

# Identification of sulfonamide compounds active on the insect nervous system: Molecular modeling, synthesis and biological evaluation

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## 1 Identification of sulfonamide compounds as modulators of insect Nicotinic

#### 2 Acetylcholine receptors: molecular modeling, synthesis and biological

- 3 evaluation
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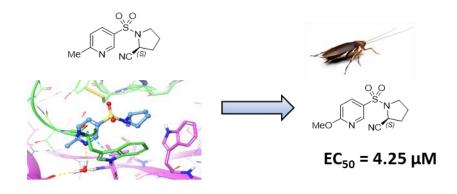
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*KEYWORDS.* Insect nicotinic acetylcholine receptor, molecular docking, sulfonamide
 compounds, electrophysiological studies, cockroach, honeybee.

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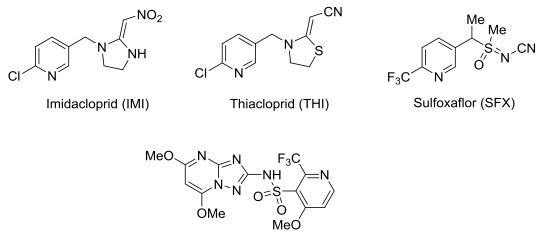
ABSTRACT. Insect nicotinic acetylcholine receptors (nAChRs) are a recognized target for 16 insecticide design. In this work, we have identified, from a structure-based approach using 17 molecular modeling tools, ligands with potential selective activity for pests versus pollinators. 18 A high-throughput virtual screening with the Openeve software was performed using a library 19 20 from the ZINC database, thiacloprid being used as the target structure. The top sixteen molecules were then docked in  $\alpha 6$  cockroach and honeybee homomeric nAChRs to check 21 from a theoretical point of view relevant descriptors in favor of pest selectivity. Among the 22 selected molecules, one original sulfonamide compound has afterward been synthesized, 23 together with various analogs. Two compounds of this family have been shown to behave as 24 activators of the cockroach cholinergic synaptic transmission. 25

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Insecticides agrochemistry is faced with the development of resistance of insect pests to 29 chemical control agents<sup>1</sup> but also with environmental concerns<sup>2</sup> such as their impacts on 30 ecologically and economically important arthropods and pollinators. Specific ligands of the 31 nicotinic acetylcholine receptors (nAChRs), one of the major neuroactive insecticide targets, 32 particularly neonicotinoids, have been identified to have harmful effects towards pollinators 33 associated with consequences on biodiversity.<sup>3</sup> If neonicotinoids have led the global 34 insecticide sales during several years, some of them such as the bestseller imidacloprid 35 (Figure 1) have been submitted to restrictions in Europe.<sup>4</sup> For sulfoxaflor, one recent 36 pesticide from the sulfoximine class (Insecticide Resistance Action Committee - Group 4C) 37 (Figure 1), despite distinct chemical features with respect to neonicotinoids, the question of 38 harmlessness in terms of environmental concerns remains a controversial subject.<sup>5</sup> In this 39 framework, alternative tools for pest management need to be found. In the case of nAChR 40 competitive modulators, several studies have been carried out in the last years to design new, 41 efficient and harmless neonicotinoids compounds<sup>6-9</sup> as well as original molecules exhibiting 42 specific synergistic activity with conventional insecticides.<sup>10,11</sup> 43



Pyroxsulam

44



#### Figure 1. Structure of some pesticides on the market

The work described herein relates to a series of original sulfonamide compounds 46 discovered from a structure-based design approach and evaluated as potential pesticides. 47 Sulfur compounds with a sulfonamide (SO<sub>2</sub>-NH-) function have been extensively developed 48 as pharmaceutical<sup>12</sup> or agricultural agents.<sup>13</sup> As therapeutic agents, sulfonamide compounds 49 were shown to possess antiinflammatory, antibacterial activities<sup>14</sup> antiviral or anti-cancer 50 properties <sup>15</sup> and were also identified as potential enzyme inhibitors such as proteases<sup>16</sup> or 51 metalloenzymes inhibitors.<sup>17</sup> Some of them have also been shown to act at the central nervous 52 system targeting neuronal enzymes<sup>18</sup> or receptors such as nAChRs.<sup>19</sup> In agrochemistry, 53 sulfonamide herbicides have emerged in the 90<sup>th</sup> and research progress has led to the 54 identification of active compounds such as any sulfonamides<sup>20</sup> or heteroaromatic 55 sulfonamides among which the triazolopyrimidine pyroxsulam (Quasar<sup>TM</sup>, Figure 1), one 56 representative commercial example.<sup>21-23</sup> If sulfur compounds bearing a sulfonimide function 57 such as sulfoxaflor (Figure 1, SFX) have also recently been developed as novel and efficient 58 pesticides, sulfonamide compounds with insecticidal activity still remain guite rare.<sup>24, 25</sup> 59 Thus, to the best of our knowledge, the novel compounds designed and synthesized here, 60

