and Winter, around 16h for Spring and Summer W and largely up to 21h for Summer D (all times are in GMT+2).

### 3.4 Dissolved $N_2O$

Dissolved $N_2O$ exhibited remarkable variations (Fig. S3). During Winter no water samples contained dissolved $N_2O$. Autumn and Spring showed similar values around 12.7 µg $N_2O$-N $L^{-1}$. During the warm Summer W, dissolved $N_2O$ level was intermediate (8.2 µg $N_2O$-N $L^{-1}$), whereas Summer D values were very low (0.32 µg $N_2O$-N $L^{-1}$). From all campaigns, the maximum values of dissolved $N_2O$ were measured in the shallow water column, while in the intermediate zone and the deep zones at inlet and outlet, the values were significantly lower.

**Table 1**. Mean values and standard errors of potential $N_2$ and $N_2O$ fluxes in sediments, $N_2O$ fluxes from open water surfaces measured with automated and manual chambers, through-flow flux of dissolved $N_2O$ (all in µg $N_2O$-N $s^{-1}$), in Rampillon CW during the measurement campaigns. Man – manual chambers.

| | N (number of measure-ments) per season A/W/S/SW/SD/AS | Autumn (A) | Winter (W) | Spring (S) | Summer W (SW) | Summer D (SD) | All seasons (AS) |
|---|---|---|---|---|---|---|---|
| *Sediments* | | | | | | | |
| $N_2$ flux (potential) | 40/40/20/20/40/160 | 5434 (574) | 3809 (319) | 4119 (979) | 5085 (909) | 1304 (412) | 3905 (281) |
| $N_2O$ flux (potential) | 40/40/20/20/40/160 | 25 (6) | 4.3 (1.3) | 85 (47) | 38 (53) | 8 (3) | 44 (20) |
| *Water* | | | | | | | |
| Dissolved $N_2O$ through-flow flux | 21/21/7/7/ 11/67 | 51.8 (17.1) | 0 | 10.1 (3.5) | 11.7 (5.8) | 0 | 19.6 (14.4) |
| *Atmosphere* | | | | | | | |
| $N_2O$ flux | 1192/178/ 584/400/168/ 2522 | 9.2 (1.5) | 2.6 (0.3) | 0.9 (0.3) | 0.1 (0.4) | 5.4 (0.3) | 3.6 (0.1) |
| $N_2O$ flux by Man | 68/70/42/44/80/304 | 11.1 (0.8) | 3.8 (0.3) | 1.6 (0.5) | 1.8 (0.4) | −0.01 (0.3) | 3.9 (0.3) |
| $N_2O$ total average | 1260/248/ 626/444/248/ 2826 | 9.7 (0.2) | 3.2 (0.1) | 0.6 (0.3) | 0.2 (0.3) | 4.8 (0.3) | 3.7 (0.1) |

The annual (mean±SE) amount of through-flow dissolved $N_2O$ (water discharge multiplied by concentration) was $20±14$ µg $N_2O$-N $s^{-1}$, being the highest in Autumn ($52±17$ µg $N_2O$-N $s^{-1}$), moderate in Spring and Summer W, and zero in Winter (no dissolved $N_2O$ in the water) and Summer D (no water flow; Table 1).

### 3.5 Potential $N_2$ and $N_2O$ flux from sediments

In the lab experiments under controlled conditions, the average potential $N_2$ emission ($N_{2potential}$) from the sediment was very high (3905 µg $N_2$-N $m^{-2}$ $h^{-1}$), showing relatively similar values throughout the seasons, except the Summer D period (1304 µg $N_2$-N $m^{-2}$ $h^{-1}$; Table 1). Besides, potential $N_2O$ flux ($N_2O_{potential}$) varied largely from −9 to 1065 µg $N_2O$-N $m^{-2}$ $h^{-1}$), showing the highest average values in Summer W and Spring and lowest ones in Winter and Summer D (Table 1). Consequently, the variation of the ratio $N_2$-$N_{potential}$ : $N_2O$-$N_{potential}$ was also remarkable, from 48:1 in Spring to 886:1 in Winter, making up the average

annual value of 89:1 (Table 1 and Fig. S4).

