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1 **Traditional irrigation practices sustain groundwater quality in a**
2 **semiarid piedmont**

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30 **Abstract**

31 In semi-arid areas, agricultural practices have been found to significantly alter groundwater
32 quality. Although significant research has been conducted on the impacts of the intensification
33 of irrigated agriculture, many pivotal questions remain relating to the impact of traditional
34 agricultural practices on groundwater quality. In this study, the results of major ions analysis
35 of 91 water samples collected in the semi-arid piedmont of the High Atlas Mountains, central
36 Morocco, are used to assess the impact of traditional irrigation practices on groundwater
37 quality. Despite the use of organic fertilizer in the irrigated area, the NO₃ groundwater
38 concentrations remain low (median = 9 mg/L) and only increase on average by 3.6 mg/L
39 during the irrigation season. All groundwater sampled in the irrigation area has an excellent
40 quality for both drinking and irrigation purposes based on the chemical indices. Overall,
41 groundwater chemistry is controlled by geogenic processes. Relationships between major ion
42 in groundwater reflects the mineral dissolution and ion exchange processes during the
43 trajectory of the streamwaters from the mountain to the alluvial plain via irrigation practices.
44 In comparison, in the non-irrigated area, halite dissolution and/or transpiration processes
45 results in increases in electrical conductivity values that were over twice the values in the
46 irrigated area. The seasonal decrease of electrical conductivity values in groundwater beneath
47 the irrigated area (on average from 841 to 692 μ S/cm) is a result of irrigation recharge, which
48 counterbalances effects of salinization mechanisms that can often characterize irrigated arid
49 zones. These results highlight the low impacts of this ancestral hydro-agro system on
50 groundwater quality. Such a traditional irrigation system provides a nexus between food
51 production, low energy costs (streamflow diversions by gravity-fed channels), and low
52 environmental impacts. In the context of accelerations in global change impacts via the
53 rapidly expanding modern irrigation practices, such traditional hydro-agro systems, where
54 possible, should be highlighted and preserved.

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56

57 **Keywords:** Traditional agriculture, alluvial aquifer, hydrochemistry, water-rock interaction,
58 ion exchange.

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61 **1. Introduction**

62 Agricultural practices have significantly altered groundwater quality (Foster and Chilton,
63 2003; Duncan et al., 2008; Foster et al., 2018). This is a worldwide issue, and has been
64 documented in many countries including in Uzbekistan (Johansson et al., 2009), Spain
65 (Merchán et al., 2015; 2020), France (e.g. Tweed et al., 2018), Japan (Nakagawa et al., 2021),
66 China (Zhang et al., 2012; Hu et al., 2019), and Tunisia (Haj-Amor et al., 2017). In particular,
67 the transition from non- or traditionally-irrigated areas to the expansion and intensification of
68 modern irrigation practices has had significant impacts on the deterioration of groundwater
69 quality (Scanlon et al., 2007). Areas equipped with modern irrigation techniques and destined
70 for intensive agricultural exploitation have globally increased from 170 to 333 Mha between
71 1965 and 2015 (FAO and IWMI, 2018). Keeping in mind that many rural areas around the
72 world use groundwater for domestic purposes without any treatment (Ravindra and Garg,
73 2007; Buschmann et al., 2008; Coyte et al., 2019), actions are required to protect groundwater
74 quality from impacts of these irrigation expansions. This is particularly the case for alluvial
75 aquifers, which can provide important water supplies (Mvandaba et al., 2018; Xiao et al.,
76 2021), but at the same time are also typically vulnerable to pollution (Wu and Sun, 2016).

77 Significant progress has been made in identifying the controlling processes of
78 groundwater quality deterioration in irrigation areas (Xiao et al., 2020), which includes both
79 geogenic and anthropogenic influences. Whilst some studies have found that irrigation by
80 surface waters has increased groundwater salinity (Rabemenana et al., 2005; Scanlon et al.,
81 2007), others have found irrigation by low salinity surface water has buffered groundwater
82 from salinization impacts (Stigter et al., 2006; Rotiroti et al., 2019; Jia et al., 2020). Irrigation
83 with diverted surface waters can result in groundwater mounds due to increased recharge.
84 This recharge and rise in the water table can result in more rapid vertical transfers of
85 agricultural chemicals from the surface to the saturated zone (Elmeknassi et al., 2021;
86 Nakagawa et al., 2021), mobilization of evaporites from the unsaturated zone (Durhan et al.,
87 2008), and an increase in the evaporative effects on the concentrations of contaminants in the
88 shallow groundwater (Cartwright et al., 2007; Scanlon et al., 2007), especially under semi-
89 arid climates. In addition, where the surface water used to irrigate is of poor quality, and the
90 hydrogeological conditions allow for rapid infiltration of contaminants that remain mobile
91 (Xiao et al., 2017), the recharge of irrigation waters can result in the widespread transfer of
92 surface contaminants to the aquifer (Exner et al., 2014; Juntakut et al., 2019). This is
93 particularly observed in agricultural regions located down-gradient from cities that are either

94 heavily populated, have high runoff events, and/or have poor infrastructure to treat
95 wastewater from domestic and industrial zones before flowing into the connecting rivers (Li
96 et al., 2016; Panda et al., 2018; Liu et al., 2020).

97 The majority of the studies regarding the impact of irrigation on groundwater quality
98 have considered the effects of modern irrigation systems, such as drip irrigation and cross-
99 regional transfers (Chen and He, 2003; Jia et al., 2020). In comparison, fewer research has
100 focused on traditional irrigation systems, such as streamflow diversion by traditional
101 irrigation channels called Seguias (Bouimouass et al., 2020) or Acequia (Fernald et al., 2015;
102 Turner et al., 2016). In semi-arid zones, streamflow diversion by traditional irrigation
103 channels is common in irrigated areas adjacent to mountain ranges. Mountains are water
104 towers for the adjacent lowlands because they receive more precipitation due to the
105 orographic effect, and the accumulation of snow and ice contribute to runoff during hot and
106 dry seasons (Immerzeel et al., 2020). Mountain streamflow sustains the downstream areas
107 (Viviroli et al. 2007; Immerzeel et al. 2010) and a recent study showed that almost 1.5 billion
108 people will depend on water contributions from mountain areas by the mid-twenty-first
109 century (Viviroli et al., 2020). However, mountains are highly sensitive to climate and
110 anthropogenic changes (Viviroli et al., 2011, Hock et al., 2019; Immerzeel et al., 2020). With
111 worldwide decreases in snow already observed over the last decades (Berghuijs et al., 2014;
112 Malek et al., 2020; Immerzeel et al., 2020), and decreases forecasted under climate change
113 (Marchane et al., 2015; Baba et al., 2018; Hajhouji et al., 2020; Immerzeel et al., 2020;
114 Viviroli et al., 2020), traditional irrigation areas and related groundwater renewal processes
115 will be affected.

