

Reduction of pulse "antinutritional" content by optimizing pulse canning process is insufficient to improve fat-soluble vitamin bioavailability

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1	Reduction of pulse "antinutritional" content by optimizing pulse canning process is
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- 24 Abstract
- 25

26 Some bioactive compounds found in pulses (phytates, saponins, tannins) display antinutritional properties and interfere with fat-soluble vitamin bioavailability (i.e., 27 28 bioaccessibility and intestinal uptake). As canned chickpeas are consumed widely, our aim 29 was to optimize the chickpea canning process and assess whether this optimization influences 30 fat-soluble vitamin bioavailability. Different conditions during soaking and blanching were 31 studied, as was a step involving prior germination. Proteins, lipids, fibers, vitamin E, lutein, 5-32 methyl-tetrahydro-folate, magnesium, iron, phytates, saponins and tannins were quantified. Bioaccessibility and intestinal uptake of vitamin D and K were assessed using in vitro 33 34 digestion and Caco-2 cells, respectively. Significant reductions of phytate, saponin and tannin 35 contents (-16 to -44%), but also of folate content (up to -97%) were observed under 36 optimized canning conditions compared with the control. However, bioaccessibility and cellular uptake of vitamin D and K remained unaffected after in vitro digestion of test meals 37 38 containing control or optimized canned chickpeas.

39

40 **Keywords:** Chickpeas; germination; phytates; tannins; saponins; vitamin D; vitamin K;

41 bioaccessibility; Caco-2 cells.

42 **1.** Introduction

43

44 Pulses display interesting nutritional properties, as they are good sources of proteins, fibers 45 and micronutrients (Margier et al., 2018). Pulses are also a significant source of bioactive 46 compounds such as phytates, saponins and tannins (Margier et al., 2018), the consumption of 47 which demonstrates some interesting lipid-lowering properties. For instance, dietary fibers 48 positively impact postprandial lipemia (Desmarchelier et al., 2019; Lee et al., 2020) and 49 cholesterolemia (Schoeneck & Iggman, 2021). Phytates at high concentrations and tannins 50 were shown to interfere with human lipase activity (Antoine et al., 2021). Finally, saponins 51 have often been described as anti-obesogenic compounds (Marrelli et al., 2016), partly 52 because they also interact with lipase activity (Antoine et al., 2021; Marrelli et al., 2016). 53 However, these compounds can also be considered "antinutritional", as they can negatively 54 interact with micronutrients such as minerals. In particular, phytates and tannins are divalent 55 cation quenchers that can decrease iron and zinc bioavailability (Antoine et al., 2021; 56 Lestienne et al., 2005). Furthermore, we recently showed that fibers, phytates, tannins and 57 saponins from pulses can reduce fat-soluble vitamin bioavailability in vitro, and more 58 precisely that they can modulate the bioavailability of vitamin K (Margier et al., 2019) and 59 vitamin D (Antoine et al., 2021).

Different strategies can be set up to decrease pulses' "antinutritional" content, one of the most straightforward being the optimization of pulse cooking/transformation methods (Margier et al., 2018). As canned pulses are consumed widely, we decided to investigate the optimization of canning process. Chickpeas were selected as the pulse to be studied. The objective of this study was thus i) to identify and optimize the parameters of the canning process that could reduce chickpeas' levels of bioactive compounds displaying antinutritional properties (phytates, saponins and tannins); and ii) assess whether the decreases in phytates, saponins

- 67 and tannins resulting from canning process optimization were sufficient to improve the
- 68 bioavailability of vitamins D and K in test meals containing canned chickpeas.

- 69 2. Material and methods
- 70

71 **2.1.** Chemicals

72 Cholecalciferol (vitamin D_3), phylloquinone (vitamin K_1), γ -tocopherol, retinyl acetate (all 73 >95% pure), phytic acid sodium salt hydrate, soyasaponin I, 4,5-dimethylthiazol-2-yl)-2,5-74 diphenyltetrazolium bromide (MTT), catechin 4-(dimethylamino)cinnamaldehyde (DMACA), 2-(N-cvclohexylamino) ethane sulfonic acid (CHES), 4-(2-hydroxyethyl)-1-piperazine 75 76 ethanesulfonic acid (HEPES), α-amylase from bacillus sp., protease from Streptomyces 77 griseus, and 5-methyltetrahydrofolate (5-CH3-H4 folate calcium salt) were purchased from 78 Sigma-Aldrich (Saint-Quentin-Fallavier, France). Lutein (>95% pure) was from 79 Extrasynthèse (Genay, France). Chicken pancreas conjugase was from R-Biopharm 80 (Darmstadt, Germany). Isio4 oil was from Lesieur (Asnières-sur-Seine, France). Iron (III) chloride hexahydrate and vanillin (all 99% pure) were from Acros Organics (Noisy-le-Grand, 81 82 France). Dulbecco's modified Eagle's medium (DMEM) containing 4.5 g/L glucose and 83 trypsin-EDTA (500 200 mg/L, respectively), non-essential and amino acids. 84 penicillin/streptomycin and phosphate-buffered saline (PBS) were purchased from Life 85 Technologies (Illkirch, France). Fetal bovine serum (FBS) came from PAA (Vélizy 86 Villacoublay, France). Chickpeas (Cicer arietinum 'Eldorado'), were produced locally and 87 provided by the GPS France cooperative (Manosque, France). All solvents were HPLC grade 88 and obtained from Carlo Erba (Peypin, France).