## 3.6 Relationship of $N_2O$ to environmental parameters

The PCA showed significant differences between groupings of sampling campaigns ($p < 0.001$; Fig. S5A) and subsites ($p < 0.01$; Fig. S5B). All seasons were significantly different from each other ($p < 0.05$), except for the low-temperature seasons (Autumn and Winter) and the high-temperature wet periods (Spring and Summer W). Although significant differences were not detected between subsites, sites from the shallow middle zone were clustered distinctly from the deeper inflow (S1 and S2) and outflow subsites (F1 and F2). The environmental parameters varied between the campaigns and sites, and the gases showed some trends (Fig. S5C). According to the PCA, the $N_2O$ emissions were the highest in Autumn and Winter, although dissolved $N_2O$ from water and potential $N_2O$ emissions from the sediment showed the highest values in Spring and Summer W. Summer D had the smallest potential $N_2$ emissions.

The FPCA (Fig. 5) revealed that gaseous $N_2O$ emissions from the water surface and potential $N_2$ flux from the topsoil were the most impacted by different physicochemical conditions showing many different and significant correlations with them. Gaseous $N_2O$ emissions from the water surface showed strong positive correlation with water $NO_3^-$ concentration, water flow and removal rate of nitrate, and strong negative correlation with water temperature, TOC_W, DOC_W and HRT. Both dissolved $N_2O$ in water and potential $N_2O$ flux from sediment were positively correlated to water temperature and DOC_S Potential $N_2$ flux was positively correlated with water ion concentrations ($Na^+$, $Ca^{2+}$, $SO_4^{2-}$, $Cl^-$, $Mg^{2+}$, $K^+$), sediment-related parameters (pH, $NH_4^+$, DOC), water flow and depth, the removal rate of nitrate, and negatively with sediment-related parameters ($NO_3^-$, OM%, N%, Mg, Ca), DOC_W, and hydraulic residence time.
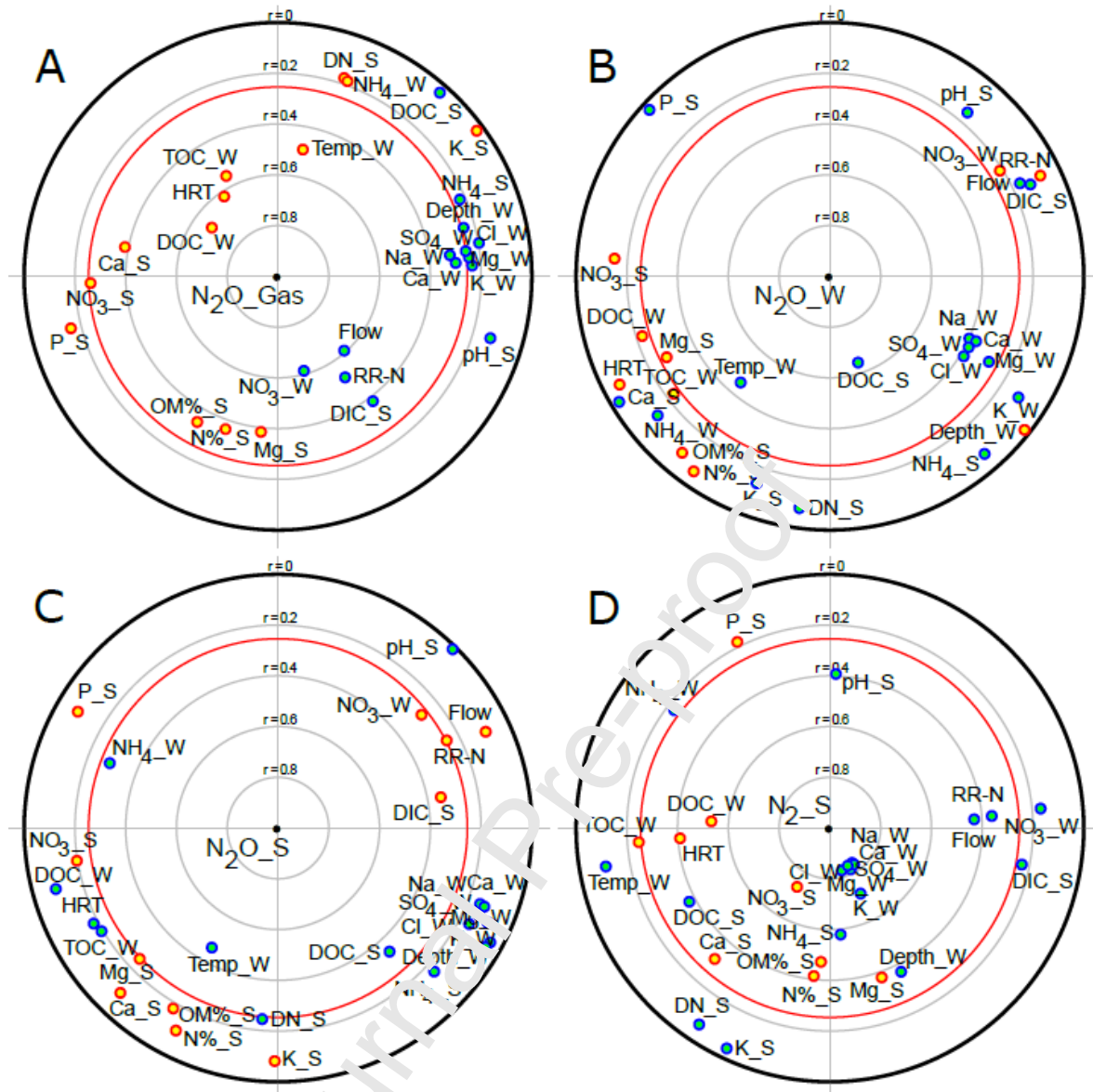
**Figure 5.** Focused PCA of physicochemical variables, with focus on: gaseous N$_2$O flux (A), dissolved in the water N$_2$O_W (B), and sediment N$_2$O_S flux (potential) (C) and N$_2$_S flux (potential) (D) (n = 60). Variables inside the red circle are significantly correlated to the dependent variable with p<0.05. Turqoise filled navy blue circles are positively correlated to the dependent variable and yellow filled red circles are negatively correlated.

