



HAL
open science

Effect of organic loading rate and effluent recirculation on biogas production of desulfated skim latex serum using up-flow anaerobic sludge blanket reactor

Marisa Raketh, Prawit Kongjan, Eric Trably, Nurta Samahae, Rattana Jariyaboon

► To cite this version:

Marisa Raketh, Prawit Kongjan, Eric Trably, Nurta Samahae, Rattana Jariyaboon. Effect of organic loading rate and effluent recirculation on biogas production of desulfated skim latex serum using up-flow anaerobic sludge blanket reactor. *Journal of Environmental Management*, 2023, 327, pp.116886. 10.1016/j.jenvman.2022.116886 . hal-03957215

HAL Id: hal-03957215

<https://hal.inrae.fr/hal-03957215v1>

Submitted on 27 Jul 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1 **Effect of Organic Loading Rate and Effluent Recirculation on Biogas Production**
2 **of Desulfated Skim Latex Serum using Up-Flow Anaerobic Sludge Blanket Reactor**

3
4 **Marisa Raketh^{1,3}, Prawit Kongjan^{2,3}, Eric Trably⁴, Nurta Samahae⁵, Rattana**
5 **Jariyaboon^{2,3,*}**

6
7 ¹ Energy Technology Program, Faculty of Engineering, Prince of Songkla University,
8 Hat Yai, Songkhla, 90112, Thailand

9 ² Department of Science, Faculty of Science and Technology, Prince of Songkla
10 University (PSU), Pattani, 94000, Thailand

11 ³ Bio-Mass Conversion to Energy and Chemicals (Bio-MEC) Research Unit, Faculty of
12 Science and Technology, Prince of Songkla University (PSU), Pattani, 94000, Thailand

13 ⁴ INRAE, Univ Montpellier, LBE, Narbonne, France

14 ⁵ Science Program in Chemistry-Biology, Faculty of Science and Technology, Prince of
15 Songkla University (PSU), Pattani, 94000, Thailand

16 *corresponding author at: Department of Science, Faculty of

17 Science and Technology, Prince of Songkla University (PSU), Pattani 94000,

18 Thailand. Tel.: +66 73 313928-50 ext1988, mobile: +66 808721260.

19 E-mail address: rattana.sa@psu.ac.th (R. Jariyaboon).

20

21

22

23

24

25 **ABSTRACT**

26 High sulfate contents in skim latex serum (SLS) can be reduced by rubber wood
27 ash (RWA). Subsequently, the desulfated skim latex serum (DSLS) can be further
28 anaerobically treated more effectively with the accompanying generated biomethane.
29 In this study, DSLS was treated using an up-flow anaerobic sludge blanket (UASB)
30 reactor operated at 10-day HRT and under mesophilic (37°C) conditions. The effect of
31 organic loading rates (OLR) at 0.89, 1.79 and 3.57 g-COD/L-reactor·d on DSLS
32 biodegradability was investigated in Phase I-IV using NaHCO₃ as an external buffering
33 agent. Maximum methane production yield of 226.35 mL-CH₄/g-COD_{added}
34 corresponding to 403.25 mL-CH₄/L reactor·d was achieved at the suitable OLR of 1.79
35 g-COD/L-reactor·d. UASB effluent recirculation which was then applied to replace the
36 NaHCO₃. It was found that with 53% effluent recirculation similar to an OLR of 2.01
37 g-COD/L-reactor·d, an average of 185.70 mL-CH₄/g-COD_{added} corresponding to 371.40
38 mL/L reactor·d of methane production was reached. The dominant bacteria in UASB
39 reactor were members of *Proteobacteria*, *Bacteroidota*, *Firmicutes*, and
40 *Desulfobacterota phyla*. Meanwhile, the archaeal community was majorly dominated by
41 the genera *Methanosaeta sp.* and *Methanomethylovorans sp.* The study clearly indicates
42 the capabilities of UASB reactor with effluent recirculation to treat DSLS anaerobically.

43

44 **Keywords:** Rubber latex wastewater, Sulfate removal, UASB reactor, Anaerobic
45 Digestion, Methane production

46

47

48

49 **1. Introduction**

50 Skim latex serum (SLS) is the wastewater generated from the concentrated latex
51 process after the sulfuric acid coagulation of skim latex which aimed at recovering the
52 rubber. SLS with a high organic matter content of sulfate (3,580-7,500 mg/L), chemical
53 oxygen demand (COD) (33.02-43.11 g/L), Volatile Solids (VS) (32.64-37.94 g/L) and
54 low pH (5.22-5.78) (Jariyaboon et al., 2015; Raketh et al., 2021).

55 Anaerobic digestion (AD) is widely used for the treatment of high strength
56 wastewaters in the rubber industry. The process is attractive for resource recovery and
57 the production of sustainable energy carrier in the form of biogas. The AD is performed
58 by a high variety of microorganisms in terms of biochemical reactions, i.e., hydrolysis,
59 acidogenesis, acetogenesis, and methanogenesis (Min et al., 2014) resulting in the
60 production of a biogas containing 40-75% CH₄, 15-60% CO₂, 5-10% water, and 0.005-
61 2% H₂S, and some amount of traces of other components such as siloxanes,
62 halogenated hydrocarbons NH₃, O₂, CO, and N₂ (Rattanaya et al., 2021; Ryckebosch et
63 al., 2011).

64 During AD process, sulfate ions contained in the substrate can be transformed to
65 hydrogen sulfide (H₂S) by sulfate-reducing bacteria (SRB) (Mu et al., 2019). H₂S is the
66 major problem for anaerobic treatment of sulfate-rich wastewater, as H₂S may lead to
67 AD process failure. In previous research, SLS was used to produce biogas without
68 reducing sulfate contents (Jariyaboon et al., 2015; Kongjan et al., 2014), resulting in the
69 inhibition of high sulfate contents during the biogas production process. Thus, reducing
70 sulfate contained in wastewaters before AD process is one of the strategies to achieve
71 successful treatment of sulfate-containing wastewaters. In the previous experiments,
72 rubber wood ash (RWA) was used to remove sulfate in the SLS. RWA can reduce

73 sulfate, as high as 42% sulfate removal efficiency at a solubility equilibrium
74 concentration of 10 g/L of added RWA (Raketh et al., 2021). Moreover, SLS with initial
75 sulfate concentration of 5,417 and 1,625 mg/L was used as substrate to produce biogas
76 in two stages and single stage AD, respectively. The results showed that the biogas
77 production yields were lower with SLS than desulfated SLS (DSLS). Using DSLS, it
78 had shown that 21% improvement of biogas production was achieved in the batch
79 reactor compared to the raw SLS (Raketh et al., 2022).

80 Many reactor configurations have been reported for anaerobically treating
81 concentrated latex wastewaters, mainly up-flow anaerobic sludge blanket (UASB),
82 anaerobic baffled reactor (ABR), and continuous stir tank reactor (CSTR). Kongjan et
83 al. (2014) reported that AD of SLS for hydrogen and methane production in separate
84 process using a two-stage digestion in a series of UASB reactor.. A yield 178.70 mL-
85 $\text{CH}_4/\text{g-COD}_{\text{added}}$ was achieved under thermophilic conditions with 9-day HRT.
86 Furthermore, single-stage AD under mesophilic conditions was studied for the treatment
87 of concentrated latex wastewater (CLW) using ABR with organic loading rate (OLR) of
88 0.60 g-COD/L·d at 10-day HRT (Saritpongteeraka and Chaiprapat, 2008). A methane
89 production yield of 242.31 mL- $\text{CH}_4/\text{g-COD}_{\text{added}}$ was obtained. Moreover, based on the
90 data collected from the concentrated latex factory in Songkhla Province, Thailand,
91 whereby the factory uses covered lagoons to treat the concentrated latex wastewaters
92 with the feed rate of 0.61 g-COD/L·d at HRT of 15.7 days. In this case the average
93 methane yield was estimated at 219.97 mL- $\text{CH}_4/\text{g-COD}_{\text{added}}$. UASB reactor is a high-
94 rate reactor, in which biological granules are formed as the anaerobic microorganism's
95 community. Thus, the solid retention time (SRT) was found to be always much higher
96 than HRT. However, the UASB reactor can productively digest organic matters in a low

97 suspension solid (SS) containing wastewater (Angenent et al., 2004) .

98 Generally, AD performances depend on various parameters, such as the substrate
99 composition, OLR, temperature, pH, C/N ratio, and HRT. Among these parameters,
100 OLR is considered as a significant parameter because it is defined as the amount of
101 COD or VS portions fed per day per unit digester's size. However, high OLR can
102 reduce both the size of digester and consequently, the capital cost. The maximal OLR
103 depends on the type of substrates fed into the digester as it imposes the level of
104 biochemical activity of the digester (Babae and Shayegan, 2011; Chandra et al., 2012;
105 Cremonez et al., 2020).

106 In addition, AD reactors require sufficient alkalinity in order to maintain an
107 optimal environment for methanogens whereby below the optimal pH range (6.7-8.0),
108 it had resulted in the inhibition of methane-producing archaea (MPA) (Deublein and
109 Steinhauser, 2011; Kongjan et al., 2014). KOH, NaOH, Na₂CO₃, and NaHCO₃ as alkali
110 solutions is usually added to maintain the pH in methanogenic reactors. However, the
111 cost of alkali chemicals is also an important element to be considered as well as the
112 additional chemicals associated with the overloading of Na⁺ and K⁺ ions which can
113 severely inhibit MPA at high concentration. One of the strategies which can be
114 employed to overcome the above limitations is the recirculation of effluent/sludge from
115 the AD process. This process can help to neutralize the pH through the dilution of
116 influent fed into the reactor with subsequent improvement in the transformation.
117 Previous studies reported that anaerobic digestion of vegetable market waste in a 4-
118 chambered anaerobic baffled reactor (ABR) with effluent recirculation (25-100%) was
119 regarded as feasible. The biogas and methane yields reached around 0.7–0.8 L
120 biogas/gVS_{added}/d and 0.42–0.52 Lmethane/gVS_{added}/d, respectively, which were among

121 the highest reported for anaerobic digestion of vegetable waste (Gulhane et al., 2016).
122 More stable performances were also observed in the reactor with recirculation
123 (Wikandari et al., 2018). Thereby, effluent recirculation is probably a good substitute for
124 alkaline compounds to maintain appropriate pH and reach the optimum range of biogas
125 production.

126 As mentioned earlier, the previous experiment has confirmed the enhancement
127 of anaerobically treatment simultaneously biogas production of the DSLS using batch
128 process. Then the process must be proved in a continuous mode before scaling up to the
129 industrial application. Therefore, the objective of this study was to investigate the effect
130 of OLR on the treatment performances of continuously treating DSLS wastewaters. The
131 strategies to maintain a sufficient alkalinity by using NaHCO_3 buffering supplement and
132 the UASB effluent recirculation were also compared. The treatment performances were
133 assessed through methane production, COD and sulfate removal efficiencies, volatile
134 fatty acids (VFA) accumulation, and microbial community.

135

136 **2. Materials and methods**

137 2.1. Substrate and Inoculum

138 Fresh raw SLS was collected from the skim latex serum coagulation baths in a
139 concentrated latex factory located in Songkhla Province, Thailand. The collected SLS
140 was stored at 4°C to minimize self-biodegradation and acidification (maximum storage
141 was 1 month). Characteristics of SLS and Desulfated SLS are presented in Table 1.
142 RWA was achieved from a high-pressure steam boiler of a glove factory situated in
143 Songkhla Province, Thailand. The collected RWA was stored in covered container at
144 room temperature.

