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1	Enzyme synergy for plant cell-wall polysaccharide degradation
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11	
12	Abstract
13	Valorising plant cell-wall, marine and algal polysaccharides is of utmost importance for the
14	development of the circular bioeconomy. This is because polysaccharides are by far the most
15	abundant organic molecules found in nature with complex chemical structures that requires a
16	large set of enzymes for their degradation. Microorganisms produce polysaccharide-specific
17	enzymes that act in synergy when performing hydrolysis. Although discovered since decades
18	enzyme synergy is still poorly understood at the molecular level and thus is difficult to harness
19	and optimize. In the last few years, more attention has been given to improve and characterize
20	enzyme synergy for polysaccharide valorisation. In this review we summarize literature to
21	provide an overview of the different type of synergy involving carbohydrate modifying enzymes
22	and the recent advances in the field exemplified by plant cell-wall degradation.
23	

24 Introduction

25 To reduce our carbon footprint and develop a more sustainable economy, it is crucial and urgent to move toward a circular bioeconomy model meaning using biotechnological tools to process 26 biological residues into products, which can be reused, recycled, or released safely to the 27 environment [1]. This implies tackling the deconstruction of chemically and structurally complex 28 molecules that are naturally resistant to degradation such as lignocellulosic biomass that compose 29 plant cell-walls [2]. Lignocellulosic biomass is mainly composed of cellulose, the most abundant 30 polysaccharide on earth, a β -1,4 linked D-glucose molecule, generating a crystalline structure; 31 32 hemicellulose, mainly xylans in cereals, which backbone sugar is composed of β -1,4-linked Dxylose units that is decorated with a variety of sugars and acetyl groups; and lignin, an aromatic 33 34 polymer [3].

Lignocellulolytic microorganisms produce an arsenal of enzymes that efficiently hydrolyse in a 35 concerted manner the different plant cell-wall polysaccharides to access the carbon source [4]. All 36 37 these enzymes cooperate to degrade or modify the complex polysaccharides by generating a 38 product that becomes the substrate of the next enzyme. When the overall activity of the concerted enzyme action is greater than the sum of individual enzyme activities, the co-operation between 39 enzymes is described as synergistic [5]. Enzyme synergy has been demonstrated not only for plant 40 cell-wall polysaccharide including cellulose, hemicelluloses, pectins, but also for complex marine 41 42 polysaccharides such as alginate, or chitin [6,7]. It is employed in advanced biorefineries to optimize enzyme cocktails that efficiently hydrolyse plant polysaccharides, thus producing their 43 component monomers [8,9]. Synergy is often described or sought for, but rarely explained at the 44 45 molecular level although its understanding will improve the efficiency of industrial biomass valorization for sustainable fuels and chemical products. In this review, we provide an 46 47 overview/update of enzyme synergies, with a focus on recent developments that provide a better

understanding of how enzymes catalyse, in space and time, plant cell-wall polysaccharidedeconstruction.

50

51 Plant cell wall enzymatic deconstruction

Lignocellulolytic microorganisms have evolved different enzyme systems or strategies to produce 52 a large set of enzymes enabling them to use plant cell-wall polysaccharides as a carbon source. 53 Aerobic microorganisms, of which fungi are seen as the most active lignocellulosic biomass 54 degraders, secrete in the extra-cellular environment a large set of enzymes [8]. Some anaerobic 55 bacteria, in particular Clostridia, produce extracellular multi-enzyme complex anchored to the cell 56 57 surface called the cellulosome [10]. The cellulosome is composed of a set of scaffolding grafting enzymes with different substrate specificities, making this complex highly efficient on 58 lignocellulose. This efficiency was attributed to a high synergy among enzymes enhanced by their 59 proximity to one another termed intermolecular synergy [11,12]. Bacteroidetes is one of the main 60 bacterial phylum in soil, ruminant and human microbiota that encodes fine-tuned gene clusters 61 dedicated to polysaccharide metabolism called Polysaccharide Utilization Loci (PULs) [13]. Each 62 PUL contains genes that encode enzymes which orchestrate the detection, sequestration, 63 enzymatic digestion, and transport of dedicated complex polysaccharides [14]. This organization 64 65 allows efficient internalization of the oligosaccharide rapidly after a first degradation, and their subsequent hydrolysis in the periplasm by diverse enzymes working in synergy [15,16]. 66