116 In traditional irrigation areas, particularly in semi-arid regions, flood irrigation is often an
117 important source of groundwater recharge (Foster and Perry, 2009; Jimenez-Martinez et al.,
118 2009; Bresciani et al., 2018; Rotiroti et al., 2019; Bouimouass et al., 2020). This recharge has
119 also been found to improve groundwater quality, for example via the dilution of groundwater
120 salinity (Rotiroti et al., 2019). In contrast, Juntakut et al., (2019) found that traditional gravity-
121 fed irrigation resulted in greater nitrate concentrations in groundwater compared with center
122 pivot-irrigation systems. Further investigation and characterization of groundwater quality
123 under such traditional irrigation practices, in the context of climate change and population
124 growth, is critical for water resources especially in semi-arid regions.

125 In the piedmont of the High-Atlas Mountains in the semi-arid Haouz plain, central
126 Morocco, traditional irrigation is the main groundwater recharge source (Bouimouass et al.,
127 2020). The area preserves a traditional agriculture where local small-scale farms have limited

128 use of organic fertilizers for growing wheat and olives trees, irrigated by diverting the
129 streamflow of the High-Atlas Mountains via channels known as Seguias. Groundwater from
130 the alluvial aquifer is the sole drinking water resource for this rural area and it is used without
131 any treatment. In this study area, we investigate the impacts of irrigation water recharge on
132 groundwater quality via the evolution of salinity, nitrate and major ion concentrations in
133 waters from the high elevation mountains to the piedmont. The results are used to determine
134 whether the current traditional irrigation practices provide a model for sustainable agricultural
135 practices and groundwater resources for a semi-arid region.

136 **2. Study area**

137 **2.1. Location, climate and hydrology**

138 As is the case for the whole of North Africa, Morocco is characterized by an arid to semi-
139 arid climate with high inter-annual variability and several periods of below average
140 precipitation (Jarlan et al., 2016). Piedmont areas along the High-Atlas in Morocco benefit
141 from a milder climate and from streamflow generated at high altitudes (Toubkal peak at
142 4167m) by rainfall and snowmelt. The study area is crossed by the Ourika stream and extends
143 from the High-Atlas Mountains to the edges of the Haouz plain, in central Morocco (Fig. 1).
144 The Ourika stream, which drains a watershed in the High-Atlas of 492 km², is one of the main
145 tributaries of the Tensift basin (Fig. 1). The climate varies from arid to semiarid in the plain,
146 and from semiarid to sub-humid in the High Atlas Mountains. The mean annual rainfall
147 ranges from 200 mm in the plain to 600 mm in the mountains as both rain and snow. In the
148 High-Atlas Mountains, streams are fed by rainfall and snowmelt (Marchane et al., 2015;
149 Hajhouji et al., 2018). The snowmelt contributes on average 15% to 30% of the streamflow
150 (Boudhar et al. 2009; Boudhar et al., 2016). Therefore, the streamflow is high during winter
151 and spring and low during summer and autumn. The average inter-annual flow-rate of the
152 Ourika stream is 4.92 m³/s at the Aghbalou gauging station (Fig. 1).

153 **2.2. Geology and hydrogeology**

154 The High-Atlas mountains extend from Morocco to Tunisia, the highest part is the High-
155 Atlas of Marrakech encompassing the study area. The High-Atlas of Marrakech is
156 characterized by a heterogeneous bedrock (Fig. 2), involving complex lithological and
157 structural variations (Delcaillau et al., 2011). The upper part of the High-Atlas Mountains is
158 dominated by a massif of Precambrian granodiorite rich silicates such as plagioclase and
159 amphibole (Juery, 1976). The middle part is formed of Triassic and Visean sandstone and
160 siltstone. The lower part of the High-Atlas, overlooking the piedmont area, is characterized by

161 the presence of Triassic conglomerate, red sandstone, silt and clay containing evaporite
162 mineral deposits, topped by infraliassic basalts, and by Eocene limestone and conglomerates
163 of Mio-pleocene (Ouanaimi, 2011). The piedmont is formed of Neogene and Quaternary
164 alluvial deposits that extend northward over the Haouz plain. They represent the main aquifer.

165 The piedmont area of the High-Atlas in the Tensift basin contains a large unconfined
166 aquifer and two confined aquifers (with very limited spatial extent). The unconfined aquifer in
167 the Haouz plain consists of alluvial fans and fluvial deposits of Neogene and Quaternary
168 age. The aquifer covers almost 6000 km² and it constitutes the principal groundwater resource
169 of the Haouz plain. It is characterized by a heterogeneous transmissivity varying from 5×10^{-5}
170 to $9 \times 10^{-2} \text{ m}^2 \text{ s}^{-1}$ with an average of $6.7 \times 10^{-3} \text{ m}^2 \text{ s}^{-1}$ (Sinan and Razack, 2006). The unconfined
171 aquifers are encompassed in the Eocene and the Turonian–Cenomanian limestone. Little
172 knowledge is available on the hydrogeological characteristics of these aquifers characterized
173 by high spatial heterogeneity. In the study area, belonging to central Haouz, the Neogene and
174 Quaternary alluvial fan are as thick as 50 m, followed by a thick basement up to 121 m
175 consisting of Miocene clay and marls (Sinan, 2000). The Eocene confined aquifer consists of
176 100 m of limestone, followed by a 60 m of impermeable marls and red sandstone (Sinan,
177 1986). The second confined aquifer of Turonian–Cenomanian limestone is 45 m thick lying
178 above a 50 m of Cenomanian clay forming its basement (Sinan, 2000). The present study
179 targeted the alluvial aquifer where groundwater levels can raise to up to 4 m during wet
180 season in the riparian zone and be as deep as 50 m away from the Ourika wadi (Bouimouass
181 et al., 2020). From the dry season to the wet season, groundwater levels substantially increase
182 due to recharge by in-channel streamflow losses and mainly by surface water irrigation
183 (Bouimouass et al., 2020).

184 **2.3. Land and water use**

185 Human settlements in the study area are in form of small villages with population often
186 not exceeding 200 person and are dispersed all over the rural area, except for some towns
187 where population can be of several thousand persons (Fig. 1). These small settlements use
188 septic tanks (up to 5 m) for their domestic wastes which are potentially threatening groundwater
189 quality especially where the aquifer is shallow. Since hundreds of years, the area hosts
190 traditional agricultural practices, consisting of a subsidence production system composed
191 mainly of olive trees (Fig. 3A) and wheat. This traditional agriculture has been secularly
192 irrigated by a large network of gravity-fed surface irrigation earth channels (locally named
193 Seguias) that divert the streamflow (Fig. 3B, C, D and E). This hydro-agro-system is similar

194 to the so-called spate irrigated systems, which are particularly found in areas where mountain
195 catchments border lowlands, in the Middle East, North Africa, West Asia, East Africa and
196 parts of Latin America (Van Steenberg et al., 2011). This type of traditional irrigation is
197 used to expand irrigation from the piedmont to the plain. During the last century, after the
198 French colonization in Morocco, this type of traditional irrigation on the plains has been
199 increasingly replaced by modernized agriculture, as it is the case for most of the plains around
200 the world (Scanlon et al., 2007), with resultant impacts on groundwater quality and quantity
201 (Hu et al., 2019). Nowadays, traditional irrigation is largely restricted to piedmont areas.