- 89
- 90 2.2. Chickpea preparation

91 2.2.1. Control canning process

92 The control canning protocol was devised by the Centre Technique de Conservation des
93 Produits Agricoles (CTCPA, Avignon, France) and implemented in their pilot plant. In brief,

chickpeas were soaked in low-ionized water (hardness titration [HT]=30 °f) at 15 °C at a ratio
of 1:3 (Chickpeas:water; w/w) for 14h. Afterwards, immersion blanching was done at 90 °C
for 5 min. Chickpeas were then conditioned in cans (200 g), and hot brine (0.5% salt, 70°C) at
a ratio of 86:114 (w:w) was added. The cans were then sterilized at 127 °C for 16 min, and
subsequently cooled to 30 °C for 10 min.

99 2.2.2. Preliminary tests

Preliminary tests were performed to address the effect of modifying single parameters.
Soaking and blanching parameters were thus modified independently (Table 1). An
"antinutritional" compound analysis was conducted after each step, but not on the final
canned products.

• *Modification of the soaking step (Table 1)*

105 Hot water soaking (water at 80°C) was done in a Stephan® Combicut TC System (Stephan 106 Bvba, Nazareth, Belgium) for 2h. The "soft" water (HT=6 °f) was Volvic water (Société des 107 eaux de Volvic - Danone group, Volvic, France) and the "hard" water (HT=184 °f) was 108 Hépar water (Nestlé Waters Vosges, Vittel, France). Chickpeas were soaked at different 109 ratios: 1:3, or 2:3, or 1:10 (chickpeas:water; w:w). An agitation step could be added either 110 with a pump (Yamada, Arlington Heights, IL, USA) during soaking at 15 °C or with paddles 111 in a Stephan® Combicut TC during soaking at 80°C. Water could be replaced at the midpoint 112 of the soaking time. Citric acid (La Littorale - Erbslöh, Servian, France) and sodium 113 bicarbonate (Cérébos, Levallois-Perret, France) were used for a soaking at pH=5 and pH=9, 114 respectively. Finally, a test was done after the addition of a germination step performed before soaking and blanching: chickpeas were immersed for 12 h in water (1:3, w:v), then kept 115 116 between thick layers of damp cotton cloth and allowed to germinate for 3 days at room 117 temperature (20-25°C).

• *Modification of the blanching step (Table 1)*

119 Tests on the blanching step were carried out on chickpeas previously soaked via the control 120 process. A longer blanching time was also tested (20 min). Moreover, the immersion 121 blanching was changed into either steaming or cascading. Steaming consisted of a 5-min 122 exposure to steam using a steam oven (Bourgeois, Faverge, France). Cascading was 123 performed using vertically flowing water in a steam-heated autoclave (Auriol, Marmande, 124 France).

125 2.2.3. Optimization tests on the global process (Table 2)

126 Once the preliminary tests were analyzed, different processes combining the most efficient 127 soaking and blanching parameters were tested. For reasons of technological practicality, hot 128 soaking was done with a high chickpea-to-water ratio (1:10) for a shorter time (2h). 129 Conditioning and sterilization parameters were not modified. Contrary to the preliminary 130 tests, bioactive compounds were only analyzed in the final products.

131

132 2.3. Texture analysis

133 The texture of the canned chickpeas was analyzed to check whether process modifications 134 induced changes in chickpea texture compared to chickpeas treated with the control process. 135 A multipurpose texture analyzer TA-plus (Lloyd Instruments, Farenham, UK) was used with 136 a method adapted from Chenoll and colleagues (Chenoll et al., 2009). The peak compression 137 force (N) was recorded as the hardness of the chickpeas.

138

139 2.4. In vitro digestion experiments

140 The test meals used for in vitro digestion experiments were made of 3.35 g of canned chickpeas and 100 mg of Isio 4 oil. Isio 4 oil contributed minor amounts of fat-soluble 141 142 vitamins to the meal, without consequences to data interpretation, as oil was present in all 143 conditions. Each test meal was supplemented with 125 µM of fat-soluble vitamin D or K dissolved in the oil. The in vitro digestion experiments were conducted as previously described (Malapert et al., 2018). Aliquots of digesta and mixed micelles were taken and stored at -80 °C under a nitrogen atmosphere until analysis. Bioaccessibility was calculated as the percentage of vitamin recovered in mixed micelles divided by the amount of vitamins recovered in the digesta at the end of the digestion.

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- 150

0 **2.5.** Fat-soluble vitamin absorption by intestinal cells in culture

151 2.5.1. Preparation of micelles from in vitro digestion

Preliminary cytotoxicity tests were conducted. Briefly, Caco-2 cells were seeded and grown for 3 days in 96-well plates, prior to a 1 h-incubation with diluted micelles obtained from in vitro digestion. The cell culture medium was then removed, and a MTT solution (1 mg/mL) was added for 2 h at 37 °C. The supernatant was then replaced by DMSO (100 μ L per well) and the absorbance was measured at 540 nm. In accordance with these tests, fat-soluble vitamin micelles obtained from in vitro digestions were diluted in DMEM at a 1:6 ratio (v:v) for further cell experiments (data not shown).