exhibiting a promising insecticide activity, have never been claimed in the literature for thisapplication.

We started the study with a high-throughput virtual screening carried out with the Openeve 63 64 software using thiacloprid, a representative of the neonicotinoid family (Figure 1, THI), as the template in the shape screening mode. The compounds emerged from this analysis were then 65 docked in the 3D models of the ligand binding domain (LBD) of cockroach (Periplaneta 66 Americana) and honeybee (Apis mellifera) of the  $\alpha 6$  homometric nAChRs, set up in our 67 previous study.<sup>26</sup> Indeed, Hawkins and collaborators have recently reported a functional 68 expression of  $\alpha 6$  homomeric insect nAChR with an EC<sub>50</sub> value of 0.88  $\mu$ M for 69 acetylcholine.<sup>27</sup> Moreover, Lu and collaborators newly studied the susceptibility of 70 Drosophila melanogaster nAChR subunit mutants to eleven known insecticides, concluding 71 that the  $\alpha 6$  mutant was sensitive to neonicotinoids such as thiacloprid or sulfoxaflor and that 72 the spynosins family could only target the  $\alpha 6$  homomeric channel.<sup>28</sup> 73

The virtual screening procedure using the lead-like subset of the ZINC database, led to 16 74 75 compounds with a Tanimoto index superior to 0.6 compared to THI. Each of these compounds were carefully examined before the docking stage. The chemical structure of most 76 of these compounds (except two molecules) have been modified to fulfill several criteria. 77 78 First, if the compounds exist under several protonation states, only the neutral form was retained for further analysis. Furthermore, when the ZINC compounds bear aromatic rings, 79 heteroatoms have been introduced in relevant positions to increase the potential of specific 80 molecular interactions (for example an aromatic benzene ring was changed into a pyridine 81 ring). Lastly, aliphatic substituents (Me, *i*-Pr) carried by heterocyclic rings have generally 82 been removed. 83

84	For the molecular docking, the ligands were prioritized according to (i) their protonation
85	state (ii) the docking scores (iii) the Glide interactions energies. In any case, the compounds
86	were docked in their neutral state, on the basis on the recognized pharmacophore of insect
87	nAChR competitive modulators.

Table 1 presents the docking scores (DS) and Glide energies (GE) obtained following the docking of the 16 compounds (**M1-M16**) considered in the binding sites of  $\alpha$ 6 cockroach and honeybee homomeric nAChRs. As recalled above, on the basis of the pharmacophore of insect nAChR competitive modulators, only compounds coming out from the virtual screening and that cannot be easily protonated at physiological *p*H were considered.

**Table 1.** Docking scores (DS) and Glide energies (GE) (see experimental section), in kJ/mol, computed following the flexible docking of the top 16 compounds coming out from the virtual screening study on  $\alpha$ 6 homomeric cockroach and honeybee nAChRs.

	a6 cockros	ach nAChR	a6 honeybee nAChR		
Compound –	DS	GE	DS	GE	
M1	-40.6	-243.9	-31.8	-176.1	
M2	-33.9	-216.3	-34.3	-264.4	
M3	-25.5	-190.8	-21.3	-117.6	
M4	-38.1	-243.5	-28.0	-136.8	
M5	-40.6	-243.9	-34.3	-167.4	
M6	-34.3	-217.1	-38.5	-215.9	
M7	-29.7	-170.7	-32.2	-204.6	
<b>M8</b>	-35.1	-179.5	-34.7	-141.8	
M9	-25.1	-97.9	-25.5	-155.6	
M10	-29.7	-170.3	-15.5	-143.5	
M11	-26.8	-83.3	-15.9	-74.1	

M12	-42.7	-236.4	-45.2	-200.0
M13	-37.2	-237.2	-41.4	-213.8
M14	-25.9	-166.9	-25.5	-182.8
M15	-38.9	-180.7	-39.7	-184.5
M16	-41.4	-247.3	a	a

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<sup>a</sup> The docking of this compound was not possible in honeybee  $\alpha 6$  homomeric nAChR.