The enzymes involved in hydrolysis are glycoside hydrolases (GH), carbohydrate esterases (CE) polysaccharide lyase (PL) and lytic polysaccharide monooxygenases (LPMOs). All these enzymes are classified in sequence-based families in the CAZy database (<u>http://www.cazy.org</u>) [17]. Many of them are multi-modular, displaying one or several catalytic modules linked to one or more carbohydrate binding modules (CBM) [18]. The main role of the latter is to address the enzyme onto its substrate and contributes to intramolecular co-operation although important in the enzyme

activity they will not be discussed here [11,19-21]. Cellulases are glycoside hydrolases that 73 74 degrade cellulose which include endo- β -1,4-glucanases, cellobiohydrolases (CBHs) (or exo- β -1,4glucanases) and β -glucosidases [22]. These enzymes are classified in different CAZy families. It 75 is now also well known that LPMOs, oxidative enzymes, classified as Auxiliary Activities (AA) 76 77 in CAZy are key players in crystalline cellulose or chitin degradation. They oxidize the C1 or C4 carbon of β -1,4-linked polysaccharides [23]. Hemicellulose is a chemically complex group of 78 polysaccharides composed of different monomeric units with different type of linkages that 79 requires the action of endoglycanases for polysaccharide backbone degradation and 80 81 exoglycosidases that hydrolyze decoration linked to the backbone [24]. Xylans, the main hemicellulose in cereals and grasses, are composed of β -D-1,4-xylose in the backbone and 82 generally decorated with α -1,2- and/or α -1,3-linked L-arabinosyl residues, α -1,2-linked D-83 glucuronosyl and 4-O-methyl-D-glucuronosyl groups and are frequently modified by acetyl groups 84 85 that are linked to the main chain D-xylosyl moieties [25]. Therefore, xylan acting enzymes are a group of β -1,4-xylanases that attack xylan backbone at random positions, β -1,4-xylosidases to 86 hydrolyze xylobiose, the product of xylanases, into xylose and α -1,2 and/or α -1,3-L-87 arabinofuranosidases, carbohydrate esterases, and α -D-glucuronisades, that hydrolyze substituents. 88 All enzymes are classified in different GH and CE enzyme families with different specificities 89 90 (Fig. 1) [26].

92 Enzyme Synergy

93 During decades, cellulose has been the center of attention in biomass conversion strategies because of its deconstruction into glucose as exemplified by 70 years of research on Trichoderma reesei 94 enzymes that showed the first examples of enzyme synergy [27]. Hemicellulases were considered 95 as accessory enzymes and were shown later to be important key players not only to valorize 96 pentoses but also to maximize cellulose degradation [28] leading to cellulase-hemicellulase 97 synergism. In order to develop efficient enzymatic cocktail for biorefinery, enzyme synergies have 98 been characterized not only between enzymes from the same organism [29,30] but also between 99 100 enzymes encoded by different microorganisms [31]. Enzyme synergy can be evaluated by 101 calculating the degree of synergy (DS) which is the ratio of the activity (or product released) of an 102 enzyme mixture to the sum of their individual activities (or product released). Synergy is observed when the DS is greater than one. Two types of synergy have been defined: homeosynergy, when 103 104 the enzymes hydrolyze the same part of the polysaccharide, i.e., both enzymes of interest hydrolyze either the main chain or the side chain of the polysaccharide, and heterosynergy when 105 enzymes hydrolyze different parts of the polysaccharide, *i.e.*, the main chain and the ramifications 106 (Fig. 2). Examples of enzyme homeosynergy and/or heterosynergy have been described for 107 108 different polysaccharides such as cellulose [32], hemicelluloses [15], pectins [33], alginate [6], or 109 chitin [34]. In this review we will further detail enzyme synergies dedicated to lignocellulose degradation. 110

111

112 Homeosynergy

113 Cellulose is a homopolymer therefore all cellulases fit into the group of homeosynergy. The 114 cellulase synergy is named after the enzyme mode of action *i.e.* endo-exo synergism (when 115 synergy occurs between endoglucanase and cellobiohydrolase), exo-exo synergism (between reducing and non-reducing end cellobiohydrolases) and, endo- β -glucosidase synergism (between

117 endoglucanase or CBH and β -glucosidase).