159 2.5.2. Cell experiments

160 Caco-2 clone TC7 cells were cultured as previously described (Goncalves et al., 2016). For 161 each experiment, cells were seeded and grown for 2 weeks in 6-well plates (24 mm diameter, 162 1 μm-pore-size polycarbonate membrane; Becton Dickinson, Le Pont-de-Claix, France). 163 Twelve hours prior to each experiment, the complete medium (i.e., DMEM supplemented 164 with 16% FBS, 1% non-essential amino acids and 1% antibiotics) was replaced by serum-free 165 complete medium. At the beginning of the experiment, cell monolayers received diluted 166 micelles obtained from in vitro digestions. Cells were incubated for 1h at 37 °C with the 167 micelles. Next, cells were washed with 0.5 mL of ice-cold PBS to eliminated adsorbed fat-168 soluble vitamins, then scraped and collected in 0.5 mL of ice-cold PBS. Vitamin absorption 169 percentage was calculated by dividing the amount of vitamin recovered in cells by the total amount of vitamin delivered to the cells (i.e., vitamin found in the medium plus vitamin remaining in the cells at the end of the experiment). All samples were stored at -80 °C until analysis.

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174 **2.6.** Chickpea lyophilization

The canned chickpeas were first lyophilized for 48h in a Cosmos freeze dryer (Cryotec, SaintGély-du-Fesc, France) and then ground into powder using a Proline 700W blender (Proline,
Paris, France).

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179

9 **2.7. Bioactive compound assays**

180 *2.7.1. Phytate assay*

181 Phytates from chickpea powder were analyzed using a spectrophotometric method derived 182 from that published by Dost & Tokul (Dost & Tokul, 2006). Briefly, phytates were extracted 183 from 200 mg of powder with 10 mL of chloridric acid (0.5 M, Fisher Scientific, Saint-184 Herblain, France) for 1 h at room temperature (20-25°C). The extract was then centrifuged 5 185 min at 800×g and the supernatant was recovered. One hundred μ L of supernatant was added 186 to 0.9 mL of water and 2 mL of iron (III) chloride hexahydrate. This mixture was mixed for 187 2h and 30 min at 40 °C and centrifuged 5min at 800×g. The absorbance of the supernatant 188 was measured at 480 nm against water using a Shimadzu UV 8000 spectrophotometer 189 (Shimadzu, Marne-la-Vallée, France). The results were expressed as phytic acid equivalent.

190 2.7.2. Saponin assay

Saponins were analyzed by a spectrophotometric method inspired by the method previously published by Li and colleagues (Li et al., 2010). The first step was the extraction of saponins from the chickpea powder. One gram of flour was added to 5 mL of methanol (80% in water), mixed for 24 h at room temperature (20-25°C) and centrifuged 5 min at 800×g. This step was repeated twice. Supernatants (0.2 mL) were completed with 0.3 mL of methanol (80% in water), 0.5 mL of vanillin (8% in water), and 5 mL of sulfuric acid (72%). The mixtures were incubated at 60°C for 10 min and placed into an ice bath. The absorbance of the mixtures was measured at 544 nm against methanol (80% in water). The results were expressed as soyasaponin I equivalent.

200 2.7.3. Tannin assay

Tannins were extracted from 0.12 g of chickpea powder with 12 mL of water. Two milliliters of a water/methanol mixture (1:1; v:v), 4 mL of tannin extract, and 1 mL of DMACA solution were mixed together. After 20 min, samples were centrifuged 5 min at 800×g and the absorbance of the supernatants was measured at 640 nm against a water/methanol mixture (1:1). The results were expressed as catechin equivalent.

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2.8. Lipid, protein, fiber and mineral assays

Lipids were extracted from the chickpea powder. Briefly, 0.5 g of powder, 2 mL of distilled water, and 5 mL of ethyl acetate were mixed for 6 min. The extract was centrifuged 15 min at 1100 \times g and the supernatant was recovered. This procedure was repeated once with the addition of ethyl acetate. To wash the extract, 3 mL of water and 0.5 mL of ethanol were added to the supernatant, and the mixture was vortexed for 5 min and centrifuged 10 min at 1100 \times g. Pooled supernatants were evaporated to dryness under nitrogen. Total lipids were measured by weighing the dry extracts.

The analysis of protein, fibers, magnesium and iron analysis was subcontracted to Phytocontrol Agrifood (Nîmes, France), which is accredited to conduct these specific analyses (Cofrac #1-1904).

218

219 **2.9.** Vitamin assay

220 2.9.1. Vitamin E (y-tocopherol) and lutein from chickpeas

221 Vitamin E (γ -tocopherol) and lutein were extracted from 250 mg of chickpea powder using 222 the following method: 1 mL of distilled water was added to the sample. The internal standard 223 (retinyl acetate) was added to the sample in 1 mL of ethanol. The mixture was extracted twice 224 with 4 mL of hexane. After centrifugation (2500×g, 10 min at 4 °C), the hexane phase was 225 evaporated to dryness under nitrogen. The dried extract was dissolved in 200 µL of methanol-226 dichloromethane (65:35, v:v). A final volume of 180 µL was used for HPLC analysis. Fat-227 soluble vitamins and carotenoids were separated as previously described (Gleize et al., 2012). 228 Luteine and γ -tocopherol were detected at 445 and 298 nm, respectively, and identified by 229 retention time in comparison with pure standards. Quantification was performed using 230 Chromeleon 6.8 software (ThermoFisher Scientific, Villebon sur Yvette, France) to compare 231 the peak area with standard reference curves.