202 The study area offers the possibility of investigating two areas of contrasting land and
203 water use. The western side of the Ourika wadi is characterized by extended cultivable lands
204 irrigated by the streamflow that is diverted via the network of irrigation channels (Fig. 1). In
205 comparison, the eastern side of the Ourika wadi is crossed by few irrigation channels, and
206 crops depend mainly on rain-fed irrigation. In this study, we refer to these two areas as
207 irrigation and non-irrigation areas respectively.

208 **3. Material and methods**

209 **3.1. Sampling and analyses**

210 In this study major ion data of precipitation, surface water, springs and groundwater are
211 compared. Precipitation (n=8) were collected in three different elevations spanning the plain,
212 piedmont and high elevations during November 2017 and February 2018. Surface water
213 (n=12) and mountain groundwater (n=12) were collected in two sites in Ourika wadi and two
214 springs with different elevations at a monthly basis from September 2017 to March 2018
215 (except for October 2017). Groundwater (n=55) collected from 27 wells were sampled during
216 four field campaigns in September 2017, November 2017, December 2017 and March 2018.
217 Sampling locations are presented in Figure 1. The sampling sites were chosen as private
218 irrigation and domestic supply wells spanning different land use (irrigated area, non-irrigated
219 area), with depth ranging from 12 to 50 m pumping water from the phreatic aquifer. Some
220 wells were sampled multiple times depending on access. Samples were collected during or
221 immediately after pumping for irrigation and conserved in 300 ml high-density polyethylene
222 bottles and preserved cold until analysis.

223 The physical parameters (temperature, pH, and EC) were measured in the field using a
224 portable pH meter sension+ for pH (Hach) and temperature and a portable conductivity meter
225 for EC (Hanna instruments). All samples were analyzed for cations (Ca^{2+} , Na^+ , Mg^{2+} , K^+)
226 after being filtered at 0.45 μm and acidified with HNO_3 and anions (Cl^- , SO_4^{2-} , NO_3^-) using

227 ion chromatography (Dionex; ICS1100 and autosampler AS-AP) at the laboratory of
228 Hydrogeology of Avignon University. The alkalinity was measured using a HACH digital
229 titrator at the end of the day. Triplicate analyses for each sample were used to assess the
230 analysis uncertainty and the relative standard deviations was around 3%. The quality of water
231 analyses is assessed using the charge balance error (CBE). Water samples having a higher
232 concentration of cations show positive CBE, while negative CBE is credited to higher
233 concentrations of anions (Bozdag et al., 2015). CBE was computed by Equation (1);

$$234 \quad \%CBE = \frac{[\sum cations - \sum anions]}{[\sum cations + \sum anions]} \times 100 \quad (1)$$

235 where ionic concentrations are expressed in milliequivalent per liter. According to the
236 standard protocols, only those water samples were accepted that had less than $\pm 5\%$ CBE
237 (Xiao et al., 2021).

238

239 **3.2. Water suitability for domestic and irrigation uses**

240 Major ions and physiochemical parameters have long been recognized as important indicators
241 for water quality (Vaiphei and Kurakalva, 2021). Common parameters include electrical
242 conductivity (EC) as a measure of salinity (Thorslund and Vliet, 2020; Banna et al., 2014), pH
243 to monitor acidity levels of drinking water (Vaiphei and Kurakalva, 2021), nitrogen species as
244 an indicator of the impacts of agricultural activities on groundwater (Teng et al., 2019), and
245 other major ions such as sodium and chloride that have be used to detect seawater intrusion,
246 evaporate dissolution, and fertiliser contamination (Bresciani et al., 2018; Rajmohan et al.,
247 2021). Such indicators of water quality are valuable in that they are relatively easy and cheap
248 to measure, and are therefore ideal to monitor water quality across regional and remote
249 hydrogeological basins and/or over long timeframes. In order to analyse the information
250 provided by these physiochemical parameters, water quality indices have been established
251 based on different statistical methods (Simões et al., 2008; Sun et al., 2016). Some of the most
252 widely used water indices used in irrigation areas are the sodium adsorption ratio (SAR),
253 residual sodium carbonates (RSC), sodium percentage (% Na) and Kelly's ratio (KR) (Kelly,
254 1993; Vincy et al., 2015). These indices have been widely used and proven to be effective in
255 assessing and monitoring groundwater quality (Celestino et al., 2019; Talib et al., 2019; Xu et
256 al., 2019).

257 The water quality index was also used in this study to analyse potential impacts of
258 irrigation in the groundwater resource quality. This is a widely used tool to assess the
259 suitability of groundwater for drinking supplies (Talib et al., 2019). The WQI relies on the

260 concentrations and the weight coefficient of the chemical elements. The coefficients are based
 261 on the impact of each element on the human health (Xiao et al., 2019). The WQI is calculated
 262 according to the following equation:

$$263 \quad WQI = \sum [W_i \times \left(\frac{C_i}{S_i}\right) \times 100] \quad (2)$$

264 Where C_i is the concentration of each parameter, S_i is the corresponding standard WHO
 265 values (WHO, 2017) and W_i is the relative weight of each parameter computed following the
 266 equation:

$$267 \quad W_i = \frac{w_i}{\sum_{i=1}^n w_i} \quad (3)$$

268 Where w_i is the weight of each parameter and n is the number of parameters. Table 1 shows
 269 the relative S_i and W_i for each parameter used (Talib et al., 2019).

270 Nitrate (NO_3^-) concentrations were also used in this study as an indicator of groundwater
 271 contamination by leaching of (mineral and organic) fertilizers or septic waste, as they are
 272 important for health concerns (WHO, 2011).