118 Endo- Exo-cellulases synergy

Since, Reese et al., first documented synergism in 1950 [35] a model of endo-exo synergy was 119 120 proposed [36,37]. Exo-acting cellobiohydrolases (CBH) start the hydrolysis from cellulose chain ends in a processive manner whereas endo-acting cellulases hydrolyze cellulose at random 121 positions and create new ends from which the exo-acting cellobiohydrolases (CBH) can release 122 cellobiose from either the reducing (GH7 and GH48) or nonreducing (GH6) ends. This model was 123 completed by Kostylev et al., (2014) which results suggested that the endocellulase from 124 125 Thermobifida fusca TfCel9A was more active on a uniform (undamaged) cellulose surface and generated eroded surface while the exocellulase TfCel48A would preferentially hydrolyze the 126 more accessible substrate and replenish the uniform surface required by the endocellulase [38]. 127 128 This finding illustrates that the substrate site generation is still a question and that it is not yet clear how the free dispersed enzyme behaves on a solid substrate. Very recently, thanks to atomic force 129 microscopy (AFM), the synergistic endo-exo synergy was observed at single-molecule resolution. 130 Zajki-Zechmeister and colleagues showed that only when Trichoderma reesei TrCel7A 131 132 exocellulase and TrCel7B endocellulase hydrolyze cellulose fibril, the dispersed enzymes move 133 to the same direction of the cellulose fibril and cluster on cellulose breaches. This phenomenon produces a molecular proximity between enzymes and an efficient multilayer-processive mode of 134 135 degradation occurs [39]. The cooperativity induces the exocellulase to move 100-fold faster than 136 when acting alone, leading to efficient cellulose hydrolysis. In this case In fine free enzymes on the substrate mimic the spatial confinement observed in cellulosome assemblies. 137

This observation is somehow similar to what high speed AFM showed earlier in exo-exo synergy with *Tr*Cel7A and *Tr*Cel6A exocellulases [40]. Although enzymes hydrolyzed the substrate from reducing- and non-reducing end respectively, when one molecule was stopped, enzymes were clustered in "traffic jams" meaning that the enzymes were productively bound to the cellulose surface and peeled of the crystalline cellulose [40]. The hypothesis was that the shorter active site of *Tr*Cel6A could open and generate nicks on the cellulose substrate like an endocellulase and that these nicks could be the starting and end point of the more processive *Tr*Cel7A like in endo-exo synergy. Recent characterization of cellulase demonstrated the important role of the substrate binding strength for interfacial enzymes irrespective to the enzyme structure or GH family [41]. In the case of synergy, the trade-off between binding and activity has also to be considered.

148

endo- β-glycosidase synergy: Synergy between Cellulases and β-glucosidase or Xylanase and β xylosidase

The full cellulose and/or hemicellulose saccharification of lignocellulosic biomass into 151 fermentable sugars requires the action of β -glycosidases, which hydrolyse oligosaccharides into 152 monosaccharides. In the case of cellulose, β -glucosidase hydrolyses cellobiose, the product of 153 CBHs, exocellulases, into glucose units. CBH and β-glucosidase act in synergy to maximize the 154 hydrolysis rate. Indeed, CBHs can be inhibited by their product, cellobiose, which causes a lower 155 conversion rate of cellulose polymers into cellobiose. β-Glucosidases produce β-D-glucose from 156 cellobiose and this monosaccharide is a lower inhibitor on CHBs compared to cellobiose [42-44]. 157 Therefore, to improve cellulose saccharification and increase synergy with cellulases, several 158 studies focused on the engineering of β -glucosidases to make the enzymes more active and/or more 159 tolerant to glucose [45–47]. Lee et al., (2012) engineered a β-glucosidase (TrBgl2, GH1) using 160 161 rational design and identified L167W mutation showing better enzyme activity and displayed an 162 improved synergism with cellulases compared to wild-type [48]. This mutation is located at the entrance of the active site and increases the affinity (K_M) of the enzyme to cellobiose. It is 163 suggested that this mutation offers a better coordination of the substrate in the active site which 164 165 leads to improve affinity and enzymatic efficiency. Similarly, the Trichoderma harzianum ThBgl