232

2.9.2. Vitamin D and vitamin K assay from in vitro digestion and cell samples

Vitamin D and vitamin K were extracted from 200 μ L of digesta and mixed micelles, and from 500 μ L of apical medium and cells. The internal standard was retinyl acetate. Both extraction and HPLC analysis were performed according to Antoine and al. (Antoine et al., 2021) and Goncalves et al. (Goncalves et al., 2014), respectively. Vitamins were identified by spectral analysis and/or retention time and co-injection in comparison with pure standards. Quantification was performed using Chromeleon 6.8 software to compare the peak area with standard reference curves.

240

2.9.3. Folate assay

Folates from chickpea powder were analyzed by a HPLC method derived from the method published by Ruggeri and colleagues (Ruggeri et al., 1999). Powder (0.75g) was added to 10.5 mL of extraction buffer in ebullition (HEPES buffer, pH 7.85, containing 2% (w:v) sodium ascorbate and 10 mM 2-mercaptoethanol). After flushing the samples with nitrogen, they were heated for 10 min in a boiling water bath and homogenized with an Ultra Turrax device 246 for 30 s at a speed of 13500 rpm. Extraction buffer was then added (qsp 15 mL). The pH was 247 adjusted to 4.9 and 600 µL of chicken pancreas conjugase, 120 µL of rat serum and 600 µL of 248 α-amylase were added. Samples were incubated under agitation at 37°C for 3 h. The pH was 249 adjusted to 7.0, then 1.2 mL of protease were added, and samples were incubated for 1h as 250 previously described. Enzymes were then inactivated by boiling for 10 min. Samples were 251 centrifuged for 30 min at 2000×g. Supernatants were transferred to a strong anion exchanger 252 (SAX) column (ThermoFisher Scientific) to purify folates (solid phase extraction). A final 253 volume of 20 µL was used for HPLC analysis. The HPLC system comprised a Dionex 254 separation module (P680HPLC Pump and ASI-100 Automated Sample Injector, Dionex, Aix-255 en-Provence, France) and a Jasco fluorimetric detector (Lisses, France). Analyses were 256 performed on a HypersilTM ODS C18 column (150×4.6 mm; 5µm) coupled with a 10×4 mm 257 C18-5µm HypersilTM ODS guard column (FisherScientific, Illkirch, France) kept at a 258 constant temperature (40 °C). The mobile phase consisted of a gradient of acetonitrile and 259 30mM phosphate potassium buffer at pH 2.6. The flow rate was 0.8 mL/min. The run time 260 was 35 min. The gradient profile of the mobile phase was set at 5-95% and changed linearly 261 to 17-83% from minute 9 to 30. The mobile phase then changed back to 5-95% from minute 262 30 to 35. 5-methyltetrahydrofolate, which is the main form of folate found in chickpeas 263 (Ruggeri et al., 1999), was detected at 356 nm after excitation at 290 nm, and identified by 264 retention time compared with a pure standard. Quantification was performed using 265 Chromeleon software to compare the peak area with standard reference curves.

266

267 2.10. Statistics

For all groups, data were expressed as mean ± SEM. Statistical analyses were performed using GraphPad Prism software, version 8.4.3 (GraphPad Software, San Diego, California, U.S.A.). Prior to ANOVA (fixed-effects models), data were tested for equality of variances. The distribution of all dependent variables was graphically checked and was considered

- 272 normal. Tukey's test was used as a post-hoc test for pairwise comparisons. Values of $p \le 0.05$
- 273 were considered significant.

- 274 **3. Results**
- 275

3.1. Chickpea contents in phytates, saponins and tannins can be decreased by modifying canning parameters

278 3.1.1. Phytate and saponin contents

Figures 1A and 1B show the results of preliminary tests. We observed a significant decrease of phytate content after hot soaking (PT1), mixed soaking (PT7) and acid soaking (PT9) (- $36.0 \pm 4.9\%$, $-25.0 \pm 7.0\%$, $-25.0 \pm 3.2\%$, p≤0.05, respectively), but also after soaking with low-ionized water (PT4), with a low chickpeas-to-water ratio (PT6) or after water replacement (PT8) (-19.0 ± 2.6\%, -18.0 ± 6.3\%, -17.0 ± 2.7\%, p≤0.05, respectively) (Figure 1A). Moreover, Figure 1B shows that phytate content was significantly reduced when the chickpeas were blanched by cascading (PT14, -7.0 ± 1.7\%, p≤0.05).

Concerning saponins, there was no significant diminution of their concentrations when soaking and blanching parameters were modified. Saponin content was slightly reduced when soaking was done with low-ionized water (PT4) or with a low chickpeas-to-water ratio (PT6) (-13.0 \pm 3.1%, -10.1 \pm 1.2%, respectively), but these values remained non-significantly different (Figure 1 A and 1B).

291 We then explored the phytate and saponin content in canned and sterilized chickpeas after 292 these different parameters were combined. As shown in Figure 1C, phytate and saponin 293 concentrations were both significantly decreased when chickpeas were canned after i) hot 294 soaking with water replacement followed by immersion blanching (T3) (-26.0 \pm 1.9%, -29.0 \pm 295 1.7%, p≤0.05, respectively), ii) hot mixed soaking followed by cascading blanching (T8) (-296 $37.0 \pm 1.6\%$, $-35.0 \pm 2.2\%$, p ≤ 0.05 , respectively), and iii) hot soaking with water replacement 297 followed by cascading blanching (T9) (-28.0 \pm 0.7%, -39.0 \pm 2.4%, p≤0.05, respectively). 298 Moreover, we observed a significant reduction of saponin content after soaking at 15 °C with i) low-ionized water and blanching by cascading (T10), ii) agitation blanching by immersion
(T5), iii) water replacement and cascading blanching (T12) and vi) germination (T13) before
soaking and blanching (-21.0 ± 2.1%, -17.0 ± 2.1%, -16.0 ± 1.7%, -29.0 ± 6.2%, p≤0.05,
respectively). These 7 conditions were chosen for further analysis.

