

The plasma membrane-associated cation-binding protein PCaP1 of Arabidopsis thaliana is a uranyl-binding protein

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- 2 Arabidopsis thaliana is a uranyl-binding protein
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30 Highlights

- Metalloproteomics analysis identified 38 UraBPs from Arabidopsis roots and/or shoots.
- Recombinant PCaP1 protein (UraBP25) binds up to 2 U(VI) ions with high affinity in vitro.
- Recombinant PCaP1 protein binds about 15 Cu(II) or 60 Fe(III) ions *in vitro*.
- U(VI) binding triggers PCaP1 oligomerization.
- U(VI) translocation capacity is reduced in an Arabidopsis *pcap1-null* mutant.
- 37

38 Abstract

39

40 Uranium (U) is a naturally-occurring radionuclide that is toxic to living organisms. Given that proteins are primary targets of U(VI), their identification is an essential step towards 41 understanding the mechanisms of radionuclide toxicity, and possibly detoxification. Here, we 42 developed an immobilized metal affinity chromatography procedure to trap protein targets of 43 uranyl from Arabidopsis thaliana. This procedure allowed the identification of 38 uranyl-binding 44 45 proteins (UraBPs) from root and shoot extracts. One of these, UraBP25, previously identified as plasma membrane-associated cation-binding protein 1 (PCaP1), was further characterized 46 47 for its ability to interact with U(VI) and other metals in vitro, by spectroscopic and structural 48 approaches, and in planta, by analyzing the fate of U(VI) in Arabidopsis lines affected in PCaP1 gene expression. Our results showed that recombinant PCaP1 binds U(VI) in vitro with affinity 49 in the nM range, as well as Cu(II) and Fe(III) in high proportions, and that Ca(II) competes with 50 U(VI) for binding. U(VI) induced PCaP1 oligomerization through binding at the interface of 51 monomers, both at the N-terminal structured domain and the C-terminal flexible region. Finally, 52 U(VI) translocation in Arabidopsis shoots was affected in a *pcap1* null-mutant, suggesting a 53 role for this protein in ions trafficking *in planta*. 54

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56 **1. Introduction**

Uranium (U) is a radionuclide naturally present in the Earth's crust. It is mainly redistributed in 57 the environment by human activities (e.g. industry, agriculture, mining) and can accumulate 58 locally to concentrations that present potential risks for ecosystems, agrosystems, and 59 eventually human health. Indeed, this element is chemotoxic and possibly radiotoxic for all 60 living organisms. In most natural settings, the di-oxo uranyl cation UO₂²⁺ is the most stable 61 62 form of U, with U in the +VI oxidation state (thereafter referred to as U(VI)). Although U(VI) is not essential for plants, it is absorbed from the soil, incorporated into the biomass, and thus 63 enters the food chain. Consequently, contamination of soils by U(VI) and its absorption by 64 plants represent a substantial health risk for humans. As predicted by the hard and soft acids 65 66 principle, uranyl cation is a hard acid and has a high propensity to interact with hard oxygen donors, primarily by electrostatic bonding (Pearson, 1963). Therefore, the biological ligands of 67 U(VI) may be as wide as metabolites, nucleic acids and polypeptides (Garai and Delangle, 68 2020). Since uranyl has the ability to bind strongly to biomolecules through carboxylate and 69 phosphate groups (two of the main hard oxygen donors), proteins may be considered as 70 primary targets of U(VI). Thus, identifying protein targets of uranyl is a crucial step in an attempt 71 72 to decipher the molecular bases of radionuclide toxicity, and perhaps their detoxification, in 73 any living being (Sarthou et al., 2020).

In the last decade, various strategies have been developed for the identification of proteins 74 with high-affinity and/or selectivity for uranyl and forming uranyl-protein complexes in vivo 75 76 and/or in vitro (Acharya and Blindauer, 2016; Basset et al., 2013; Bucher et al., 2014; Creff et al., 2019; Eb-Levadoux et al., 2017; Gallois et al., 2021; Huynh et al., 2015; Pardoux et al., 77 78 2012; Qi et al., 2014; Safi et al., 2013; Vidaud et al., 2019; Xu et al., 2014). Most of them were 79 metalloproteomic approaches combining efficient protein separation techniques with powerful and sensitive protein and metal identification devices, namely protein tandem mass 80 81 spectrometry (MS/MS) and inductively coupled plasma mass spectrometry (ICP-MS), respectively. These investigations identified a wide range of proteins that bind uranyl in various 82 organisms including archaea, bacteria, animals and humans. Their discovery was an important 83 milestone for understanding the cellular toxicity of the radionuclide (Basset et al., 2013; 84 Cvetkovic et al., 2010; Dedieu et al., 2009; Gallois et al., 2021; Huynh et al., 2016; Xu et al., 85 2014). Uranyl-binding proteins (UraBPs) are implied in calcium homeostasis (for example 86 calmodulin, fetuin-A and osteopontin), iron homeostasis (ferritin and transferrin), glucose 87 metabolism, oxidative stress response, microtubule and actin cytoskeleton regulation, or 88 proteolysis. Despite the large number of studies published to date on the impact of U(VI) 89 exposure on plant physiology and development, and on its fate upon absorption (Aranjuelo et 90 91 al., 2014; Berthet et al., 2018; Doustaly et al., 2014; Ebbs et al., 1998; Lai et al., 2020; Laurette et al., 2012; Misson et al., 2009; Saenen et al., 2014; Saenen et al., 2015; Sarthou et al., 2020; 92

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93 Serre et al., 2019; Tewari et al., 2015; Vanhoudt et al., 2014), no plant UraBP has been 94 identified so far. Yet, such discoveries can help to better understand the molecular 95 mechanisms that govern the fate of U(VI), and understanding how plants manage to thrive in 96 contaminated environmental ecosystems. This knowledge is also required for a future 97 sustainable management of U(VI) in polluted soils and waters as well as in the food chain 98 (Berthet et al., 2018).

99 In this study, we developed a successful two-stage enrichment procedure (ion-exchange 100 chromatography prefractionation step followed by affinity capture) to identify protein targets of 101 uranyl in Arabidopsis thaliana. This procedure enabled the purification of 38 UraBP from root 102 and/or shoot extracts by shotgun proteomics. One of these proteins, named UraBP25 (25th 103 U(VI)-interacting protein of our list), hitherto-identified as the plasma membrane-associated cation-binding protein 1 (PCaP1), was of particular interest. PCaP1 was further characterized 104 for its ability to interact with U(VI) and other metals in vitro, by a combination of biochemical 105 106 and structural approaches, and *in planta*, by analyzing the fate of U(VI) in Arabidopsis lines affected in the expression of the corresponding gene. Taken together, our results showed first, 107 that recombinant PCaP1 binds U(VI) in vitro with high affinity and Cu(II) and Fe(III) in high 108 proportions, and that Ca(II) competes with U(VI) for binding. Second, we showed that U(VI) 109 110 induces PCaP1 oligomerization through binding to both the N-terminal structured protein domain and the C-terminal flexible region. Last, we observed that U(VI) translocation in 111 Arabidopsis shoots is affected in a *pcap1* null-mutant suggesting a role *in planta* for this protein 112 113 in cations mobility.

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2. Materials and Methods

2.1. Plant growth conditions for protein fractionation studies

Sterilized and stratified seeds of wild-type *Arabidopsis thaliana* ecotype Columbia (Col-0) were sown into homemade thermoprinted plates, as described previously (Sarthou et al., 2022). Plants were grown in hydropony for 5 weeks in the 'Gre medium' nutrient solution, as detailed in (Sarthou et al., 2022), with weekly solution changes. Growth was performed in a controlled environment, with alternation of 8 h light period at 22°C (light intensity of 110 µmol of photons m⁻²s⁻¹) and 16 h dark period at 20°C.

Total soluble proteins from plant shoot and root tissues were extracted by grinding 15 g powdered samples in 10 mM Tris-HCl, pH 7.5, 1 mM dithiothreitol, and a cocktail of protease inhibitors (Roche Applied Science). Samples were clarified by centrifugation, desalted on PD10 Sephadex G25 (M) columns (GE Healthcare), and 30-40 mg soluble proteins were then

^{124 2.2.} Preparation of protein extracts and anion exchange chromatography (AEC)
125 fractionation

fractionated by chromatography onto Q-Sepharose High Performance columns (1.6 x 5 cm), 130 as described previously (Sarthou et al., 2020). Stepwise elution was performed using 131 discontinuous increasing NaCl concentrations in the extraction buffer (from 0 to 1 M NaCl; 132 133 Sub-fractions F0 to F1000). The salt concentrations were chosen to obtain a homogenous 134 repartition of the quantity of proteins in each sub-fraction (Supplementary Figure S1). Collected sub-fractions were stored at -80°C until used for further analyses. Four independent 135 136 fractionation experiments from independent 8-plant cultures each were performed (Series 1 to 4). 137

138 2.3. Protein determination

Proteins from plant extracts were measured by the Bradford method using Bio-Rad protein assay reagent, with bovine serum albumin as a standard (Bradford, 1976). Protein AEC subfraction aliquots (30 µg) were analyzed by SDS-PAGE 12% and staining with Coomassie Brilliant Blue R250, as described (Sarthou et al., 2020).

143 2.4. Protein capture by immobilized metal affinity chromatography (IMAC)

Anion exchange chromatography (AEC) eluted proteins from leaf and root extracts were 144 divided in a rational way into 2 and 3 main homogenous fractions, respectively, covering the 145 whole polypeptide profiles (herein named leaf LA3, LB3 and LA4, LB4 fractions and root RA3, 146 147 RB3, RC3 and RA4, RB4 and RC4 fractions, from Series 3 and 4, respectively). Protein fraction aliquots were then adjusted to 0.05 to 0.1 mg/mL with binding buffer (50 mM Hepes, pH 7.4, 148 0.5 M NaCl, 0.05% (w/v) Brij) and extensively dialyzed overnight against the binding buffer. 149 150 High salt concentration and surfactant supplementation in the binding buffer aimed at minimizing non-specific interactions with the IMAC matrix (Dedieu et al., 2009). After addition 151 152 of 5 µL/mL of 0.2 M NaHCO₃ (a chelating agent for uranyl cations which acts as a competitor 153 for proteins with low uranyl affinity, when used at low concentration), to prevent weak 154 interactions, 1-2.5 mL fraction aliguots were put in contact with 50 µL freshly prepared and 155 conditioned U(VI)-loaded and U(VI)-free (chromatographic blanks) Duolite C467 beads (Basset et al., 2008; Dedieu et al., 2009), and stirred gently for 1 h at room temperature. The 156 157 supernatants (flow through) were put aside after centrifugation and the beads washed 158 extensively with binding buffer and dialyzed overnight at 4°C to remove any non-specifically bound proteins. Then, the proteins were eluted with 600 µL of 0.2 M NaHCO₃, for 90 min 159 160 (Vidaud et al., 2019). The eluates were collected, concentrated to a final volume of 60 to 200 µL using a micro-concentrator system (Amicon® Ultra, Ultracel® - 3K, 3-kDa cutoff; Millipore) 161 and stored at -20°C until use for further protein identification. Flow through fractions were 162 concentrated and stored in the same way. 163