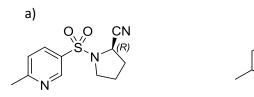
98 Among these sixteen compounds, seven appeared promising competitive candidates (M1, M3, M4, M5, M10, M11, M16) since their docking parameters (DS and GE) were 99 significantly more favorable for  $\alpha 6$  cockroach nAChR compared to  $\alpha 6$  honeybee nAChR. 100 Indeed, for M2, M6, M8-9 and M14-15, one or both of the discriminating parameters have 101 very similar values for both insect models, no selectivity emerging from these results. For two 102 compounds (M12, M13), no clear conclusions could be given from their results since the 103 trends suggested with the docking scores and the Glide energies are opposite. Finally for 104 compound M7, both values were in favor of the  $\alpha$ 6 honeybee nAChR, suggesting a possible 105 selectivity for this specie. 106

107 To go further with this structure-based design study, among the seven relevant molecules (M1, M3, M4, M5, M10, M11, M16), we have focused on the sulfonamide M3 because of its 108 relative simple synthesis. As shown in Figure 2a, it is worth noticing that M3 has an 109 110 asymmetric center and can therefore interact with nAChRs through two stereoisomers (R or S). We have only investigated in this work the binding of the (S) enantiomer since a molecular 111 fitting of the chemical structures of the most promising compounds have shown that only this 112 113 enantiomer gave the optimal orientations of the relevant chemical fragments in the binding site of the  $\alpha 6$  homomeric nAChRs (see SI; Figure S1). 114

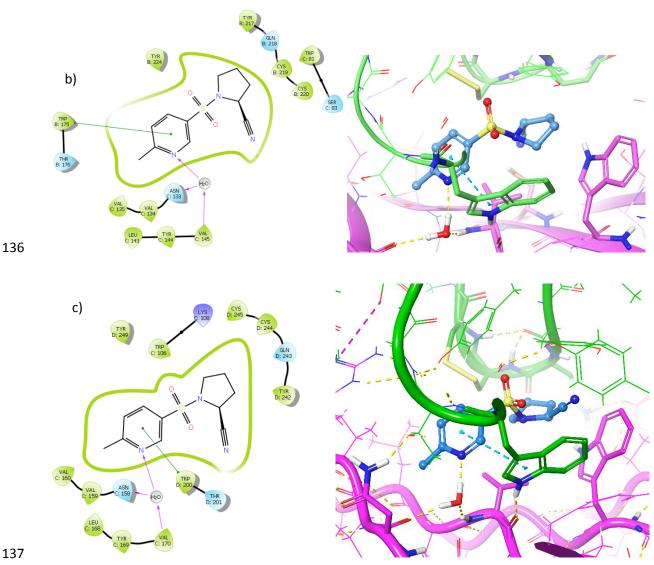
115 The interactions of the (S)-M3 enantiomer in the binding site of  $\alpha 6$  cockroach and honeybee nAChRs are shown in Figure 2b. It can first be seen that a hydrogen-bond 116 117 interaction involves the pyridine nitrogen of the ligand and main chain CO and NH groups of  $\alpha$ 6 nAChR residues through a water molecule in both insect species. This feature agrees with 118 the role played by a water molecule that appears conserved in several cocrystallized ligand-119 nAChR complexe models<sup>29, 30</sup> and has been suggested to be incorporated in the binding 120 pocket for the construction of a pharmacophore and the design of new ligands. In both cases, 121 Trp residues (Trp 175 and 200 for α6 cockroach and honeybee nAChRs, respectively) have a 122 pivotal contribution in the binding of the ligand, as well as with the five membered saturated 123 ring and the pyridine ring. These trends are in line with the prominence of this residue pointed 124 out in the literature by experimental studies<sup>31, 32</sup> and rationalized by computational 125 investigations.33-35 126