303 *3.1.2. Tannin content*

Our results show that tannin content was reduced in 3 conditions among 7 (Figure 1D): i) hot soaking with water replacement followed by immersion blanching (T3), ii) hot soaking with water replacement followed by cascading blanching (T9), and iii) germinated chickpeas (T13) $(-44.1 \pm 8.6\%, -40.0 \pm 7.6\%, -43.0 \pm 6.8\%, p \le 0.05$, respectively). However, tannin content increased significantly relative to the control when chickpeas were soaked in hot mixed water and blanched by cascading (T8) (+74.0 ± 20.4\%, p \le 0.05).

310

311 3.2. The texture of optimized canned chickpeas is different from that of control 312 chickpeas

313 The texture of the different batches of canned chickpeas was analyzed to check if process 314 modifications changed the chickpea texture compared to the control condition. Table 3 shows 315 that there was a significant increase in chickpea hardness after hot soaking followed by 316 immersion blanching: $+43.9 \pm 3.8\%$ for hot soaking followed by immersion blanching (T1), 317 +48.4 \pm 3.8% for mixed and hot soaking followed by immersion blanching (T2) and +41.9 \pm 318 5.2% for hot soaking with water replacement followed by immersion blanching (T3). 319 Moreover, we observed a significant decrease in chickpea hardness after soaking at 15 °C and 320 blanching by cascading: $-33.1 \pm 3.1\%$ after soaking at 15 °C with low-ionized water followed 321 by cascading blanching (T10) and $-26.2 \pm 4.1\%$ after mixed soaking at 15 °C followed by 322 cascading blanching (T11).

324 3.3. Protein, mineral and vitamin contents can be modified after optimized canning 325 processes compared with control canning.

326 Table 4 shows the concentrations of proteins, fibers, iron and magnesium, lipids and vitamins 327 in the 7 optimized canned chickpeas and in control chickpeas. There was a reduction in 328 protein content after soaking at 15 °C with i) low-ionized water followed by cascading 329 blanching (T10), ii) agitation followed by an immersion blanching (T5) and iii) a germination 330 step (T13) (-35.8 \pm 0.1%, -19.5 \pm 0.1%, -23.6 \pm 0.1%, p \leq 0.05, respectively). The magnesium 331 content decreased similarly (-35.9 \pm 0.04%, -21.5 \pm 0.05%, -27.6 \pm 0.06%, p \leq 0.05, 332 respectively), but increased after hot soaking when water is replaced, followed by immersion blanching (T3, +12.8 \pm 0.05%, p \leq 0.05). Moreover, our results show that relative to the 333 334 control, iron content was increased after hot soaking with i) water replacement and immersion 335 blanching (T3), ii) agitation and cascading blanching (T8) and iii) water replacement and 336 cascading blanching (T9): +30.7 ± 1.1%, +24.5 ± 0.8%, +40.0 ± 1.6%, p≤0.05, respectively. 337 We also found a higher iron content in chickpeas soaked in water at 15 °C with either i) low-338 ionized water (T10) or ii) water replacement (T12), both followed by a cascading blanching 339 $(+26.7 \pm 0.8\%, +67.3 \pm 1.1\%, p \le 0.05, respectively).$

Additionally, we noticed that germination induced a decrease in lipid content (-36.1 \pm 4.6%, 341 p≤0.05) (Table 4).

Vitamin E content significantly increased for 4 conditions relative to the control: +94.2 \pm 16.7% for hot soaking with water replacement and cascading blanching (T9), +82.1 \pm 17.3% for soaking at 15 °C with low-ionized water followed by cascading blanching (T10), +130.2 \pm 14.3% for mixed soaking at 15 °C and immersion blanching (T5), and +102.4 \pm 11.2% for soaking at 15 °C with water replacement followed by cascading blanching (T12). Conversely, a significant decrease in lutein content occurred in the germination condition relative to the control condition (-63.8 \pm 3.8%, p≤0.05). Finally, 5-methyltetrahydrofolate content decreased significantly for all conditions (from -12.1 ± 4.7 % to -97.1 ± 0.1 %), except for a mixed hot soaking followed by a cascading blanching (T8).

351

352 3.4. Vitamin D and K bioavailability is not improved in meals containing optimized 353 canned chickpeas

As shown in Figure 2A, there was no difference in vitamin D and K bioaccessibility between meals containing control chickpea and meals containing optimized canned chickpeas. Similar results were obtained regarding vitamin D and K uptake by differentiated Caco-2 cells (Figure 2B).

359 **4. Discussion**

360

Pulse components such as phytates, saponins and tannins have previously been shown to modulate fat-soluble vitamin bioavailability (Antoine et al., 2021; Margier et al., 2019). The objective of this study was to find technological solutions to address this negative effect. We thus modified several parameters of the canning process to decrease phytate, saponin and tannin content in chickpeas.