164 2.5. Identification of the captured proteins by high-resolution tandem mass spectrometry
165 Ten μL aliquots of proteins (concentrated flow through or eluted fractions) were supplemented
166 with 20 μL of 1.5X Laemmli LDS buffer (Invitrogen). Samples were heated for 5 min at 99°C,

centrifuged for 1 min, and the resulting supernatants loaded onto a NuPAGE 4-12% Bis-Tris 167 gel (Invitrogen). After a short denaturating electrophoresis of 4 min at 200 V, gel bands 168 containing the LDS-soluble proteomes were excised and processed for an in-gel trypsin 169 proteolysis, as previously described (Hartmann et al., 2014). The resulting peptides were 170 171 analyzed using a LTQ Orbitrap XL mass spectrometer (Thermo Fischer Scientific) coupled to an Ultimate 3000 nanoRSLC nano LC system (Dionex-LC Packings). A volume of 10 µL (out 172 173 of 50 µL) of peptides was injected and desalted on-line on a reverse-phase Acclaim PepMap100 C18 micro precolumn (5 µm bead size, 100 Å pore size, 300 µm internal diameter, 174 175 5 mm length, Thermo Fisher Scientific) and then separated on an Acclaim PepMap100 C18 176 nano column (3 µm bead size, 100 Å pore size, 75 µm internal diameter, 15 cm length, Thermo 177 Fisher Scientific) at a 0.2 µL/min flow rate with a 4-48% gradient of solvent B (99.9% CH₃CN, 0.1% HCOOH) against solvent A (99.9% H_20 , 0.1% HCOOH). Each run was preceded by at 178 least one blank run avoid cross-contamination. Data-dependent acquisition was done on 179 180 peptides with potential charges of either 2+ or 3+. Full-scan mass spectra were recorded at a resolution of 30,000 from 300 to 1800 m/z with an automatic gain control target set at 5 x 10^5 181 ions. MS/MS scans were acquired on the three most abundant ions, selecting ions with a 182 dynamic exclusion of 60 s of all previously fragmented peptide ions. MS/MS spectra were 183 assigned to peptide sequences (p value ≤ 0.05) using the MASCOT search engine (version 184 2.5.1, Matrix Science) against the protein sequences of the SwissProt database (SwissProt 185 2015_02 download), selecting Arabidopsis thaliana as organism (13,640 protein sequences). 186 187 MS and MS/MS mass tolerances were set to 5 ppm and 0.5 Da, respectively. The other search of 188 parameters were: а maximum two trypsin miss-cleavages, systematic 189 carbamidomethylation of cysteine, as well as deamidation of glutamine and asparagine and 190 oxidation of methionine as three possible dynamic modifications. Proteins were identified when 191 at least two different peptide sequences were assigned. The abundance of the identified 192 proteins in each fraction was assessed by their spectral counts (number of MS/MS spectra 193 assigned per protein).

194 2.6. Proteomics data deposition

The proteomics data have been deposited to the ProteomeXchange Consortium via the PRIDE (Perez-Riverol et al., 2019) partner repository with the dataset identifier PXD033354 and 10.6019/PXD033354. [The dataset is available with the Username: reviewer_pxd033354@ebi.ac.uk and Password: 62dQFqjg].

199 2.7. Cloning of PCaP1 cDNA forms for recombinant proteins production

Sequences of primers used in this study are listed in Supplementary Table S1. Full-length *PCaP1* (At4g20260) and splicing variant *PCaP1v* cDNA coding sequences were amplified by RT-PCR using total RNA isolated from 3-week-old Arabidopsis leaves with the RNeasy plant mini extraction kit (Qiagen) and reverse-transcribed with the ThermoScript RT-PCR system

(Invitrogen). The cDNA fragments were amplified by PCR with primers introducing Ncol and 204 Sall restriction sites upstream the initiation codon and downstream the stop codon, 205 206 respectively (Supplementary Table S1). The PCR products were sequenced (Eurofins) and 207 ligated into the pET28b+ plasmid (Novagen) between the *Ncol* and the *Sall* sites to obtain the 208 pET28-PCaP1 and pET28-PCaP1v recombinant plasmids, respectively. The cDNA encoding the truncated PCaP1 form, PCaP1_A, was obtained by PCR amplification using pET28-PCaP1 209 plasmid as a matrix and specific primers (Supplementary Table S1). The PCR product was 210 ligated into the pET28b+ plasmid to obtain the pET28-PCaP1∆ recombinant plasmid. The 211 212 resulting pET28-PCaP1 constructs were amplified into *Escherichia coli* DH5 α cells and then 213 introduced into the E. coli overexpression host Rosetta 2 (DE3) (Stratagene).

214 2.8. Production and purification of the recombinant PCaP1 forms

The transformed cell cultures were grown at 37°C in lysogeny broth (LB) medium 215 supplemented with the appropriate antibiotics until A_{600} was 0.6. Then, isopropylthio β -D-216 galactoside was added to a final concentration of 0.4 mM and incubations pursued for 15 h at 217 37°C. Cell pellets from 2-L cultures were resuspended in 50 mL of extraction buffer containing 218 20 mM Tris-HCl, pH 8, 10% (w/v) glycerol, 1 mM dithiothreitol and a cocktail of complete 219 220 protease inhibitors (Roche Applied Science), and then disrupted by sonication with a Vibra-221 Cell disrupter (Branson Ultrasonics). Cell debris were eliminated by centrifugation at 40,000 x 222 g for 30 min and soluble proteins recovered from supernatants were subjected to ammonium 223 sulphate precipitation at 4°C with crystalline ammonium sulphate to 50% saturation. The resulting precipitates were collected by centrifugation (40,000 x g, 15 min, 4 °C), resuspended 224 in 10 mM potassium phosphate buffer, pH 7 supplemented with protease inhibitors and 225 dialyzed overnight at 4°C against 4 L of 10 mM phosphate buffer, pH 7. Protein samples were 226 227 applied onto a Bio-Gel HTP hydroxylapatite column (2.6 x 7 cm, Bio-Rad) equilibrated with 10 mM potassium phosphate buffer pH 7. Proteins were eluted with a linear gradient of phosphate 228 229 buffer from 10 to 350 mM potassium phosphate (320 mL) at a flow rate of 0.5 mL/min. Five-230 mL fractions were collected. Fractions containing recombinant PCaP1 proteins were pooled 231 and desalted by successive concentrations/dilutions with 10 mM Tris-HCI buffer, pH 7.5 containing 10% (w/v) glycerol, using Amicon® Ultra-15 centrifugal filtration units (3 kDa cut-232 off; Millipore). Samples were then loaded onto a Fractogel EMD-DEAE (650 M) column (1.6 x 233 5 cm; Merck) equilibrated with 10 mM Tris-HCI, pH 7.5 buffer containing 10% (w/v) glycerol. 234 235 After extensive wash of the column with five volumes of buffer, proteins were eluted using a 236 150-mL linear gradient from 0 to 500 mM NaCl in this buffer, at a flow rate of 0.5 mL/min. 237 PCaP1-containing fractions were pooled and concentrated before being applied onto a 238 Hiload® Superdex 75 column (1.6 X 60 cm, Cytiva) equilibrated with 10 mM Tris-HCl buffer, pH 7.5 containing 10% (w/v) glycerol and 150 mM NaCl. Elution was conducted in the same 239

buffer at a flow rate of 1 mL/min. Purified recombinant proteins were finally concentrated and stored aliquoted at -80°C, until used. Protein purity and integrity were monitored by SDS-PAGE and mass spectrometry under denaturing conditions. Purified protein concentration was determined by recording UV absorption spectra, using a NanoDrop 2000 spectrophotometer (Thermo Fischer Scientific) (Mass Extinction Coefficients at 280 nm were $E_{1\%} = 4.66 \text{ Lg}^{-1} \text{ cm}^{-1}$ for PCaP1; 5.70 for PCaP1v and 7.32 for PCaP1 Δ , as calculated from the Molar Extinction Coefficient $\varepsilon_{280} = 11460 \text{ M}^{-1} \text{ cm}^{-1}$, ProtParam ExPASy).

Production and purification of recombinant PCaP1 protein uniformly labelled with ¹⁵N and ¹³C ($U^{-15}N$, ¹³C-PCaP1) or labelled with ¹⁵N and ²H (²H, ¹⁵N, -PCaP1) was performed similarly as for unlabeled protein, except that growth was conducted in 2 L of a minimal M9 medium, containing antibiotics and supplemented with either ¹⁵N-NH₄Cl (15N, 99%) and ¹³C-glucose (U-13C6, 99%) or ¹⁵N-NH₄Cl (15N, 99%) and D₂O (99.9%) (Cambridge Isotope Laboratories, Inc.), respectively, after progressive medium acclimatization from pre-cultures conducted in LB medium.

254 The presence of recombinant PCaP1 in eluted fractions during the purification process was 255 assessed by western-blotting, initially using a commercially available polyclonal rabbit antibody 256 raised against KLH-conjugated peptide derived from the Arabidopsis PCaP1 sequence (1:2000 dilution) (Agrisera; Ide et al. (2007)), and subsequently using a more sensitive custom-257 258 made rabbit polyclonal PCaP1 antibody raised against the whole purified recombinant protein 259 (1:50000 dilution) (Covalab). Proteins were separated by SDS-PAGE, electro-transfered to nitrocellulose membrane and probed using the PCaP1 antibody and horseradish peroxidase-260 conjugated anti-rabbit IgGs (Bio-Rad). Protein detection was achieved using the ECL Plus[™] 261 262 Western Blotting detection reagents and a Typhoon 9400 imager (Cytiva).

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2.9.

Native mass spectrometry

Mass Spectrometry under native conditions was performed to determine the oligomeric state 264 of PCaP1. Prior to analysis, purified protein (a 100-µL 50 µM aliquot) was extemporaneously 265 266 chromatographed onto a Superdex 200 Increase 10_300 GL column equilibrated with 250 mM 267 ammonium acetate, pH 8, for buffer exchange. Protein was injected in the spectrometer at a 7 268 µM final concentration. A nanoflow electrospray (nano-ESI) source and Nanoflow platinum-269 coated borosilicate electrospray capillaries (Thermo Electron SAS, Courtaboeuf, France) were used to generate ions (Boeri Erba et al., 2018; Puglisi et al., 2020). A quadrupole time-of-flight 270 mass spectrometer (Q-TOF Ultima, Waters Corporation, Manchester, U.K.) modified for the 271 272 detection of high masses was operated with the following parameters: capillary voltage = 1.2-1.3 kV, cone potential = 40 V, RF lens-1 potential = 40 V, RF lens-2 potential = 1 V, aperture-273 1 potential = 0 V, collision energy = 30–140 V, and microchannel plate (MCP) = 1900 V. The 274 275 mass spectrometer was calibrated externally with a solution of cesium iodide at 6 mg/mL 276 dissolved in 50% isopropanol. Mass spectra were processed using the Masslynx 4.0 software (Waters Corporation, Manchester, U.K.), Massign software package (Morgner and Robinson,
2012) and UniDec (Marty et al., 2015).