145 DSLS was prepared following the method described in Raketh et al. (2021)
 146 (Raketh et al., 2021). A ratio of 10 g-RWA to 1 L SLS was used to remove sulfate from
 147 SLS. The mixer was continuously stirred at 150 rpm for 10 minutes at room
 148 temperature. Then, the ash residue was immediately separated from the mixed solutions
 149 and a desulfated solution, so-called DSLS was obtained.

150 Anaerobic granules used in this study were obtained from the UASB reactor of a
 151 frozen food factory in Songkla Province, Thailand. The mesophilic methane inoculum
 152 was sampled from a biogas plant using palm oil mill effluent as substrate in a palm oil
 153 mill factory located in Surat Thani Province, Thailand.

154

155 **Table 1** Characteristics of skim latex serum (SLS) and Desulfated SLS

| Parameters | Unit | Value | |
|-------------------------------|-------------------------|---------------|---------------|
| | | SLS | DSLS |
| pH | | 5.24 - 5.54 | 5.99 - 6.45 |
| total Solids (TS) | g/L | 38.89 - 41.01 | 39.82 - 41.99 |
| Volatile Solids (VS) | g/L | 32.45 - 34.42 | 33.23 - 35.25 |
| Ash | g/L | 6.44 - 6.59 | 6.59 - 6.75 |
| Chemical Oxygen Demand (COD) | g/L | 36.00 - 38.40 | 37.01 - 39.48 |
| Total Organic Carbon (TOC) | g/L | 14.25 - 15.12 | NA |
| Sulfate | mg/L | 4,452 - 4,728 | 2,793 - 2,979 |
| Alkalinity | mg-CaCO ₃ /L | 2,890 - 2,953 | 3,108 - 3,267 |
| Total Kjeldahl Nitrogen (TKN) | mg/L | 1,548 - 1,588 | NA |

NA denoted not analyzed

156

157

158 2.2. Reactor set-up and operation

159 Firstly, to enrich the microorganisms, 100 mL basic anaerobic (BA) medium
 160 (supplemented with 3 g/L glucose) (Angelidaki and Sanders, 2004), 100 mL DSLS at a
 161 concentration of 33.23 g VS/L, and 1800 mL of anaerobic granules and methane
 162 inoculum mixture (70:30 by volume).were mixed in a batch reactor. The reactor was

163 purged by 2 L/min nitrogen gas for 10 min to ensure anaerobically condition and
164 incubated at ambient temperature (30-33 °C). The volume and composition of biogas
165 were monitored daily. It was found the biogas production was steady within 14 days.

166 In this experiment, UASB reactor was operated with 1,200 mL working volume
167 and maintained at 37 °C by circulating hot water inside a water jacket surrounding the
168 reactor. The UASB began by adding 840 mL of the enriched microorganism's solution
169 (70% of working volume) and 360 mL of 33.23 g VS/L DSLS. Nitrogen gas at 2 L/min
170 was used to purge the reactor for 10 min to ensure anaerobically conditions. For the
171 start-up phase, the reactor was operated at HRT of 20 days by feeding a mixture of
172 DSLS and the 2.6 g/L final NaHCO₃ solution which corresponding to the OLR of 1.1 g-
173 COD/L-reactor·d. The feed mixture of 30 mL was transferred to the UASB twice a day
174 using a peristaltic pump. The methane production at start-up phase was continuously
175 performed for 25 days.

176 According to the methane production profile of DSLS in batch mode reported by
177 Raketh et al. (2022), it indicated that 90% of maximum methane production was
178 obtained within 10 days. Thus, the HRT was decreased to 10 days in order to increase
179 the methane production rate in a continuous process. The effect of OLR at 0.89, 1.79,
180 and 3.57 g-COD/L-reactor·d on biogas production were carried out in phase I-IV,
181 respectively. NaHCO₃ solution which was prepared from tap water was used to dilute
182 DSLS and obtain the desired COD concentration for each OLR, with the 2.6 g/L final
183 NaHCO₃ concentration in the feed. The feed mixture of 60 mL was transferred to the
184 UASB twice a day using a peristaltic pump. In Phase V, the NaHCO₃ solution was
185 substituted by adding 53% in volume of the UASB effluent, mixed with DSLS before
186 feeding corresponding to OLR 2.1 g-COD/L-reactor·d was operated.

187 During UASB operation, biogas volume and composition were daily analyzed.
188 pH and alkalinity of the effluent were also daily monitored. COD, sulfate content and
189 VFAs in the effluent were analyzed at the steady-state of each phase. The steady-state
190 was considered when the variation of biogas production was less than 10% as suggested
191 in (Kongjan et al., 2014). Phase I-V conditions were operated for approximately three
192 times of HRT.

193 Anaerobic granules samples were taken from the effluent at steady state. 10-15
194 whereby the granules were randomly separated from the effluent after 10 minutes of
195 sedimentation, and their diameter were measured using a Vernier caliper. For microbial
196 community analysis, the sediment granule samples were also taken from the effluent at
197 steady state and stored at -20°C before the analysis.

198

199 2.3. Analytical methods

200 The volume of produced biogas was recorded using a laboratory water
201 displacement set. Biogas main composition of CH_4 and CO_2 were analyzed using gas
202 chromatography equipped with a 2.5 m Porapak Q column and a thermal conductivity
203 detector (Shimadzu GC 14A). A 30 mL/min. Helium was used as a carrier gas at a flow
204 rate of 30 ml/min. The temperature of injection port, oven, and detector were set at 100,
205 60, and 110 $^{\circ}\text{C}$, respectively. A 0.5 mL sample of the gas was injected in triplicate.
206 While, H_2S concentration in the biogas was measured using a gas chromatography fitted
207 with a 2.5 m Porapak S column with Hayesep Q (80/100) and a flame photometric
208 detector (Shimadzu GC 14A). Helium at a flow rate of 30 mL/min was used as the
209 carrier gas. The injection port and detector were set at the same temperatures of 150 $^{\circ}\text{C}$.
210 A 0.2 mL sample of the gas was injected in triplicate.

211 VFAs (acetic, propionic, and butyric acid) in the liquid sample were measured
212 by using the gas chromatograph connected with a flame ionization detector (Shimadzu
213 GC 8A). A 30 m capillary column packed with fused silica (Stabiwax® column) was
214 used. The inlet temperature of 230°C and detector temperatures of 250°C were set. The
215 running temperature of the column were set as 60 °C for 35 min, 2 °C/min to 110 °C,
216 10 °C/min to 200 °C, and hold for 1 min.

217 Total alkalinity, COD, Total Kjeldahl Nitrogen (TKN), TS, VS, ash, pH and
218 sulfate content of the liquid sample were analyzed according to the standard methods
219 (APHA, 2012). TOC-Liquid: multi N/C 3100 TOC analyzer (Analytik Jena) was used to
220 determine the total Organic Carbon (TOC).

221 The microbial communities were analyzed by using the Next Generation
222 Sequencing (NGS) technology. Total genome DNA from the samples was extracted
223 using CTAB/SDS method. DNA concentration and purity was monitored on 1% agarose
224 gels. According to the concentration, DNA was diluted to 1ng/μL using sterile water.
225 16S rRNA/18SrRNA/ITS genes of distinct regions which were amplified using specific
226 barcode. All PCR reactions were carried out with Phusion® High-Fidelity PCR Master
227 Mix (New England Biolabs). PCR products quantification and qualification was carried
228 out by mixing the same volume of 1X loading buffer (containing SYB green) with PCR
229 products and with an operated electrophoresis on 2% agarose gel for detection. Samples
230 with bright main strip between 400bp-450bp were selected for further experiments.
231 PCR products were mixed at equal density ratios. The mixed PCR products were
232 purified with Qiagen Gel Extraction Kit (Qiagen, Germany). The libraries generated
233 with NEBNext® Ultra™ DNA Library Prep Kit for Illumina and quantified via Qubit
234 and Q-PCR, were analyzed by Illumina platform. Statistically significant differences in

235 the results were determined using the one-way analysis of variance (ANOVA) of SPSS
236 v26.0 software (IBM, USA).

237

238 **3. Results and discussion**

239 3.1. Performances of UASB reactor fed with NaHCO₃-supplemented DSLS

240 Daily methane production rates, methane yields and methane contents of the
241 biogas in the UASB reactor fed with NaHCO₃-supplemented DSLS are presented in
242 Fig. 1. A summary of the reactor performances at steady state are given in Table 2. For
243 the start-up phase fed with DSLS at OLR 1.11 g-COD/L-reactor·d and HRT of 20 days,
244 an average methane production rate of 174.52 mL/L-reactor·d was observed.

245 After the start-up phase, HRT was reduced to 10 days and the OLR was also
246 reduced to 0.89 g-COD/L-reactor·d in Phase I. The higher feed flow rate with lower
247 feed concentration has let the system to slowly acclimate to the higher shearing force.
248 An average methane production rate at steady state of 148.98 mL/L-reactor·d slightly
249 lower than the start-up phase was observed. The average methane yield was 166.40 mL-
250 CH₄/g-COD_{added} and the average methane content in the biogas was 66.23%.

251 In Phase II, where the OLR was twice higher (1.79 g-COD/L-reactor·d), the
252 average methane production rate reached 403.25 mL/L-reactor·d. This result indicates
253 that a higher substrate density could enhance the activities of microorganisms present in
254 the reactor, reaching also to a higher methane yield of 226.35 mL-CH₄/g-COD_{added}. A
255 slightly higher average methane content in biogas of 67.19% was also observed. In
256 addition, the methane yield in this phase achieved 77.27% of the theoretical
257 yield (350 mL-CH₄/g-COD) which was 270.75 mL-CH₄/g-COD_{removed}.

258 In phase III, the OLR was increased to 3.57 g-COD/L-reactor·d. In this phase,

259 methane production rate achieved 467.61 mL/L-reactor·d which was 16% higher than
260 phase II. However, the methane yield and average methane concentration has
261 significantly decreased to 130.50 mL-CH₄/g-COD_{added} and 51.81%, respectively. The
262 methane yield was 158.03 mL-CH₄/g-COD_{removed} which was only 45.15 % of the
263 theoretical yield. Meanwhile, an increase in VFA concentration was observed in Phase
264 III. The remaining VFA concentrations in the effluent are presented in Fig.2. In all
265 phases, butyric acid had the lowest concentration in the effluent. Acetic and propionic
266 acids were detected in nearly amounts in the effluent. The VFA concentrations in phase
267 III were higher than during the other phases operated at a lower OLR as shown in Fig.2
268 (1.54 g/L acetic acid, 1.48 g/L propionic acid, and 0.58 g/L butyric acid). Higher
269 influent COD concentration had therefore led to higher VFAs concentration which
270 possessed the potential to partly inhibit the methanogenic activity, hence lowering the
271 methane yield.

272 The optimum concentration of acetic acid, propionic acid, and butyric acid to
273 achieve the maximum cumulative methane yield is 1.6, 0.3, and 1.8 g/L, respectively, as
274 reported by Wang et al. (2009). A concentration level of 2.0 g/L of acetic, 0.9 g/L of
275 propionic acid, and 4.5 g/L of butyric acid were reported as inhibition threshold levels
276 of VFAs acid (Demirel and Yenigün, 2002).