366 To this aim, we first modified some process parameters independently of each other. Our 367 results indicate that phytate content can be significantly decreased by modifying soaking or 368 blanching parameters. Under the conditions of our study, this is likely due to an improved 369 release of these compounds, caused by the agitation of chickpeas during soaking, the increase 370 of chickpea coat permeability at 80°C (Tabekhia & Luh, 1980), osmosis for a soaking with 371 low-ionized water (Vijayakumari et al., 1996), a low chickpeas-to-water ratio or with water 372 replacement/ cascading. Moreover, phytates can also be degraded by phytase during either hot 373 or acid soaking, as phytase activity is increased by heat (Khokhar & Chauhan, 1986) and 374 optimized at a pH of 4.6 (Wodzinski & Ullah, 1996).

According to our data, saponin content was not significantly reduced relative to the control condition when a single parameter was changed. A mild reduction in saponin content was observed after soaking with low-ionized water and with a low chickpeas-to-water ratio, reflecting previously published results (Shi et al., 2009), but these differences remained nonsignificant, likely because of the variability of the results. Yoshiki and colleagues showed that saponin content was decreased after blanching for 30 min (Yoshiki et al., 1998). Our blanching conditions (10 or 20 min) may thus not have been long enough.

382 After performing preliminary tests using single modifications, we then combined the most 383 effective processes (13 conditions compared with the control process). The decreases in

384 phytate and saponin content were greater than those in preliminary tests because of the 385 combinations of different parameters, followed by the use of hot brine and sterilization, in 386 accordance with previous data (Jood et al., 1986; Tabekhia & Luh, 1980).

387 It is noteworthy that no difference in phytate and saponin contents were observed after 388 germination. According to previous results (Kyriakidis et al., 1998)(El-Adawy, 2002), 389 phytates were expected to decrease due to a rise in phytase activity during germination. We 390 hypothesize that we failed to observe a decrease in phytates due to the method used - as we 391 did not perform a purification step, we may also have detected the product of phytic acid 392 hydrolysis (phosphoric acid) together with the remaining phytic acid. Saponin assay may also 393 detect the hydrolysis product of saponins: sapogenin (Hiai et al., 1976; Price et al., 1986) 394 together with remaining saponins. In addition, Ruiz and colleagues showed a reduction in 395 saponin content 6 days after germination (Ruiz et al., 1996), while our germination period was 396 of only 3 days. Ayet and colleagues observed a decrease in saponin content 3 days after 397 germination, but it was followed by an increase after 6 days (Ayet et al., 1997). Different 398 germination times should thus be tested to fully comprehend these phenomena, and other 399 analytical methods such as mass spectrometry detection should be set up for further 400 investigations.

Following these intermediate results, 7 conditions inducing the greatest decreases in both phytate and saponin content were ultimately selected to be fully characterized. Changes in texture were observed between these conditions, but further investigations including a sensory analysis are needed to confirm whether consumers would perceive these changes.

The first additional compound evaluated were tannins. A decrease in tannin content was observed for the 2 conditions incorporating water replacement during the soaking step. As tannins are water-soluble, they are likely better released by osmosis during soaking when the water is replaced (Deshpande et al., 1986). A decrease in tannin content was also observed 409 after germination. This is consistent with previous works (Khandelwal et al., 2010)(Rao &
410 Deosthale, 1982): polyphenol oxidase activity increases during germination and induces
411 tannin hydrolysis. Conversely, an increase in tannin content occurred after hot (80 °C) mixed
412 soaking and cascading blanching. We suggest this is due to an inactivation of polyphenol
413 oxidase (Buckow et al., 2009).

We then checked whether compounds of interest remained unaltered after modifying the canning process. Fiber content was not modified. Although the presence of fibers can decrease vitamin bioavailability (Margier et al., 2019), its preservation is important because most people lack fiber in their diets (Lairon et al., 2003). In accordance with previous data (El-Adawy, 2002), lipid content decreased after germination only, as sprouting chickpeas use lipids for growth. Lutein content also decreased during germination, presumably for the same reason.

Water-soluble compounds such as minerals or proteins were expected to decrease when soaking or blanching were optimized. Protein content decreased moderately under 3 conditions (immersion blanching, cascading blanching and germination) that likely increased chickpea moisture and softness. These increases may facilitate the release of water-soluble compounds. Our results differ from previous data (Chitra et al., 1996)(El-Adawy, 2002), but authors compared raw chickpeas with germinated chickpeas.

427 Surprisingly, iron content significantly increased under 4 conditions (as did magnesium 428 content in 1 condition). We suggest that this effect may be due either to i/ lower phytate 429 content, leading to fewer quenched cations, or to ii/ greater disruption of the chickpea matrix, 430 which may favor mineral extraction prior analysis. Vitamin E content also increased under 4 431 conditions. Much as with iron, we suggest that greater softening of the vegetal walls would 432 make vitamin E more extractable, as has been observed in kidney beans (Margier et al., 2018). 433 Finally, 5-methyltetrahydrofolate contents were reduced after most of the optimized processes relative to the control process. Although heat does not appear to be the main factor in
inducing folate loss (Delchier et al., 2013), folates are known to be highly susceptible to
leaching during food transformation processes (Scott et al., 2000).

In a second set of experiments, we assessed vitamin bioavailability after in vitro digestion of test meals made with the different batches of canned chickpeas. We focused especially on vitamin D_3 (cholecalciferol) and vitamin K_1 (phylloquinone), as these vitamins have very different structures and thus may behave differently, as previously observed (Antoine et al., 2021; Margier et al., 2019).