279 2.10. Metal-binding affinities measured by fluorescence titration

Fluorescence spectroscopy was performed using a MOS450 Bio-Logic spectro-fluorimeter (Bio-Logic-Science Instruments, Claix, France) set at 277 nm for excitation. The slit widths for emission and excitation were set at 8 nm. The spectro-fluorimeter was driven by Bio-Kine 32 4.1 software. The measurements were performed at 25°C, in a 1 cm path cell.

For fluorescence titrations of PCaP1 in the presence of copper (CuCl₂), ferric iron (Fe(NO₃)₃) 284 or calcium (CaCl₂), we used a 0.8 µM protein solution in 10 mM Tris-HCl pH 7.5 or 10 mM 285 286 MES pH 6 buffer with 100 mM NaCI. The fluorescence was transmitted through a long pass 287 filter (FF01-300/LP-25, Semrock) and emission was collected at 342-348 nm. The apparent dissociation constant (K_d app) and the stoichiometry of metal binding to PCaP1 (n) were 288 determined using the fluorescence titration curves. The variation of PCaP1 fluorescence 289 290 intensity as a function of the metal concentration was analyzed by using the following equation (Eq. 1) as described (Nagasaki-Takeuchi et al., 2008): 291

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$$(F_0 - F) = \frac{\Delta Fmax \times [Metal]}{K_dapp + [Metal]}$$

where, F_0 and F are fluorescence intensities measured at 342-348 nm without or with the presence of metal, respectively; ΔF_{max} is the maximum fluorescence change observed when the protein is fully saturated at the specific metal-binding sites with an apparent affinity of K_d app in a binding-site dependent manner; [Metal] and [PCaP1] are concentrations of free metal and PCaP1 in the assay, respectively. We also applied the following equation (Eq. 2) to determine the ligand-binding number (*n*) with the postulate that PCaP1 has *n* binding sites with identical affinity (Nagasaki-Takeuchi et al., 2008):

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(2)

303

$$\frac{1}{F_0 - F} = \frac{2n}{\Delta Fmax} \times \frac{1}{[Metal]/[PCaP1]} + \frac{1}{\Delta Fmax}$$

Fitting of the data was performed with KaleidaGraph 4.51 software (Synergy Software, PA,USA).

For fluorescence titrations of PCaP1 in the presence of U(VI), we used a 0.8 µM protein solution in 10 mM MES pH 6 buffer with 100 mM NaCI. Iminodiacetate (IDA) was added to the protein solutions at a IDA:protein ratio of 12.5:1 to prevent the appearance of hydroxo complexes and to control U(VI) speciation (Pardoux et al., 2012). Fluorescence emission spectra were acquired from 280 to 470 nm. Competition experiments between the PCaP1 proteins and IDA were performed to calculate the conditional stability constants of the U(VI)-

protein complexes at pH 6. IDA binds uranyl with mild affinity and forms the UO2IDA, 312 $[UO_2(IDA)_2]^{2-}$ and $[(UO_2)_2(IDA)_2(OH)_2]^{2-}$ complexes. The conditional stability constants of these 313 three species at 25°C and 0.1 M ionic strength were calculated from published global constants 314 315 (Jiang et al., 2002; Starck et al., 2015). These values were set up in the spectral data analyses, which were performed using the SPECFIT software (Binstead et al., 2003; Gampp et al., 1985). 316 Several models for the UO₂²⁺-protein complexes were assayed, and only the one predicting a 317 UO₂(Protein)₂ specie only gave a good fit for the experimental data. The model considering a 318 non-luminescent complex was the most reliable. Experiments were repeated to ensure 319 320 reproducibility.

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2.11. Metal quantification in metal-protein complexes by Size Exclusion Chromatography (SEC) and Inductively Coupled Plasma-Mass Spectrometry (ICP-MS)

Metal-protein complexes were prepared by incubating 10 µM purified PCaP1 solutions for 15 323 min at 25°C, in 20 mM Tris-HCl pH 7.5, or 20 mM MES pH 6 buffer containing 150 mM NaCl 324 and 2 mM CuCl₂, 2 mM Fe(NO₃)₃, 10 mM CaC₂ or 50 µM UO₂(NO₃)₂, as indicated. Formed 325 complexes were separated from unbound metals by SEC, through centrifugation for 2 min at 326 724g of the samples (150 µL), on MicroSpin[™] G-25 columns (Cytiva), equilibrated with the 327 appropriate working buffer (Tris-HCI/NaCI or MES/NaCI). Controls without protein were run in 328 329 parallel for background correction. Binding assays were performed at least in triplicate for each metal. Protein in the eluates was quantified by recording A₂₈₀. Eluates (50 µL aliquots) were 330 then incubated with 10% (v/v) HNO₃ (Suprapur, Merck) for 2h at 60°C to ensure protein 331 332 denaturation and release of bound elements. Precipitated proteins were removed by centrifugation and the mineralized samples were then diluted in 0.5% (v/v) HNO₃ and analyzed 333 334 by ICP-MS using an iCAP RQ quadrupole mass instrument (Thermo Fisher Scientific GmbH, 335 Germany), as detailed previously (Sarthou et al., 2020). Elements (⁴⁴Ca, ⁵⁶Fe, ⁵⁷Fe, ⁶³Cu, ⁶⁵Cu, ²³⁸U) were analyzed using the standard mode and the collision mode with helium as a cell gas. 336 337 Elements concentration was determined using standard curves and corrected using a solution of ⁴⁵Sc, ¹⁰³Rh and ¹⁷²Yb, as an internal standard. Data integration was performed using the 338 339 Qtegra software (Thermo Fisher Scientific GmbH, Germany).

340 2.12. Solution-state Nuclear Magnetic Resonance (NMR) studies

NMR experiments were performed on Bruker Avance IIIHD spectrometers operating at magnetic field strengths of 16.5, 20, and 22.3 T (¹H Larmor frequency of 700, 850 and 950 MHz). All spectrometers were equipped with helium-cooled triple-resonance probes (HCN TCI 5mm) and pulsed z-field gradients. NMR assignments were obtained from a set of 3D Bandselective Excitation Short-Transient Transverse Relaxation-Optimized Spectroscopy (BEST-TROSY) (Favier and Brutscher, 2011) type correlation experiments: HNCO, HNCACO, HNCA, HNCOCA, HNCACB, and HNCOCACB. These experiments were performed on uniformly $^{13}C/^{15}N$ labeled PCaP1 samples (in 10 mM Tris-HCl, pH 7.5 buffer, at 300K) and sample concentrations ranging from 100 to 500 μ M.

- 350 Translational diffusion constants of the proteins in solution were measured by 1D ¹H Diffusion
- 351 Ordered Spectroscopy (DOSY) experiments (Johnson, 1999) focusing either on the methyl or
- amide spectral region. A series of 1D spectra was recorded with varying gradient strength and
- a total acquisition time of about 15 min.
- ^{15}N relaxation experiments (T₁ and T₂) were performed on uniformly ^{15}N -labeled full-length
- 355 PCaP1 on the free form and in complex with U(VI) using standard NMR pulse sequences
- 356 (Farrow et al., 1994). The ¹⁵N T₁ signal decay was sampled for 12 time points varying between
- 357 0 to 1.8 s while ${}^{15}N$ T₂ relaxation experiments were recorded with relaxation delays ranging
- 358 from 0 to 280 ms.
- NMR titration experiments were performed by adding up to 5-fold excess of U(VI) to either full length PCaP1 or PCaP1∆ and recording at each titration step a 2D ¹H-¹⁵N BEST-TROSY
 spectrum.
- All NMR experiments handled in this study were implemented in the NMRlib pulse sequence library (Favier and Brutscher, 2019) that can be freely downloaded from the IBS website (<u>http://www.ibs.fr/research/scientific-output/software/pulse-sequence-tools</u>). The experiments were processed and analyzed using Bruker Topspin 3.5 and CCPNMR V3 software (Collaborative Computing Project for NMR; https://ccpn.ac.uk/software/downloads/).
- 367 2.13. Characterization, genetic complementation and U(VI) treatment of the Arabidopsis
 368 pcap1.4 mutant

Seeds of the Arabidopsis *pcap1.4* knockout mutant (Li et al., 2011; Nagata et al., 2016) were 369 obtained from the Arabidopsis Biological Resource Center (registration no. SAIL_241_A08). 370 Homozygous status of the T-DNA insertion line was checked by PCR, using gene-specific and 371 T-DNA border primers (Supplementary Table S1). The T-DNA insertion sites within exon 4 of 372 373 PCaP1 gene (Li et al., 2011) were confirmed by sequencing. For genetic complementation 374 experiments, the complete coding sequence of *PCaP1* was fused to a double enhanced 35S 375 promoter using the plasmid pFP101 containing a green fluorescent protein (GFP) reporter 376 gene (Bensmihen et al., 2004). This construct was introduced into homozygous pcap1.4 plants 377 by the Agrobacterium-mediated floral-dip transformation method (Clough and Bent, 1998). Transformed seeds expressing the GFP marker were selected by fluorescence detection. All 378 primer sequences are available in Supplementary Table S1. PCaP1 expression in Arabidopsis 379 lines was checked by western blot analysis of total proteins extracted from leaves with 10 mM 380 Tris-HCl, pH 7.5 buffer, containing 1 mM dithiothreitol, 1% SDS and a cocktail of protease 381 inhibitors (Roche Applied Science), using custom-made PCaP1 antibody. 382

For U(VI) treatment, plants were first grown in hydropony for 5 weeks in Gre medium with weekly solution changes until treatment, and transferred to Gre medium depleted with

phosphate and complemented with 20 µM uranyl nitrate (UO₂(NO₃)₂). Plants were harvested 385 1 to 11 days after U(VI) treatment. To remove U(VI) weakly bound on the root surface, the 386 plants were first rinsed once with 10 mM Na₂CO₃, then twice with distilled water. Photosystem 387 388 II efficiency in dark-adapted leaves (Fv/Fm) was assessed using a FluorPen (FP100/D, Photon 389 Systems Instruments, Brno, Czech Republic) and used as an indicator of plant health. Roots 390 and shoots were separated, quickly dried on absorbing paper and fresh biomass was then 391 determined. Finally, roots and shoots were dehydrated at 80°C for 24 h and dried samples were mineralized in 0.5-1 ml of 65% (w/v) HNO₃ (Suprapur, Merck), diluted in distilled water 392 393 and analyzed by ICP-MS, as described previously (Sarthou et al., 2022).

394 Statistical analyses were performed using the R Studio software (RStudio Team, 2015) and 395 the *nparcomp* package (Konietschke et al., 2015). Statistical significance was determined 396 using non-parametric Tukey tests conducted with a confidence level set at 99% (p < 0.01).