277 Since the raw SLS has also exhibited a high sulfate concentration, thus the
278 reduction of sulfate, to H₂S by sulfidogenesis is unavoidable and considered as a major
279 concern for an effective anaerobic treatment. RWA was used to reduce sulfate in SLS,
280 but since RWA also release some sulfate, thus a maximum of 10 g/L RWA loading was
281 suggested (Raketh et al., 2021). Sulfate concentration in the high OLR phases are
282 shown in Table 2. The DSLS in phase III contained the highest sulfate concentration.

283 This is also the reason that could have led to lower methane production yield. H₂S

284 concentration during methane production in UASB reactor shown in Fig.1c. Trends of

285 H₂S concentration in the biogas production fully correlated with the sulfate loading rate.

286 Phase I with the lowest sulfate loading influent produced very low H₂S in the biogas.

287 There was significant difference of sulfate removal efficiency between Phase I and III.

288 In Phase III the highest H₂S generation in a range of 16,826-36,661 ppm was exhibited.

289 Then, the feed rate was reduced to OLR of 1.79 g-COD/L-reactor·d similarly to

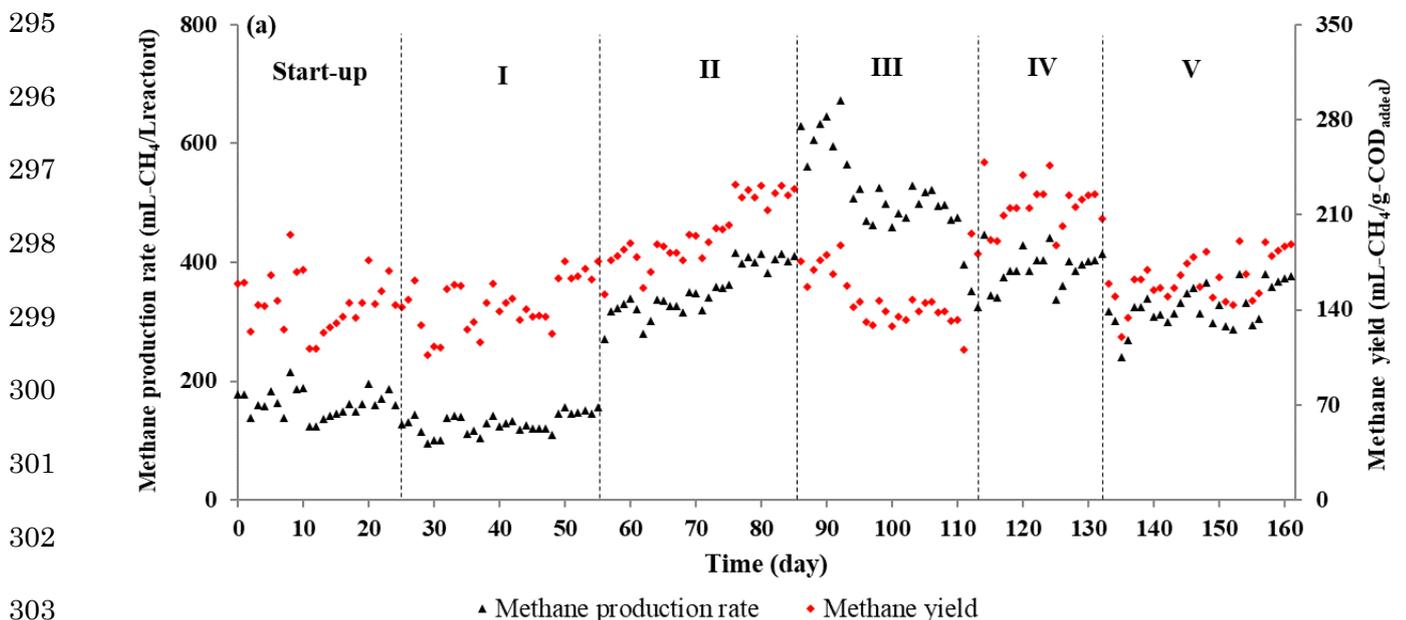
290 Phase II for the purpose of confirming the optimum feed rate to attain the highest

291 methane yield. The methane production rate was however slightly lower (3% lower)

292 than in Phase II, while the average methane content in biogas rebounded to 61.60 %.

293 Furthermore, methane yield showed no significant difference between phases II and IV.

294



304

305

306

307

308

309

310

311

312

313

314

315

316

317

318

319

320

321

322

323

324

325

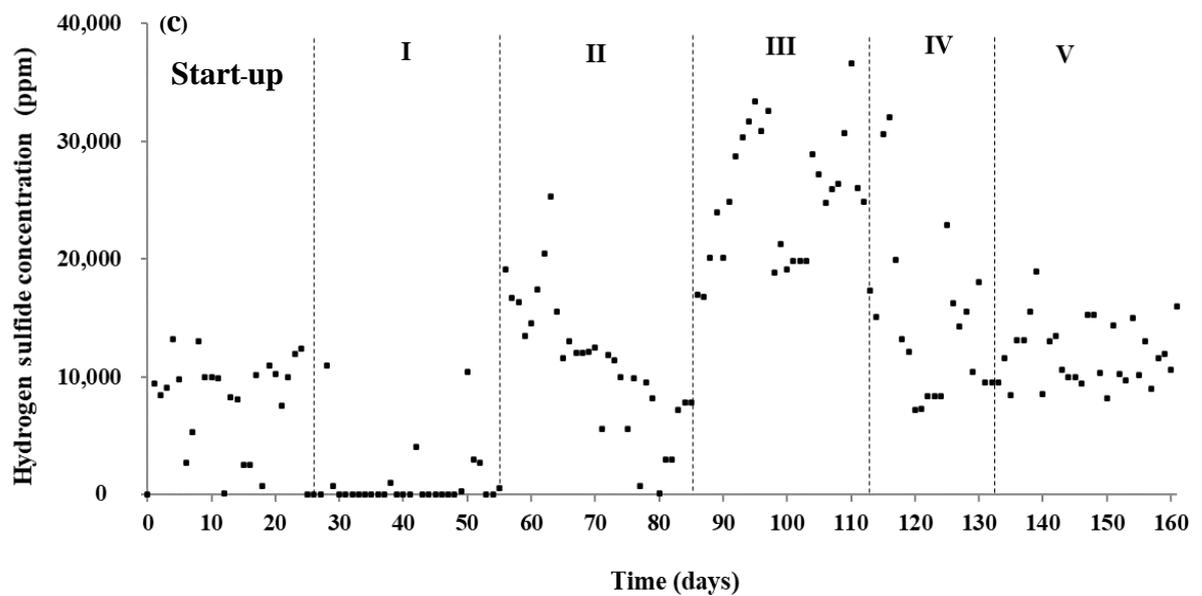
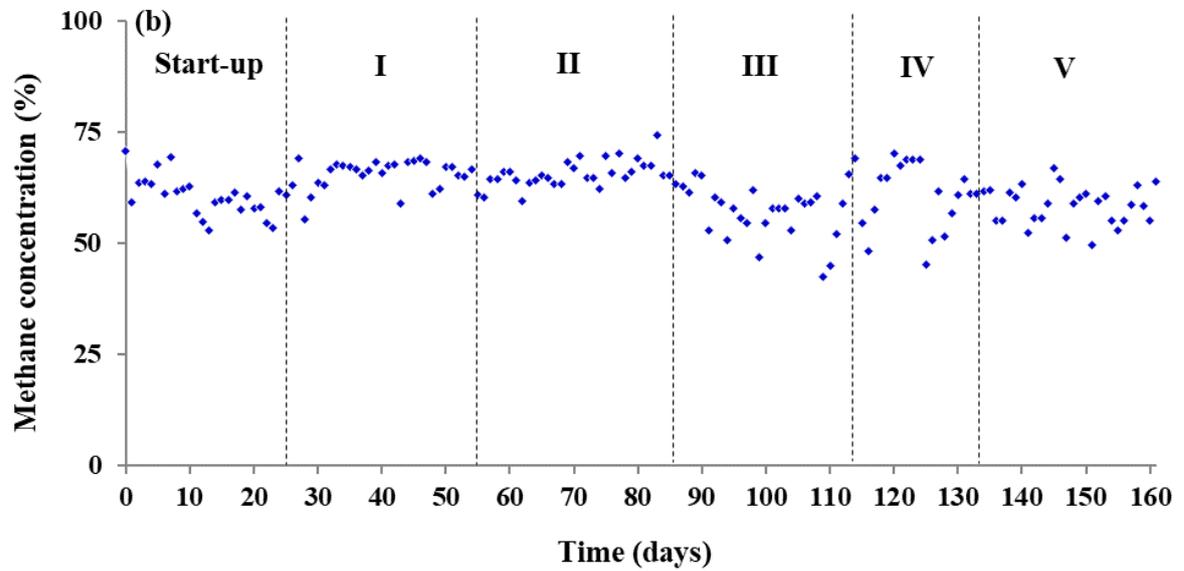
326

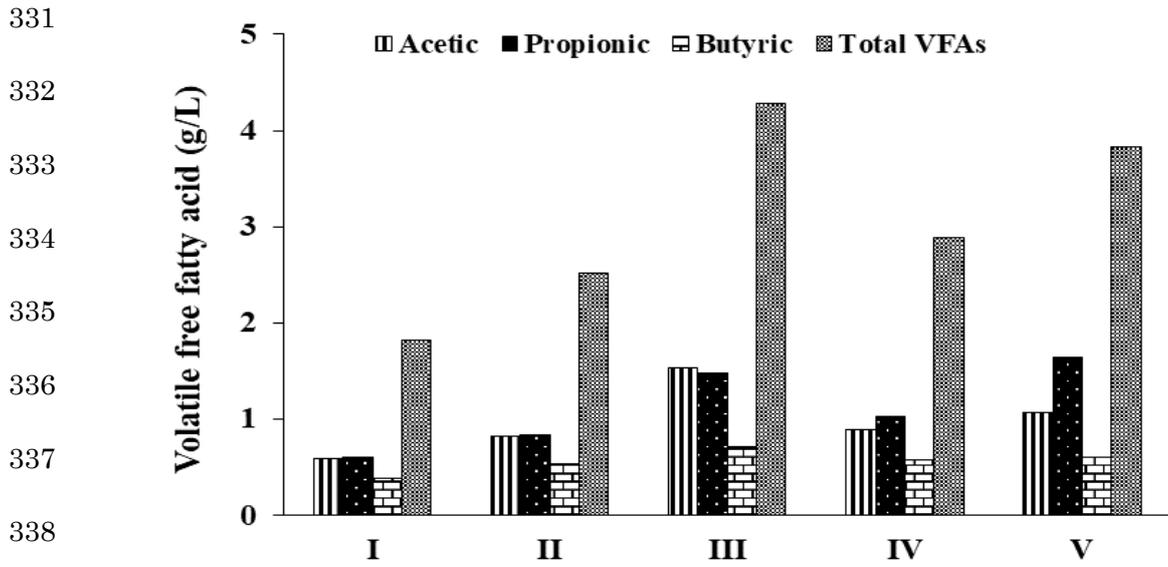
327 **Fig. 1.** UASB reactor performance of methane production: (a) methane production rate

328 and methane yield, (b) methane concentration, and (c) hydrogen sulfide concentration,

329 for various levels of OLR (Phase I=0.89, II=1.79, III=3.57, IV=1.79, and V= 2.01 g-

330 COD/L-reactor·d) at 10-day HRT





341 **Fig. 2.** Volatile fatty acids of effluent at steady-state in UASB reactor during methane
 342 production, for various levels of OLR (Phase I=0.89, II=1.79, III=3.57, IV=1.79, and
 343 V= 2.01 g-COD/L-reactor·d) at 10-day HRT.