Lastly, it is worth noticing that the key cysteine residues (Cys219-Cys220 or Cys244-Cys245) are in both binding sites ( $\alpha$ 6 cockroach and honeybee nAChRs, respectively) in close vicinity with the ligand, a sulfur atom being in close contact with the oxygen atom of the sulfonamide group. This feature highlights the potential of interaction of the sulfone moiety, found in recent nAChRs modulators acting as insecticides, in particular the sulfonimide compound sulfoxaflor, designed by Dow Agrosciences.<sup>36</sup>

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139 Figure 2. (a) M3 (R) and (S)-enantiomers chemical structures. 2D ligand interaction plots and 3D views of the interactions between the (S) enantiomer of M3 (compound 3d) and (b) 140 cockroach  $\alpha$ 6 nAChR or (c) honeybee  $\alpha$ 6 nAChR. 141

From this analysis, no clear difference of interactions appeared therefore for the (S)142 enantiomer of M3 for cockroach and honeybee  $\alpha 6$  nAChRs binding sites. In fact, a further 143 examination of the interaction energies rationalizes its better affinity for  $\alpha 6$  cockroach 144 nAChR. Indeed, table 2 shows that for the main amino acid residues involved in the binding 145 and discussed above, the stabilization is significantly greater for cockroach  $\alpha$ 6 nAChRs. The 146 present molecular modeling results, validated by their good agreement with known 147

148 (experimental) structural features<sup>37</sup> for the interaction of nAChRs modulators and their target,

149 are therefore promising for the insecticide activity of **M3** and its potential selectivity for pests.

**Table 2.** Interaction energies (kJ/mol) computed by the Glide program for the main components of the  $\alpha$ 6 cockroach and honeybee nAChRs binding sites with the *S* enantiomer of **M3**. Docking scores (DS) and Glide energies (GE) (kJ/mol) are reminded for clarity.

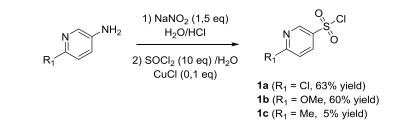
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		a6 cockro	ach nACl	nR model		
Trp81	Val145	Trp175	Tyr217	Cys219	DS	GE
-10.9	-21.3	-23.0	-20.1	-10.5	-25.5	-190.8
α6 honeybee nAChR model						
Trp106	Val170	Trp200	Tyr242	Cys245	DS	GE
>0	-9.6	-18.8	-12.1	-8.4	-21.3	-117.6
	-10.9 Trp106	-10.9 -21.3 Trp106 Val170	Trp81         Val145         Trp175           -10.9         -21.3         -23.0           α6 honey           Trp106         Val170         Trp200	Trp81         Val145         Trp175         Tyr217           -10.9         -21.3         -23.0         -20.1           α6 honeybee nACh           Trp106         Val170         Trp200         Tyr242	-10.9       -21.3       -23.0       -20.1       -10.5         α6 honeybee nAChR model         Trp106       Val170       Trp200       Tyr242       Cys245	Trp81         Val145         Trp175         Tyr217         Cys219         DS           -10.9         -21.3         -23.0         -20.1         -10.5         -25.5           α6 honeybee nAChR model           Trp106         Val170         Trp200         Tyr242         Cys245         DS

154

The heteroaromatic compound M3 (or 3d) and analogs 3a-c displaying a sulfonamide 155 function, could be simply prepared from nucleophilic substitution of a sulfonyl chloride 156 157 precursor 1 with the corresponding pyrrolidine 2 as depicted in scheme 2. We started with the synthesis of the selected sulforyl chlorides **1a-c** bearing a chloride, a methoxy or a methyl 158 group on the C6 position of the pyridine ring. Access to such heteroaromatic sulfonyl chloride 159 compounds could be achieved starting from halogeno heterocycles, by nucleophilic 160 substitution with methylthiolate followed by oxidative chlorination of thiol intermediate with 161 chlorine<sup>38</sup> or with 2,4-dichloro-5,5-dimethylhydantoin.<sup>39</sup> Woolven and collaborators also 162 proposed to prepare those sulfonyl chloride reagents, starting from Grignard derivatives and 163 the DABSO (DABCO-bis(sulfur dioxide complex) a stable sulfur dioxide equivalent, 164 followed by addition of sulfuryl chloride.<sup>40</sup> For our study we chose to apply the one-pot two 165

steps Sandmeyer-sulfonylation approach starting with 3-amino pyridines. This process was proved to work well with thionyl chloride in aqueous conditions and at low temperature.<sup>41</sup> Starting with commercial 3-amino-6-substituted pyridines, the targeted chlorosulfonyl reagents **1a-c** were obtained with good yields excepted for compound **1c** which failed to precipitate in this aqueous medium (Scheme 1).