397 398

3. Results

399 3.1. Identification of 38 UraBPs from Arabidopsis roots and shoots captured by IMAC 400 on U(VI)-loaded beads

To isolate UraBPs from plant extracts, we used the efficient uranyl-loaded Duolite C467[™] 401 matrix previously developed for the identification of UraBPs from human serum, kidney-2 and 402 dopaminergic SH-SY5Y cells (Basset et al., 2008; Dedieu et al., 2009; Vidaud et al., 2019). 403 The experimental design is depicted in Figure 1. Soluble proteomes isolated from roots and 404 405 leaves from hydroponically grown Arabidopsis thaliana plants were first pre-fractionated by anion exchange chromatography (AEC) on High Performance Q-Sepharose column. Protein 406 407 concentration profiles and SDS-PAGE analysis of the sub-fractions obtained from four 408 independent biological experiments are presented in Supplementary Figure S1. This 409 fractionation procedure reduced the complexity of the protein samples in a very reproducible 410 way and enriched the less abundant proteins facilitating their capture and identification. As an example, in the case of the leaf proteome, the method allowed to separate the very abundant 411 412 Rubisco protein (present in the LB fraction) from less abundant proteins displayed in the LA 413 fraction (Supplementary Figure S1). In the case of the root proteome, the method led to the 414 separation of three enriched homogenous protein fractions, RA, RB and RC.

Two of the four-fractionation experiments conducted with Arabidopsis root and shoot tissues (Series 3 and 4, Supplementary Figure S1) were used as starting biological materials (biological duplicate, experiments 1 and 2) to capture UraBPs by immobilized metal affinity chromatography (IMAC) on U(VI)-loaded beads in a batch system, and to subsequently identify them by mass spectrometry (Figure 1). Fractionation experiments from series 1 and 2 were set aside for analyses unrelated to this work (unpublished data). All the specifically captured proteins obtained from the ten IMAC experiments (five from experiment 1 and five from

experiment 2), as well as the corresponding flow through supernatants were submitted to 422 nanoLC-ESI MS/MS analysis for identification (Supplementary Table S2). The IMAC process 423 424 including the capture conditions and the washing steps was highly efficient and selective since 425 we did not detect any protein in the fractions eluted from the different control runs on U(VI)-426 free beads, except for the highly abundant large Rubisco subunit from leaf extracts, which was thus not considered as a U(VI)-affine protein (Supplementary Table S2). Taken together, our 427 428 analyses identified a total of 338 different proteins from leaves and roots, among which 38 429 were specifically captured on U(VI)-loaded beads, and were named UraBP1 to 38 430 (Supplementary Table S2). The overall low number of total proteins identified in our analyses 431 was the consequence of the high dilution and the small amounts of protein fractions exposed 432 to U(VI)-loaded beads, which was a prerequisite for high selectivity and very low background noise. We considered as genuine UraBPs only proteins captured from identical AEC fractions 433 in experiments 1 and 2, or proteins captured from different AEC fractions from the same organ 434 435 (leaf or root) (highlighted in red and pink, respectively in Supplementary Table S2). Single occurrences were not considered as proteins of interest. 436

Among the 38 UraBPs identified, 30 were specifically captured from leaf protein extracts, 4 437 from root protein extracts and 4 from both tissues (Supplementary Table S2). Interestingly, 10 438 out of the 38 UraBPs (UraBP3, 15, 18, 21, 25, 28-31 and 33) were detected only in captured 439 440 fractions and not in the flow through fractions, or were highly enriched on the captured fractions 441 (on a spectral count basis) (Supplementary Table S2). The other UraBPs were detected 442 similarly in both fractions and could either correspond to abundant UraBPs saturating U(VI)-443 loaded beads or to proteins with low uranyl binding affinity with a free/bound equilibrium being 444 established. In order to select the most interesting UraBPs for further analyses, we performed gene ontology (GO) and semantic terms enrichment analyses (biological process and/or 445 446 molecular function in relation with metals) as well as analyses of some of their physico-447 chemical characteristics (Supplementary Table S2). This included their content in some amino 448 acid residues known to coordinate with U(VI) (the oxygen of aspartate, glutamate, and tyrosine and the nitrogen of histidine are both hard Lewis bases and are able to bind U(VI)). Also, 449 450 phosphorylated amino acid residues that are known to interact with U(VI) and have been 451 shown to strengthen the binding of U(VI) to aspartate and glutamate were analysed (Garai and Delangle, 2020; Laporte et al., 2019; Pardoux et al., 2012; Sauge-Merle et al., 2017; Starck et 452 al., 2017). Firstly, these analyses indicated that most of the Arabidopsis UraBPs (27/38 i.e. 453 71% of the protein data set) are described as metal-binding proteins and/or are metal-454 responsive proteins. The phosphoproteins were also greatly represented (29/38 i.e. 76% of 455 456 the protein data set), which is quite higher than in global proteomes (about 40-47% in 457 Arabidopsis) (Mergner et al., 2020; van Wijk et al., 2014). Among phosphoproteins, 20 out of 29 were poly-phosphorylated. Particularly, UraBP17 (PSBP1), UraBP18 (BCA2), UraBP23
(RCA), UraBP25 (PCaP1), UraBP28 (PSBQ2), UraBP33 (RBG7) and UraBP38 (CATA3)
contain more than five phosphorylation sites (Supplementary Table S2). Noticeably, 38% of
the phosphoproteins present one or more phosphotyrosine group, which is much higher than
in global *Arabidopsis* phospho-proteome (< 5%) (Mergner et al., 2020; Reiland et al., 2009;
van Wijk et al., 2014).

With all these criteria taken into consideration, we decided to focus our attention more 464 particularly onto UraBP25 for more detailed characterization. Indeed, UraBP25 is a Ca²⁺ 465 466 regulatory protein that binds Ca²⁺ and Cu²⁺ ions that was previously reported as the Plasma 467 membrane-associated Cation-binding Protein 1 (PCaP1), or the Microtubule Destabilizing Protein of 25 kDa (MDP25) (Ide et al., 2007; Li et al., 2011; Nagasaki-Takeuchi et al., 2008; 468 Nagasaki et al., 2008; Qin et al., 2014). UraBP25 accumulates in response to Cu²⁺ and Mg²⁺ 469 ions excess and its abundance is lowered in response to Fe³⁺ ions exposure (Ide et al., 2007; 470 Nagata et al., 2016). Moreover, UraBP25 was isolated from both leaf and root extracts on 471 472 U(VI)-loaded beads and was highly enriched in captured fractions from both R3A, R3B and L3A, L4B AEC fractions (Supplementary Table S2). Finally, UraBP25 is a poly-phosphorylated 473 protein, with 9 identified sites, and presents a high frequency of glutamate residues (44/225 474 *i.e.* 19.6% of total residues) (Supplementary Table S2), compared to 6.65% on average in the 475 476 Arabidopsis proteome (Tsuji et al., 2010).

477

3.2. PCaP1 binds U(VI) with high affinity and Cu(II) and Fe(III) in high proportions 478 In order to confirm the capacity of PCaP1 from Arabidopsis thaliana to bind U(VI) and to further 479 480 characterize its metal-binding properties, we cloned the cDNA encoding the full length PCaP1 protein by RT-PCR and produced the recombinant protein devoid of any tag sequence in E. 481 *coli*. We used a three-step chromatographic procedure to purify recombinant PCaP1 protein 482 483 to near homogeneity (Figure 2). The integrity of the preparations was confirmed by LC-ESI 484 TOF MS under denaturing conditions. The experimental mass of PCaP1 was 24,453.06 Da, matching the amino acidic sequence 2-225 (with loss of the N-terminal methionine) 485 (Supplementary Figure S2A). As previously observed (Ide et al., 2007), the purified protein 486 487 was detected as a 36 kDa polypeptide on SDS-PAGE instead of 24.5 kDa due to its peculiar 488 amino acid composition (Figure 2A), PCaP1 being particularly rich in glutamate, lysine and valine residues. Also, recombinant PCaP1 protein behaved as an apparent ~90 kDa globular 489 490 protein by size exclusion chromatography (SEC) onto a Superdex 200 Increase 10/300 GL 491 column, as calculated from a calibration curve obtained by measuring elution volumes of spherical folded calibration proteins (Figure 2B). However, native ESI MS analysis indicated 492

that PCaP1 was essentially a monomer (1mer) in solution under our assay conditions, with a small proportion of dimer (2mer) also being detected (Figure 2C). The unexpected migration of recombinant protein by SEC analysis could be explained by the occurrence of an intrinsically unstructured region in PCaP1 structure, as suggested previously by far-UV circular dichroism spectroscopy analyses (Nagasaki-Takeuchi et al., 2008).

In a previous biochemical and biophysical study, Nagasaki-Takeuchi et al. (2008) 498 499 demonstrated by fluorescence titration and circular dichroism analyses that a recombinant A. 500 thaliana PCaP1 protein was able to bind Cu(II) ions. These analyses indicated that binding of 501 Cu(II) altered the structure of PCaP1 locally at proximity of aromatic amino acid residues, and 502 particularly of Tryptophan 4 (Trp4), the unique Trp residue within the protein sequence. Here, 503 taking advantage of this property, we checked for U(VI) binding to PCaP1 by Trp fluorescence titration. We observed that the fluorescence emission peak at 342 (typical for Trp residue 504 exposed to a water solution) when excited at 277 nm, was also markedly guenched by the 505 addition of U(VI) ions at low concentrations to recombinant PCaP1 protein (up to 4 equivalents 506 on a monomer basis) buffered at pH 6, suggesting interaction of the metal with the protein 507 (Supplementary Figure S3). Addition of an excess of EDTA restored almost completely the 508 original fluorescence spectrum indicating reversibility of U(VI) binding and testifying of protein 509 510 integrity maintenance upon U(VI) exposure. Trp fluorescence quenching was followed by titration of a 0.8 µM PCaP1-buffered solution at pH 6 containing a 12.5-fold excess of 511 iminodiacetic acid (IDA), a well-known low affinity competitor for U(VI) binding, to control uranyl 512 513 speciation (Figure 3A). The stability constants of complexes formed with IDA at pH 6 being known (Jiang et al., 2002), one can infer the conditional stability constant of the uranyl 514 515 complexes with PCaP1. Under our assay conditions, the fluorescent peak at 342 nm gradually 516 decreased with the increase in the U(VI) concentration from 0.1 to 2 μ M and reached less than 517 20% of the original fluorescence (Figure 3A). The evolution of fluorescence intensity with the 518 quantity of uranyl added was characteristic of the formation of a (UO_2^{2+}) -(Protein)₂ complex, since the extrapolation of the initial slope converged to 0.5 uranyl equivalent in the complex. 519 520 This was clearly confirmed in the fitting procedure (Figure 3B). Indeed, the best fit to the 521 experimental data was obtained with a model considering a non-fluorescent $(UO_2^{2^+})$ -(Protein)₂ complex only, involving two proteins for one single uranyl cation. The calculated conditional 522 523 stability constant for the formation of this species deduced from the deconvolution of 524 fluorescence data was $\log \beta_{12} = 14.5 \pm 0.4$. This constant was defined for a 1:2 complex so its value has not the same order of magnitude than a $1/K_d$ value. Thus, this value cannot be 525 compared directly with literature data reporting β_{11} constants (1:1 complex) or K_d values. To 526 make such a comparison possible, we simulated competitions with a (UO₂²⁺)-P' complex with 527 a theoretical competing P' protein of 1 nM affinity ($\log \beta_{11} = 9$), 10 nM ($\log \beta_{11} = 8$) or 100 nM 528 $(\log \beta_{11} = 7)$. This simulation indicated that the protein has an affinity for uranyl similar to a 529