## 4    Discussion

### 4.1    Dynamics of gaseous N$_2$O emissions

The average annual emission of gaseous N$_2$O emissions in Rampillon was unexpectedly low: 3.7 µg N$_2$O-N m$^{-2}$ h$^{-1}$. It represents the lowest range of emission values in the CW, which varies between −3 to 650 µg N$_2$O-N m$^{-2}$ h$^{-1}$ (Mander et al., 2014). Our results on seasonal average N$_2$O emission values (0.1 to 9.2 µg N$_2$O-N m$^{-2}$ h$^{-1}$) are lower than those measured in

an analogous CW for treating nitrate-polluted river water in Columbus, OH, USA: averaged values from 7 to 22 µg $N_2O$-N $m^{-2}$ $h^{-1}$ by Hernandez & Mitsch (2006) and from 38 to 415 µg $N_2O$-N $m^{-2}$ $h^{-1}$ by Batson et al. (2012). Furthermore, Glaz et al. (2017) reported low $N_2O$ emission in waste stabilization ponds with likely complete denitrification: from 0.06 µg $N_2O$-N $m^{-2}$ $h^{-1}$ in Western Australia to 0.84 µg $N_2O$-N $m^{-2}$ $h^{-1}$ in Quebec, Canada.

Compared to $N_2O$ emissions from riverine riparian zones in a similar agricultural and geological context, the values found in Rampillon CW are more than one order of magnitude lower (Billen et al., 2020).