β-glucosidase from GH1 was engineered and the resulted double mutant L167W/P172L narrowed 166 the entrance of the active-site and prevented product inhibition [45]. Mixed with cellulases, this 167 mutant enhanced significantly biomass conversion to glucose. By mixing different enzymes, 168 synergy experiments are not always performed under optimal conditions for each enzyme therefore 169 improving stability or activity in defined conditions can improve enzyme synergy. Cao et al., 170 discovered high glucose tolerant β -glucosidases (Bgl6 and Bgl15) and engineered these enzymes 171 172 for improved stability. The resulting engineered Bgl6 β-glucosidase mutant displays better stability 173 and activity while maintaining a high glucose tolerance and significantly improves sugar cane bagasse hydrolysis when added to commercial cellulases [49]. Mutations at position 167 in Bgl15, 174 175 close to the active site, showed that increasing hydrophobicity of this region by introducing a 176 valine or alanine improved the catalytic performance of the enzymes and their glucose tolerance [47]. Altogether, results indicated that improving activity, glucose tolerance and stability of β -177 glucosidases boosted synergy with cellulases and maximized the total saccharification of cellulose 178 polymers. 179

 β -Xylosidases are a group of enzymes that removes xylose units from xylo-oligosaccharides, and these enzymes are mainly found in GH3, 39,43 and 52 families and act in synergy with endoxylanases [50,51]. Indeed, xylanases produce xylo-oligosaccharides (XOS), mainly xylobiose and xylotriose, and studies have shown that adding β-xylosidase clearly improved the activity of endoxylanases [52–54]. β-Xylosidase, usually high xylose tolerant enzymes, hydrolyses XOS into xylose units and decreases the inhibition effect of XOS on xylanase activity [51]. This synergy leads to acceleration of the deconstruction of hemicellulose polymers [55].

187 Synergy between oxidative LPMOs and cellulases or xylanases

188 LPMOs are cooper-containing enzymes classified in CAZy database into auxiliary activities 189 families (AA9, 10, 11, 13, 14, 15, 16 and 17) [56]. They oxidatively cleave the β -1,4-glycosidic 190 linkage of polysaccharides. Several studies show that LPMOs and cellulases synergistically depolymerize crystalline cellulose and it was reported that this oxidative-hydrolytic synergism improves the saccharification yield of recalcitrant biomass [32,57,58]. For instance, the exoglucanase *Tr*Cel6A and the LPMO *Ta*AA9A (*Thermoascus aurantiacus*) maximized the degradation of Avicel, and bacterial cellulose, with a synergy index equal to 2 and 2.5, respectively [32].

A large part of the described LPMOs acts on crystalline cellulose and cause local disruption of the 196 ordered cellulose structure [57,59]. LPMOs are adsorbed into crystalline cellulose and induce 197 198 fibrillation by impacting the cellulose fiber architecture. Then LPMOs catalyze the oxidation of glucose units which leads to creation of new reducing and non-reducing oxidized termini [60–62]. 199 In such manner, LPMOs generate new binding and acting sites for processive cellulases. A single-200 201 molecule study by AFM demonstrated the positive effect of LPMOs activities on TrCel7A 202 exocellulase [63]. The treatment with LPMOs increased the fibrillation of crystalline cellulose which boosted CBH activity and the release of soluble products. In addition, LPMOs also 203 204 increased the dynamics of cellulose-cellulase interactions. The adsorption – desorption events of cellulases is improved and their mobility on cellulose surface is enhanced in the presence of 205 LPMOs. This finding demonstrates the positive effects of LPMOs on cellulases adsorption and 206 activity [63]. 207

In order to improve LPMOs – cellulases synergy, the AA10 LPMO form *Hahella chejuensis* (*Hc*LPMO10) was engineered and one mutation improved catalytic activity due to an improved binding capacity of the protein toward cellulosic polymers [64]. Combined with cellulases, this mutant boosted the hydrolysis level of microcrystalline cellulose. More recently, Srivastava et al., focused on the ability of multimodular LPMO (AA9) to bind recalcitrant biomass and the synergy between LPMO and cellulases. Authors demonstrate the positive effects of LPMO-cellulose binding on LPMO-cellulase synergy [65]. Altogether, these results suggest the crucial role of

LPMOs on the deconstruction of cellulose and improving their adsorption leads to increase theiractivities and their synergy with cellulases during cellulose hydrolysis.