273 Similar to drinking water, there are several indices used to assess the suitability of
 274 groundwater and surface water for irrigation (Talib et al., 2019; Xu et al., 2019; Xiao et al.,
 275 2020). The indices sodium adsorption ratio (SAR), residual sodium carbonates (RSC), sodium
 276 percentage (% Na) and Kelly's ratio (KR) were used in this study to consider the
 277 sustainability of the current irrigation practices, and were computed using the following
 278 equations:

$$279 \quad SAR = \frac{Na^+}{\sqrt{(Ca^{2+} + Mg^{2+})/2}} \quad (4)$$

$$280 \quad RSC = (CO_3^- + HCO_3^-) - (Ca^{2+} + Mg^{2+}) \quad (5)$$

$$281 \quad \%Na = \left[\frac{(Na^+ + K^+)}{(Na^+ + K^+ + Ca^{2+} + Mg^{2+})} \right] \times 100 \quad (6)$$

$$282 \quad KR = \frac{Na^+}{Ca^{2+} + Mg^{2+}} \quad (7)$$

283 where all the ions are expressed in milliequivalent per liter

284 **4. Results**

285 **4.1. Hydrochemical properties from the mountain to the piedmont**

286 The electrical conductivity (EC) is very low for mountain streamflow (average = 273
 287 $\mu\text{S/cm}$), increases slightly for mountain springs (average = 328 $\mu\text{S/cm}$), and more
 288 substantially for piedmont groundwater (average = 807 $\mu\text{S/cm}$) (Table 2). For groundwater,

289 only 14 of the 55 groundwater samples exceed 1000 $\mu\text{S}/\text{cm}$, and two samples exceed 2000
290 $\mu\text{S}/\text{cm}$ (well W17). The groundwater EC is lower in the irrigation area (mean = 727 $\mu\text{S}/\text{cm}$
291 and max = 1094 $\mu\text{S}/\text{cm}$) than the non-irrigation area (mean = 994 and max = 2620 $\mu\text{S}/\text{cm}$).
292 From the dry (September to November) to the wet (January to March) season the EC
293 decreased in streamflow from 324 $\mu\text{S}/\text{cm}$ to 244 $\mu\text{S}/\text{cm}$ in average, and in the groundwater
294 beneath the irrigation area from 841 $\mu\text{S}/\text{cm}$ to 692 $\mu\text{S}/\text{cm}$ (Table 2). The pH is higher in the
295 mountain stream water (an average of 7.94 and a maximum of 9.1) than in piedmont
296 groundwater (an average of 7.58 and a maximum of 8.37).

297 The major ion compositions are similar in streamflow and springs, but undergoes a
298 concentration increase in the piedmont groundwater (Fig. 4, table 2), which is enriched
299 particularly of Ca^{2+} , Mg^{2+} and HCO_3^- . The irrigation area mostly has lower major ion
300 concentrations compared with groundwater from the non-irrigated area (Fig. 4); the largest
301 differences are observed for Na^+ and Cl^- , whose mean values increase by 46 and 44 %
302 respectively in the non-irrigated area.

303 On the Piper diagram (Fig. 5), the hydrochemical facies of the streamflow is of Ca-HCO_3
304 type similar to that of atmospheric precipitation (rain and snow). Compared to precipitation, a
305 slight enrichment of Na^+ , Mg^{2+} and SO_4^{2-} is observed. The majority of the groundwater in the
306 piedmont also has a Ca-HCO_3 facies (Fig. 5). It shows in addition two secondary facies, Ca-
307 Mg-Cl and Na-Cl related to specific enrichment in Cl^- , Na^+ , and Mg^{2+} . Indeed, 7 wells located
308 close to the Ourika wadi exhibit Ca-Mg-Cl facies. 3 wells in the non-irrigated area, W17,
309 W19 and W27, have Na-Cl facies; they are close to a tributary, Elmaleh wadi (Elmaleh in
310 Arabic means the salty), which drains the low-altitude halite-rich terrains.

311 Figure 6 shows that most of the major ions (Cl^- , HCO_3^- , Na^+ , Ca^{2+} , and K^+) have
312 decreased from the dry to wet season in terms of median, maximum and minimum. This is
313 also reflected in the decrease of EC from the dry to wet season, and these results indicate that
314 a dilution in the wet season has occurred following irrigation using streamflow diversion. In
315 contrast, SO_4^{2-} and Mg^{2+} concentrations increased in terms of the median during the wet
316 season, and NO_3^- showed a significant increase from the dry to the wet season.

317 4.2. Nitrates

318 Nitrogen is an important plant nutrient that is naturally present in the environment, but
319 often in very low concentrations (< 10 mg/L; Hill, 1996). The multiple natural origins of
320 nitrates are related to evaporative enrichment of dry and wet deposition, biogenic sources
321 through bacterial activity in soil, or to a geogenic origin (Stadler et al., 2008). In the study

322 area, additional sources of nitrogen might also originate from fertilizers (from animal waste)
323 or animal and domestic sewage.

324 The results of the nitrate (NO_3^-) in this study show low concentrations in the streamflow
325 (average = 3.2 mg/l, and a max of 8.2 mg/L) and springs (average = 3.8 mg/l, and a max of
326 7.2 mg/L), and in piedmont groundwater, the nitrate concentrations are higher (10.1 mg/l in
327 average). Furthermore, nitrates increase slightly from the dry season (average = 8.4 mg/l) to
328 the wet season (average = 12 mg/l). This is explained by an accumulation of NO_3^- in the soil
329 during the dry period and its leaching from the soil by rainfall and irrigation returns that are
330 more important during the wet period. Despite this increase in NO_3^- during irrigation
331 recharge, the groundwater NO_3^- concentrations in the irrigation area remains well below the
332 WHO standards fixed at 50 mg/L (WHO, 2017).

333 **4.3. Quality of drinking and irrigation water**

334 Domestic water needs in the study area are supplied by groundwater. The small towns,
335 that are host to tens to hundreds of inhabitants, use community-managed wells and some
336 inhabitants still use their own wells to meet their domestic needs. Groundwater from these
337 wells is usually used without further treatment, but occasional chlorination, which constitutes
338 a potential health risk if the quality of this water is or becomes unsuitable for human use.
339 Groundwater is also used for irrigation in the study area during extended dry periods. In the
340 irrigated area, groundwater is used during summer and autumn months when streamflow is
341 absent. In the non-irrigated area, rain-fed agriculture is dominant but some farmers use
342 groundwater to irrigate olive fields.

343 The overall quality of groundwater can be assessed by the water quality index (WQI)
344 (Talib et al., 2019, Sadat-Noori et al. 2014). WQI values are classified into five categories
345 (calculated using the parameters ...): excellent (<50), good (>50), poor (>100), very poor
346 (>200) and water unsuitable for drinking (>300). The WQI values ranges from 3 to 95 with a
347 mean of 24. According to this classification, almost all groundwater and surface water are
348 “excellent” for drinking purposes. Only the 03 samples of the non-irrigated area and
349 contaminated by salts are close to the “poor “quality category. Although groundwater is of
350 good quality from a mineral perception, additional biological analyses are needed to fully
351 assess the adequacy of groundwater to human use.

352 The suitability of groundwater for irrigation is assessed through the Wilcox diagram (Fig.
353 7) and various irrigation water quality indices, such as SAR, RSC, %Na and KR (Table 3). As
354 a conclusion, most of the water is suitable for irrigation. Those few presenting unsuitable

355 characteristics due the Na⁺ content correspond to the samples collected from the well W17,
356 which is characterized by high Na⁺ and Cl⁻ concentrations and located outside of the irrigated
357 area. Compared to the others wells in the study area, the well W17 is influenced by the
358 dissolution of evaporites.