In Rampillon, the mean value of annual emission factor (EF; $N_2O$-N emitted / inflow $NO_3^-$-N) was 0.0065%, varying from 0.0025% in Winter to 0.025% in Spring. This is 20 times less than the values reported by Mander et al. (2014) for FWS CWs for wastewater treatment (based on inflow TN values) but comparable with the indirect EFs given by IPCC (2002) and Sawamoto et al. (2005): 0.015% and 0.0024%, respectively. This can be explained by nitrate as the main nitrogen form in agricultural runoff, leading to reduced nitrification or DNRA. This indicates that the source of $N_2O$ is mainly denitrification.

Except for the Summer D period of no water flow (Fig. 2), the average $N_2O$ emissions from automated chambers were slightly lower than those from the manual chambers (Fig. S1). This is likely due to the missing nights from the manual measurements. However, diurnal variability was smaller than the interdiurnal one (Fig. S6). Similar results have been reported from agricultural soils (Benoit et al., 2015). Averaged across the four campaigns, the manual chambers led to a minor overestimate of fluxes compared to the high-frequency measurements with the automated chambers. Also, the range of individual values measured with the automated chambers was larger than from manual ones: from −9 to 138 and from −4 to 31 µg $N_2O$-N $m^{-2}$ $h^{-1}$, respectively. Certainly manual chambers are less adapted for capturing ebullitive fluxes, at least in the deeper water zones and during warm periods (see Fig. S6). However, during our campaigns,, most of the gaseous $N_2O$ emissions in automated chambers were formed through the diffusion process and only 3.6% of emissions (those above 20 µg $N_2O$-N $m^{-2}$ $h^{-1}$) were emitted by ebullition (see Fig. S7). Likewise, the minor role of ebullition in the total release of $N_2O$ from wetlands has been shown by Gao et al. (2013). For a monitoring strategy in remote regions without power supply, manual chambers can be used with confidence.

## 4.2 Dynamics of dissolved $N_2O$

According to several experiments (Garnier et al. 2009; Audet et al., 2017; Marescaux et al., 2018), dissolved $N_2O$ concentrations have been converted into $N_2O$ gas emissions, whereas

higher streamflow significantly increases evasion of $N_2O$ (Aho & Raymond, 2019). Nevertheless, this phenomenon happens at water velocity >0.1 m s$^{-1}$, but in our CW, water velocity was that high only in the Autumn and Spring. That is why we found no significant correlation between the gaseous $N_2O$ emissions and dissolved $N_2O$ in the water nor between the dissolved $N_2O$ and $NO_3^-$ in the water. For comparison, in an agricultural river basin in Eastern China, a significant positive correlation was found between the concentrations of $NO_3^-$-N and dissolved $N_2O$-N. In our case, the characteristics did not help us evaluate $N_2O$ emissions from concentrations in the water directly. The dissolved $N_2O$ during the different campaigns ranged between 0 and 30 $\mu gN_2O$-N L$^{-1}$ leading to a ratio of $N_2O$-N : $NO_3^-$-N between 0 and 0.85%. Likewise, in earlier studies on riverine $N_2O$ budgets, dissolved $N_2O$ saturation levels exhibited a strong seasonal pattern reaching the highest, supersaturated values in the summer and the lowest equilibrium values in the winter (Beaulieu et al., 2010). Positive correlation between dissolved $N_2O$ in the water and potential $N_2O$ flux from the sediment (together with water temperature and DOC_S) reflects the $N_2O$ production at the sediment interface well. However, degassing of dissolved $N_2O$ into the atmosphere is controlled by several other factors such as redox potential of the water column (Beaulieu et al., 2011).

Based on the discharge and dissolved $N_2O$ concentration values in the inlet, we calculated the throughflow of dissolved $N_2O$, which resulted in 20±14 $\mu g$ N s$^{-1}$. At the annual scale, the average throughflow of dissolved $N_2O$ was 0.62 kg $N_2O$-N yr$^{-1}$. Likewise, Bruun et al. (2017) showed that dissolved $N_2O$ flow was higher than gaseous $N_2O$ emission. In further perspective, specific measurements of gas transfer velocity are required to enable an evaluation of gaseous $N_2O$ emissions dissolved in stagnant waters.