442 No difference was observed between the different conditions in term of bioaccessibility (i.e., 443 the vitamin amount transferred from the solid phase of the meal to the aqueous phase of the 444 digesta during in vitro digestion). No difference was observed between the different 445 conditions in term of cellular uptake by Caco-2 cells in culture, either. This means that the 446 decreases in phytate, saponin and/or tannins contents in optimized canned chickpeas were not 447 important enough to modulate micelle formation, vitamin incorporation in mixed micelles or 448 vitamin membrane transporter functioning. In a previous study, we observed significant 449 differences in terms of vitamin K bioaccessibility and bioavailability when we compared 450 household cooked chickpeas vs canned chickpeas (Margier et al., 2019). The decreases in 451 phytate, saponin, tannin and fiber content were of -24%, -4%, -17,5% and -22%, respectively. 452 We suggest that the difference between the two studies may be due partly to a decrease in 453 fiber content not observed in the present study.

In conclusion, we identified different soaking and blanching parameters in the present study that promoted decreases in phytate, tannin and saponin contents of canned chickpeas (-16 to -456 44%). Further experiments are required to validate these data in other chickpea varieties and/ 457 or in other pulse species. Additionally, when using these chickpeas in tests meals, we did not 458 observe any improvement in the bioaccessibility and the cellular uptake of vitamins D and K

459 as compared with the control conditions. Although further modification of the canning 460 process could lead to greater decreases in bioactive amounts, these modifications may be 461 more expensive and less environmentally friendly. We thus suggest that upstream solutions, 462 such as improved cultivar choice, should also be considered.

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464

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- 615 Figure legends
- 616

617 Figure 1: Bioactive compound contents in canned chickpeas.

618 A: Phytates and saponin content under different soaking conditions after preliminary tests. B: 619 Phytate and saponin content under different blanching conditions after preliminary tests. 620 During preliminary tests (PT), canning process parameters were modified independently. 621 Phytate (white bars) and saponin (black bars) contents were analyzed in chickpea powder 622 after each step, but not in the final product. Soaking was done with hot (80°C), acidic (pH 5), 623 alkaline (pH 9), mixed, low- (HT = 6 °f) and high-ionized (HT = 184 °f) water, but also in 624 different chickpeas-to-water ratios (2:3 or 1:10), with water replacement and for a longer time 625 (24h). Blanching was tested for a longer time (20 min) and with different methods (cascading 626 or steaming). Data are expressed as mean \pm SEM (n = 3).

627 *C: Phytate and saponin content in canned chickpeas.* Different processes combining the most 628 efficient soaking and blanching parameters were tested (T). Contrary to preliminary tests, 629 phytate (white bars) and saponin (black bars) contents were analyzed on the final product 630 powders, after conditioning and sterilization. Data are expressed as mean \pm SEM (n = 9).

631 *D: Tannin content in canned chickpeas.* Once optimized conditions (low phytate and saponin 632 content) were chosen, tannins were analyzed in chickpea powders. Data are expressed as 633 mean \pm SEM (n = 9).

A difference of p ≤ 0.05 was considered significant. Cp: Control Process; Germ: Germination.
dm = dry matter.

636

637

638 Figure 2. Vitamin D and K bioavailability in the presence of optimized chickpeas

A: Vitamin D and K bioaccessibility. Test meals were made with the different canned
chickpeas and supplemented with vitamin D or vitamin K. Bioaccessibility was calculated as
the percentage of the vitamins recovered in mixed micelles divided by the vitamin amount
recovered from the digesta at the end of the digestion.

B: Vitamin D and K intestinal absorption. Mixed micelles obtained from in vitro digestions were delivered to Caco2-TC7 cells after appropriate dilution. The apical side and the well monolayers were collected after 1h of incubation. The vitamin absorbed was calculated as the percentage of the vitamins recovered from the cells divided by the total vitamin amount delivered to the cells .

- 648 Vitamin D: white bars; vitamin K: black bars; CP: Control Process; Germ: Germination. Data
- 649 are expressed as mean \pm SEM (n = 6). A difference of p \leq 0.05 was considered significant.

650 Tables

	SOAKING								BLANCHING								
Parameters	СР	PT 1	PT 2	PT 3	PT 4	PT 5	PT 6	PT 7	PT 8	РТ 9	PT 10	PT 11	Parameters	СР	PT 12	PT 13	PT 14
Temperature (°C)	15	80	15	15	15	15	15	15	15	15	15	15	Temperature (°C)	90	90	90	90
Time (h)	14	14	24	14	14	14	14	14	14	14	14	14	Time (min)	5	20	5	5
Water hardness (°f)	30	30	30	184	6	30	30	30	30	30	30	30	Mode	Immersion		Steaming	Cascading
Chickpeas-to- water ratio	1:3	1:3	1:3	1:3	1:3	2:3	1:10	1:3	1:3	1:3	1:3	1:3	-	-	-	-	-
Agitation	-	-	-	-	-	-	-	Yes	-	-	-	-	-	-	-	-	-
Water replacement	-	-	-	-	-	-	-	-	Yes	-	-	-	-	-	-	-	-
Additive	-	-	-	-	-	-	-	-	-	Acid	Base	-	-	-	-	-	-
Germination	-	-	-	-	-	-	-	-	-	-	-	Yes	-	-	-	-	/

Table 1: Chickpea soaking and blanching parameters for preliminary experiments

654 Different preliminary tests (PT) were done by modifying single soaking and blanching parameters of the control process (CP).