- 530 protein having a 10 nM affinity ($\log\beta_{11} = 8$) in a (UO_2^{2+})-P' complex, using a similar approach 531 (Supplementary Figure S4). Trp fluorescence quenching was also followed by titration of a 532 PCaP1-buffered solution at pH 7.5 (Supplementary Figure S5A). The evolution of fluorescence 533 intensity with the quantity of uranyl added could not be easily interpreted since there was no 534 plateau with an addition of a large amount of U(VI) up to 20 µM (Supplementary Figure S5B). 535 So, fits were not appropriate under these conditions.
- To study the stoichiometry of U(VI) complexation to PCaP1 at pH 6 and 7.5, we determined 536 U(VI) content by ICP-MS in U(VI)-PCaP1 complexes formed after incubation of 10 µM PCaP1-537 buffered samples in the presence of 50 µM uranyl nitrate (5 U(VI) equivalents). Separation of 538 539 complexes from unbound metal was achieved through centrifugation onto Sephadex 540 MicroSpin G-25 columns. We did not perform such analyses at higher U(VI) to protein ratios since under large U(VI) excess conditions, protein samples were somewhat affected, as 541 monitored by UV-Vis absorption, and a slight precipitate was observed, particularly when 542 543 titration was done in MES buffer at pH 6 (not shown). Under optimized conditions, data obtained revealed a ligand-binding number of 0.50 ± 0.03 U(VI) ions per PCaP1 monomer at 544 pH 6 (Supplementary Table S3), in line with fluorescence spectroscopy studies (Figure 3B). 545 Since PCaP1 was found to be mainly a monomer in solution (Figure 2C), these data could 546 547 suggest dimerization of the protein upon U(VI) exposure under our experimental conditions, with one uranyl cation located at the interface of two monomers (compatible with formation of 548 a 1:2 complex, in line with fluorescence titration experiments; Figure 3B). When performed at 549 pH 7.5 in Tris-HCl buffer, binding experiments revealed a U(VI) content of the resulting U(VI)-550 protein complex of 2.11 ± 0.07 U(VI) ions per PCaP1 monomer (Supplementary Table S3). 551
- 552 As mentioned before, Nagasaki-Takeuchi et al. (2008) characterized Cu(II)-binding properties 553 of PCaP1 by fluorescence titration of Trp4 with CuCl₂ increasing concentrations and monitoring 554 of changes in fluorescence spectra. The obtained values of apparent K_d (K_d app) and ligand-555 binding number, as determined by indirect fluorescence and absorption spectroscopy methods, were 10 to 16 µM and 6 to 7 Cu(II) ions per monomer, respectively (Nagasaki-556 557 Takeuchi et al., 2008). In our hands, under similar assay conditions, the fluorescence emission 558 peak at 342 nm was also markedly quenched when Cu(II) was added in the protein solution, at concentrations up to 300 µM (Supplementary Figure S6A). From the collected data we 559 calculated a K_{d} app for Cu(II) binding of 29 μ M, and a ligand-binding number of 13 ions per 560 561 monomer (Supplementary Table S4).
- PCaP1 has been shown to interact with calcium and calmodulin in a calcium-dependent manner and has both *in vitro* and *in vivo* calcium-dependent microtubule destabilizing and actin severing activities (Li et al., 2011; Nagasaki et al., 2008; Qin et al., 2014). Despite PCaP1 being a genuine Ca(II)-binding protein, thus requiring calcium for its activities, we observed that the addition of CaCl₂ has a limited effect on fluorescence emission, at pH 7.5. However,

fluorescence changes were significant at pH 5.5, yielding a K_d app of 241 μ M (Supplementary 567 Figure S6B and Supplementary Table S4). Interestingly, Fe(III) but not Fe(II) ions also affected 568 569 Trp fluorescence emission in a similar way as Cu(II) ions (K_dapp of 26 μ M and ligand-binding number of about 19 ions per monomer; Supplementary Figure S6C and Supplementary Table 570 571 S4). Finally, none of the other metal ions tested had any impact on fluorescence emission (Supplementary Table S4). In order to confirm the ability of PCaP1 to bind Cu(II), Ca(II) and 572 573 Fe(III) ions, we used SEC on MicroSpin G-25 columns and ICP-MS analyses to measure 574 metals in protein complexes resulting from the incubation of 10 µM PCaP1 with large excess 575 of any of the metal ions. Data obtained revealed an apparent stoichiometry for Cu(II)-PCaP1 576 protein complexes of 15:1, consistent with the values estimated by fluorescence spectroscopy. 577 Surprisingly, much higher values were observed by direct ICP-MS determination for Fe(III)-578 PCaP1 protein complexes (stoichiometry of 60:1 compared to 19:1 by fluorescence 579 spectroscopy; Supplementary Table S4). Unfortunately, we were unable to measure 580 accurately the calcium content of Ca(II)-PCaP1 protein complexes because of unexplained high background values of unbound metal ion in MicroSpin G-25 column eluates. 581

582 PCaP1 is known to bind the physiological ion Ca(II) both in vivo and in vitro (Li et al., 2011; Nagasaki et al., 2008; Qin et al., 2014). However, we have shown that at physiological pH (pH 583 7.5), Ca(II) alone did not significantly guench PCaP1 fluorescence (Supplementary Table S4), 584 therefore a competition between UO_2^{2+} and Ca(II) could be evidenced by an increase in Trp 585 fluorescence intensities of U(VI)-PCaP1 protein complexes upon addition of increasing Ca(II) 586 587 concentrations (Supplementary Figure S7). Thereby, addition of 8 mM Ca(II) fully restored Trp fluorescence intensity of PCaP1 protein initially guenched to half its initial value by 20 µM 588 589 U(VI), which suggested identical or close binding-sites for both metals and indicated a much 590 higher affinity of the protein for U than for calcium, even at physiological pH.

591

592 3.3. U(VI)-binding to PCaP1 induces oligomerization

593 In order to obtain information about the *in vitro* interaction of U(VI) with PCaP1 at the molecular 594 level, we initiated structural studies of the complex. The recombinant protein was predicted to 595 be an instable protein (Instability Index of 53.55 using ProtParam tool; Supplementary Table 596 S2) due to the presence of a long intrinsically disordered C-terminal domain (residues 140-225, as identified by computational disorder prediction tools) (Supplementary Figure S8A). All 597 our attempts to crystallize PCaP1 were unsuccessful, most likely because of this high degree 598 599 of flexibility. Consequently, structure determination based on X-ray crystallographic data was not pursued. As an alternative strategy, we have undertaken solution-state NMR spectroscopic 600 601 analyses of this protein. Solution-state NMR spectroscopy is a well-adapted technique for low 602 molecular weight proteins (< 30 kDa), which is the case for PCaP1, providing information about structure and conformational dynamics of proteins over a wide range of time scales and degrees of order. NMR experiments require samples at high protein concentration (typically > 100μ M), for which U(VI) exposure triggers precipitation of PCaP1 at pH 6, but not at pH 7.5. Thus, all NMR experiments were performed at pH 7.5 with U(VI):protein ratios in the range between 0:1 to 5:1.

The 1D ¹H NMR spectrum of PCaP1 shows a few backbone amide (HN) and side-chain methyl (CH₃) NMR resonances outside the typical random-coil region, indicative of the presence of well-folded protein segments (Supplementary Figure S8B). In addition, a more quantitative assessment of the degree of tertiary structure in PCaP1 using the HETSOFAST approach (Schanda et al., 2006) confirmed the presence of both well-folded domains and flexible protein regions (Supplementary Figure S8C).

In order to obtain site-specific information on the structural and dynamic properties of 614 recombinant PCaP1, we produced uniformly ¹³C, ¹⁵N-labeled PCaP1 and performed a set of 615 616 3D NMR assignment experiments (Best-TROSY HNCO, HNCACO, HNCA, HNCOCA, HNCACB and HNCOCACB) (Favier and Brutscher, 2011; Solyom et al., 2013). Based on this 617 data set, 85% of the non-proline PCaP1 residues were unambiguously assigned (Figure 4A; 618 Supplementary Figure S9A). The assigned chemical shifts revealed the presence of a N-619 terminal structured domain (residues 29-129), mainly formed by α -helices, and a highly flexible 620 621 (intrinsically disordered) C-terminal domain (residues 132-225) (Figure 4B; Supplementary 622 Figure S9A). We did not intend an atomic-resolution structure determination by NMR, but rather computed a structural model with AlphaFold2, a DeepMind Al-based computational tool 623 that was proven to produce accurate structural models for the majority of the protein targets 624 625 tested (Jumper et al., 2021). The resulting AlphaFold2 structural model (Figure 4C) presents 626 all the secondary structural features (α -helical structure) that have been derived from our NMR data, providing an independent validation of this computational model. 627

628 The 2D ¹H-¹⁵N correlation spectra of PCaP1 recorded in the presence of increasing amounts 629 of U(VI) resulted in a progressive decrease of peak intensities for residues located in the 630 structured N-terminal part, while no such intensity loss was observed for the highly flexible C-631 terminal segment (Figure 5A). In contrary, some amide sites in this flexible part even gained 632 in NMR signal intensity. Interestingly, no changes in peak positions were observed during this NMR titration experiment. These findings suggest that the structured N-terminal part of PCaP1 633 oligomerizes upon U(VI) binding. In order to obtain a more quantitative idea about the average 634 635 particle size in PCaP1:U(VI) mixed samples, we performed NMR-based translational diffusion experiments. The intensity decay of observed ¹H signals, mainly corresponding to the flexible 636 C-terminal domain, as a function of the applied gradient-field strength are plotted in Figure 5B 637 for U(VI):PCaP1 ratios of 0:1; 1:1 and 5:1. The translational diffusion constants extracted from 638 639 these curves are 21% (1:1) and 42% (5:1) smaller in the presence of U(VI) as compared to the

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diffusion of the free PCaP1 protein. This apparent slower diffusion corresponds to an average
 particle size that is increased 2 and 5-fold for the 1:1 and 5:1 mixtures, respectively.