359 **5. Discussion**

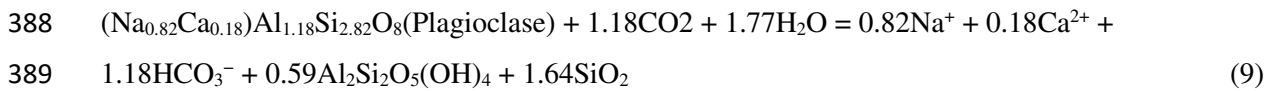
360 **5.1. Origins of ions in mountains water and groundwater**

361 The natural chemical composition of water is potentially influenced by atmospheric
362 precipitation, evaporation, transpiration, evaporite dissolution and rock weathering. In
363 mountain waters Ca²⁺, HCO₃⁻, Na⁺, Mg²⁺, SO₄²⁻ ions are likely sourced firstly from
364 precipitation and secondly from the weathering of silicate minerals in the crystalline rocks of
365 the High-Atlas massif. The piedmont groundwater samples show similar facies with mountain
366 water, however their relatively higher concentrations of Na⁺, Cl⁻, and in a lesser proportion of
367 Ca²⁺, Mg²⁺ and HCO₃⁻ suggest additional sources of mineralization.

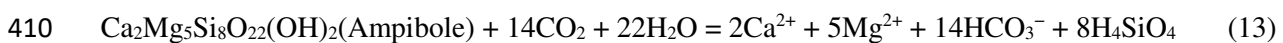
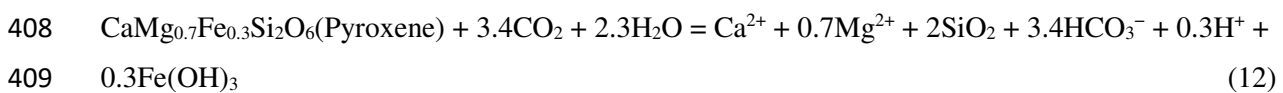
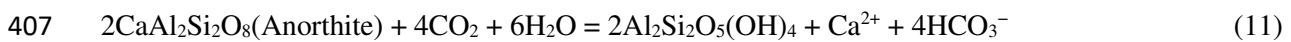
368 On the Na⁺/Cl⁻ versus Cl⁻ scatter diagram (Fig. 8), samples with Na/Cl ratios close to 1
369 with increasing Cl⁻ concentrations indicate either halite dissolution (Hao et al., 2020; Talib et
370 al., 2019), evaporation or transpiration is the major process increasing Na⁺ and Cl⁻
371 concentrations. The groundwater samples from the non-irrigated area exhibit little changes in
372 the Na⁺/Cl⁻ ratio with high Cl⁻ concentrations (> 10 meq/L). A previous study by Bouimouass
373 et al. (2020) using the stable isotope values of groundwater for the same samples, showed no
374 evidence of evaporation effects. Therefore, high Na⁺ and Cl⁻ concentrations in groundwater
375 are driven by either Triassic halite dissolution and/or transpiration processes. For Cl⁻
376 concentrations lower than 10 meq/L, there are greater variations in the Na/Cl ratio that
377 indicates either contributions from rock weathering reactions (Na/Cl > 1) or ion exchange
378 effects (Na/Cl < 1). Based on the Na/Cl ratio, the piedmont groundwater is divided into two
379 groups (Fig. 6). The sodium enriched group n° 1 (where Na/Cl > 1), which is similar to the
380 mountain streamflow Na/Cl ratios, is mostly located in the irrigation area. The sodium
381 depleted group n° 2 (where Na/Cl < 1) is predominantly composed of samples located along
382 the Ourika wadi.

383 For group n° 1 (Na/Cl > 1) representing the irrigation area, the enrichment of Na could be
384 due to weathering by the streamflow in the mountains of the Na-silicate such as albite and
385 plagioclase (equations 8 and 9) (Gao et al., 2020) since Na-silicates are largely present in the
386 High-Atlas as granodiorite and granite.





390 In the HCO_3^- versus Na^+ plot (Fig. 9a), water samples falling along the 1:1 and 1.18:0.82
 391 lines indicate albite and plagioclase weathering might increase HCO_3^- and Na^+ in water (Kim
 392 et al., 2002; Zhang et al., 2020). All streamflow and mountain groundwater lie along and
 393 above the line 1.18:0.82 indicating contribution from weathering of albite and plagioclase to
 394 Na^+ and HCO_3^- in these waters. The majority of the piedmont groundwater samples lie on or
 395 between the two lines (Fig. 9a) implying that weathering of albite and plagioclase is
 396 responsible for amounts of Na^+ and HCO_3^- . The groundwater samples contaminated by salts
 397 plot far to the right of the 1:1 line. The additional sources of HCO_3^- ion could originate from
 398 various sources including calcite dissolution (equation 10), and Ca-silicates weathering such
 399 as anorthite (equation 11), pyroxene (equation 12) and amphibole (equation 13) (Zhang et al.,
 400 2020). Based on chemical reactions 10, 11, 12 and 13, samples falling along the lines 1:1, 2:1,
 401 1.7:1, 7:2 in figure 5b are due to the dissolution of carbonates such as calcite and dolomite,
 402 and the weathering of Ca-silicates such as anorthite, pyroxene and amphibole, respectively.
 403 All mountains streamflow and springs samples, and most of the piedmont groundwater
 404 samples fall between the lines 1:1 and 1.7:1 (Fig. 9b) indicating that the dissolution of calcite
 405 and some Ca-silicates such as pyroxene contributes to HCO_3^- concentrations in these waters.



411 For group n° 2 (where $\text{Na}/\text{Cl} < 1$) representing the piedmont groundwater along the wadi,
 412 the ion exchange processes and anthropogenic sources of Cl^- (Celestino et al., 2012) could be
 413 responsible of Na depletion or Cl^- enrichment. The plot of $\text{Na}^+ + \text{K}^+ - \text{Cl}^-$ versus $\text{Ca}^{2+} + \text{Mg}^{2+} -$
 414 $\text{SO}_4 - \text{HCO}_3^-$ (Fig. 9b) shows a linear relationship with a slope of -0.83, close to -1, indicating
 415 the occurrence of ion exchange (Garçia, 2001; Yang et al., 2016). Cation exchange processes
 416 were analysed using the chloro-alkaline indices (CAI-I and CAI-II), which are widely used as
 417 indicators of ion exchange occurring in aquifers (Schoeller, 1965; Li et al., 2013; Talib et al.,
 418 2019). The indices, expressed by the equations 14 and 15, are negative when the cation
 419 exchange occurs, meaning that Na^+ is released from the medium in exchange with Ca^{2+} and