The mean $N_2O$-N$_{dissolved}$: $NO_3^-$-N ratio was 0.16%, with a peak of 0.58% in the Autumn. This shows that the proportion of dissolved $N_2O$-N in Rampillon CW is relatively small and the potential degassing downstream is low. Correspondingly, these values were up to 15 and 58 times lower than those found in a study conducted within riparian zones along the agriculturally impacted San Joaquin River, CA, USA (Hinshaw & Dahlgren, 2016). However, the higher throughflow of dissolved $N_2O$ than the emission of gaseous $N_2O$ could mean that the $N_2O$ produced in the CW will be emitted downstream (Outram & Hiscock, 2012).

To avoid this, we advise a low $N_2O$-N$_{dissolved}$: $NO_3^-$-N concentration ratio when designing a CW for nitrate removal.

**4.3 Relationship between gaseous $N_2O$ fluxes and environmental factors**

Seasonal variations of gaseous $N_2O$ fluxes are controlled by abiotic factors such as temperature and hydrological regime but also by DOC in the water. $N_2O$ emission increased with higher $NO_3$ water concentration and amplified with high temperature, high flow and consequently short HRT. Indeed, Autumn and Winter, albeit with similar water flows, led to strongly different $N_2O$ emissions, due to differences in temperature, and nitrate concentrations. Dissolved $N_2O$ also confirmed the tendency peaked in Autumn. One explanation can be different availability of DOC due to $NO_3^-$ reduction in the water. DOC content was the lowest in Autumn, which was probably depleted in denitrification. It is well-known from previous studies that below some critical threshold of C:N ratio the $N_2O$ emission from soils increases exponentially (Chung & Chung, 2000; Klemedtsson et al., 2005). In our study, significant positive nonlinear correlation ($R^2 = 0.56$) was found between the DOC:$NO_3^-$-N ratio and $N_2O$ emissions: the emission rapidly increases below a ratio of less than 5 (Fig. 6). In mid-Autumn (November), after a dry period, we can expect that most of internal DOC (formed from vegetation decay in the CW) and external DOC (possibly leached from soils in the catchment) was already consumed by denitrification and could not fuel the complete denitrification process. Therefore, denitrification stopped at the stage of $N_2O$ emissions. In Autumn, we observed the highest $N_2O$ emissions.
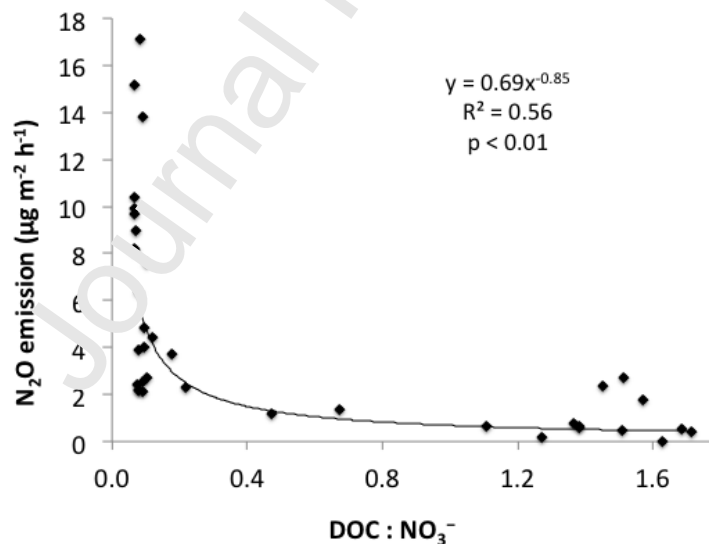


$$y = 0.69x^{-0.85}$$
$$R^2 = 0.56$$
$$p < 0.01$$

**Figure 6.** Relationship between $N_2O$ emissions and DOC : $NO_3^-$-N as C : N ratio in the inlet of Rampillon CW.

Spring and Summer W with the highest HRT left enough time to complete the denitrification with less $N_2O$ emissions (Table 1). It even caused consumption of $N_2O$ (negative emission values for 52% of the time of Spring up to −8.8 µg $N_2O$-N m² h⁻¹) of $N_2O$, which dropped