642 In the case of a 1:1 stoichiometric ratio, we observed a decrease in signal intensity in the ¹H-643 ¹⁵N spectrum by a factor 2 for the structured part of the protein (Figure 5C). In view of our 644 translational diffusion results, this reduced signal intensity may originate either from the presence of U(VI)-bound PCaP1 dimers, or free PCaP1 in slow exchange (high binding affinity) 645 646 with larger PCaP1 oligomers that are invisible by NMR. In order to discriminate the two 647 scenarios, we performed ¹⁵N spin relaxation experiments that provide quantitative information 648 about changes in rotational tumbling correlation times upon interaction with U(VI). Site-specific 649 ¹⁵N relaxation rate constants T₂ and T₁, measured for free PCaP1 and a 1:1 U(VI):PCaP1 650 mixture (Figure 5C), did not show any significant difference for the structured N-terminal part. 651 This observation is in agreement with a model of coexisting monomeric and oligomeric species 652 with only the free PCaP1 monomers contributing to the detected NMR signal. In an attempt to detect additional NMR signals in the ¹H-¹⁵N correlation spectra, arising from U(VI)-bound 653 PCaP1 oligomers, we produced ²H,¹⁵N-labeled PCaP1 protein (deuterated to about 70%) to 654 reduce spin-relaxation-induced NMR signal loss. However, even under these isotope-labeling 655 conditions favorable for the detection of more slowly tumbling molecules, no additional peaks 656 657 could be observed in an overnight experiment, strongly suggesting that the undetected species correspond to relatively large oligomers (at least trimers or tetramers). 658

- Surprisingly, for some peptide regions in the flexible C-terminal part of PCaP1, notably around 659 residue 160, we observed a significant increase in NMR signal intensity upon uranyl binding 660 (Figures 5A, C). Note that the corresponding ¹H-¹⁵N correlation peaks for these residues were 661 662 of (very) low intensity in the spectra of free PCaP1, most likely due to conformational 663 exchange-induced line broadening effects (dynamics on the µs – ms time scale). ¹⁵N relaxation 664 rates (T₁ and T₂) indicated only minor changes in the local dynamics on the fast (sub-ns) time 665 scale. In particular, the high T₂ and low T₁ values confirmed that this domain remains highly flexible in the presence of uranyl. The observed increase in signal intensity upon uranyl binding 666 667 for the peptide region around residue 160 can thus be explained either by a side effect of U(VI)-668 binding to the structured N-terminal part, altering the dynamic properties of this peptide region, or by additional binding site(s) in the C-terminal part of PCaP1, again inducing a change in the 669 local dynamics of the peptide chain. 670
- 671
- 672 673

3.4. U(VI) binds to the structured N-terminal and disordered C-terminal domains of PCaP1

In an attempt to better understand the interaction of U(VI) with the structured and disordered parts of the protein, we produced two truncated forms of PCaP1, deleted from all or part of the flexible region of the protein (Supplementary Figure S2B, C). The first one was a minor splicing 677 variant of PCaP1, formed by cryptic intron splicing within exon 4 (herein named PCaP1v), that was isolated in the course of PCR-cloning of the full-length ORF (this form represented 1/10th 678 679 of the clones obtained by PCR cloning). This cDNA encodes a PCaP1 protein form lacking 680 residues 172-212 (Supplementary Figure S9B). The second PCaP1 form produced (herein named PCaP1₍) was designed to remove the whole C-terminal flexible domain (residues 143-681 225). The structural integrity of the PCaP1^Δ construct was checked by NMR spectroscopy, 682 with no (significant) spectral changes in the ¹H-¹⁵N correlation spectrum of PCaP1_Δ with 683 684 respect to N-terminal residue signals detected in the spectrum of the full-length protein (Supplementary Figure S10A). No degradation of recombinant proteins was also noted, as 685 judged by SDS-PAGE analysis and mass spectrometry in denaturing conditions 686 (Supplementary Figure S2). 687

The stoichiometry of U(VI) complexation to both PCaP1v and PCaP1^Δ forms was analyzed by 688 MicroSpin G-25 chromatography coupled to ICP-MS, at pH 6 and 7.5 under the same 689 690 conditions as for full-length PCaP1 (Supplementary Table S3). Data obtained for PCaP1v were very close to those observed for PCaP1, (i.e. 0.5 U(VI)/monomer at pH 6 and ~2 691 U(VI)/monomer at pH 7.5), demonstrating full U(VI)-binding capacity of this truncated protein. 692 693 Concerning PCaP1 Δ , data obtained at pH 6 were also similar to those measured for the other 694 two forms (0.56 U(VI)/monomer). At pH 7.5, however, the U(VI)-binding capacity of PCaP1 Δ was at least reduced by 2 times ($\leq 1 \text{ U(VI)}/\text{monomer}$), as compared to that of the other forms. 695 Trp-fluorescence titration of PCaP1v by up to 2 U(VI) equivalents at pH 6 gave guenching 696 spectra profiles similar to those of the full-length PCaP1 protein (Supplementary Figure S11A), 697 yielding a calculated conditional stability constant for U(VI) complex (UO₂(P₂) model) log β_{12} = 698 14.2 ± 0.7, close to that measured for the full-length PCaP1 protein (Supplementary Table S5). 699 700 In contrast, the fluorescence spectrum of PCaP1^Δ was altered and fluorescence intensities were too low to determine a conditional stability constant (Supplementary Figure S11B; 701 702 Supplementary Table S5).

NMR titration experiments adding increasing amounts of U(VI) to a sample of PCaP1 Δ showed 703 704 a shift in binding stoichiometry with respect to full-length PCaP1 (Supplementary Figure S10B. 705 C). At sub-stoichiometric amounts of U(VI) the NMR signal intensity decreased, without 706 changing the translational diffusion properties of the NMR-observable species (similar to full 707 length PCaP1), while no NMR signal was detected for U(VI):PCaP1_Δ ratios as soon as uranyl 708 was provided in excess. A SEC analysis of PCPa1 Δ protein exposed to a 5-fold excess U(VI), 709 using a Superdex 200 10 30 Increase GL column, showed that protein completely eluted as large oligomers under these conditions, while the free protein behaved as a monomer 710 711 (Supplementary Figure S12). The NMR signal was partly recovered by adding EDTA (5 mM), suggesting that U(VI) induced PCaP1∆ oligomerization can be reversed by U(VI)-EDTA
complexation (not shown).

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3.5. U(VI) translocation in Arabidopsis shoots is affected in a pcap1 null-mutant 715 716 Our data showed that PCaP1 has the capacity to bind U(VI) with high affinity at various sites, 717 in vitro. In order to determine a possible involvement of PCaP1 on the fate of U(VI) in planta, we obtained Arabidopsis seeds of an homozygous T-DNA insertion line of PCaP1 (mdp25; Li 718 719 et al. (2011); also known as pcap1-4; Nagata et al. (2016); herein named knock-out (ko) 720 mutant). Also, we generated transgenic lines expressing the full-length PCaP1 under the 721 control of a 35S promoter in the ko mutant line background (ko/35S::PCaP1 lines). Three 722 independent homozygous transgenic lines carrying the 35S::PCaP1 construct, among the 9 obtained, were selected for subsequent U(VI) uptake and translocation analyses (#2; #10 and 723 #30 lines). A western blot analysis of total shoot proteins using a PCaP1 antibody showed no 724 725 detectable PCaP1 protein in the ko mutant, confirming that it is a genuine null-mutant (Figure 726 6A; Nagata et al. (2016)). Transgenic lines exhibited nearly wild-type levels of PCaP1 727 expression. To estimate the importance of PCaP1 synthesis on the accumulation of U(VI), 728 plants were grown for 4 weeks in a complete standard Gre medium, then transferred in a Gre 729 medium depleted of Pi and exposed to 20 µM uranyl nitrate for short (1 or 2 days) and long (11 days) periods of time. No obvious difference was observed on plant growth between the 730 different lines, for the same U(VI) exposure time (not shown). We used the photosynthetic 731 732 parameter Fv/Fm (measurement of photosystem II efficiency in dark-adapted leaves) as an 733 estimate of plant fitness, and found it was unchanged in all cases (comprised between 0.79 ± 0.01 and 0.81 ± 0.01). Root and shoot U(VI) content was determined by ICP-MS following 734 careful washing of plant tissues with sodium carbonate and deionized water to eliminate the 735 736 loosely-bound radionuclide. In agreement with previous observations made under comparable conditions (Berthet et al., 2018; Doustaly et al., 2014; Saenen et al., 2014; Sarthou et al., 2022; 737 738 Vanhoudt et al., 2008), the amount of U(VI) in shoots was low compared to roots in wild-type (Col-0) control plants, with an highest translocation factor (shoot to root ratio) of 6 x 10⁻³, 739 740 illustrating the low mobility of U(VI) in Arabidopsis (Figure 6B, C). Data presented in Figure 6B 741 showed that U(VI) biosorption in roots was poorly affected in *pcap1 ko* plants or in transgenic 742 lines as compared to Col-0 control plants, throughout the experiment. In contrast, U(VI) 743 translocation in Arabidopsis shoots was significantly affected in ko plants after 11 days of U(VI) 744 exposure, as compared to Col-0 controls, showing a 2 to 3-fold decrease in U(VI) translocation (Figure 6C). No difference was observed for short periods of U(VI) exposure (days 1 or 2). 745 Finally, ectopic expression of PCaP1 cDNA in *pcap1 ko* plants (#2, #10, #30 transgenic lines) 746

restored wild-type levels of U(VI) translocation in shoots, in line with PCaP1 protein expression
levels.

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750 **4. Discussion and conclusions**

The cellular toxicity of U(VI) in prokaryotic and eukaryotic organisms is not yet fully described 751 752 but it likely results from its ability to bind strongly to biomolecules, mainly proteins that contain 753 numerous uranyl-binding groups. Over the past few years, significant progress has been made 754 in identification of UraBPs from various organisms, including the hyperthermophilic archaea Pyrococcus furiosus (Cvetkovic et al., 2010), the crayfish Procambarus clarkii (Xu et al., 2014), 755 the zebrafish Danio rerio (Bucher et al., 2014; Eb-Levadoux et al., 2017), rat (Frelon et al., 756 2009) and humans (Basset et al., 2008; Dedieu et al., 2009; Huynh et al., 2015; Vidaud et al., 757 758 2005; Vidaud et al., 2019). These proteins were identified by a combination of non-denaturing 759 electrophoretic and/or chromatographic approaches combined with proteomic analyses, 760 allowing to isolate proteins with affinity for U(VI) either in vitro, in cellulo or both.