420 Mg^{2+} in water. However, positive indices mean that Na^+ is adsorbed in the medium
 421 simultaneously with the release of Ca^{2+} and Mg^{2+} , known as reverse ion exchange.

$$422 \quad CAI - I = \frac{Cl^- - (Na^+ + K^+)}{Cl^-} \quad (14)$$

$$423 \quad CAI - II = \frac{Cl^- - (Na^+ + K^+)}{HCO_3^- + SO_4^{2-} + CO_3^{2-} + NO_3^-} \quad (15)$$

424 Piedmont groundwater samples are almost equally distributed between positive and
 425 negative CAI-1 and CAI-2 values (Fig. 9c) (25 samples with positive values, 23 with negative
 426 values and 04 with CAI-I and CAI-II equal to 0). This indicates that both base cation
 427 exchange (positive CAI-1 and CAI-2 values) and reverse ion exchange (negative CAI-1 and
 428 CAI-2) potentially occur. Along the wadi (group 2), the base cation exchange dominates
 429 implying depletion of Na^+ in groundwater. Reverse ion exchange characterizes groundwater
 430 beneath the irrigation area (Group 1), it induces a release of Na^+ of the aquifer matrix in
 431 groundwater and adsorption of Ca^{2+} and Mg^{2+} (Carol et al., 2012; Zaidi et al., 2015).

432 Ca^{2+} in water could also be due to gypsum dissolution, alongside SO_4^{2-} , as indicated by
 433 the equation 17. A Ca^{2+}/SO_4^{2-} ratio around 1 indicates that these ions are derived from gypsum
 434 dissolution. The majority of samples (Fig. 9f) are plot to the left of the 1:1 line indicating an
 435 excess of Ca^{2+} over SO_4^{2-} ; the Gypsum would be responsible of low amounts of Ca^{2+} and
 436 Mg^{2+} .



438

439 **5.2. Hydrochemical evolution from the mountain to the piedmont**

440 In the high-Atlas Mountains, the streamflow and springs are supplied by rainfall and
 441 snowmelt, and flow through large outcrops of silicates rocks formed of Precambrian
 442 Granodiorite and Gneiss containing plagioclase, amphibole and micaschist, as well as Visean
 443 and Triassic siltite and sandstone rocks. They have a Ca- HCO_3 facies similar to that of
 444 atmospheric precipitation, and a slight enrichment in Na^+ , Mg^{2+} and SO_4^- . In mountains
 445 streamflow and springs, Na^+ and HCO_3^- are strongly correlated ($R^2 = 0.84$, table 4), their main
 446 source is Na-silicates weathering. The Ca-silicate weathering also contributes to the loading
 447 of Ca^{2+} and Mg^{2+} . The pyroxene, amphiboles and calcic feldspar are common minerals in
 448 basic rocks and are easily weathered (Jacks 1973; Bartarya 1993; Rajmohan and Elango,
 449 2004). Therefore, these minerals are also likely to contribute to the ionic composition of the
 450 waters in the mountains. The elevated SO_4^{2-} might also be due to the dissolution of gypsum.

451 In the piedmont, groundwater is generally fresh (EC average = 713 $\mu\text{S}/\text{cm}$), with a
452 dominant Ca-HCO_3 facies similar to the streamflow. This similarity results from the recharge
453 of the alluvial aquifer in the piedmont via irrigation of the streamflow waters, and also from
454 in-stream losses (Bouimouass et al., 2020). Therefore, piedmont groundwater primarily
455 inherits the hydrochemical characteristics of the high-altitude mountain water. When evolving
456 in the piedmont and compared to streamflow, groundwater acquires a higher total
457 mineralization and exhibits two secondary facies, Ca-Mg-Cl and Na-Cl .

458 The generally higher ion content of the groundwater in the piedmont compared to the
459 recharging streamflow could be explained by transpiration processes and water-rock
460 interactions in the alluvial aquifer. In this semi-arid study area, even though the evaporative
461 demand is high (around 1600 mm/year), evaporation effects on piedmont groundwater were
462 not detected from the stable isotope data presented in Bouimouass et al., (2020). This could be
463 explained by the rapid infiltration of irrigation water, the recharge of unevaporated mountain
464 streamflow, and by a deep unsaturated zone (4-50 m depth) that increases during the summer
465 months (Bouimouass et al., 2020).

466 The secondary Na-Cl facies in piedmont groundwater was observed in the right side of
467 the wadi in the non-irrigation area, and is either due to halite dissolution from Triassic
468 deposits or transpiration. The Ca-Mg-Cl facies was observed near the wadi, along with the
469 Na^+ depletion; this can be ascribed to cation exchange inducing Na^+ adsorption and, Ca^{2+} and
470 Mg^{2+} release in groundwater. Beneath the irrigated area, reverse ion exchange was related to
471 Na^+ enrichment; it seems that the irrigation recharge transfers Ca^{2+} and Mg^{2+} that are adsorbed
472 by clay minerals whilst Na^+ is released in the groundwater. The occurrence of this process
473 might be explained by the dynamics of the seasonal recharge of the diverted streamflow from
474 the Ourika wadi. Traditional irrigation by flooding relatively large lands with streamflow
475 induces more interaction between surface water and sediment matrix during percolation.

476 **5.3. Effects of the traditional irrigation on the groundwater quality**

477 In our piedmont study area, the irrigation area has the prime location of being at the foot
478 of a major mountain range that is lowly populated and cultivated, and where streamflow is
479 supplied by rainfall and snowmelt generally from winter to early summer. The diverted
480 streamflow by a network of irrigation channels managed by community-driven systems is of
481 low EC and NO_3 (273 $\mu\text{S}/\text{cm}$ and 3.2 mg/L respectively), and such features constitute
482 favorable factors of groundwater quality since the streamflow is the main groundwater
483 recharge source. The piedmont groundwater beneath the irrigated area is characterized by

484 excellent to good chemical quality regarding both drinking and irrigation use indices. The
485 substantial induced recharge from irrigation flow leads to seasonal groundwater renewal, and
486 counterbalances effects of salinization mechanisms that often characterize irrigated arid zones
487 (Foster et al., 2018). Almost all the ions showed a dilution from the dry to the wet season. The
488 results of this study highlight a system where the irrigation practices have not adversely
489 impacted groundwater resource quality. The irrigation practices have not significantly
490 adversely altered processes controlling the major ion composition, nor are there significant
491 transfers of nitrogen-based fertilizers to the shallow groundwater. Thus, the traditional
492 irrigation practices in this mountain piedmont, which have been in place for hundreds of
493 years, are considered a low impact practice in terms of groundwater resource quality and even
494 an enhancing factor of groundwater quality.