below the saturation value in the water leading to dissolution from the atmosphere to the water column. Although this kind of consumption has been observed in several aquatic ecosystems (Beaulieu et al., 2011; Mander et al., 2014; Soued et al., 2016), it seems to be the first time that a CW is shown to turn to a sink of $N_2O$ during warm temperature and low flow. A similar seasonal pattern to Rampillon was demonstrated in a study of Lake Neusiedl, Austria, where the highest emission rates from the water (25-48 µg $N_2O$ $m^{-2}$ $h^{-1}$) were observed in late autumn whereas $N_2O$ was taken up mainly in the spring and occasionally in late summer and even early autumn (Soja et al., 2014).

Although nitrate concentrations in Rampillon water are high, gaseous $N_2O$ emission is relatively low, which can be explained by a complex of environmental factors and water regime control. One of the basic determinants is the high pH and high $Ca^{2+}$ concentration in both water and sediments (Table S1). These conditions are beneficial for the transformation of $N_2O$ to $N_2$, i.e., a complete denitrification process.

Water flow, water table and HRT are important factors influencing gaseous $N_2O$ fluxes in Rampillon CW (Fig. 4 and Fig. S4). From previous studies, it is known that the fluctuating water table and pulsing water regime are enhancing $N_2O$ emissions from various wetlands (Jia et al., 2011; Mander et al., 2011; Kaushal et al., 2014). Most likely, this is due to the better aeration of temporarily inundated wetland sediments, which inhibits the full path of denitrification and enhances $N_2O$ release from other processes such as nitrification (Ji et al., 2011), DNRA (Jahangir et al., 2017; Wang et al., 2018) and nitrifier denitrification (Masta et al., 2020). Expectedly, one could see that phenomenon in Rampillon where the water table frequently fluctuates. However, here the impact of pulsing was not so clear and appeared only at extremely low water periods when half of the bed was dry and the water stayed only in deeper parts (Summer D and also the beginning of Summer W sessions; Fig. 1, Table 1). Assumably, such short-term aeration in the dry parts could not inhibit the denitrification process (see Jørgensen & Elberling 2012) and might enhance the fast growth of vegetation. Young shoots of macrophytes and also filamentous algae mats covered the dried parts in a few days. Thus, plants might compete for N source with denitrifiers, even in such an N-saturated system ($NH_4^+$ content in sediment was highest during Summer D). This is supported by the fact that in Summer D, the formation of $N_{2potential}$ in the sediments was the lowest among all the campaigns (1304 µg N $m^{-2}$ $h^{-1}$, i.e., 34% of the mean annual value). Likewise, $N_2O_{potential}$ was very low (8 µg N $m^{-2}$ $h^{-1}$; i.e., 18%; Table 1, Fig. S4).

Several studies have analysed the influence of vascular plants in $N_2O$ emissions from wetlands. It depends on many factors in an equilibrium between the oxygen-ruled microbial

processes (first of all denitrification vs. nitrification), plant uptake and oxygen supply to the rhizosphere, which depends on the presence of aerenchymous tissue, and age, biomass and roots development of the plants. Thus, depending on the equilibrium, wetland plants can increase (Chen et al., 2011; Jørgensen et al., 2012) or decrease (Hu et al., 2016) $N_2O$ emissions.

As expected, the gaseous $N_2O$ emissions were strongly linked to nitrate concentration and nitrate removal rate, DOC concentrations, and pH (Figs. 4 and 5), high pH favoring $N_2O$ reduction (Hénault et al., 2019; Billen et al., 2020).

The annual average $N_2O$ emission and the estimated flux of dissolved $N_2O$ in through-flow water were 0.32 and 0.62 kg $N_2O$-N ha$^{-1}$ yr$^{-1}$, respectively. Comparing the gaseous flux value to 938 kg $NO_3^-$-N ha$^{-1}$ yr$^{-1}$ denitrified, we get a low $N_2O$-N:$NO_3^-$-N ratio of 0.033%.