761 Despite U(VI) is known to be taken up by plants and to have cellular toxic effects, very few of 762 the proteins involved in these processes have been identified so far (Berthet et al., 2018; Doustaly et al., 2014; Laurette et al., 2012; Misson et al., 2009; Serre et al., 2019; Vanhoudt 763 764 et al., 2011). Recently, several evidences were provided that U(VI) is taken up by the roots of 765 Arabidopsis plants through MCA1 and ANN1 Ca²⁺-permeable cation channels (Sarthou et al., 2022). However, the fate of U(VI) within plant cells is still largely unknown. In particular, 766 identification and characterization of UraBPs have not been done so far in higher plants or any 767 768 photosynthetic organism. Here, in order to identify UraBPs in Arabidopsis, we used an immobilized metal affinity chromatography (IMAC) method, based on the cation-exchange 769 properties of aminophosphonate groups for U(VI) binding (Basset et al., 2008). Such a 770 771 technique was developed because uranyl cations have singular physico-chemical 772 characteristics that prevent them from being immobilized on commercially available metal 773 chelating supports. This IMAC technique was a powerful tool for capturing UraBPs in vitro, 774 from human serum (Basset et al., 2008), kidney-2 cell extracts (Dedieu et al., 2009) and 775 dopaminergic SH-SY5Y cells (Vidaud et al., 2019). Adapted to plant soluble protein extracts, 776 this technique allowed us to capture and identify 38 UraBPs from Arabidopsis shoots and/or roots. Interestingly, several proteins identified in our study are homologous to UraBPs 777 identified in other organisms. For example, the glycolytic enzymes fructose-bis phosphate 778 779 aldolases 2 and 1 (UraBP7 and 13, in this study) are homologous to several aldolases identified in human urano proteomes (Dedieu et al., 2009; Vidaud et al., 2019). Also, 780 peroxiredoxin Q (UraBP29), the plasma-membrane associated cation-binding protein, PCaP1 781 782 (UraBP25) and the glycine-rich RNA-binding protein 7 (UraBP33) are related to peroxiredoxin 1, the numerous actin network regulatory proteins, and various ribonucleoproteins, identified 783

in the human urano proteomes, respectively (Dedieu et al., 2009; Vidaud et al., 2019). Finally,
the beta carbonic anhydrase 2 (UraBP18) and the glutathione S-transferase F2 and F7
(UraBP15 and 35) are homologous to carbonic anhydrase and glutathione S-transferases,
respectively, identified as UraBPs in zebrafish and the crayfish (Bucher et al., 2016; EbLevadoux et al., 2017; Xu et al., 2014).

Among the 38 UraBPs identified in this study, we focused on PCaP1 for a detailed 789 790 characterization. PCaP1 was previously shown to bind calcium possibly at a unique site and 791 copper at multiple sites, both in vitro and in planta (Ide et al., 2007; Nagasaki-Takeuchi et al., 792 2008; Nagasaki et al., 2008; Nagata et al., 2016; Qin et al., 2014). Here, we established that 793 recombinant PCaP1 was indeed able to bind copper but also ferric ion to multiple sites. The 794 large number of glutamate residues in the protein sequence and particularly in the intrinsically unstructured region probably accounted for this high-capacity metal binding. We also 795 796 demonstrated that PCaP1 was able to bind U(VI) with high affinity, and that calcium somewhat 797 competed with U(VI) for binding. At pH 6, PCaP1 displayed high U(VI)-binding affinity with a conditional dissociation constant in the vicinity of 10 nM. By comparison, binding constants for 798 U(VI) of the same order (from the sub-nanomolar to the sub-micromolar range) were reported 799 for peptides or proteins with native or artificial metal-binding sites, using similar experimental 800 801 approaches (Acharya and Blindauer, 2016; Basset et al., 2013; Gallois et al., 2021; Laporte et al., 2019; Le Clainche and Vita, 2006; Lebrun et al., 2014; Michon et al., 2010; Pardoux et al., 802 2012; Qi et al., 2014; Sauge-Merle et al., 2017; Starck et al., 2017; Starck et al., 2015; Wegner 803 804 et al., 2009), and also reviewed by Lin (2020).

In a more singular way, our results are consistent with U(VI) exposure inducing PCaP1 oligomerization. Structural impacts of U(VI) binding to proteins reported in the literature vary from slight secondary structural to large conformational changes, but to our knowledge, no study describes a change of protein oligomeric state (Acharya and Blindauer, 2016; Bal et al., 2013; Basset et al., 2013; Gallois et al., 2021; Hoarau et al., 2019; Lin et al., 2011; Pardoux et al., 2012; Qi et al., 2014; Sun et al., 2014; Vidaud et al., 2007; Wan et al., 2012; Zhou et al., 2014).

812 Structural evidences indicate that the presence of two oxo groups in uranyl cation promotes 813 the coordination of four to six additional ligands to U(VI) in its equatorial plane perpendicular 814 to the O-U-O axis (Garai and Delangle, 2020; Van Horn and Huang, 2006). In this context, 815 based on the results obtained on the full-length and truncated versions of PCaP1 by fluorescence titration, U(VI) quantification by ICP-MS, solution-state NMR and SEC analyses, 816 817 we can suggest structural models for the interaction with uranyl (Figure 7). At pH 6, our data 818 are consistent with the occurrence of one U(VI)-binding site at the interface of two monomers 819 and coordinated by residues from the two monomers, yielding a stoichiometry of the complex 820 of 0.5 U(VI)/monomer (in line with the 1:2 model fitting the fluorescence titration data) (Figure

7A). Such a stoichiometry in U(VI)-PCaP1 complex was found in all the protein forms analyzed 821 (PCaP1, PCaP1v and PCaP1A) (Supplementary Table S3), suggesting that the U(VI)-binding 822 823 site is located at the surface of the structured domain. At pH 7.5, the situation is more complex. According to our data, the well-folded N-terminal domain of PCaP1 possesses (at least) two 824 825 high-affinity uranyl binding sites. Uranyl binding occurs at the interface between two monomers, thus allowing the formation of oligomers of different size and shape. The average 826 stoichiometry U(VI):PCaP1 of such large oligomers is 1:1 (Figure 7B). The disordered C-827 terminal domain also possesses one or several uranyl binding sites of similar affinity than the 828 829 N-terminal domain. Our study highlights the peptide segment around residue 160 to be involved in uranyl binding, as PCaP1v, retains the uranyl binding features of the full-length 830 protein, and NMR shows altered conformational dynamics in this region upon uranyl binding. 831 In the full-length protein, the flexible C-terminus is responsible for scavenging part of the 832 833 available U(VI), which explains our results showing that about twice as much U(VI) is required to observe the complete disappearance of NMR signals from the well-folded monomeric N-834 terminal part of the full-length PCaP1 protein than of the isolated N-terminal domain. Uranyl-835 binding by the disordered C-terminal domain most likely induces the formation of a highly 836 837 dynamic mesh of peptide chains, which contributes to the larger apparent particle size of U(VI)-838 PCaP1 complexes (Figure 5B). Such a configuration would be consistent, as observed, with a 839 mean global U(VI) content for large complexes close to or higher than 2 U(VI)/monomer of full-840 length protein (about 1 U(VI) equivalent per structured domain and 1 U(VI) equivalent per unstructured domain) (Figure 7 B). 841

The capacity of PCaP1 to bind U(VI) in vitro does not provide insight into the functional, 842 physiological and developmental consequences of protein-metal interactions in planta. PCaP1 843 844 is located on the inner side of the plasma membrane via N-myristoylation of the glycine 2 residue (Ide et al., 2007; Nagasaki et al., 2008; Nagata et al., 2016). At first glance, one would 845 expect PCaP1 to prevent the diffusion of U(VI) into the cells and protect the plant from its toxic 846 effects. This prediction is strengthened by the fact that PCaP1 is poly-phosphorylated in vivo 847 (Supplementary Table S2; (Mattei et al., 2016; Rayapuram et al., 2014)), a situation known to 848 amplify the affinity of proteins for U(VI) (Laporte et al., 2019; Qi et al., 2014; Starck et al., 2017; 849 850 Starck et al., 2015). Such scavenging of U by a high-affinity protein to prevent its toxic 851 outcomes would resemble that described for the UipA protein in U(VI)-tolerant Microbacterium species. UipA is a single-pass transmembrane protein exposed to the external phase of the 852 plasma membrane; it is induced by U(VI) and protects *Microbacterium* from U(VI) entry thanks 853 854 to the high affinity of its large soluble domain for U(VI) (Gallois et al., 2021). However, the role 855 of PCaP1 does not resemble that of UipA. Indeed, the absence of PCaP1 in pcap1-null mutants 856 has no impact on U(VI) accumulation in roots as compared with wild-type plants. On the 857 contrary, the presence of wild-type levels of PCaP1 seemes to promote translocation of U(VI)

from roots to aerial parts (Figure 6). In Arabidopsis, PCaP1 was found to regulate the actin 858 (severing activity) and the microtubule (depolymerizing activity) cytoskeleton in a calcium-859 dependent manner (Li et al., 2011; Qin et al., 2014). PCaP1 was also proposed to play a role 860 861 in calcium signaling through its binding to calmodulin and phosphatidylinositol phosphates 862 (Kato et al., 2010; Nagasaki et al., 2008). In connection with these activities, PCaP1 negatively regulates hypocotyl elongation (Li et al., 2011), pollen-tube growth (Qin et al., 2014), and the 863 864 root hydrotropic response (Tanaka-Takada et al., 2019). Moreover, it is involved in plasmodesmata aperture associated to plant immunity (Cheng et al., 2020; Giovannoni et al., 865 2021; Vijayapalani et al., 2012) and stomatal closure (Nagata et al., 2016). These structures 866 have a key role in the transport of small molecules and ion for cell-to-cell communication 867 868 (Roeder et al., 2022) and in water and metabolites fluxes into the xylem through the driving force of transpiration (Hetherington and Woodward, 2003). It is conceivable that perturbation 869 of these mechanisms by inactivation of the *PCaP1* gene might be responsible for the observed 870 871 defect in U(VI) translocation. Further work is needed to determine precisely by which mechanism PCaP1 favors U(VI) translocation to shoots and if the peculiar U(VI)-binding 872 properties of the protein is involved in this process, particularly given the apparent much higher 873 affinity of the protein for U(VI) than for calcium at the slightly alkaline pH of plant cell cytosol. 874

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876 Statement of environmental implication

Uranium (U) is a harmful radionuclide that may be absorbed by plants from soil and, consequently, contaminate the food chain, with human and animals health hazards. The cellular toxicity and detoxification mechanisms of U are still poorly described but likely result from its ability to bind strongly to biomolecules, mainly proteins. Our study describes for the first time the large-scale identification of U-binding proteins from plant tissues and the further characterization of one of them both *in vitro* and *in planta*. Our findings represent an important milestone to advance knowledge in the fields of radionuclide toxicology and phytoremediation.

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885 Data availability

All data supporting the findings of this study are included in this published article (and its Supplementary Information files), or are available from the corresponding authors upon reasonable request. NMR chemical shifts of PCaP1 have been deposited with the BMRB (http://www.bmrb.wisc.edu/) under accession number 51607.

- Sequence data from this article can be found in the GenBank/EMBL, Arabidopsis Genome
 Initiative and/or UniProt databases under the following accession numbers: PCaP1
 (At4g20260; Q96262); PCaP1v (splicing variant; ON209435).
- The proteomics data have been deposited to the ProteomeXchange Consortium via the PRIDE (Perez-Riverol et al., 2019) partner repository with the dataset identifier PXD033354 and

89510.6019/PXD033354.[The dataset is available with the Username:896reviewer_pxd033354@ebi.ac.uk and Password: 62dQFqjg].

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898 CRediT authorship contribution statement

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916 **Declaration of Competing Interest**

917 The authors declare that they have no known competing financial interests or personal 918 relationships that could have appeared to influence the work reported in this paper.

919

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932 Figure legends

Figure 1: Strategy for the isolation and identification of uranyl-binding proteins from*Arabidopsis thaliana*.