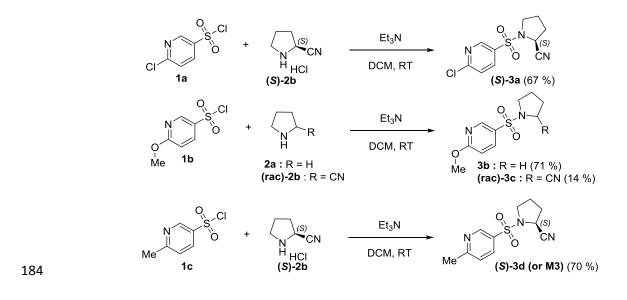
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172 Scheme 1. Synthesis of the selected pyridine-3-sulfonyl chloride reagents 1a-c

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To access to the racemic 2-cyano pyrrolidine (Scheme 2, (rac)-2b, R = CN), some synthetic ways were proposed in the literature, such as the oxidation of pyrrolidine with aqueous sodium peroxodisulfate to furnish the corresponding trimer being hydrocyanated with HCN <sup>42</sup> or TMSCN.<sup>43</sup> Applying this last method we prepared racemic pyrrolidine (rac)-2b in 34% overall yield in two steps after purification by distillation.

The expected sulfonamides **3a** to **3d** were finally prepared by nucleophilic substitution of the heteroaromatic sulfonyl chlorides **1a-c** with the commercially available pyrrolidines **2a** and **(S)-2b** or synthesized racemic pyrrolidine **(rac)-2b**, in classical reaction conditions (Scheme 2).<sup>44</sup> They were obtained with good yield excepted for compound **3c** which yield was not optimized.

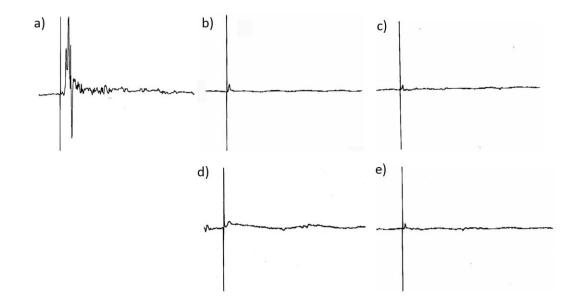




### Scheme 2. Synthesis of compounds 3a-3d

These sulphonamide compounds **3** were then evaluated on the cholinergic synaptic transmission in cockroach *Periplaneta americana*. Experiments were performed on the cercal nerve giant interneuron synapses located within the cockroach sixth abdominal ganglion (A6 (abbreviated SAG); see SI, Figure S2) using the mannitol-gap method pioneered by Callec.<sup>45</sup>

The electrophysiological properties of the sulfonamide compounds **3** emerged from 190 the modeling studies (M3=3d) are presented in Figure 3. We first compared the 191 pharmacological properties of the four compounds 3a-3d on insect cholinergic synaptic 192 transmission with imidacloprid (IMI, Figure 1), the forerunner of neonicotinoid insecticides. 193 We aimed to demonstrate if our compounds were able to depolarize the sixth abdominal 194 ganglion as found with IMI. Indeed, previous studies demonstrated that bath applications of 195 neonicotinoid insecticides such as IMI<sup>46</sup> and clothianidin<sup>47</sup> on the sixth abdominal ganglion 196 197 induced its depolarization through activation of postsynaptic nAChRs. Here, we found that bath application of the four compounds blocked the amplitude of the excitatory postsynaptic 198 potentials (EPSPs) evoked by electrical stimulation of the ipsilateral cercal nerve XI (see SI, 199 Figure S3) when the depolarization reached a peak (Figure 3), suggesting that as found with 200 IMI, these compounds activated postsynaptic nAChRs.<sup>46</sup> 201



202

Figure 3. Effect of 100 μM of compounds 3a, 3b, 3c and 3d on electrical stimulations
of the ipsilateral nerve XI. Stimulation of the nerve XI induces a EPSP (a: control condition)
which is blocked under bath application of the four compounds (b: 3a, c: 3b, d: 3c and e: 3d)
The dose-response curves plotted according to the known equation (see SI)
demonstrated that the four compounds (3a, 3b, 3c and 3d) induced a depolarization of the

sixth abdominal ganglion (Figure 4). Note that, for the four compounds, their effect onsynaptic depolarization was not reversed after wash-out.