344

345 **Table 2** Summary of the steady-state methane production with DSLs at 10-day HRT.

| Parameters | Values | | | | |
|---|---------------------|---------------------|---------------------|---------------------|---------------------|
| Phase | I | II | III | IV | V |
| OLR (g-COD/L-reactor·d) | 0.89 | 1.79 | 3.57 | 1.79 | 2.01 |
| Day at the condition (day) | 26-55 | 56-86 | 87-112 | 113-132 | 133-162 |
| Day at steady state (day) | 50-55 | 77-86 | 108-112 | 127-132 | 158-162 |
| CH ₄ yield (mL/g-COD _{added}) | 166.40 ^a | 226.35 ^b | 130.50 ^c | 218.70 ^b | 185.70 ^d |
| CH ₄ yield (mL/g-COD _{removed})* | 210.45 ^a | 270.45 ^b | 158.03 ^c | 269.49 ^b | 224.83 ^a |
| CH ₄ production rate (mL/L-reactor·d) | 148.98 ^a | 403.25 ^b | 467.61 ^c | 389.60 ^b | 371.40 ^d |
| CH ₄ composition (%) | 66.23 ^a | 67.19 ^a | 51.81 ^b | 61.60 ^c | 59.79 ^c |

| | | | | | |
|--|--------------------|--------------------|--------------------|--------------------|--------------------|
| H ₂ S composition (%) | 0.54 ^a | 0.65 ^a | 2.91 ^b | 1.24 ^c | 1.08 ^c |
| Biogas yield (mL/g-COD _{added}) | 265.34 | 365.00 | 212.79 | 363.57 | 300.00 |
| Biogas production rate (mL/L-reactor·d) | 233.33 | 647.00 | 738.00 | 656.25 | 596.94 |
| Influent pH | 7.13 | 6.57 | 6.56 | 6.73 | 7.02 |
| Effluent pH | 7.48 | 7.70 | 7.63 | 7.62 | 7.71 |
| Influent COD (mg/L) | 8.87 | 17.85 | 35.50 | 17.85 | 20.05 |
| Effluent COD (mg/L) | 2.60 | 4.65 | 9.43 | 4.99 | 5.37 |
| COD removal efficiency (%) | 70.66 ^a | 73.95 ^a | 73.44 ^a | 72.04 ^a | 73.24 ^a |
| Influent sulfate (mg/L) | 695.10 | 1390.20 | 2780.40 | 1390.20 | 1488.45 |
| Effluent sulfate (mg/L) | 346.91 | 274.54 | 212.21 | 184.19 | 194.98 |
| Sulfate removal efficiency (%) | 50.09 ^a | 80.25 ^b | 92.37 ^c | 86.75 ^c | 86.90 ^c |
| Energy recovery (kJ/g-COD _{added}) | 5.30 | 7.14 | 4.14 | 6.94 | 5.89 |

All the value is in average,

*The CH₄ yield was calculated at STP,

^{a-d} are the statistically significant difference ($p \leq 0.05$).

346

347 3.2. Performances of UASB reactor fed with DSLS with effluent recirculation

348 To maintain the buffering capacity, the NaHCO₃ solution was replaced by a
349 recirculation of the UASB effluent during Phase V. In this phase, OLR of 2.01 g-
350 COD/L-reactor·d was applied, which was slightly higher than in Phase IV due to the
351 remaining COD in the effluent. In Phase V, methane yield and methane production rate
352 were lower than in the optimal feed rate (Phase II and IV) which was 185.70 mL-
353 CH₄/g-COD_{added} and 371.40 mL/L-reactor·d. The average CH₄ composition was
354 59.79 % which is slightly lower than that in Phase IV. However, there was no
355 significant difference of average CH₄ composition between phase IV and V.

356 Nonetheless, methane yield in this phase ($185.70 \text{ mL-CH}_4/\text{g-COD}_{\text{added}}$) was higher than
357 in Phase I and III as show in Table2.

358

359 3.3. Monitoring of alkalinity and pH

360 After removal of sulfate with RWA, the pH of DSLS increased from 5.24-5.54 to
361 5.99-6.45. The alkalinity of DSLS ($3,108 - 3,267 \text{ mg-CaCO}_3/\text{L}$) was also higher than in
362 the raw SLS ($2,890-2,953 \text{ mg-CaCO}_3/\text{L}$), most likely due to the the alkaline leachate
363 from the metal oxide of RWA, released into desulfated SLS. The desulfated SLS used
364 in AD process was diluted by the NaHCO_3 solution, hence the initial alkalinity of
365 influent decreased during Phase I, II, III, and IV at 1,590, 2,110, 3,080, and 2,650 mg-
366 CaCO_3/L , respectively. In Phase V, the DSLS was diluted by the rich alkalinity effluent,
367 thus higher alkalinity of $4,930 \text{ mg-CaCO}_3/\text{L}$ was obtained.

368 Alkalinity is the parameter referred to the buffer capacity of the AD system. The
369 fact regarding this matter is that it should have high alkalinity enough to maintain the
370 system pH. A digester should be kept higher than $2,000 \text{ mg-CaCO}_3/\text{L}$ of alkalinity o to
371 resist to the changes of pH in the system (Reungsang., 2019). Alkalinity of the effluent
372 during all operation phases was higher than $2,000 \text{ mg-CaCO}_3/\text{L}$ (Fig.3a), indicating that
373 the digester was kept within the desired range of alkalinity by adjusting the buffering
374 capacity in the feed. The start-up period showed a higher effluent alkalinity than in
375 phases I and II due to a longer HRT. The effluent alkalinity increases when OLR
376 increased from Phase I to Phase III because the influent alkalinity was also increased.
377 The trend of effluent alkalinity in Phase III showed the highest trend due to higher COD
378 loading in influent than the other phases. Moreover, a sharp increase in effluent
379 alkalinity can also be used as a parameter to monitor and control the AD systems. In

380 Phase III, the trend of effluent alkalinity was increasing which can negatively affect the
381 system if the value is too high, hence the operation in Phase III was switched back to
382 lower OLR before the three times HRT operation time (at day 26 of Phase operation).
383 The effluent alkalinity in Phase V was maintained at about 8,000 mg/L. This indicated
384 that using the effluent recirculation strategy was successful for maintaining high
385 alkalinity without external chemical addition to efficiently produce biogas from DSLS.

386 The pH of influent and effluent during methane production are presented in
387 Fig.3b, which indicated that pH significantly affects the performance of AD system, and
388 also considered as a crucial factor influencing the growth of diverse microorganisms.
389 The optimal pH range for producing methane is 6.7-8.0 (Chandra et al., 2012;
390 Cremonoz et al., 2020; Kongjan et al., 2014). The result indicated that the pH of
391 influent was maintained by using NaHCO_3 in Phase I to IV and by using effluent
392 recirculation in Phase V. Influent pH fluctuated depending on the OLR of DSLS and the
393 characteristics of the raw SLS which was collected from the factory once a month.
394 During Phase I, the pH influent ranged 6.93-7.22. Then, change of OLR during Phase II
395 had caused a slight decrease of influent pH (6.44-6.85) due to higher influent COD
396 concentration. As expected, the lowest pH (6.24-6.62) was obtained at the highest OLR
397 in Phase III. The other promising method to raise the pH of DSLS is mixing with the
398 alkaline rich effluent. As shown in Phase V, the influent pH was increased to 7.20-7.43.
399 It is worth to note that a rather stable pH of effluent in the range of 7.48-7.90 were
400 obtained in Phase I – V. This indicated that all operating phases were run at the
401 condition which provide sufficient buffering capacity to properly maintain the system
402 pH.

403

404 In addition, monitoring the ratio of VFA and alkalinity (VFA/ALK) produced in
405 the reactor is a more valuable tool in following the performances of the AD process. The
406 VFA/ALK ratio is the specific bicarbonate alkalinity level that can help furnish insight
407 into the reactor stability. In this study, the VFA/ALK ratio during operation varied
408 between 0.02-0.31, which was lower than the 0.40 imposed in the literature as the
409 inhibition thresholds. Most probably this was due to an excessive concentration of VFA
410 which led to the process of acidification (El Gnaoui et al., 2020; Kim and Kim, 2020;
411 Wilawan et al., 2014). Hence, the VFA/ALK ratio during operation remained in the
412 optimum range and the reactor could maintain the stability of the buffering capacity for
413 optimal methane production throughout the experiment.

414 During the AD, carbon to nitrogen (C/N ratio) has effects on methane
415 production, and it is an essential factor for stable operation. The C/N ratio of the SLS in
416 the current study is approximately 9.5, which is a low value when compared to the
417 optimum C/N ratio (20-30) (Fu et al., 2012), due to the SLS contains a high
418 concentration of nitrogenous compounds such as ammonia and protein. When C/N ratio
419 of substrate is low, nitrogen will be rapidly consumed for growing most microbes,
420 although this has a positive effect on methane production rate. However, the form of
421 ammonium ions that increases the pH can adversely affects biogas production (Yen and
422 Brune, 2007). This is one of the reasons which might probably described the product
423 inhibition from the overloading in Phase III.