		Control	T 1	Т2	Т3	Т4	Т 5	T 6	T 7	T 8	Т9	T 10	T 11	T 12	T 13
	Temperature (°C)	15	80	80	80	15	15	15	80	80	80	15	15	15	15
	Time (h)	14	2	2	2	14	14	14	2	2	2	14	14	14	14
Soaking	Water hardness (°f)	30	30	30	30	6	30	30	30	30	30	6	30	30	30
	Chickpeas-to- water ratio	1/3	1/10	1/10	1/10	1/3	1/3	1/3	1/10	1/10	1/10	1/3	1/3	1/3	1/3
	Mixing	No	No	Yes	No	No	Yes	No	No	Yes	No	No	Yes	No	No
	Water replacement	No	No	No	Yes	No	No	Yes	No	No	Yes	No	No	Yes	No
	Additive	No	No	No	No	No	No	No	No	No	No	No	No	No	No
	Germination	No	No	No	No	No	No	No	No	No	No	No	No	No	Yes
ing	Temperature 90														
nch	Time (min)		5												
Bla	Mode	Immersion							Cascading Ir si					Immer- sion	
	Conditioning in a can (200 g), and hot brine (0.5% salt, 70 °C) at a ratio of 86:114 (w:w)														
	Autoclave sterilization at 10 rpm and 127 °C for 6 min														

Table 2: Optimization tests on global process

Different tests (T) based on the results of preliminary tests and combining the most efficient
soaking and blanching parameters were performed. Six soaking conditions were defined,
followed either by immersion blanching or cascading blanching. Conditioning and
sterilization parameters remained unchanged.

	Texture
	Maximal applied force (N)
Control	9.58 ± 0.5
T1	13.77 ± 0.4 *
T2	$14.02 \pm 0.4*$
T3	$13.60 \pm 0.5^*$
T4	8.33 ± 0.4
T5	8.60 ± 0.6
T6	12.10 ± 0.9
T7	10.14 ± 0.3
T8	10.00 ± 0.4
Т9	9.50 ± 0.4
T10	$6.53 \pm 0.3*$
T11	$7.07 \pm 0.4*$
T12	8.50 ± 0.55

666

A compression force was applied on the pulses using an Uniaxial Microsystem TA XT texture analyzer to determine the texture (compression forces are positively correlated with seed hardness). Data are expressed as mean \pm SEM (n = 30). A difference of p \leq 0.05 was considered significant.

671

674 Table 4. Protein, lipid, fiber, mineral, fat-soluble vitamin and folate content in canned

675 chickpeas

	Control process	Т3	Т5	Т8	Т9	T10	T12	T13
Proteins (g/100g dm)	23.1 ± 1.0	23.5 ± 1.1	18.6 ± 0.9*	21.4 ± 0.1	22.2 ± 1.0	$14.8 \pm 0.7*$	22.1 ± 1.0	17.6 ± 0.8*
Total lipids (g/100g dm)	5.0 ± 0.1	4.8 ± 0.2	6.4 ± 1.6	6.0 ± 0.6	5.40 ± 0.05	5.1 ± 0.1	4.0 ± 0.3	$3.2 \pm 0.2*$
Total fibers (g/100g dm)	14.9 ± 2.0	15.7 ± 1.9	14.8 ± 2.1	16.6 ± 2.0	14.1 ± 1.9	16.6 ± 2.0	14.9 ± 2.01	14.6 ± 1.9
Vitamin E (mg/100g dm)	2.2 ± 0.4	3.0 ± 0.3	$5.2 \pm 0.3*$	3.1 ± 0.1	$4.3 \pm 0.4*$	$4.1 \pm 0.4*$	$4.6 \pm 0.3*$	2.3 ± 0.3
Lutein (µg/100g dm)	226.7± 36.0	198.1 ± 20.8	$321.4 \pm 34,1$	158.7 ± 5.1	140.7 ± 18.8	255.3 ± 38.5	220.0 ± 17.1	82.1 ± 8.6*
5-methyl- tetrahydro-folate (µg/100g dm)	32.1 ± 9.6	6.4 ± 1.4*	$0.9 \pm 0.1*$	28.2 ± 1.5	1.8 ± 0.3*	3.5 ± 0.8*	1.8 ± 0.4*	4.9 ± 1.3*
Magnesium (mg/100g dm)	131.6 ± 9.5	148.5 ± 10.7*	$103.3 \pm 7.3*$	122.9 ± 8.9	140.7 ± 10.1	84.4 ± 6.0*	116.6 ± 8.3	95.2 ± 6.7*
Iron (mg/100g dm)	32.2 ± 1.7	$42.0 \pm 1.6^*$	32.5 ± 1.7	$40.0 \pm 1.7*$	45.0 ± 3.2*	$40.7 \pm 1.7*$	53.8 ± 3.5*	27.2 ± 1.9
Dry matter (g/ 100g)	34.2	35.7	33.9	34.9	35.5	34.4	33.4	35.7

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The nutritional characterization of canned chickpeas was performed on chickpeas processed according to either the control process (CP) or our 7 optimized conditions (See Table 2). dm = dry matter. Data are expressed as mean \pm SEM (n = 3). A difference of p \leq 0.05 was considered significant.

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