## 5   Conclusions

Our first hypothesis was not supported: in contrast to the expected high $N_2O$ emissions, the Rampillon CW was a weak source of $N_2O$ emitted to the atmosphere. The ratio of $N_2O$-N emitted to $NO_3^-$-N removed was 0.034%. Likewise losses of dissolved $N_2O$ were relatively small in the system: the ratio of dissolved $N_2O$-N in the outflow to $NO_3^-$-N removed was 0.066%..The second hypotheses was supported, our analysis showed a high denitrification potential reflected by large potential $N_2$ fluxes and a high $N_2$:$N_2O$ ratio in the sediments. All the results indicate complete denitrification in the CW. Since there was no significant diurnal pattern in the $N_2O$ emission, the high and low frequency measurements of $N_2O$ emission gave comparable results. We found clear seasonal variation of $N_2O$ emissions showing the highest values in the autumn when organic carbon was mineralised, fuelling denitrifiers. This variation correlated with the temporal pattern of $NO_3^-$-N concentrations in inlet water. In addition, streamflow and nitrate reduction rate were strongly positively correlated with the gaseous $N_2O$ fluxes. Water temperature, TOC and DOC in the water and hydraulic retention time showed strong negative correlations with $N_2O$ emissions.

Our study represents an off-stream CW where high denitrification potential in sediments and a complete denitrification pathway ensure satisfactory nitrate removal but very low $N_2O$ emissions, still showing a seasonal pattern. The overall $N_2O$ balance is not significantly affected by the flow of dissolved $N_2O$ from the wetland. The results convince us that climate-friendly design of off-stream CWs for treating nitrate-polluted agricultural run-off should include both deep and shallow zones to maintain nitrogen removal processes in fluctuating hydrological conditions. Further studies on $N_2O$ fluxes in analogous CWs should focus on the role of macrophytes, algae and microorganisms in the N transformation processes.

Author Agreement

ÜM, KS, JT, CC and MM (Mart Muhel) designed the study. ÜM, KS, JT, CC, ME, MM (Martin Maddison), MM (Mart Muhel), JG, JDL, EU, and KR performed the study. ÜM and JT wrote the article, KS, ME JG and JP commented on draft versions of the article. All authors have approved the final article.

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**Declaration of competing interests**

X The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

☐The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:
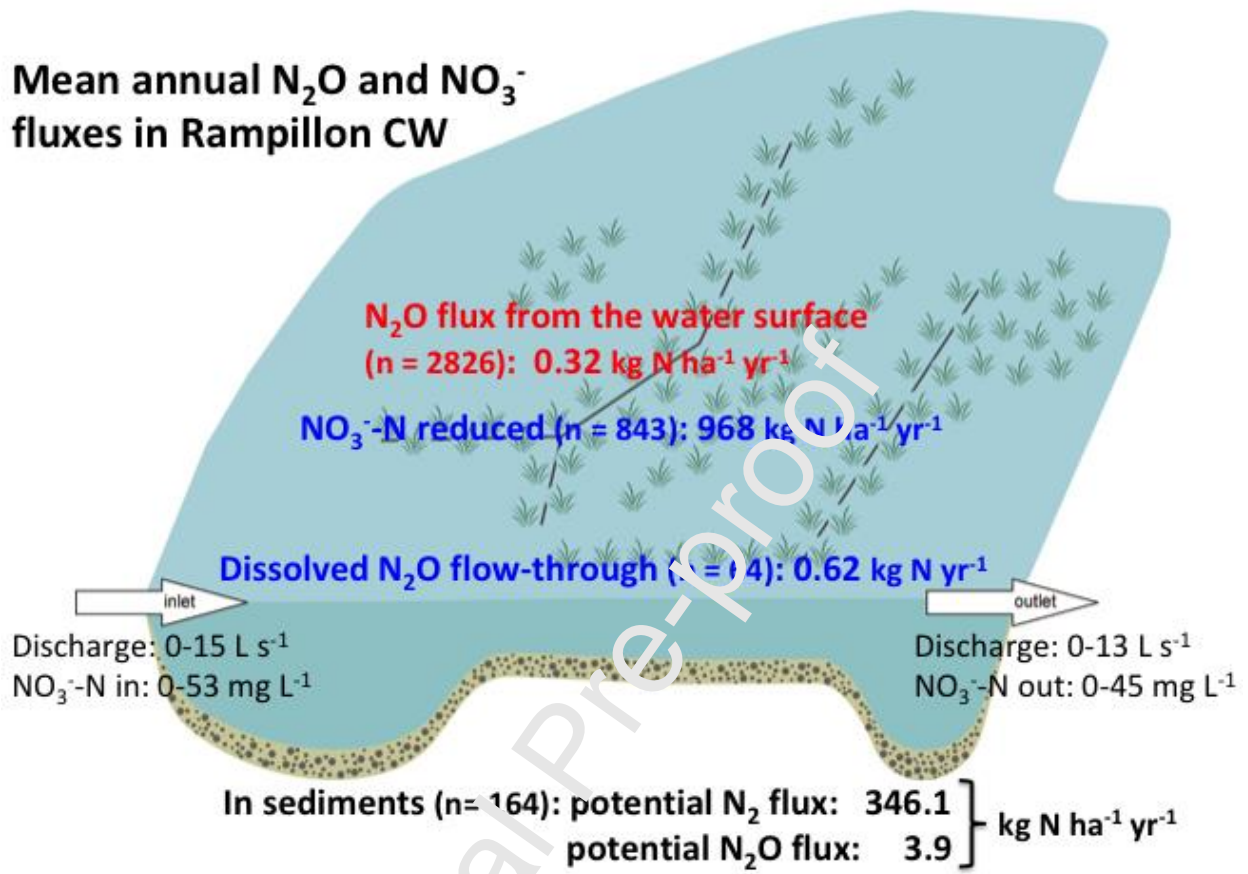
**CRediT author statement**