495 More intensive and modern agricultural and irrigation practices are currently observed in
496 the neighboring plain. Within this plain, the intensive modern agriculture has had severe
497 impacts on groundwater quantity, with hydraulic heads decreasing by 1 to 3 m/year (Fakir et
498 al., 2015, Le Page et al., 2021), the groundwater nitrate concentrations are increasing
499 (Boukhari et al., 2015), and the soil salinity as well (Sefiani et al., 2019). This expansion of
500 irrigated agriculture threatens the existence of the traditional agriculture system of the piedmont
501 and disturb its ecosystem. The studied traditional irrigation piedmonts, irrigated by high
502 mountain streamflow, is one of the rare sustainable human exploits that is likely to remain as
503 a low environmental impact practice. Such systems represent a strong connection and a long
504 history between water and users to ensure sustainability and drought survival, and have
505 constituted a main driver of socio-economic activities for centuries. A key success of survival
506 of such ancestral systems is the sense of mutualism between the users. This mutualism,
507 defined by Gunda et al., (2018), is the feeling of collective well-being including social
508 identity, pride of place, and maintenance of historical traditions. For those piedmonts, the
509 current traditional agricultural practices should be maintained and enhanced in order to
510 preserve their groundwater resources sustainability as they provide a nexus between food
511 production, low energy costs (streamflow diversions by gravity-fed channels), and low
512 environmental impacts. This will also benefit groundwater resources in down-gradient areas
513 since piedmonts are generally favorable recharge zones in (semi)arid basins (Wilson and
514 Guan, 2004; Liu and Yamanaka, 2012). The traditional agriculture could be improved by
515 introducing organic agriculture (FAO, 2013) that could increase the financial incomes of
516 farmers, better optimize irrigation water and increase its value.

517 **6. Conclusion**

518 The present study investigated the impact of groundwater recharge from a traditional
519 irrigation system on groundwater quality in a semiarid piedmont. Groundwater
520 hydrochemistry is controlled mainly by water-rock interactions rather than anthropogenic
521 activities. Groundwater chemistry is influenced by both the chemical characteristics of the
522 streamflow acquired in the mountains, and by local hydrochemical processes that occur
523 during infiltration and recharge via the alluvial plain. Groundwater benefits from substantial
524 seasonal recharge from diverted streamflow irrigation, which counterbalances the potential
525 effects of salinization mechanisms common in irrigated areas of (semi)arid zones, such as
526 evaporation and leaching of saline soils. Nitrate concentrations slightly increase with
527 irrigation season but remain well below the permissible limits for drinking water. Almost all
528 groundwater and surface water are “excellent” for drinking and irrigation purposes. The
529 current state of groundwater chemistry in the piedmont of the High-Atlas is a heritage of
530 hundreds of years of exploitation. The hydrological processes, the irrigation practices and the
531 use by local farmers of organic fertilization have preserved the traditional hydro-agro-systems
532 that are close to sustainable agricultural systems. Protection measures of these irrigation
533 systems should be considered in the framework of adaptive strategies for sustainable
534 management and for water heritage, as this system is undergoing severe pressure climate
535 change effects and by the expansion of intensive modern agricultural practices overexploiting
536 the water resources.

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Tables

Table 1: WQI index.

Parameter	S _i (mg/l)	Weight	Relative weight
TDS	1000	5	0.13
Cl ⁻	250	5	0.13
SO ₄ ²⁻	250	5	0.13
F ⁻	1	5	0.13
NO ₃ ⁻	50	5	0.13
Na ⁺	200	4	0.10
Mg ²⁺	150	3	0.08
Ca ²⁺	200	3	0.08
K ⁺	12	2	0.05
HCO ₃ ⁻	250	1	0.03
		$\sum W_i = 38$	$\sum W_i = 1$

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Table 2: Statistics of the field parameters and major ions.

Hydrochemical variables (mg/l)	Mountain streamflow ST (n=11)			Mountain springs SP (n=14)			Groundwater irrigation area (n=39)			Groundwater non-irrigation area (n=16)			WHO (2017)
	Min	Max	Mean	Min	Max	Mean	Min	Max	Mean	Min	Max	Mean	
pH	6.64	9.05	7.94	7	9.1	7.72	7.00	8.2	7.5	7.0	8.4	7.8	6.5-8.5
EC	104.0	563.0	273.1	77.0	943.0	328.2	293.0	1945.0	726.7	258.0	2620.0	994	
TDS	79.0	423.0	194.5	54.0	705.0	241.3	216.0	1130.0	491.2	198.0	1615.0	655.3	500
Na ⁺	3.3	24.7	11.7	2.6	55.4	15.39	18.1	257.7	53.6	10.2	425.5	100.0	200
K ⁺	0.5	2.1	1.2	0.3	3.6	1.42	0.6	1.4	3.0	0.2	2.9	1.1	200
Mg ²⁺	3.0	9.5	7.2	1.9	21.5	7.5	6.6	16.6	30.5	7.8	46.9	25.4	NA
Ca ²⁺	11.5	69.7	29.5	7.4	38.4	21.49	30.5	129.9	64.4	27.0	111.4	59.3	200
Cl ⁻	3.2	32.2	14.8	2.4	79.9	19.53	17.4	462.8	93.4	7.8	719.3	167.4	250
SO ₄ ²⁻	6.9	27.1	16.3	5.6	70.4	23.03	11.0	91.5	29.1	2.5	149.8	39.9	200
HCO ₃ ⁻	49	256.0	110.5	27.0	371.0	132	112.2	370.9	222.8	131.8	398.9	247.3	NA
NO ₃ ⁻	0.5	8.1	3.2	1.9	7.1	3.8	1.2	22.5	10.1	2.4	22.1	10.1	50

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Table 3: Statistics of the irrigation water quality indices in the study area.

Indices	Groundwater			Surface water			Permissible limit	Unsuitable samples
	Min	Max	Mean	Min	Max	Mean		
SAR	0.2	13.4	1.6	0.2	0.8	0.5	≤ 18	-
RSC	-4.0	1.1	-0.8	-0.5	-0.1	-0.3	≤ 2.5	-
%Na	12	83	30	14.3	27.5	19.6	≤ 60	3
KR	0.1	4.9	0.5	0.1	0.3	0.2	≤ 1	3

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971 *Table 4: Linear relationship (R2) between various parameters of groundwater in the study area.*

	pH	EC	TDS	Na	K	Mg	Ca	Cl	SO4	HCO3	NO3
pH	1										
EC	-0,32	1									
TDS	-0,34	0,99	1								
Na	-0,25	0,96	0,93	1							
K	-0,2	0,53	0,52	0,46	1						
Mg	-0,18	0,51	0,59	0,34	0,05	1					
Ca	-0,47	0,49	0,54	0,25	0,54	0,41	1				
Cl	-0,29	0,97	0,94	0,98	0,52	0,35	0,39	1			
SO4	0,05	0,7	0,7	0,63	0,57	0,5	0,38	0,64	1		
HCO3	-0,38	0,39	0,5	0,21	0,12	0,77	0,57	0,19	0,19	1	
NO3	0,26	-0,11	-0,05	-0,2	-0,26	0,47	-0,02	-0,22	0,02	0,28	1

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Figure 1: Map of the study area and sampling points. (a) Morocco, (b) Tensift basin, and (c) sampling points in the Ourika watershed, the wadi gauging station and some principal irrigations channels (Seguias).