In 2014, Agger et al, reported a new LPMO (NcLPMO9C) form Neurospora crassa acting on 217 xyloglucan and glucomannan, two hemicellulosic compounds [66,67]. This finding opened the 218 door to the study of LPMOs acting on hemicellulose and their potential synergism with 219 hemicellulases during the degradation of hemicellulosic compounds [68]. Recently, a LPMO 220 221 acting on xylan (PcAA14B from Pycnoporus coccineus) was tested with a xylanase GH11-M4 (GH11) from Aspergillus niger [68] and significantly increased the saccharification level releasing 222 high levels of xylobiose and xylotriose from beechwood cellulosic fibers. Understanding the 223 224 synergic interaction at the molecular level between LPMOs and their interplay with other hemicellulases is still rare. In the same publication, PcAA14B was assayed with the 225 xylobiohydrolase TtXyn30A from Thermothelomyces thermophilaa acting on xylan polymers. 226 227 Despite low activity on different pretreated biomasses, PcAA14B and TtXyn30A synergy index reached up to 5.70 on H₂O/acetone pretreated beechwood substrates suggesting that PcAA14B 228 degrades the recalcitrant hemicellulose chain adsorbed on crystalline cellulose and constantly 229 creates substrates sites available for TtXyn30A. The degree of synergism is inversely proportional 230 231 to the recalcitrant hemicellulose content, probably due to less accessibility of PcAA14B to 232 adsorbed xylan polymers.

Overall LPMOs act in synergy with cellulases and hemicellulases and this can be further boosted by laccases that release phenolic molecules able to donate electrons to LPMOs [70]. However, the structural determinants, physical and kinetics parameters involved in hemicellulosic LPMOs synergy remain unexplored.

237 Heterosynergy

238 Synergy between Cellulases and Hemicellulases

The use of hemicellulases with cellulases has been shown to improve the hydrolysis of 239 240 lignocellulosic biomass, because they remove hemicellulose polymers that interact with cellulose making cellulose microfibers more accessible to cellulase attack while hydrolyzing xylo-241 oligosaccharides shown to inhibit cellulases [28]. Synergy between TrCel7A and xylanases or 242 mannanases, suppressed the inhibitory effect of soluble and adsorbed hemicellulosic compounds 243 on cellulases activity [71]. Cellulases-hemicellulases synergism was observed during the 244 245 hydrolysis of several lignocellulosic biomass such as hardwood and softwood, corn stover and wheat straw [72–74]. Nevertheless, resulting global hydrolysis can be controversial as it was 246 observed with cellobiohydrolyses, xylanases and α -L-arabinofuranosidases. Although xylanase 247 248 and α-L-arabinofuranosidase together enhanced the hydrolysis of hemicellulose their unsubstituted 249 oligosaccharide products inhibit the activity of cellobiohydrolyses [63]. It is suggested that arabinose substituents on xylooligosaccharides in absence of arabinofuranosidase could prevent 250 251 their accommodation hemicellulosic compounds in the cellobiohydrolase active site thus displaying no inhibition. 252

253

254 Synergy between xylanases and accessory enzymes

255 The complete degradation of xylan requires several enzyme activities (Fig. 1), the main ones being 256 xylanases and xylosidases, as described above. However, accessory enzymes are important players 257 to liberate substitutions and make the polymer more accessible to depolymerizing enzymes. α -L-Arabinofuranosidase (Abf) is a debranching class of enzymes that removes arabinose substitutions 258 259 from arabinoxylan backbone and enhances the hydrolysis of this polymer by synergetic action with xylanases, they are classified in GH 3, 5, 43, 51, 54, 62 and 159 [64-66]. Xylanases, in GH10 and 260 GH11 families, often display a low tolerance to substitutions [79]. Thus, removal of arabinose 261 substitutions leads to an optimal activity on arabinoxylan and synergy between xylanases and α -262 L-arabinofuranosidases [80]. Synergy index of 1.30, 1.39 and 1.24 were reported when XynA1 263

264 (GH10 endoxylanase) was tested with AbfA (GH51 arabinofuranosidase) on oat spelt xylan,
265 Birchwood xylan and Beechwood xylan, respectively [52].

α-D-Glucuronidases remove glucuronic acids substitutions from glucuronoarabinoxylan and make 266 the polymer more accessible to depolymerizing enzymes [81]. Indeed, the large part of 267 characterized a-glucuronidases belongs to GH67 and removes glucuronic acids from short 268 xylooligosaccharides [82]. More recently a new family of α-glucuronidases was discovered 269 270 (GH115) and biochemical characterization indicates that these enzymes act on long chain of arabinoglucoronoxylan [83–85]. This family was tested in mixture with the endoxylanase Xyn10C 271 from Clostridium thermocellum during the hydrolysis of softwood arabinoglucoronoxylan and it 272 273 was shown that GH115 has a positive effect on xylanase activity, the amount of oligosaccharides 274 released by the xylanase increases dramatically thanks to GH115 [81].