424

425

426

427

428

429

430

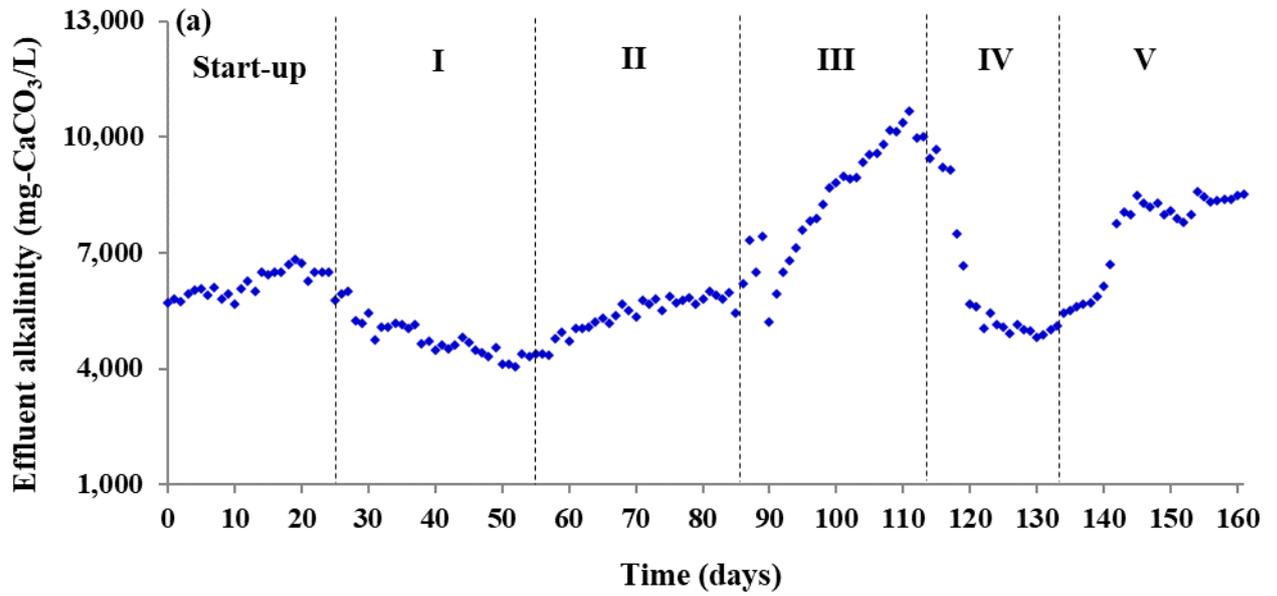
431

432

433

434

435



436

437

438

439

440

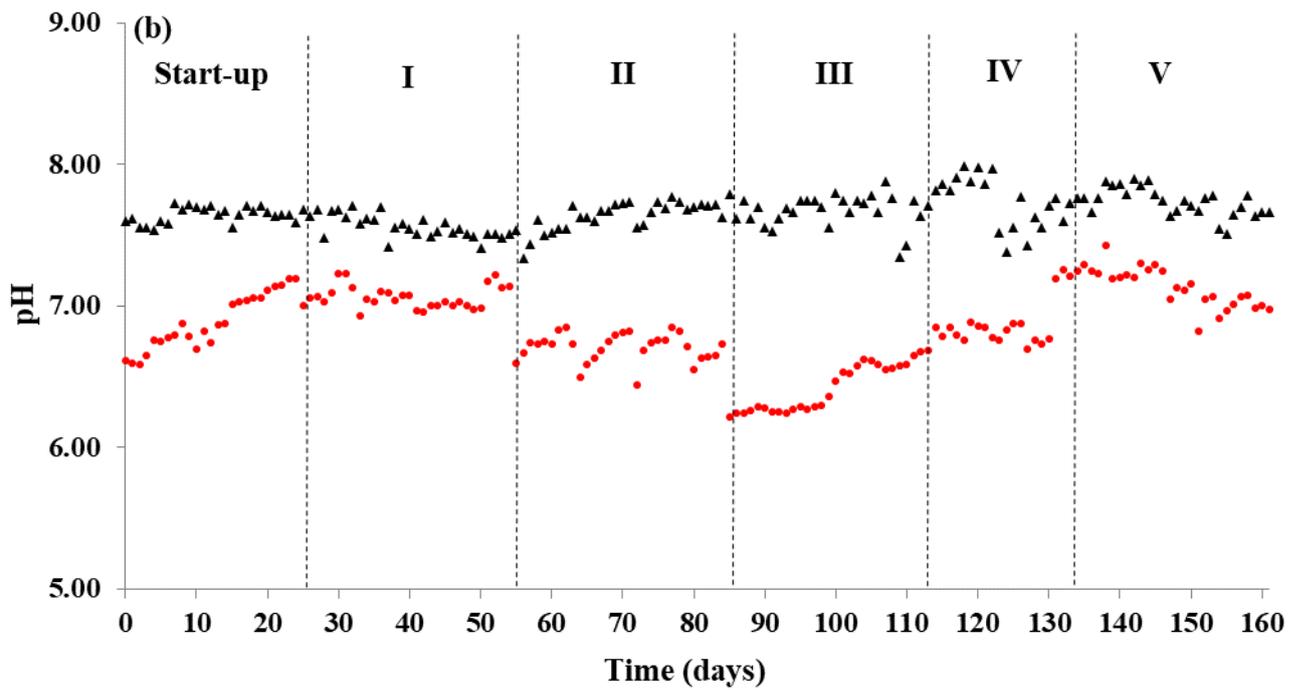
441

442

443

444

445



446

447

• Influent ▲ Effluent

448

449 **Fig. 3** UASB reactor performance of methane production: (a) effluent alkalinity, and (b)

450 pH of influent and effluent, for various levels of OLR (Phase I=0.89, II=1.79, III=3.57,

451 IV=1.79, and V= 2.01 g-COD/L-reactor·d) at 10-day HRT.

452 3.4. Anaerobic biodegradability

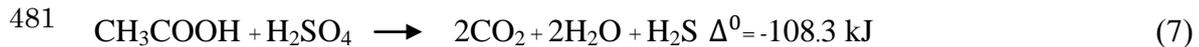
453 COD removal efficiencies in Phase I - V ranged between 70.66 and 73.95 %.

454 The operation at Phase I obtained lower COD removal efficiency than other phases due
455 to a lower COD concentration. When COD concentration was increased to 17.85 mg/L
456 in Phase II, the COD removal efficiency slightly increased and was then constant when
457 COD was increased to 35.50 mg/L in Phase III. For the result presented that there were
458 no significant differences in COD removal efficiencies. Although the pathway of
459 methanogenesis was changed due to different COD and sulfate concentration was feed
460 during Phase I-V but COD removal efficiencies of all phases was not different.

461 In addition, The COD distribution in the effluent was also calculated to
462 check the reliability which is presented in Table S1. The total of main VFAs were
463 contributed to the effluent ranged from 54.80-88.24%. While the rest of the organic
464 matter such as sugar, lactic acid, formic acid, and ammonium group were left in the
465 effluent ranging in 11.76-45.20 %. Phase III had the highest total VFA concentration
466 due to the highest COD concentration fed. However, when considering the COD
467 contribution in the effluent, it was found that the other organic matter beside the main
468 VFAs were also presented in higher portion compared to another phase with lower OLR
469 loading. This observation confirms that the lower biodegradation causing by lower
470 microorganism favorable at the overloading of OLR at Phase III.

471 Typically, during AD, organic substances are converted to biogas with the main
472 composition of 40-75% CH₄, 15-60% CO₂, 5-10% water, and 0.005-2% H₂S. Higher
473 CH₄ production yield is expected along with higher COD removal. However, in systems
474 containing high sulfate concentrations, Sulfate-Reducing Bacteria (SRB) are also able to
475 use the organic substances to generate H₂S, outcompeting MPA in using organic

476 substances to produce methane. This resulted in a decrease of the methane production
477 yield and higher H₂S production when sulfate concentration was increased. For
478 instance, the VFAs can be converted to H₂S as illustrated in the following Equation (6)-
479 (8) (Jariyaboon et al., 2015) .



483 In this study, the average sulfate removals ranged between 50.09 and 92.37 %
484 and the highest sulfate removals were achieved at Phase III. The organic loading
485 corresponding to initial sulfate loading in influent, the higher OLR was fed resulting in
486 higher sulfate fed too. The result presented that higher sulfate removal efficiencies were
487 obtained correspondingly with higher H₂S concentration observed in the biogas.
488 Normally, when higher OLR was fed not only sulfate was higher but also the metal ions
489 which was leached from RWA during sulfate removal process. The metal ion leached
490 from RWA such as Ca, Mg, Fe, Ni, P, and K. Each ion had a limit value for the optimum
491 condition for methane production in the AD process as mentioned in our previous
492 research (Raketh et al., 2021). Therefore, metal ions leaching was also a one of the
493 reasons which was affected to the AD process.

494 The energy production in this process was also assessed and the values are shown
495 in Table 2. Only methane heating value was used to convert the produced biogas to
496 energy. The energy yield from DSLs ranged 4.14-7.14 kJ/g-COD_{added}. Phases II and IV
497 had high energy yield with 7.14 and 6.94 kJ/g-COD_{added}, respectively, while the highest
498 OLR (Phase III) led to the lowest energy yield (4.14 kJ/g-COD_{added}) due to the
499 production of minimum methane yield. Phases II and IV showed higher potential to

500 recover energy from DSLS than Phases I and III. Energy recovered from Phase II was
501 even 42.02% higher than the energy generated in Phase III. However, a higher COD
502 concentration of DSLS was loaded causing the inhibition to possibly occur more
503 significantly, and achieved a lower energy recovery yield. Phase V achieved higher
504 energy recovery (5.89 kJ/g-COD_{added}) than Phase I and III while attaining 17.5 % of
505 energy recovery, lower than Phase II.

506

507 3.5. Microbial community

508 Fig. 4 presents the relative abundance of the microbial community kingdom
509 namely: (a), bacteria in phylum level (b), archaea in genus level (c) the microbial
510 community in the UASB reactor for various levels of OLR.

511 The highest relative abundance of total bacteria was observed in Phase III,
512 operated at the highest OLR, while Phases I and II showed similar quantities. The
513 percentages of relative abundance present the indication of bacterial amounted for
514 77.86%, 78.25%, and 86.05% in Phase I, II, and III, respectively (Fig.4a). The microbial
515 consortium exhibited the highest capability in Phase III due to the highest organic
516 loading applied in this phase whereby the higher concentration of both organic
517 compounds and nutrients in the system could drive the rate of the biochemistry
518 reactions leading to more growth of bacteria community.

519 Sequences retrieved from Phase I showed dominant phyla within the bacterial
520 community with *Bacteroidota* (23.46%), *Desulfobacterota* (17.73%), and *Chloroflexi*
521 (15.45%). Simultaneously, the bacterial community in Phase II was dominated by
522 *Bacteroidota* (25.69%), *Desulfobacterota* (18.04%), and *Synergistota* (14.83%) phyla.
523 On the contrary, Phase III was illustrated among three main bacterial community

524 phylum which was dominated by *Proteobacteria* (42.24%), *Firmicutes* (17.77%), and
525 *Bacteroidota* (13.03%) as shown in Fig.4b. *Bacteroidetes* and *Firmicutes* represent
526 important contributors for the degradation of saccharides and proteins. As well as
527 enriching at an expeditious multiplication rate in a growth environment, it also indicated
528 the high concentration of a soluble organic substance. Additionally, VFAs such as
529 butyrate were reported to be biodegraded by *Firmicutes* as fermentative and syntrophic
530 bacteria (Garcia-Peña et al., 2011; Kabisch et al., 2014) .

531 Similarly, *Proteobacteria* and *Chloroflexi* are also important bacteria involved in
532 hydrolysis and acidification. These bacteria made the overall transformations that
533 underpin the function of AD systems (Dai et al., 2016; Petriglieri et al., 2018).
534 *Desulfobacterota* is a phylum known to harbor sulfur-cycling bacteria (Bell et al.,
535 2022). The members affiliated with *Desulfobacterota* were the third most abundant
536 phylum of Phase I and II. *Desulfobacterota* is mostly composed of a diversity of SRB
537 (Yang et al., 2022). The presence of SRB in this study was found to be resulted from the
538 use of sulfate-rich wastewater in AD process.