Figure 2: Purification of recombinant PCaP1 and determination of its oligomerization state. A-935 Documentation of PCaP1 purification. Polypeptides were separated by SDS-PAGE 12% and 936 stained with Coomassie Brilliant Blue R250. Lane 1: soluble proteins (25 µg) from E. coli 937 938 Rosetta cells harboring pET28-PCaP1 construct grown in the presence of IPTG; Lane 2: 939 ammonium sulfate 50% of saturation precipitating fraction (25 µg); Lane 3: BioGel HTP 940 hydroxyapatite column pool (15 µg); Lane 4: EMD-DEAE column pool (10 µg); Superdex 75 941 column pool (8 µg); M, molecular mass markers. B- Apparent molecular mass estimation of 942 native recombinant PCaP1 by gel filtration. Purified protein (150 µg) was resolved by size exclusion chromatography (SEC) onto a Superdex 200 Increase 10/300 GL column. Eluted 943 fractions were analyzed by SDS-PAGE. Standard proteins for column calibration (inset) were 944 945 ferritin (440 kDa), covalbumin (75 kDa), carbonic anhydrase (29 kDa), and ribonuclease A (13.7 kDa). $K_{av} = (Ve-Vo)/(Vt-Vo)$; Ve, elution volume; Vo, void volume; Vt, total volume. C-946 Native mass spectrum of recombinant PCaP1. 1mer (monomer) = 24453 Da; 2mer (dimer) = 947 48906 Da. 948

- **Figure 3:** Fluorescence emission spectrum analysis of uranyl binding to PCaP1 at pH 6. A-Fluorescence titration of recombinant PCaP1 (0.8μ M) with UO₂²⁺ (0 to 2.5 equivalents) at pH 6 in MES buffer 10 mM, NaCl 100 mM, in the presence of IDA 10 μ M, with excitation at 277 nm. B- Variation of the intensity at the peak maximum (342 nm) upon UO₂²⁺ addition. Circles represent experimental (Exp) values and the solid line the best fit with the program SPECFIT, taking into consideration the known conditional stability constants of the uranyl complexes with IDA. n_{eq}(U), number of U(VI) equivalents ([U(VI)]/[PCaP1]).
- Figure 4: NMR assignment and structural features of PCaP1. A- Amide ¹H-¹⁵N correlation 956 957 spectrum of free PCaP1. Resolved cross peaks are annotated by their residue type (one letter amino acid code) and residue number. Sample: PCaP1 300 µM in Tris-HCl 10 mM pH 7.5, 958 NaN₃ 0.03%. B- Helical propensity along the PCaP1 backbone (only the structured part is 959 960 shown) computed from the measured ¹³C chemical shifts using the TALOS-N server (https://spin.niddk.nih.gov/bax/software/TALOS-N/). The helical elements derived from this 961 plot (red bars) are in agreement with the structure predicted by Alpha Fold v2.0 962 (https://alphafold.ebi.ac.uk/) shown in (C). Model confidence (pLDDT score) for the structured 963 domain was >90 (highly confident). The image was produced using PyMOL (DeLano Scientific, 964 San Carlos, CA, USA) as a ribbon model colored in the "chainbows" mode. 965
- **Figure 5:** NMR characterization of PCaP1-U(VI) interaction. A- Overlay of 2D ${}^{1}H{}^{15}N$ correlation spectra of free ${}^{15}N{}^{-13}C$ labelled PCaP1 (100 μ M) alone (black) or in complex (100 μ M) with U(VI) (500 μ M) (red) in Tris-HCl 10 mM pH 7.5. Only correlation peaks in the random

coil ¹H spectral region (indicated by dashed lines) remain visible upon U(VI) binding. B-969 Diffusion ordered spectroscopy (DOSY) measurements of PCaP1 in the absence or in the 970 presence of 1 or 5 U(VI) equivalents. The observed NMR signal intensity (integrated over the 971 972 1D ¹H spectrum) is plotted as a function of the applied gradient strength. Fitting of decay curves yields the translational diffusion constant. Translational diffusion of PCaP1 decreases U(VI) 973 with increasing amounts of uranyl, indicating the formation of larger particles. C-¹⁵N spin 974 975 relaxation rate constants (T₁ and T₂) measured for free PCaP1 (black) and PCaP1 in the 976 presence of an equal amount of U(VI) (red). In addition, the intensity ratios of ¹H-¹⁵N correlation 977 peaks detected in the absence and presence of an equal amount of U(VI) are plotted (top 978 panel).

979 Figure 6: Uranium content in wild type, *pcap1*-null mutant and transgenic lines expressing PCaP1. Wild type (Col-0), pcap1-null mutant (ko) and transgenic lines expressing PCaP1 980 under the control of a 35S promoter (#2; #10 and #30 lines) were grown hydroponically for 5 981 weeks in Gre medium, and then challenged with 20 µM uranyl nitrate for 1, 2 and 11 days in 982 Gre medium with no phosphate. A-Western blot analysis of total shoot proteins with a PCaP1 983 antibody. B- Accumulation of U in roots. C- Accumulation of U in shoots. Samples were 984 dehydrated, digested with HNO₃, and ²³⁸U was measured by ICP-MS. Data distribution is 985 displayed in Tukey's boxplots with the median as the solid line inside the box, and the first and 986 third quartiles as the bottom and top lines of the box, respectively. Outliers are shown as 987 individual dots. Each distribution in boxplots represents n = 4 samples for roots and n = 4-8988 989 samples for shoots. Statistical significance determined using non-parametric comparisons with Tukey's test is shown: *, p < 0.05; **, p < 0.01. 990

991 **Figure 7:** Models for *in vitro* U(VI)-PCaP1 complexes formation at pH 6 and pH 7.5.

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994 Appendix A. Supporting information

995 Supplementary data associated with this article can be found in the online version.

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Figure 1: Strategy for the isolation and identification of uranyl-binding proteins from *Arabidopsis thaliana*.



Figure 2 (legend on the next page)

Figure 2: Purification of recombinant PCaP1 and determination of its oligomerization state. A- Documentation of PCaP1 purification. Polypeptides were separated by SDS-PAGE 12% and stained with Coomassie Brilliant Blue R250. Lane 1: soluble proteins (25 µg) from E. coli Rosetta cells harboring pET28-PCaP1 construct grown in the presence of IPTG; Lane 2: ammonium sulfate 50% of saturation precipitating fraction (25 µg); Lane 3: BioGel HTP hydroxyapatite column pool (15 μg); Lane 4: EMD-DEAE column pool (10 μg); Superdex 75 column pool (8 μg); M, molecular mass markers. B- Apparent molecular mass estimation of native recombinant PCaP1 by gel filtration. Purified protein (150 µg) was resolved by size exclusion chromatography (SEC) onto a Superdex 200 Increase 10/300 GL column. Eluted fractions were analyzed by SDS-PAGE. Standard proteins for column calibration (inset) were ferritin (440 kDa), covalbumin (75 kDa), carbonic CaP1.1mer anhydrase (29 kDa), and ribonuclease A (13.7 kDa). $K_{av} = (Ve-Vo)/(Vt-Vo)$; Ve, elution volume; Vo, void volume; Vt, total volume. C- Native mass spectrum of recombinant PCaP1. 1mer (monomer) = 24453 Da; 2mer (dimer) = 48906 Da.



Figure 3: Fluorescence emission spectrum analysis of uranyl binding to PCaP1 at pH 6. A- Fluorescence titration of recombinant PCaP1 (0.8 μ M) with UO₂²⁺ (0 to 2.5 equivalents) at pH 6 in MES buffer 10 mM, NaCl 100 mM, in the presence of IDA 10 μ M, with excitation at 277 nm. B- Variation of the intensity at the peak maximum (342 nm) upon UO₂²⁺ addition. Circles represent experimental (Exp) values and the solid line the best fit with the program SPECFIT, taking into consideration the known conditional stability constants of the uranyl complexes with IDA. n_{eq}(U), number of U(VI) equivalents ([U(VI)]/[PCaP1]).

Figure 3



Figure 4 (legend on the next page)

Figure 4: NMR assignment and structural features of PCaP1. A- Amide ¹H-¹⁵N correlation spectrum of free PCaP1. Resolved cross peaks are annotated by their residue type (one letter amino acid code) and residue number. Sample: PCaP1 300 µM in Tris-HCl 10 mM pH 7.5, NaN₃ 0.03%. B- Helical propensity along the PCaP1 backbone (only the structured part is shown) computed from the measured ¹³C chemical shifts using the TALOS-N server (https://spin.niddk.nih.gov/bax/software/TALOS-N/). The helical elements derived from this plot (red bars) are in agreement with the structure predicted by Alpha Fold v2.0 (https://alphafold.ebi.ac.uk/) shown in (C). Model confidence (pLDDT score) for the structured domain was >90 (highly confident). The image was produced using PyMOL (DeLano Scientific, San Carlos, CA, USA) as a ribbon model colored in the "chainbows" mode. h.



Figure 5: NMR characterization of PCaP1-U(VI) interaction. A- Overlay of 2D ¹H-¹⁵N correlation spectra of free ¹⁵N-¹³C labelled PCaP1 (100 μ M) alone (black) or in complex (100 μ M) with U(VI) (500 μ M) (red) in Tris-HCI 10 mM pH 7.5. Only correlation peaks in the random coil ¹H spectral region (indicated by dashed lines) remain visible upon U(VI) binding. B- Diffusion ordered spectroscopy (DOSY) measurements of PCaP1 in the absence or in the presence of 1 or 5 U(VI) equivalents. The observed NMR signal intensity (integrated over the 1D ¹H spectrum) is plotted as a function of the applied gradient strength. Fitting of decay curves yields the translational diffusion constant. Translational diffusion of PCaP1 decreases U(VI) with increasing amounts of uranyl, indicating the formation of larger particles. C- ¹⁵N spin relaxation rate constants (T₁ and T₂) measured for free PcaP1 (black) and PCaP1 in the presence of an equal amount of U(VI) (red). In addition, the intensity ratios of ¹H-¹⁵N correlation peaks detected in the absence and presence of an equal amount of U(VI) are plotted (top panel).



Figure 6: Uranium content in wild type, *pcap1*-null mutant and transgenic lines expressing PCaP1. Wild type (Col-0), *pcap1*-null mutant (ko) and transgenic lines expressing PCaP1 under the control of a 35S promoter (#2; #10 and #30 lines) were grown hydroponically for 5 weeks in Gre medium, and then challenged with 20 µM uranyl nitrate for 1, 2 and 11 days in Gre medium with no phosphate. A- Western blot analysis of total shoot proteins with a PCaP1 antibody. B- Accumulation of U in roots. C- Accumulation of U in shoots. Samples were dehydrated, digested with HNO₃, and ²³⁸U was measured by ICP-MS. Data distribution is displayed in Tukey's boxplots with the median as the solid line inside the box, and the first and third quartiles as the bottom and top lines of the box, respectively. Outliers are shown as individual dots. Each distribution in boxplots represents n = 4 samples for roots and n = 4-8 samples for shoots. Statistical significance determined using non-parametric comparisons with Tukey's test is shown: *, p < 0.05; **, p< 0.01.



Figure 7: Models for in vitro U(VI)-PCaP1 complexes formation at pH 6 and pH 7.5.

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