**Ülo Mander**: Conceptualization, Methodology, Investigation (field studies), Validation, Writing, Reviewing, Editing, Supervision; **Julien Tournebize**: Conceptualization, Methodology, Investigation (field studies) , Validation, Writing, Editing, Supervision; **Mikk Espenberg**: Software, Data curation, Validation, Vizualization, Writing; **Cedric Chaumont**: Methodology, Investigation, Validation; **Raili Torga**: Investigation (field studies), Data curation; **Josette Garnier**: Methodology, Investigation (laboratory analyses), Data curation, Validation, Writing; **Mart Muhel**: Methodology, Investigation (field studies); **Martin Maddison**: Methodology, Investigation (laboratory analyses); **Jérémie D. Lebrun**: Software, Data curation, Writing; **Emmanuelle Uher:** Software, Data curation; **Kalle Remm**: Software, Data curation; **Jaan Pärn**: Writing, Reviewing, Editing; **Kaido Soosaar**: Methodology, Investigation (field studies), Validation, Writing, Reviewing.

**Graphical Abstract**



Mean annual $N_2O$ and $NO_3^-$ fluxes in Rampillon CW

$N_2O$ flux from the water surface (n = 2826): 0.32 kg N ha$^{-1}$ yr$^{-1}$

$NO_3^-$-N reduced (n = 843): 968 kg N ha$^{-1}$ yr$^{-1}$

Dissolved $N_2O$ flow-through (n = 14): 0.62 kg N yr$^{-1}$

inlet

outlet

Discharge: 0-15 L s$^{-1}$
$NO_3^-$-N in: 0-53 mg L$^{-1}$

Discharge: 0-13 L s$^{-1}$
$NO_3^-$-N out: 0-45 mg L$^{-1}$

In sediments (n= 164): potential $N_2$ flux: 346.1
potential $N_2O$ flux: 3.9 } kg N ha$^{-1}$ yr$^{-1}$

**Highlights**

- Low $N_2O$ fluxes were found in an off-stream wetland treating $NO_3^-$ polluted run-off
- A high potential of $N_2$ formation is supported by high $N_2$-N : $N_2O$-N ratio in sediments
- From the potential $N_2O$ emission in sediments, only 9% was emitted to the atmosphere
- Analogous wetlands can be efficient in nitrate removal from agricultural runoff