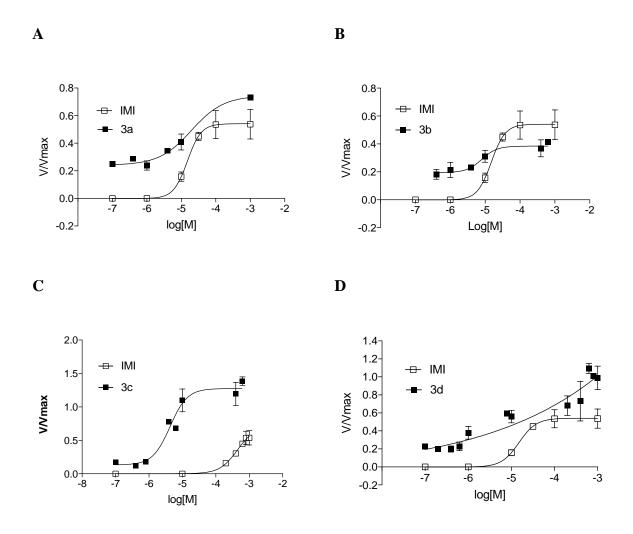


Figure 4. Dose-response curves of the synaptic depolarization induced by the four compounds **3a** (A) **3b** (B) **3c** (C) and **3d** (D), compared with IMI-induced depolarization. Concentration ranges from 0.1  $\mu$ M to 1 mM. Data are mean values of the amplitude of the peak depolarization. Each point represents n = 8 recordings.

The EC<sub>50</sub> values for compounds **3a**, **3b** and **3c** were evaluated as 19  $\mu$ M, 8.02  $\mu$ M and 4.25  $\mu$ M respectively. For compound **3d**, we were unable to determine the EC<sub>50</sub> value. The 6methoxy-pyridinyl sulfonamide compounds **3b** and **3c** as well as the 6-chloro analog **3a** gave EC<sub>50</sub> values in the same range. Thus, among this series, the most effective compounds appeared to be **3b** (EC<sub>50</sub> = 8.02  $\mu$ M) and **3c** (EC<sub>50</sub> = 4.25  $\mu$ M), bearing the methoxy substituent on the pyridine ring. We could also observe that the addition of a cyano group at the 2-position of the pyrrolidine ring (compound 3c) very slightly improved the EC<sub>50</sub> value compared to 3b. Than, in comparison with our hit candidate M3 (3d) bearing a methyl group on the pyridine ring, electrophysiological assays on the cockroach nervous system showed that the three analogs 3a-3c, bearing a methoxy or chloro substituent on the pyridine ring, appeared as better activators of the cholinergic synaptic transmission.

In summary, a new family of potential competitive modulators of insect nAChRs has been 225 designed in the present work. From a cross-disciplinary approach, a sulfonamide compound 226 227 M3, identified by molecular modeling, has been synthesized (M3 or 3d) together with three relevant analogs (3a, 3b and 3c). The cholinergic synaptic effect of these four compounds on 228 nAChRs has been evidenced through electrophysiological measurements in cockroach, and 229 230 was found to be similar to imidacloprid one, the neonicotinoid forerunner. Thus, the compounds designed and evaluated in the present work can be regarded as a promising 231 chemical tool to help to characterize and study insect nAChRs. Further investigations are also 232 in progress in our laboratories with the best sulfonamide compounds, particularly lethality 233 experiments with pests compared to bees, to complement the promising data presented herein. 234

235

#### 236 **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personalrelationships that could have appeared to influence the work reported in this paper.

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#### 246 A. Supplementary data.

Complementary figures, tables, for molecular modeling and experimental parts: synthesis and
NMR data of the compounds (1a-c and 3a-d) and biological tests experimental data. These
data are free of charge via the Internet.

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- 265 Supervision: Chemistry: M.M-A, J.L.; Electrophysiology: S.H.T.; Molecular Modeling: J.-
- Y.L.Q.; A.L. J.G. Writing: J.-Y.L.Q., M.M.-A., B.S., S.H.T.J. G., A.L.; Experiments: B.S.,
  E.L., A.C.

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