539 The abundance of archaea community is presented in Fig.4c. Archaea were more
540 dominant in phases I and II than in Phase III. This corresponds to the decreasing in
541 methane production yield in Phase III as mentioned previously. The third main genus
542 of archaea in Phase I was affiliated to *Methanosaeta* (87.14%),
543 *Candidatus_Methanofastidiosum* (3.40%), and *Methanomethylovorans* (3.15%).
544 Furthermore, Phase II, and III were found to be majorly dominant by *Methanosaeta*
545 and *Methanomethylovorans* which were 44.73%, 45.24% of Phase I, and 23.44%,
546 72.47% of Phase II, respectively. The results indicated that the *Methanomethylovorans*
547 appeared to be relatively higher in Phase II and III. *Methanolinea* and

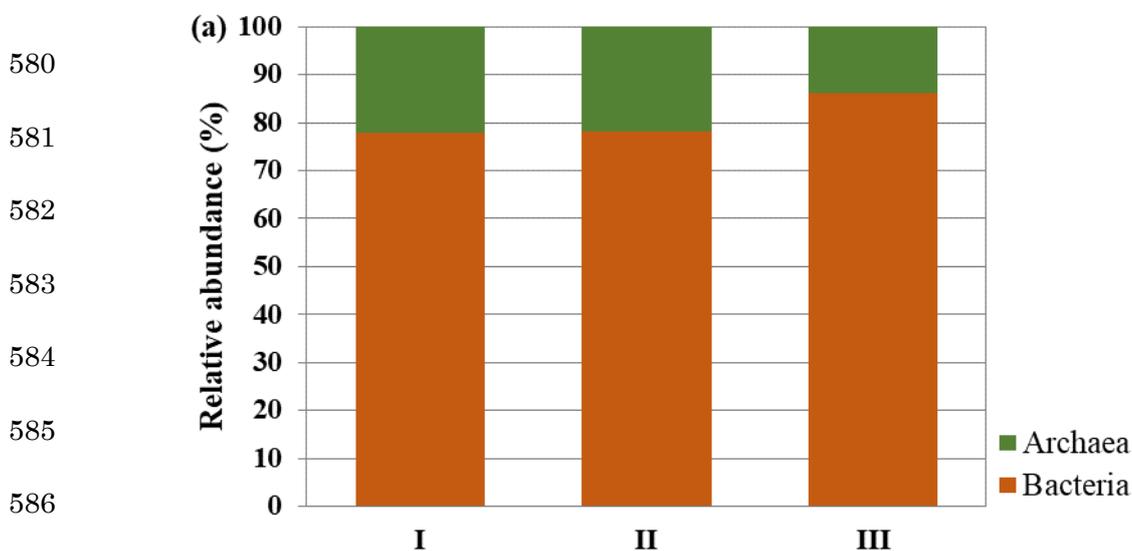
548 *Methanobacterium* species were also detected but at low level (<3%).

549 The *Methanosaeta* genus gathers acetoclastic methanogens utilizing acetate as a
550 substrate for methane production (Dai et al., 2016). *Methanomethylovorans sp.* are
551 methylotrophic methanogens and competent to grow and achieve methanogenesis from
552 methanol, mono-, di-, and trimethylamine. Hydrogen and acetate are not utilized (Kim
553 and Rhee, 2015; Whang et al., 2015). Methylotrophic methanogenesis is often presented
554 to be responsible for methane production in sulfate-rich environments (Xiao et al.,
555 2018). According to the SLS was used as a substrate which was sulfate-rich wastewater,
556 resulting *Methanomethylovorans sp.* was found in all phases. Particularly in Phase III
557 the highest *Methanomethylovorans sp.* appeared due to the highest OLR and sulfate
558 were loaded. The results confirms that although the diverse genus-aerchare was found, it
559 meant that the pathway to produce methane was different, but the COD removal
560 efficiencies were still obtained in similar value which was mentioned in previous
561 section.

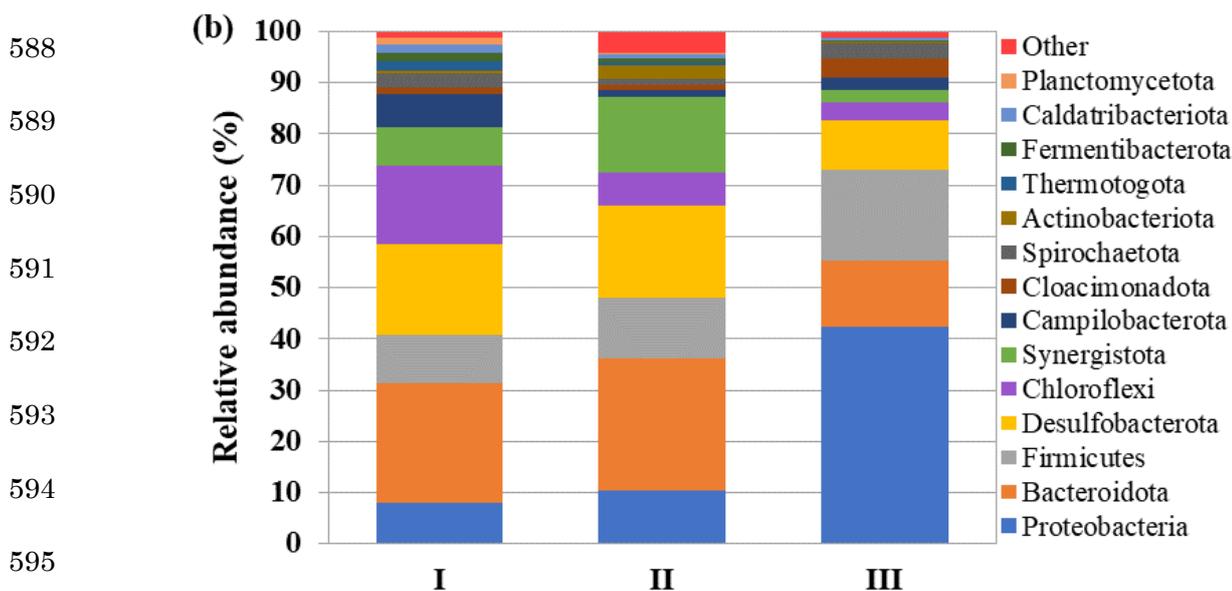
562 The composition of the microbial community in the reactor is related to methane
563 production performances, and seed organisms or inoculum type that could also have a
564 large impact on reactor dynamics (Rajendran et al., 2020). A variety of anaerobic
565 bacteria and methanogenic archaea were observed in this study with differences in their
566 relative abundance. The relatively high abundance of *Proteobacteria*, *Bacteroidota*,
567 *Firmicutes*, and *Desulfobacterota* were found in all phases in the AD process.
568 Interestingly, it was also observed that a relative dominance of *Desulfobacterota*
569 phylum was likely due to the use of sulfate-rich wastewater as substrate. An effective
570 metabolism was achieved from archaeal community majorly dominated by the genera
571 *Methanosaeta sp.* and *Methanomethylovorans sp.* Nonetheless, the microbial

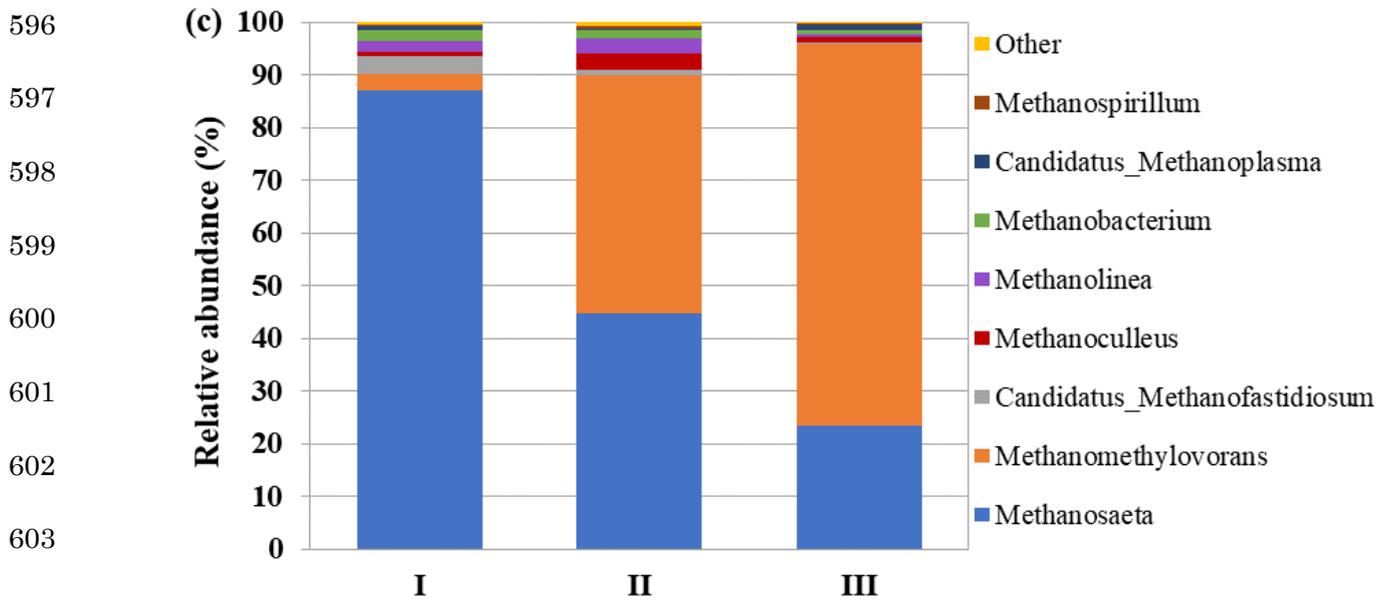
572 community analysis was performed to provide a better understanding of the granules
 573 functioning and support the macroscopical observations. Interestingly, a clear shift of
 574 the archaeal community was observed, providing new insight into the microbial
 575 community in granular systems. In practice, the efficiency and stability of the AD
 576 process could be monitored by the microbial community (Lim et al., 2020). A stable AD
 577 process desires an exquisite balance of microbial population dynamics and metabolic
 578 activities among the different guilds or trophic groups of the microorganism.

579



587

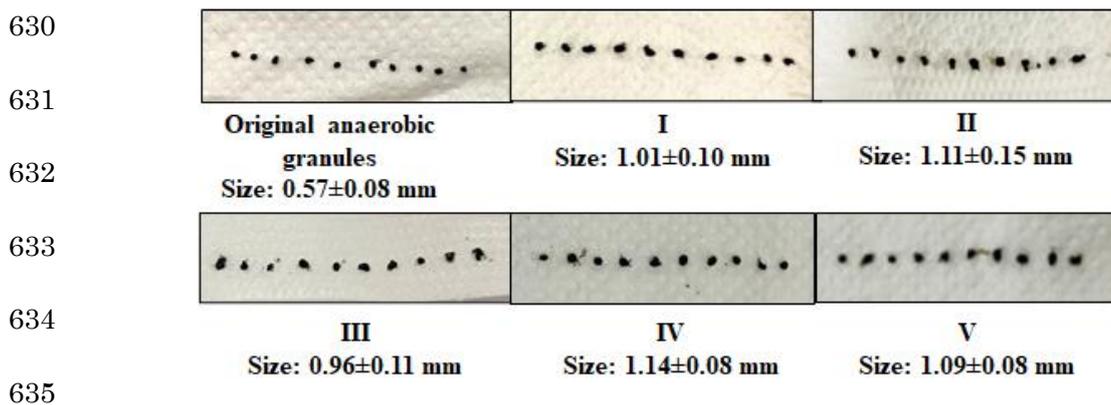




605 **Fig.4** Relative abundance of microbial community in UASB reactor performance of
 606 methane production: (a) Kingdom, (b) phylum-bacteria, and (c) Genus-archaea, for
 607 various levels of OLR (Phase I=0.89, II=1.79, III=3.57 g-COD/L-reactor·d) at 10-day
 608 HRT.

609 Anaerobic granules are particulate biofilms, spontaneously formed by auto-
 610 immobilization of anaerobic bacteria without other additional support material
 611 (McHugh et al., 2003). These particles comprised of an intertwined mixture of the
 612 symbiotic anaerobic microorganisms that operate together in methane fermentation. The
 613 size of the anaerobic granules in the UASB reactor were examined as illustrated in
 614 Fig.5. The range of anaerobic granule size was 0.57-1.14 mm. The granule size also
 615 increases with higher CH₄ production yields. Consistently, the anaerobic granule size
 616 decreases following the declination of CH₄ yield at overload OLR in Phase III. This
 617 obviously indicates that the feed conditions favored the growth of anaerobic
 618 microorganisms and yielded higher auto-immobilization resulting in a bigger size of
 619 anaerobic granules and higher methane production.