Xylan acetylations play a crucial role in the interaction between cellulose and hemicellulose 275 276 compounds and it modulates the adsorption of xylan on cellulose surface [86]. Acetylation of xylan inhibits the activity of endoxylanases. It alters the hydrophobicity of xylans and impacts the 277 binding efficiency of hydrolytic enzymes [87–89]. Carbohydrate esterases are a class of enzymes 278 that remove acetyl groups from xylan and make this polymer more accessible to depolymerizing 279 xylanases [90]. These enzymes enhance the activity of xylanases and promote the hydrolysis of 280 281 xylan polymers [91]. All these observations highlight the relationship between enzyme specificity and synergy. Feruloyl esterases, part of CE1 family, are important key players by removing ferulic 282 acid and many examples of synergy were evidenced with xylanases and cellulases [92,93]. 283

Synergy between debranching enzymes is less explored and studied compared to synergy 284 involving main chain acting enzymes. You et al., investigated the synergy between 285 arabinofuranosidases and glucuronidases and confirmed the positive effect of 286 arabinofuranosidases on the activity of one glucuronidase. This effect is probably due to the 287 architecture of the active site of the glucuronidase which is, in some cases, not able to 288

accommodate substituted arabinoglucuronoxylan [94]. Recently, the same positive effect was
observed between glucuronidase and carbohydate esterase. The carbohydrate esterase boosts the
activity of glucuronidase by more than 50% on beechwood glucuronoxylan [95].

292

Overall, facing the complexity of plant cell-wall and, to a wider extent, to polysaccharides from 293 various origins nature has evolved efficient enzymes and systems for their degradation. 294 Nethertheless, the complete deconstruction of polysaccharides is still not mastered and 295 understanding how enzymes operate in synergy to degrade such molecules is highly important. 296 Within the last few years, many PULs have been characterized thus unraveling complete 297 298 enzymatic cascades where synergy and enzyme co-operation is important [96] constituting an interesting source to find new enzyme synergies [16]. More recently AFM experiments showed 299 that enzymes tackle the cellulose by forming enzyme clusters, where enzyme proximity enhances 300 301 synergy and seems to be a key stone for the degradation. In this respect developing designer cellulosomes [97] controlling spatial organization [98] or characterizing multimodular enzymes 302 303 that display high levels of synergy [99,100] could lead to a better understanding of enzyme synergy. 304

305

307 Concluding remarks

Plant Polysaccharide hydrolysis requires enzyme synergy for efficient valorization. Although
 synergy has been widely described, the molecular determinants that influence enzyme synergy
 are not yet mastered.

Applying visualization technics to follow enzyme synergy on insoluble substrates and or in
conditions that mimic the substrate complexity and concentration could be highly relevant.
Furthermore, exploring systems to engineer enzyme spatial organization appears to be an
interesting approach.

Enzyme synergy does not always exist between two enzymes and is difficult to predict. However,
enzyme engineering can allow increasing synergy when molecular determinants are known. This
is why the combination of structural and functional characterization of enzymes with approaches
that consider the polysaccharide complexity, will enable progress in this field.

319

320

321 Abbreviations

GH, Glycoside Hydrolase; CE, Carbohydrate Esterase; PL, Polysaccharide Lyase; LPMO, Lytic
Polysaccharide Monooxygenases; CAZy database, Carbohydrate-Active Enzymes database;
CBM, Carbohydrate-Binding Module; CBH, cellobiohydrolase; AA, Auxiliary Activities; DS,
Degree of Synergy; PUL, Polysaccharide Utilization Locus; AFM, Atomic Force Microscopy.

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655 Figure Caption

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- Figure 1: Schematic representation of enzyme activities degrading cellulose and xylan. On the
- left, enzymes acting on cellulose polymers and β -glucosidase acting on cellobiose. On the right,
- schematic representation of xylan polymers with several ramifications, enzymes acting on xylan and β -xylosidase hydrolyzing a xylobiose.
- and p-xylosidase hydrolyzing a xyl

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- Figure 2: Synergic interactions of cellulosic and xylanolytic enzymes. Green boxes indicate
- 664 cellulosic enzymes and pink boxes indicate xylanolytic enzymes. Only GH, AA and CE families
- discussed in the review are mentioned. GH : Glycoside Hydrolases, AA : Auxiliary Activities,
- 666 CE: Carbohydrate esterases

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669 Figure 1