Figure 2: Pictures from the study area. A: a panoramic view of the agricultural crops over the piedmont, B: a Seguia diverting water directly from the Ourika stream, C, D and E: different types of Seguias in the piedmont.

Figure 3: Geology and cross section of Ourika watershed.

Figure 4: Boxplots of the major ions in groundwater (GW), spring (SP) and surface water (SW).

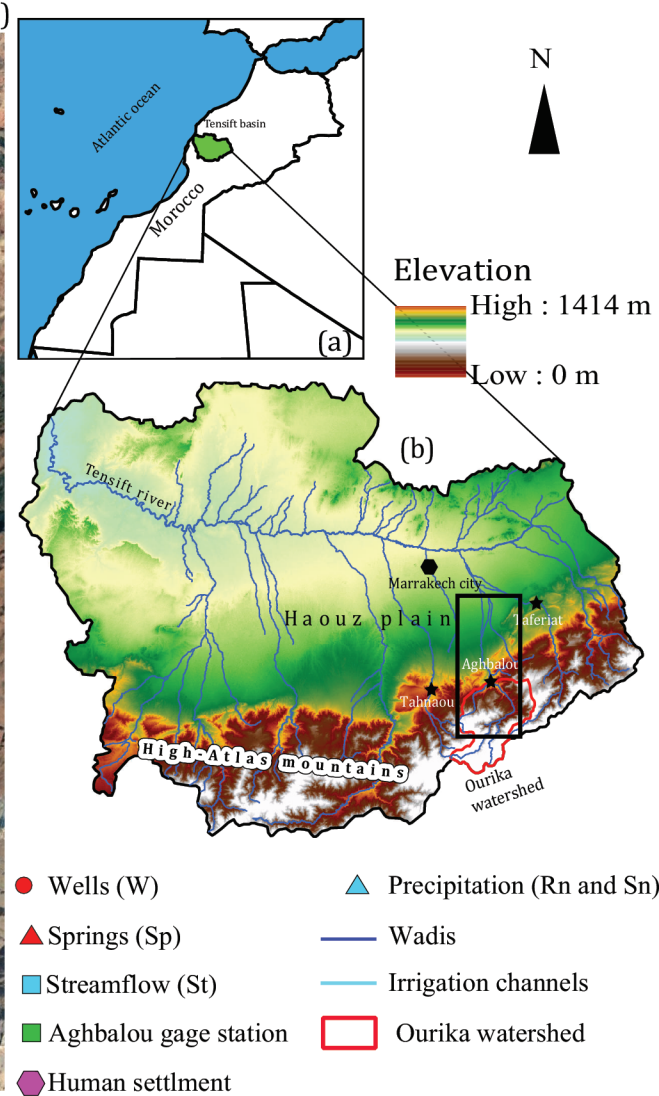
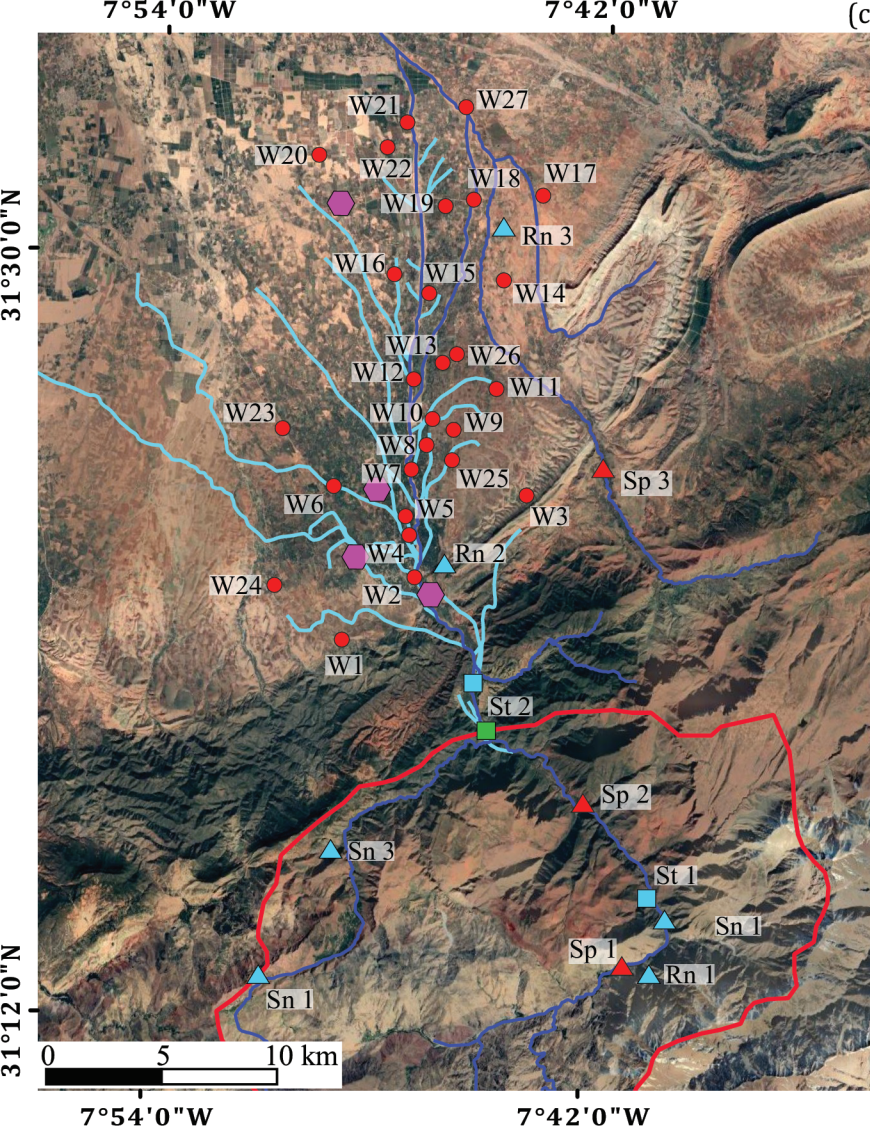
Figure 5: Piper diagram. Pink color refers to dry season and blue color refers to wet season.

Figure 6: Seasonal variation in EC and major ions concentrations.

Figure 7: Wilcox diagram. Red color refers to dry season and blue color refers to wet season.

Figure 8: Plot of Na/Cl ratio versus Cl. Red color refers to dry season and blue color refers to wet season.

Figure 9: Scatter plots of ions. Red color refers to dry season and blue color refers to wet season.



A**B****C****D**