620 After Phase I, the granule size increased from the original anaerobic granules,
 621 increasing from 0.57 ± 0.08 mm to 1.01 ± 0.10 mm of average anaerobic granules. Phases
 622 II and IV achieved the highest average anaerobic granule sizes of 1.14 ± 0.08 and
 623 1.11 ± 0.15 mm, respectively. This result corresponded to the methane production yield
 624 which showed the highest production in both phases (II and IV), the bigger anaerobic
 625 granules size was also achieved producing higher methane yield. While the anaerobic
 626 granules size of Phase III decreased (0.96 ± 0.11 mm) which was most probably due to
 627 the overload of organic substance, resulting in the disability of anaerobic granules to
 628 auto-mobilise on granules. Therefore, the size of the granules was significantly different
 629 in each phase.



636 **Fig. 5** Anaerobic granules in UASB reactor during methane production, for various
 637 levels of OLR (Phase I=0.89, II=1.79, III=3.57, IV=1.79, and V= 2.01 g-COD/L-
 638 reactor·d) at 10-day HRT.

639

640 3.6 Perspective for methane recovery from DSLS by using UASB reactor

641 Table 3 shows a comparison of single-stage AD and two-stage AD process
 642 performances on using wastewater of concentrated latex industry as a substrate.

643 Methane production yield of single-stage AD in this study had higher significance

644 compared to the Two-stage AD by Kongjan et al, 2014 which used the same type of
645 wastewater as substrate (SLS) (Kongjan et al., 2014). This condition was likely due to a
646 lower OLR and also reduced sulfate taken place before the AD process.

647 For other types of wastewaters issued from the concentrated latex industry,
648 wastewaters at a latex mill were investigated with two-stage AD using acid tank and
649 UASB reactor by Jawjit,2013. A methane yield of 95.12 mL-CH₄/g- COD_{added} was
650 observed which is lower than the methane yield in this study. They used 3 days of HRT
651 and 1.4 g-COD/L·d of OLR which is also lower than this study (Jawjit, 2013). This was
652 possibly due to the fast feed flow rate that did not provide enough time for the
653 completion of the biochemical reaction during AD process.

654 Furthermore, some previous studies reported a higher methane production than
655 this study. Saritpongteeraka and Chaiprapat, 2008 represents that single-stage AD by
656 using ABR demonstrated high performance for decomposing organic substances in
657 concentrated latex wastewater (CLW) (Saritpongteeraka and Chaiprapat, 2008), which
658 was 7.66% of methane production yield higher than this study. In addition, the data
659 achieved from concentrated latex factory in Songkhla, Thailand found that production
660 of methane from CLW by using anaerobic pond feed rate 0.61 g-COD/L·d for 15.7-day
661 HRT achieved 219.97 mL-CH₄/g- COD_{added} of methane yield, which is lower than this
662 study. This phenomenon might be due to a lower OLR and varying characteristics of
663 substrate and reactor operation.

664 In practice, NaHCO₃ as an alkali solution is usually applied for pH control in
665 AD process. The cost of alkali chemicals is relatively high. The effluent recirculation
666 which has the potential of replacing the alkali solution is an attractive choice, although
667 the yield is lower than using NaHCO₃ 17.5 %. For the industrial scale of concentrated

668 latex factory, machine-washing wastewater was generated during the concentrated latex
669 processing. This wastewater can use to dilute the effluent (replacing the tap water in the
670 current study) before mixing with the substrate to be influent wastewater, then feeding
671 into UASB reactor. Hence, this strategy in the experiment not only reduced the alkali
672 chemicals cost but also reduced the volume of tap water for dilution. In addition, UASB
673 technology is a good choice to replace the cover lagoon, the most popular low price-
674 digester for concentrated latex wastewater. The UASB reactor is preferred for high
675 organic loading rates application. Hence, it is possible to use compact UASB for
676 treating large volumes or highly concentrated organic wastes.

677

678 **Table 3** Comparison to previous reports on methane production in AD process from wastewater of concentrated latex industry.

| Substrate | AD | Reactor type | Methane production | | | | | Reference | |
|----------------------------|------|----------------|-----------------------------|-----------------|------------|---|--|-----------|--|
| | | | Temperature (°C) of reactor | OLR (g-COD/L·d) | HRT (days) | CH ₄ Yield (mL-CH ₄ /g-COD _{added}) | CH ₄ production rate (mL/L-reactor·d) | | % CH ₄ Content |
| SLS | TSAD | UASB-UASB | 55 | 4.47 | 9 | 178.70 | 712.00 | 57-65 | (Kongjan et al., 2014) |
| Wastewater at a latex mill | TSAD | Acid tank-UASB | 35 | 1.4 | 3 | 95.12 | NS | 60-70 | (Jawjit, 2013) |
| CLW | SSAD | ABR | 35 | 0.60 | 10 | 242.31 | NS | 65-75 | (Saritpongteraka and Chaiprapat, 2008) |
| CLW | SSAD | Anaerobic Pond | 33 | 0.61 | 15.7 | 219.97 | 131.70 | 59.8 | * |
| Desulfated SLS | SSAD | UASB | 35 | 1.79 | 10 | 226.35 | 403.25 | 67.19 | This study |

SLS = Skim Latex Serum
CLW = Concentrated latex wastewater
NS = No Show

TSAD = Two stages anaerobic digestion
SSAD = Single stage anaerobic digestion
*Data achieve from Concentrated Latex Factory in Songkhla, Thailand

UASB-UASB = Up-flow anaerobic sludge blanket- Up-flow anaerobic sludge blanket
ABR = Anaerobic baffled reactor
UASB = Up-flow anaerobic sludge blanket

679

680

681

682 **4. Conclusions**

683 This study demonstrates that treatment of DSLS by using UASB reactor in
684 single stage AD has the capability and efficiency for producing methane while using
685 effluent recirculation method can replace the external buffering, NaHCO₃ solution. Due
686 to the nature characteristic of DSLS which still contain some sulfate and has low C/N
687 ratio, the investigation on suitable OLR is needed. The average maximal methane
688 production yield of 226.35 mL/g COD_{added} was achieved and 7.14 kJ/g-COD_{added} of
689 energy can be recovered in which OLR of 1.79 g-COD/L-reactor·d was fed (Phase II).
690 Although the effluent recirculation is a practical and economical method to keep
691 sufficient alkalinity for the stable system. However, with 53% of the effluent
692 recirculation 18% decrease of methane yield than in Phase II was obtained.

693 For organic wastewater treatment, one of the main purposes is organic reduction.
694 COD removal efficiency was within the range of 70.66-73.95% while the range of
695 sulfate removal efficiency was within 50.09-92.37%. This indicated that post treatment
696 of the effluent is still needed. The current study has demonstrated the capabilities of the
697 UASB reactor in AD for the treatment of SLS. However, to enhance the methane
698 productivity of SLS, other low cost method for reducing more sulfate content and co-
699 digestion for increasing C/N ratio are suggested.

700

701 **Acknowledgements**

702 This study was granted by the Royal Golden Jubilee Ph.D. Program (Grant No.
703 PHD/0216/2559), Thailand Research Fund. The authors would like to thank the
704 concentrated latex factory, the glove factory, the frozen food factory in Songkhla
705 Province, and the palm oil mill factory in Surat Thani Province, Thailand for providing

706 materials to our research. In addition, the authors are also grateful to the Language
707 Center, Prince of Songkla University, Pattani campus, for proofreading a manuscript
708 draft.

709

710 **References**

711

712 Angelidaki, I., Sanders, W., 2004. Assessment of the anaerobic biodegradability of
713 macropollutants. *Rev. Environ. Sci. Biotechnol.* 3, 117–129.

714 <https://doi.org/10.1007/s11157-004-2502-3>

715 Angenent, L.T., Karim, K., Al-Dahhan, M.H., Wrenn, B.A., Domínguez-Espinosa, R.,

716 2004. Production of bioenergy and biochemicals from industrial and agricultural

717 wastewater. *Trends Biotechnol.* 22, 477–485.

718 <https://doi.org/10.1016/j.tibtech.2004.07.001>

719 Babæ, A., Shayegan, J., 2011. Effect of organic loading rates (OLR) on production of

720 methane from anaerobic digestion of vegetables waste, in: *World Renewable*

721 *Energy Congress-Sweden; 8-13 May; 2011; Linköping; Sweden. Linköping*

722 *University Electronic Press, pp. 411–417.*

723 Bell, E., Lamminmäki, T., Alneberg, J., Qian, C., Xiong, W., Hettich, R.L., Frutschi, M.,

724 Bernier-Latmani, R., 2022. Active anaerobic methane oxidation and sulfur

725 disproportionation in the deep terrestrial subsurface. *ISME J.* 1–11.

726 Chandra, R., Takeuchi, H., Hasegawa, T., 2012. Methane production from

727 lignocellulosic agricultural crop wastes: A review in context to second generation

728 of biofuel production. *Renew. Sustain. Energy Rev.* 16, 1462–1476.

729 <https://doi.org/10.1016/j.rser.2011.11.035>

730 Cremones, P.A., Sampaio, S.C., Teleken, J.G., Meier, T.W., Frigo, E.P., de Rossi, E., da
731 Silva, E., Rosa, D.M., 2020. Effect of substrate concentrations on methane and
732 hydrogen biogas production by anaerobic digestion of a cassava starch-based
733 polymer. *Ind. Crops Prod.* 151. <https://doi.org/10.1016/j.indcrop.2020.112471>

734 Dai, X., Li, X., Zhang, D., Chen, Y., Dai, L., 2016. Simultaneous enhancement of
735 methane production and methane content in biogas from waste activated sludge
736 and perennial ryegrass anaerobic co-digestion: The effects of pH and C/N ratio.
737 *Bioresour. Technol.* 216, 323–330.
738 <https://doi.org/10.1016/j.biortech.2016.05.100>

739 Demirel, B., Yenigün, O., 2002. The effects of change in volatile fatty acid (vfa)
740 composition on methanogenic upflow filter reactor (ufaf) performance. *Environ.*
741 *Technol. (United Kingdom)* 23, 1179–1187.
742 <https://doi.org/10.1080/09593332308618336>

743 Deublein, D., Steinhauser, A., 2011. *Biogas from waste and renewable resources: an*
744 *introduction.* John Wiley & Sons.

745 El Gnaoui, Y., Karouach, F., Bakraoui, M., Barz, M., El Bari, H., 2020. Mesophilic
746 anaerobic digestion of food waste: Effect of thermal pretreatment on improvement
747 of anaerobic digestion process. *Energy Reports* 6, 417–422.

748 Fu, B., Zhang, J., Fan, J., Wang, J., Liu, H., 2012. Control of C/N ratio for butyric acid
749 production from textile wastewater sludge by anaerobic digestion. *Water Sci.*
750 *Technol.* 65, 883–889.

751 Garcia-Peña, E.I., Parameswaran, P., Kang, D.W., Canul-Chan, M., Krajmalnik-Brown,
752 R., 2011. Anaerobic digestion and co-digestion processes of vegetable and fruit
753 residues: Process and microbial ecology. *Bioresour. Technol.* 102, 9447–9455.

754 <https://doi.org/https://doi.org/10.1016/j.biortech.2011.07.068>

755 Gulhane, M., Khardenavis, A.A., Karia, S., Pandit, P., Kanade, G.S., Lokhande, S.,
756 Vaidya, A.N., Purohit, H.J., 2016. Biomethanation of vegetable market waste in an
757 anaerobic baffled reactor: Effect of effluent recirculation and carbon mass balance
758 analysis. *Bioresour. Technol.* 215, 100–109.
759 <https://doi.org/https://doi.org/10.1016/j.biortech.2016.04.039>

760 Jariyaboon, R., O-Thong, S., Kongjan, P., 2015. Bio-hydrogen and bio-methane
761 potentials of skim latex serum in batch thermophilic two-stage anaerobic digestion.
762 *Bioresour. Technol.* 198, 198–206. <https://doi.org/10.1016/j.biortech.2015.09.006>

763 Jawjit, S., 2013. Anaerobic treatment of concentrated latex processing wastewater in
764 two-stage up flow anaerobic sludge blanket 8, 805–813.
765 <https://doi.org/10.1139/L10-029>

766 Kabisch, A., Otto, A., König, S., Becher, D., Albrecht, D., Schüler, M., Teeling, H.,
767 Amann, R.I., Schweder, T., 2014. Functional characterization of polysaccharide
768 utilization loci in the marine Bacteroidetes ‘Gramella forsetii’ KT0803. *ISME J.* 8,
769 1492–1502. <https://doi.org/10.1038/ismej.2014.4>

770 Kim, M.-J., Kim, S.-H., 2020. Conditions of lag-phase reduction during anaerobic
771 digestion of protein for high-efficiency biogas production. *Biomass and Bioenergy*
772 143, 105813.

773 Kim, S., Rhee, S., 2015. *Methanomethylivorans*. *Bergey’s Man. Syst. Archaea Bact.* 1–
774 5.

775 Kongjan, P., Jariyaboon, R., O-Thong, S., 2014. Anaerobic digestion of skim latex
776 serum (SLS) for hydrogen and methane production using a two-stage process in a
777 series of up-flow anaerobic sludge blanket (UASB) reactor. *Int. J. Hydrogen*

778 Energy 39, 19343–19348. <https://doi.org/10.1016/j.ijhydene.2014.06.057>

779 Lim, J.W., Park, T., Tong, Y.W., Yu, Z., 2020. The microbiome driving anaerobic
780 digestion and microbial analysis, in: *Advances in Bioenergy*. Elsevier, pp. 1–61.

781 McHugh, S., O'Reilly, C., Mahony, T., Colleran, E., O'Flaherty, V., 2003. Anaerobic
782 granular sludge bioreactor technology. *Rev. Environ. Sci. Biotechnol.* 2, 225–245.
783 <https://doi.org/10.1023/B:RESB.0000040465.45300.97>

784 Min, H., Hyun, J., Hyub, J., Moon, J., 2014. Bioresource Technology Bacterial and
785 methanogenic archaeal communities during the single-stage anaerobic digestion of
786 high-strength food wastewater. *Bioresour. Technol.*
787 <https://doi.org/10.1016/j.biortech.2014.02.028>

788 Mu, T., Xing, J., Yang, M., 2019. Sulfate reduction by a haloalkaliphilic bench-scale
789 sulfate-reducing bioreactor and its bacterial communities at different depths.
790 *Biochem. Eng. J.* 147, 100–109. <https://doi.org/10.1016/j.bej.2019.04.008>

791 Petriglieri, F., Nierychlo, M., Nielsen, P.H., McIlroy, S.J., 2018. In situ visualisation of
792 the abundant Chloroflexi populations in full-scale anaerobic digesters and the fate
793 of immigrating species. *PLoS One* 13, e0206255.

794 Rajendran, K., Mahapatra, D., Venkatraman, A.V., Muthuswamy, S., Pugazhendhi, A.,
795 2020. Advancing anaerobic digestion through two-stage processes: Current
796 developments and future trends. *Renew. Sustain. Energy Rev.* 123, 109746.
797 <https://doi.org/10.1016/j.rser.2020.109746>

798 Raketh, M., Jariyaboon, R., Kongjan, P., Trably, E., Reungsang, A., Sripitak, B.,
799 Chotisuwan, S., 2021. Sulfate removal using rubber wood ash to enhance biogas
800 production from sulfate-rich wastewater generated from a concentrated latex
801 factory. *Biochem. Eng. J.* 173, 108084. <https://doi.org/10.1016/j.bej.2021.108084>

802 Raketh, M., Kongjan, P., Sani, K., Trably, E., Cheirsilp, B., Jariyaboon, R., 2022.
803 Biodegradation Efficiencies and Economic Feasibility of Single-stage and Two-
804 Stage Anaerobic Digestion of Desulfated Skim Latex Serum (SLS) by Using
805 Rubber Wood Ash. *Process Saf. Environ. Prot.*

806 Rattanaya, T., Manmeen, A., Kongjan, P., Bunyakan, C., Reungsang, A., Prasertsit, K.,
807 Lombardi, L., Jariyaboon, R., 2021. Upgrading biogas to biomethane using
808 untreated groundwater-NaOH absorbent: Pilot-scale experiment and scale-up
809 estimation for a palm oil mill. *J. Water Process Eng.* 44, 102405.

810 Ryckebosch, E., Drouillon, M., Vervaeren, H., 2011. Techniques for transformation of
811 biogas to biomethane. *Biomass and Bioenergy* 35, 1633–1645.
812 <https://doi.org/https://doi.org/10.1016/j.biombioe.2011.02.033>

813 Saritpongteeraka, K., Chaiprapat, S., 2008. Effects of pH adjustment by parawood ash
814 and effluent recycle ratio on the performance of anaerobic baffled reactors treating
815 high sulfate wastewater. *Bioresour. Technol.* 99, 8987–8994.
816 <https://doi.org/10.1016/j.biortech.2008.05.012>

817 Whang, L.-M., Hu, T.-H., Liu, P.-W.G., Hung, Y.-C., Fukushima, T., Wu, Y.-J., Chang,
818 S.-H., 2015. Molecular analysis of methanogens involved in methanogenic
819 degradation of tetramethylammonium hydroxide in full-scale bioreactors. *Appl.*
820 *Microbiol. Biotechnol.* 99, 1485–1497.

821 Wikandari, R., Millati, R., Taherzadeh, M.J., Niklasson, C., 2018. Effect of effluent
822 recirculation on biogas production using two-stage anaerobic digestion of citrus
823 waste. *Molecules* 23, 3380.

824 Wilawan, W., Pholchan, P., Aggarangsi, P., 2014. Biogas production from co-digestion
825 of *Pennisetum purpurem* cv. Pakchong 1 grass and layer chicken manure using

826 completely stirred tank. *Energy Procedia* 52, 216–222.

827 Xiao, K.-Q., Beulig, F., Røy, H., Jørgensen, B.B., Risgaard-Petersen, N., 2018.

828 Methylophilic methanogenesis fuels cryptic methane cycling in marine surface

829 sediment. *Limnol. Oceanogr.* 63, 1519–1527.

830 [https://doi.org/https://doi.org/10.1002/lno.10788](https://doi.org/10.1002/lno.10788)

831 Yang, S., Xue, W., Liu, P., Lu, X., Wu, X., Sun, L., Zan, F., 2022. Revealing the

832 methanogenic pathways for anaerobic digestion of key components in food waste:

833 Performance, microbial community, and implications. *Bioresour. Technol.* 347,

834 126340.

835 Yen, H.-W., Brune, D.E., 2007. Anaerobic co-digestion of algal sludge and waste paper

836 to produce methane. *Bioresour. Technol.* 98, 130–134.

837 Reungsang, A., 2019. *Thailand: Microbial-Derived Biofuels and Biochemicals*, first

838 ed., Khon Kaen University, Thailand.

839 APHA., 2012, *Standard methods for the examination of water and wastewater*, 22th

840 ed. Washington DC. USA.

841 Rubber Intelligence Unit, 2020, *Rubber in Thailand*. <http://rubber.oie.go.th/> (accessed

842 25.4.20).

843

844

845

846

847

848

849

850

851

852

853

854

855

856

857
858
859
860
861
862
863
864
865
866
867
868
869
870
871
872
873
874
875
876
877
878
879
880
881
882
883
884
885
886
887
888
889
890
891
892
893
894
895
896
897
898
899
900
901
902
903
904

Supplementary material for

Effect of Organic Loading Rate and Effluent Recirculation on Biogas Production of Desulfated Skim Latex Serum using Up-Flow Anaerobic Sludge Blanket Reactor

Marisa Raketh^{1,3}, Prawit Kongjan^{2,3}, Eric Trably⁴, Nurta Samahae⁵, Rattana Jariyaboon^{2,3,*}

¹ Energy Technology Program, Faculty of Engineering, Prince of Songkla University, Hat Yai, Songkhla, 90112, Thailand

² Department of Science, Faculty of Science and Technology, Prince of Songkla University (PSU), Pattani, 94000, Thailand

³ Bio-Mass Conversion to Energy and Chemicals (Bio-MEC) Research Unit, Faculty of Science and Technology, Prince of Songkla University (PSU), Pattani, 94000, Thailand

⁴ INRAE, Univ Montpellier, LBE, Narbonne, France

⁵ Science Program in Chemistry-Biology, Faculty of Science and Technology, Prince of Songkla University (PSU), Pattani, 94000, Thailand

*corresponding author at: Department of Science, Faculty of Science and Technology, Prince of Songkla University (PSU), Pattani 94000, Thailand. Tel.: +66 73 313928-50 ext1988, mobile: +66 808721260.

E-mail address: rattana.sa@psu.ac.th (R. Jariyaboon).

905
906
907

Table S1 COD distribution in the effluent

| | Phase I | Phase II | Phase III | Phase IV | Phase V |
|--|----------------|-----------------|------------------|-----------------|----------------|
| OLR (g-COD/L-reactor·d) | 0.89 | 1.79 | 3.57 | 1.79 | 2.01 |
| VFA in effluent (g/L) | | | | | |
| Acetic | 0.60 | 0.82 | 1.54 | 0.90 | 1.07 |
| Propionic | 0.60 | 0.83 | 1.48 | 1.03 | 1.65 |
| Butyric | 0.39 | 0.53 | 0.71 | 0.58 | 0.61 |
| VFA in effluent (g-COD/L) | | | | | |
| Acetic | 0.64 | 0.88 | 1.64 | 0.96 | 1.14 |
| Propionic | 0.91 | 1.26 | 2.23 | 1.56 | 2.49 |
| Butyric | 0.70 | 0.96 | 1.29 | 1.06 | 1.10 |
| Total VFA (g-COD/L) | 2.25 | 3.10 | 5.17 | 3.58 | 4.73 |
| Total COD in effluent (g-COD/L) | 2.60 | 4.65 | 9.43 | 4.99 | 5.37 |
| Other organic matter (g-COD/L) | 0.35 | 1.55 | 4.26 | 1.42 | 0.63 |
| % Contribution in Effluent | | | | | |
| Total VFA | 86.57 | 66.68 | 54.80 | 71.65 | 88.24 |
| Other organic matter | 13.43 | 33.32 | 45.20 | 28.35 | 11.76 |

908
909
910